# Focused Feasibility Study for Surface Soil at the Main Pits and Pushout Area, J-Field Toxic Burning Pits Area, Aberdeen Proving Ground, Maryland

**Environmental Assessment Division Argonne National Laboratory** 



Operated by The University of Chicago, under Contract W-31-109-Eng-38, for the

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by

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June 1996

Work sponsored by U.S. Army Aberdeen Proving Ground, Directorate of Safety, Health, and Environment



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#### **FOREWORD**

Argonne National Laboratory acknowledges Mr. Larry Thebeau and Mr. Prakash Ramaswamy of ICF Kaiser Engineers, Inc., for the development of Alternative 2 (Limited Removal and Disposal, and In-Situ Containment) for this focused feasibility study. Mr. Thebeau and Mr. Ramaswamy prepared Sections 4.2.2, 5.2.2, 6.2, 7.2.7, and F.1.

#### **NOTATION**

The following is a list of acronyms, initialisms, chemical abbreviations, and units of measure used in this document.

#### ACRONYMS AND INITIALISMS

ADD applied daily dose AE assimilation efficiency

ANL Argonne National Laboratory

AOC area of concern

APG Aberdeen Proving Ground

ARAR applicable or relevant and appropriate requirement

ART Alternative Remedial Technologies, Inc.
ASTM American Society for Testing and Materials

AWQC ambient water quality criteria

BRA baseline risk assessment

CAMU Corrective Action Management Unit

CCMAS Construction Cost Management Analysis System

CERCLA Comprehensive Environmental Response, Compensation, and

Liability Act (as amended)

CFR Code of Federal Regulations
CLP Contract Laboratory Program
CMS corrective measures study
COC contaminant of concern

COEC contaminant of ecological concern
COMAR Code of Maryland Regulations
CRC Chemical Rubber Company

CRQL contract-required quantitation limit CSTA Combat Systems Test Activity

CWA chemical warfare agent

DANC decontaminating agent, noncorrosive
DNAPL dense, nonaqueous-phase liquid
DOE U.S. Department of Energy

DOT U.S. Department of Transportation DRE destruction/removal efficiency

DSHE Directorate of Safety, Health, and Environment (U.S. Army)

DW dry weight

EEQ environmental effects quotient

EMD Environmental Management Division (U.S. Army)

EO Executive Order EP extraction procedure

EPA U.S. Environmental Protection Agency

ERA ecological risk assessment
ERT Emergency Response Team

FEMA Federal Emergency Management Agency

FFS focused feasibility study

FR Federal Register FS feasibility study

GPR ground-penetrating radar

HAZMAT hazardous materials
HE high explosives

HFA Human Factors Applications, Inc.

HW hazardous waste

IRA interim remedial action IRD interim remedial design

LDR land disposal restriction LEL lower explosive limit

MCL maximum contaminant level MCR maximum credible release

MDNR Maryland Department of Natural Resources

MSL mean sea level

NCP National Oil and Hazardous Substances Pollution Contingency Plan

NOAA National Oceanic and Atmospheric Administration

NOAEL no observed adverse effects level

NPDES National Pollutant Discharge Elimination System

NPL National Priorities List NR National Resources

O&M operation and maintenance O&P overhead and profit

OB/OD open burning/open detonation

ODC other direct cost

OSAS on-site analytical suite

OSHA Occupational Safety and Health Administration
OSWER Office of Solid Waste and Emergency Response

PL Public Law

PRG preliminary remediation goal

QC quality control

RACER Remedial Action Cost Engineering and Requirements System

RCRA Resource Conservation and Recovery Act

RFI RCRA facility investigation RI remedial investigation

RME reasonable maximum exposure

ROD Record of Decision

SARA Superfund Amendments and Reauthorization Act of 1986

SPTF soil-to-plant transfer factor SVE soil-vapor extraction

SWMU solid waste management unit

TAL target analyte list

TBC to-be-considered requirement

TBP Toxic Burning Pits
TCL target compound list

TCLP toxicity characteristic leaching procedure

TDS total dissolved solids

TSCA Toxic Substances Control Act treatment, storage, and disposal

UCL upper confidence limit

USATHAMA U.S. Army Toxic and Hazardous Materials Agency

USC United States Code
USGS U.S. Geological Survey

UV ultraviolet

UXO unexploded ordnance

WPP white phosphorus burning pits

XRF X-ray fluorescence

#### CHEMICAL ABBREVIATIONS

BHC benzene hexachloride

BTEX benzene, toluene, ethylbenzene, and xylenes

CAH chlorinated aliphatic hydrocarbon

DCE dichloroethylene
11DCE 1,1-dichloroethylene
12DCE 1,2-dichloroethylene

DDD 4,4'-dichlorodiphenyldichloroethane
DDE 4,4'-dichlorodiphenyldichloroethylene
DDT 4,4'-dichlorodiphenyltrichloroethane
DNAPL dense nonaqueous-phase liquid
EDTA ethylenediamine tetraacetic acid

GA ethyl N,N-dimethyl phosphoramidocyanidate (tabun)

GB isopropyl methylphosphonofluoridate (sarin)
GD pinacolyl methylphosphonofluoridate (soman)

HD sulfur mustard

LNAPL light nonaqueous-phase liquid
PAH polyaromatic hydrocarbon
PCB polychlorinated biphenyl
PEH petroleum hydrocarbon
PETN pentaerythritol tetranitrate

RDX hexahydro-1,3,5-trinitro-1,3,4-triazine

SVOC semivolatile organic compound

111TCE 1,1,1-trichloroethane
112TCE 1,1,2-trichloroethane
TCLEA 1,1,2,2-tetrachloroethane
TCLEE tetrachloroethylene

TNT trinitrotoluene

TPH total petroleum hydrocarbon

TRCLE trichloroethylene

VOC volatile organic compound VX methylphosphonothioic acid

#### UNITS OF MEASURE

cm	centimeter(s)	L	liter(s)
°F	degree(s) Fahrenheit	$m^3$	cubic meter(s)
ft	foot (feet)	mg	milligram(s)
$ft^2$	square foot (feet)	mi	mile(s)
gal	gallon(s)	mm	millimeter(s)
gpm	gallon(s) per minute	ng	nanogram(s)
h	hour(s)	μg	microgram(s)
in.	inch(es)	ppm	part(s) per million
kg	kilogram(s)	S	second(s)
km	kilometer(s)	V	volt(s)
kW	kilowatt(s)	$yd^3$	cubic yard(s)

# FOCUSED FEASIBILITY STUDY FOR SURFACE SOIL AT THE MAIN PITS AND PUSHOUT AREA, J-FIELD TOXIC BURNING PITS AREA, ABERDEEN PROVING GROUND, MARYLAND

#### **SUMMARY**

#### S.1 INTRODUCTION

The Environmental Management Division of Aberdeen Proving Ground (APG), Maryland, is conducting a remedial investigation and feasibility study of the J-Field area at APG pursuant to the Comprehensive Environmental Response, Compensation, and Liability Act, as amended (CERCLA). J-Field is located within the Edgewood Area of APG in Harford County, Maryland. Since World War II, activities in the Edgewood Area have included the development, manufacture, testing, and destruction of chemical agents and munitions. These materials were destroyed at J-Field by open burning/open detonation. Portions of J-Field continue to be used for the detonation and disposal of unexploded ordnance (UXO) by open burning/open detonation under authority of the Resource Conservation and Recovery Act.

J-Field is almost flat and is covered by open fields, woods, and nontidal marshes. It encompasses about 460 acres at the southern end of Gunpowder Neck Peninsula. The peninsula is surrounded by tidal estuaries on three sides — Gunpowder River to the west and Chesapeake Bay to the south and east.

The Toxic Burning Pits (TBP) area, one of the designated areas of concern at J-Field, is located on about 9 acres in the southern portion of J-Field. This area was used to dispose of bulk chemical wastes, drummed chemical wastes, high explosives, various chemical agents, and chlorinated solvents. The TBP location contains several potential source areas of contamination, including five burning pits, a pushout area, a storage area, a disposal area, and a demolition area.

Results from the remedial investigation (Yuen et al. 1996) indicate that the principal contaminant sources at the TBP area are surface and subsurface soils associated with the two main pits and the pushout area. Surface soil is the environmental medium of greatest concern; soil contaminants include heavy metals and organic compounds. These contaminants may be released by surface runoff, precipitation infiltration, and gaseous emission of volatile organic compounds. Contaminant transport from these sources has resulted in contamination of sediments and surface water in the marsh-pond ecosystem that borders the pushout area and the southern boundary of the site. Surface water and sediments are contaminated primarily with heavy metals; some organic compounds have also been detected.

This report presents the focused feasibility study (FFS) for the TBP area at J-Field. The purpose of this study is to develop and evaluate potential remedial action alternatives to address surface soil contamination at three source areas within the J-Field TBP area: the northern main pit, the southern main pit, and the pushout area. Other source areas and contaminated media at the TBP area are outside the scope of this interim remedial action and will be addressed as part of the long-term remediation for J-Field.

#### S.2 FOCUSED FEASIBILITY STUDY APPROACH

The initial task of the FFS was to develop remedial action objectives and define the scope of the proposed action given the information available from the remedial investigation, human health and ecological risk assessments, and earlier investigations. Interim preliminary remediation goals (PRGs) were developed for each contaminant of concern (COC) on the basis of risk to human and ecological receptors and probable future land use. The objectives and scope (Section S.3) guided the subsequent tasks of identifying and screening remedial action technologies (Section S.4), assembling and screening preliminary remedial action alternatives (Sections S.5 and S.6), conducting treatability studies, and evaluating the final remedial action alternatives. The last step in the evaluation process was a comparative analysis of alternatives (Section S.7).

#### S.3 INTERIM REMEDIAL ACTION OBJECTIVES

#### S.3.1 Objectives and Scope

The overall objectives of the proposed interim remedial action at the TBP area of J-Field are to:

- Reduce exposure of human and environmental receptors to surface contamination in three source areas: the northern main pit, the southern main pit, and the pushout area;
- Minimize the potential for contaminant migration via sediment transport from these areas;
- Minimize the potential for contaminant migration via downward leaching through these areas; and
- Support long-term site remediation.

Three source areas of contamination at the TBP area are the subject of this FFS: the northern main pit, the southern main pit, and the pushout area. Two additional pits in the pushout area have been filled in and are partially covered by soil pushed out from the main pits. Preliminary results of human health and ecological risk assessments (ICF Kaiser Engineers 1995a; Hlohowskyj et al. 1996) indicate that contaminated surface soil in these areas should be addressed to protect human health and the environment. An interim remedial action could be implemented that would achieve the objectives stated above.

Preliminary area and volume estimates for these areas are presented in Table S.1. Figure S.1 shows the areas with metal concentrations that exceed the mean regional background as reported by ICF Kaiser Engineers (1995b). The figure also includes areas with levels of contaminants that exceed the interim PRGs reported in Table S.2. The interim PRGs developed for surface soils at the TBP area (Section S.5) were used to determine which source areas should be cleaned up as part of the interim action and which could be appropriately left in their current condition until further action is taken at the site as part of the long-term remediation for J-Field.

# S.3.2 Preliminary Remediation Goals

For the TBP area, human health-based interim PRGs for soil were developed by using the risk-based soil concentrations derived by the U.S. Environmental Protection Agency (EPA) Region III. EPA Region III derives the risk-based values by following the Superfund risk assessment guidelines (EPA 1989a,b) and standard default assumptions (EPA 1991a); the values are based on carcinogenic and noncarcinogenic effects of ingesting "industrial" soil (as opposed to residential soil).

Two approaches were employed to develop ecological risk-based interim PRGs for the TBP area: (1) use of the contaminant uptake models developed for the ecological risk assessment to back-calculate acceptable surface soil concentrations and (2) use of the human health-based interim PRGs for COCs for which uptake modeling was not performed. In this latter approach, risk reduction for ecological resources is considered to be directly correlated with the reduction of contaminant concentrations or with removal of contaminated media. Thus, reducing surface soil contaminant concentrations to the levels of the human health risk-based interim soil PRGs would also reduce risk for ecological resources. In both approaches, each derived interim soil PRG was screened against the regional background soil concentration of that contaminant, and the greater of the two values was selected as the interim PRG.

Table S.2 integrates the ecological and human health risk-based PRGs and presents an overall list of preliminary interim soil PRGs for the TBP area. The identified interim soil PRG for each COC is the lower of the human health and ecological risk-based interim PRGs. These interim PRGs are initial guidelines and are not intended to set final cleanup levels or to establish that cleanup to meet these goals is warranted (EPA 1991b, 1994).

TABLE S.1 Estimated Areas and Volumes of Contaminated Surface Soil at the Main Pits and Pushout Area

Contaminated Surface Soil Interval <sup>a</sup>	Contaminants of Concern	Area (ft <sup>2</sup> )	Volume (yd³)
0-6 in.	Metals, organics	218,300	4,043
6-24 in.	Metals, organics	208,600	11,590

<sup>&</sup>lt;sup>a</sup> See Figure S.1 for location of excavation area.

#### S.4 POTENTIALLY APPLICABLE TECHNOLOGIES

Remedial technology types and process options that are potentially applicable to addressing contaminated soil at the TBP area were identified and screened on the basis of effectiveness, implementability, and cost and for applicability to specific site conditions in accordance with EPA guidance (1988). The current understanding of contaminants and conditions at the site suggested that the general response actions that could be implemented to achieve the project objectives (Section S.3) are institutional controls, in-situ containment, removal, treatment, short-term storage, and disposal. Specific application of these technologies to site conditions was evaluated to determine which would be most appropriate for interim remedial action at the TBP area. The technology types retained through the screening process include institutional controls (access restriction, ownership and use or deed restriction, and monitoring); in-situ containment (surface control diversions, vertical barriers, and caps); removal (excavation, and clearing and grubbing); ex-situ treatment (stabilization/fixation, soil washing, soil leaching, incineration, and thermal desorption); short-term storage (on-site open and closed structures); and disposal (off-site land-based facility). These technologies were combined to develop preliminary interim remedial action alternatives for the TBP area.

#### S.5 DESCRIPTION OF ALTERNATIVES

Preliminary alternatives for remediating surface soil at the TBP area were developed and screened in accordance with CERCLA, EPA guidance (1988), and the National Oil and Hazardous Substances Pollution Contingency Plan (EPA 1990). Five preliminary alternatives were developed and screened on the basis of implementability, effectiveness, and cost. These alternatives are described briefly in the following subsections.

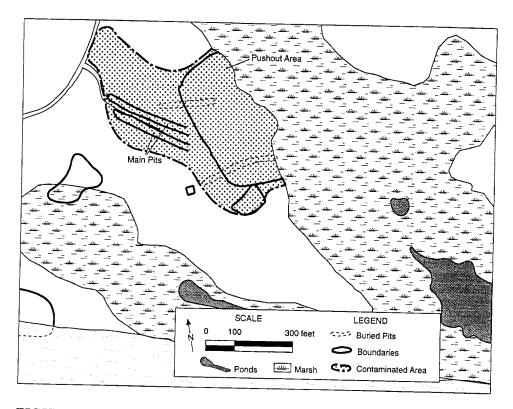


FIGURE S.1(a) Contaminated Surface Soil, 0-6 in., to Be Addressed by the Proposed Action at the TBP Area

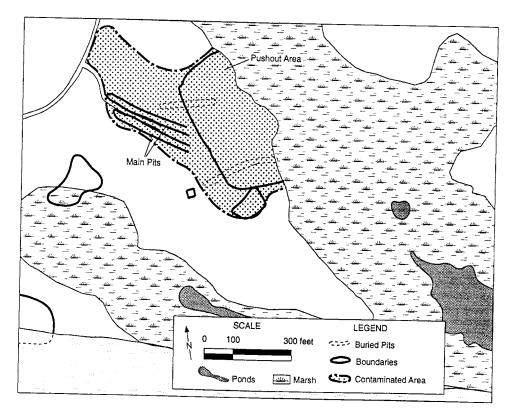


FIGURE S.1(b) Contaminated Surface Soil, 6-24 in., to Be Addressed by the Proposed Action at the TBP Area

TABLE S.2 Risk-Based Interim Preliminary Remediation Goals (PRGs) for Surface Soil at the TBP Area

Contaminant	Interim PRG (mg/kg)
Volatile and semivolatile organic compounds	
Trichloroethene	1.3 <sup>a</sup>
2-Methylnaphthalene	NA <sup>b</sup>
Explosive	
Nitroglycerin	NA
Inorganic materials	
Aluminum	7,940 <sup>c</sup>
Antimony	$3.7^{a}$
Arsenic	$3.8^{d}$
Barium	64.4 <sup>a</sup>
Beryllium	1.3 <sup>d</sup>
Cadmium	0.34 <sup>c</sup>
Chromium	16.7 <sup>c</sup>
Cobalt	25.0 <sup>a</sup>
Copper	100°
Cyanide	$1.0^{a}$
Lead	58.7/400 <sup>e</sup>
Mercury	0.04 <sup>c</sup>
Selenium	0.20 <sup>c</sup>
Silver	$2.0^{a}$
Zinc	37.0°

<sup>&</sup>lt;sup>a</sup> Interim PRG is ecological risk-based (see Table 3.5).

b NA = No interim PRG developed. Insufficient data to develop ecological risk-based interim PRG, and no EPA Region III human health-based PRG has been developed (see Table 3.2).

<sup>&</sup>lt;sup>c</sup> Ecological risk-based PRG value is below mean background concentration; interim PRG is mean background concentration.

<sup>&</sup>lt;sup>d</sup> Interim PRG is human health-based (see Table 3.2).

<sup>&</sup>lt;sup>e</sup> The ecological risk-based PRG is 58.7 mg/kg (see Table 3.5). EPA currently recommends a soil cleanup value of 400 mg/kg for residential land use (see Table 3.2). The cleanup goal for lead has been as high as 1,000 mg/kg at other APG sites (Wrobel 1995).

#### S.5.1 Alternative 1: No Action

Alternative 1 is included to provide a baseline for comparison with the other action alternatives. Under this alternative, the site would continue to operate under restricted access. Current conditions of the contaminated surface soil would continue. The baseline condition of the J-Field TBP area would include only minor maintenance activities, such as mowing and monitoring.

# S.5.2 Alternative 2: Limited Removal and Disposal, and In-Situ Containment

Under Alternative 2, contaminated surface soil in the two main pits and pushout area would be contained in place with a protective soil cover. Limited excavation and disposal would be implemented to address contaminated soil in the main pits (specifically arsenic and polychlorinated biphenyls [PCBs]).

The "risk-reduction cover" would involve placing a geotextile fabric over the vegetation-cleared surface (about 5 acres) and covering it with a layer of soil sufficient to raise the site topography above the 100-year floodplain. The soil layer would be stabilized with a vegetative cover. Runoff from the area would be diverted by surface grading controls. Erosion of contaminated soil from the portion of the pushout area in contact with the marsh would be mitigated by installing a vertical barrier. Because the soil layer would be emplaced incrementally in 2-ft layers, UXO screening most likely would not be required for this component. A UXO survey would be required for the limited excavation component, especially if conventional equipment were used. The area of excavation would be limited, so hand removal methods could be used (if the density of metal contacts present is high). The limited disposal component under Alternative 2 would involve shipping about 400 yd<sup>3</sup> of soil to a permitted off-site facility for treatment (stabilization and possibly incineration) and disposal.

Stabilization of the shoreline to the south of the TBP area would also be implemented as part of this alternative. Installation of a boulder riprap berm would reduce the potential for long-term erosion along the shoreline.

# S.5.3 Alternative 3: Removal and Short-Term Storage

Under Alternative 3, contaminated surface soil in the two main pits and pushout area that exceeds soil cleanup criteria would be removed and transported to an on-site, short-term storage facility, where it would be stored until final remediation is addressed at J-Field.

Soil would be excavated to a depth of 2 ft from the source areas with standard earthmoving equipment, such as front-end loaders, scrapers, and backhoes. The excavation area would be

surveyed for UXO in the upper 2 to 4 ft. Approximately 16,000 yd³ of soil would be removed. The short-term storage facility would be constructed adjacent to the Prototype Building to minimize the distance the material would have to be transferred. The short-term facility would be a modular building consisting of fabric walls. The base of the structure would require the construction of a bermed and drained concrete or asphalt pad equipped with a leachate collection system. An air infiltration system might also be required.

#### S.5.4 Alternative 4: Removal, On-Site Treatment, and Limited Disposal

Under Alternative 4, contaminated soil in the two main pits and pushout area that exceeds soil cleanup criteria would be removed and treated by soil washing/leaching to reduce the volume of waste for off-site disposal.

The removal component of Alternative 4 would be similar to that of Alternative 3; however, the excavated soil would be transported to an on-site soil washing/leaching treatment facility, which would be constructed adjacent to the Prototype Building. For the conceptual design, it was estimated that 16,000 yd³ of soil would be treated. After removal and treatment, the cleaned material would either remain on-site to be used as fill (pending a treatability variance or CERCLA waiver) or be shipped to an off-site solid waste disposal facility, in accordance with the State of Maryland's preference. Limited disposal under Alternative 4 would involve shipping recovered lead, scrap metal, and other (nonhazardous) refuse, as necessary, to an off-site facility (smelter and/or municipal landfill). A portion of the soil from the southern main pit, contaminated with PCBs, would also be sent off-site for treatment and disposal.

# S.5.5 Alternative 5: Removal, Off-Site Treatment, and Disposal

Under Alternative 5, contaminated soil in the two main pits and pushout area that exceeds soil cleanup criteria would be removed and transported to an off-site treatment and disposal facility.

The removal component of Alternative 5 would be similar to that of Alternative 3; however, the excavated soil would be temporarily staged on-site following excavation, then transported to an off-site treatment and disposal facility.

#### S.6 IDENTIFICATION OF FINAL ALTERNATIVES

On the basis of the screening analysis for preliminary alternatives, all five of the preliminary alternatives were retained for detailed analysis. Although the no-action alternative (Alternative 1) would not be protective of human health and the environment in the long term, it was

retained through this screening to provide a basis for comparison with the remaining action alternatives during the subsequent detailed analysis. Alternatives 2 and 3 were retained because they offer low-cost options for reducing risk; however, Alternative 3 it is not considered effective in the long term and would thus require an additional action at a future date. Alternatives 4 and 5 were retained because they offer permanent solutions to surface soil contamination; however, these alternatives are fairly costly.

# S.7 EVALUATION SUMMARY AND COMPARATIVE ANALYSIS

Table S.3 provides the comparative analysis of the five remedial action alternatives. The analysis is summarized as follows.

All of the final remedial action alternatives for the TBP area, except for the no-action alternative (Alternative 1), satisfy the threshold criteria for protecting human health and the environment and complying with regulatory requirements, with waivers as appropriate. Overall protectiveness would be comparable for Alternatives 2, 4, and 5. Alternative 3 is protective in the short term but would require an additional component (e.g., off-site treatment and disposal) to be as protective as Alternatives 4 and 5.

Only Alternatives 2, 4, and 5 are expected to provide a permanent solution that would ensure protection for a very long time; however, long-term effectiveness under Alternative 2 could be affected by the potential for flooding. It is possible that the soil washing/leaching treatment under Alternative 4 would be more protective than Alternative 5 if, at some future date, the stabilized/solidified waste were to be exposed to the environment and the contaminants leached. However, appropriate design and good engineering practices would minimize the likelihood of such an occurrence.

Each action alternative would reduce contaminant mobility. Under Alternative 2, waste toxicity and volume would be only somewhat reduced. Under Alternative 3, waste volume would not be affected. Under Alternative 4, waste volume would decrease. Waste volume would increase under Alternative 5 because of the addition of cement and fly ash to stabilize the waste. Treatment methods under Alternatives 2, 4, and 5 would reduce contaminant toxicity; under Alternative 4, metals would be recovered and recycled.

The short-term effectiveness of Alternatives 2 through 5 would be comparable. The overall risk to the general public would be higher for Alternative 5 because it involves off-site transport of a large volume of contaminated soil. Environmental impacts at the TBP area from excavation and construction activities would be common for all action alternatives, and comparable impacts would be expected. Mitigative measures would be used to minimize potential short-term impacts.

Emplacement of the "risk-reduction cover" under Alternative 2 would be straightforward, as would construction of the storage facility under Alternative 3. Implementing Alternative 4 would also be fairly straightforward, although additional studies would be required to refine the soil washing/leaching treatment system design and cost estimates. The chemical stabilization/solidification treatment that would be performed off-site under Alternative 5 is fairly well established and would not require further development before implementation.

Alternative 2 has the lowest overall estimated cost (\$1.8 million) of the action alternatives; however, this estimate only includes projected costs for the next 30 years and is not directly comparable to Alternatives 3, 4, and 5 (for excavation to 2 ft). Alternative 5 is considered more cost-effective than Alternative 4 for site cleanup. The estimated total cost of Alternative 5 is \$10.5 million (for excavation to 2 ft), and it would provide a similar level of overall effectiveness as Alternative 4, which would cost an estimated \$13.1 million.

TABLE S.3 Comparative Analysis of Alternatives

Alternative 1: No Action	Limited Removal and Disposal, and In-Situ Containment	Alternative 3: Removal and Short-Term Storage	Alternative 4: Removal, On-Site Treatment,	Alternative 5: Removal, Off-Site Treatment,
Overall Protection of Human Health and the Environment	alth and the Environment		medera parille pin	and Disposal
Would not ensure protection of human health and the environment in the long term. Source areas would not be removed or treated, and exposures could increase over time.	Engineering and mitigative measures would be employed during the remedial action period so that no significant adverse impacts would occur to the general public or environment. Worker exposures would be similarly controlled to levels within health-protective limits. Long-term exposures would be minimized by removing "hot spots" (arsenic and PCBs) from the main pits and by covering remaining contaminated soil with a protective soil cover.	Generally similar to Alternative 2.  Long-term exposures would be minimized by removing and storing contaminated soil from the main pits and pushout area for about 5 years. At the end of that time, an additional component (e.g., off-site treatment and disposal) would be needed to ensure long-term protection.	Generally similar to Alternatives 2 and 3. Long-term exposures would be minimized by removing and treating contaminated soil from the main pits and pushout area.	Generally similar to Alternatives 2, 3, and 4. Treatment and disposal of a large volume of contaminated soil would be conducted off-site; therefore, the overall risk of exposure to the general public would be higher than that for Alternatives 2, 3, and 4.
Compliance with ARARs				
Would not meet all ARARs, including the corrective action requirements of RCRA and TSCA. In addition, would not satisfy the CERCLA-mandated preference for remedies that reduce contaminant toxicity, mobility, or volume through treatment.	ARs, Would meet all pertinent ARARs action (with waivers as appropriate), and including those that address protection of endangered species and habitats, andated floodplains, and wetlands. Implementing Alternative 2 might require preparation of a sediment and erosion control plan.	Same as Alternative 2. The on-site storage facility would be operated to ensure compliance with RCRA.	Same as Alternatives 2 and 3. The on-site treatment facility would be operated to ensure compliance with RCRA.	Same as Alternatives 2, 3, and 4, with additional disposal requirements that would be met, including RCRA hazardous waste generator.

Alternative 1: No Action	Alternative 2: Limited Removal and Disposal, and In-Situ Containment	Alternative 3: Removal and Short-Term Storage	Alternative 4: Removal, On-Site Treatment, and Limited Disposal	Alternative 5: Removal, Off-Site Treatment, and Disposal
Long-Term Effectiveness and Permanence	гтапепсе			
Current exposures and impacts would continue and could increase over time because of continued contaminant migration. Existing institutional controls would continue to limit access to site workers and other authorized personnel, thereby reducing risk to the general public.	More protective than Alternatives I and 3 because contaminated soil would be partially removed, the remaining soil then would be covered to reduce exposure of human and environmental receptors to surface contamination. Soil cleanup criteria would be applied to the removal of soils and to the delineation of the area to be covered.	More protective than Alternative 1 because contaminated soil would be removed and stored in an enclosed facility to provide an interim solution for risk posed by contaminated surface soil. Soil cleanup criteria would be applied as appropriate to the removal of soils, so the action would reduce contaminant concentrations remaining in soil to the most protective levels practicable.	More protective than Alternatives I and 3 because contaminated surface soil would be removed and treated (soil washing/leaching) to provide a permanent solution for risk posed by contaminated surface soil. Soil cleanup criteria would be applied as appropriate to the removal of soils, so the action would reduce contaminant concentrations remaining in soil to the most protective levels practicable.	Similar to Alternative 4, except that exposure of stabilized/solidified waste to the environment could result in contaminants leaching from the waste over time.
Reduction of Toxicity, Mobility, or Volume	Reduction of Toxicity, Mobility, or Volume			
Toxicity, mobility, and volume would significantly reduce contant of contaminated soil would inant mobility. Contaminant volu would be somewhat reduced by the limited removal and disposal component. Contaminant toxicity would not be reduced.	The in-situ containment component would significantly reduce contaminant mobility. Contaminant volume would be somewhat reduced by the limited removal and disposal component. Contaminant toxicity would not be reduced.	Short-term storage of soil in an on-site facility would significantly reduce contaminant mobility. An estimated 16,000 yd³ (27,200 tons) of soil would be stored on-site for about 5 years. Contaminant toxicity and volume would not be reduced.	Chemical treatment of soil by soil washing/leaching would significantly reduce contaminant mobility and volume. An estimated 16,000 yd³ (27,200 tons) of soil would be treated on-site by soil washing/leaching. This process could reduce original soil volume by about 94%; the remaining 6% could then be sent off-site for disposal. Toxicity of PCBs would be reduced. Lead toxicity would not be reduced, bowever, the metal would be removed and recycled.	Chemical treatment of soil with the stabilization/solidification technology would significantly reduce contaminant mobility. An estimated 16,000 yd <sup>3</sup> (27,200 tons) of soil would be treated off-site by chemical stabilization/solidification. This process could increase the original soil volume and weight by as much as 30% and 60%, respectively. This would result in a total treated volume of about 20,800 yd <sup>3</sup> and total weight of about 35,360 tons to be placed in a landfill. Contaminant toxicity would

# TABLE S.3 (Cont.)

Alternative 5: Removal, Off-Site Treatment,	Similar to Alternatives 3 and 4, except that overall risk to the general public would be higher because this alternative would involve the off-site transport of a large volume of contaminated soil. Additional mitigative measures would be implemented to reduce related impacts.
Alternative 4: Removal, On-Site Treatment, and Limited Disposal	Similar to Alternative 3.
Altemative 3: Removal and Short-Term Storage	Exposures could be higher than Alternatives I and 2 in the short term because of particulate and VOC emissions associated with removal, transportation, and storage activities. Mitigative measures would be implemented to minimize potential human health and environmental impacts. Risks to a site worker would increase compared to Alternative I; however, worker health and safety precautions would be used to control exposures. No adverse impacts to the general public are expected from contaminant releases during implementation of this alternative. Soil, vegetation, and wildlife disturbances would be significant during soil excavation activities. An estimated 5 acres would be affected. Most impacts to biota would be temporary. Activities are not expected to affect threatened or endangered species. Mitigative measures would be employed to minimize impacts to surface water resources (including the adjacent marsh), air quality, and those associated with noise levels.
Alternative 2: Limited Removal and Disposal, and In-Situ Containment	Exposures would be lower than for Alternatives 3, 4, and 5, which involve large-scale excavation. Mitigative measures would still be required in the short term because of particulate and VOC emissions associated with removal and transportation, a particulate and VOC emissions associated with removal and transportation, and impacts to the general public are expected from contaminant releases during implementation of this alternative. Soil, vegetation, and wildlife disturbances would be significant during soil excavation activities. An estimated 5 acres would be affected. Most impacts to biota would be temporary. Surface water impacts associated with the construction of a berm along the pushout-marsh boundary would be impacts associated with the construction in erosion and transport of soils into the marsh ecosystem.  Mitigative measures would be a benefit to the marsh ecosystem. Mitigative measures would be implemented to minimize impacts to biota water input would be reduced, but reduction in erosion and transport of soils into the marsh ecosystem. Mitigative measures would be a diacent marsh, air quality and those associated with noise levels.
Alternative 1: No Action	Short-Term Effectiveness  Current exposures and adverse impacts would continue. Existing institutional controls would continue to limit access to site workers and other authorized personnel, thus reducing risks to the general public. Short-term risks to site workers and the environment as a result of monitoring and maintenance activities are considered low.

# TABLE S.3 (Cont.)

Alternative 1: No Action	Alternative 2: Limited Removal and Disposal, and In-Situ Containment	Alternative 3: Removal and Short-Term Storage	Alternative 4: Removal, On-Site Treatment, and Limited Disposal	Alternative 5: Removal, Off-Site Treatment, and Disposal
Implementability				
Minimum site operations (monitoring and maintenance) would continue with the use of readily available resources.	Fairly straightforward to implement. Resources are readily available for UXO screening, soil removal, in-situ containment, and shoreline stabilization.	Fairly straightforward to implement. Resources are readily available for UXO screening, soil removal, and on-site storage. Short-term storage facilities are available as kits and can be customized to meet project specifications.	Similar to Alternative 3. Soil washing/leaching has been used at several hazardous waste sites and probably would not require further development before it could be implemented at the TBP area. Further bench-scale testing would be required to refine and optimize the design of the treatment system.	Similar to Alternative 3. Soil  washing/leaching has been used at several hazardous waste sites and probably would not require further development before it could be implemented.  Similar to Alternatives 3 and 4.  Chemical stabilization/solidification is an established technology and would an established technology and would implemented at the TBP area.  Further bench-scale testing would be required to refine and optimize. The design of the treatment system.
Cost	Cost		,	
The total cost would be the lowest in the short term (about \$3 million over a 10-year period), but the comparative level of effectiveness would be low. In addition, the cost could be potentially higher than the action alternatives over the long term because conditions could worsen over time, necessitating an expensive expanded response in the future.	The total cost would be about \$1.8 million, the lowest of all the action alternatives. This cost, however, reflects only O&M costs for the next 30 years. Because contaminated soil would remain in place, O&M costs would likely extend indefinitely into the future. Repairs associated with the catastrophic effects of a 100-year flood (should such occur) are not reflected in this estimate.	The total cost would be about \$5.0 million (for 16,000 yd³ of soil), the second lowest of all the action alternatives. This cost, however, does not include the cost of an additional component (off-site treatment and disposal) that would be needed to ensure long-term protectiveness and permanence. Therefore, while this alternative would achieve project objectives of reducing exposures and minimizing contaminant migration at a low cost, it is considered the least costeffective because additional components would be required when the 5-year storinge period ended.	The total cost would be about \$13.0 million (for 16,000 yd³ of soil), which is the highest of the action alternatives. The soil washing/leaching component would account for 86% of the total direct cost of this alternative.	The total cost would be about \$10.5 million (for 16,000 yd³ of soil), which is lower than Alternative 4 for the same overall level of effectiveness. Off-site treatment and disposal would account for 85% of the total direct cost of this alternative.

## S.8 REFERENCES FOR SUMMARY

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#### 1 INTRODUCTION

#### 1.1 BACKGROUND

The Environmental Management Division (EMD) of Aberdeen Proving Ground (APG), Maryland, is conducting a remedial investigation and feasibility study (RI/FS) of the J-Field area at APG pursuant to the Comprehensive Environmental Response, Compensation, and Liability Act, as amended (CERCLA). J-Field is located within the Edgewood Area of APG in Harford County, Maryland (Figure 1.1). Since World War II, activities in the Edgewood Area have included the development, manufacture, testing, and destruction of chemical agents and munitions. These materials were destroyed at J-Field by open burning/open detonation (OB/OD). Portions of J-Field continue to be used for the detonation and disposal of unexploded ordnance (UXO) by OB/OD under authority of the Resource Conservation and Recovery Act (RCRA).

J-Field is almost flat and is covered by open fields, woods, and nontidal marshes. It encompasses about 460 acres at the southern end of the Gunpowder Neck Peninsula (Figure 1.1). The peninsula is surrounded by tidal estuaries on three sides — Gunpowder River to the west and Chesapeake Bay to the south and east.

The Toxic Burning Pits (TBP) area, one of the designated areas of concern (AOCs) at J-Field, is located on about 9 acres in the southern portion of J-Field (Figure 1.2). This area was used to dispose of bulk chemical wastes, drummed chemical wastes, high explosives, various chemical agents, and chlorinated solvents. The TBP location contains several potential source areas of contamination, including five burning pits (two main pits and the methylphosphonothioic acid [VX], mustard, and liquid smoke pits), a pushout area, a storage area, a disposal area, and a demolition area.

## 1.2 PURPOSE OF THIS REPORT

The purpose of this focused feasibility study (FFS) is to develop and evaluate potential remedial action alternatives to address surface soil contamination at three source areas within the J-Field TBP area: the northern main pit, the southern main pit, and the pushout area. The FFS report documents this process; its purpose is to provide sufficient information to support an informed decision regarding an appropriate interim remedial action for these areas.

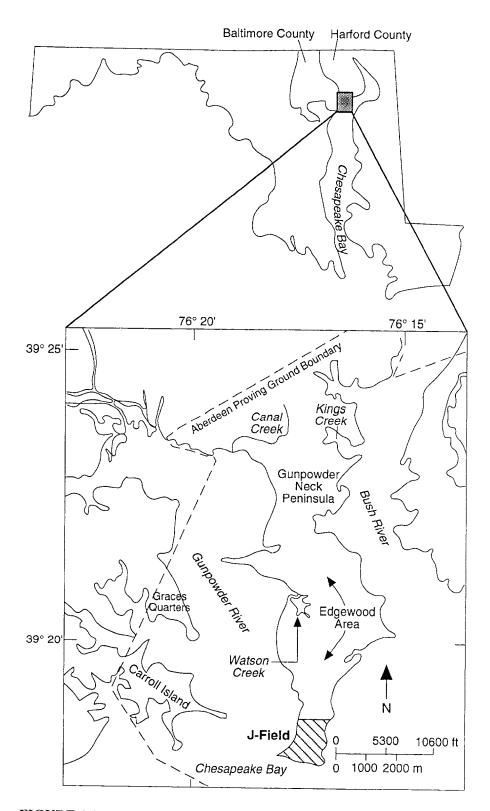


FIGURE 1.1 Location of J-Field in the Edgewood Area at APG

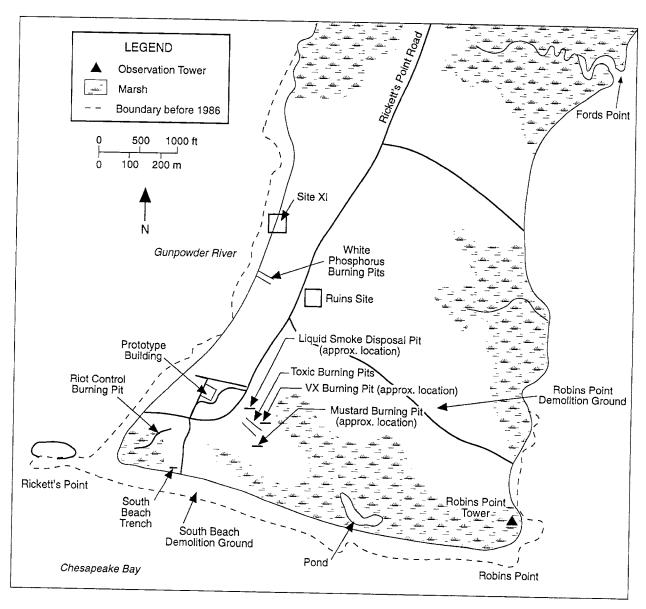


FIGURE 1.2 Locations of Principal Site Features at J-Field

# 1.3 ORGANIZATION OF THIS REPORT

Section 1 of this report presents background information and summarizes the purpose of the proposed action within the context of the RI/FS process. Section 2 describes the site history and environmental setting. The environmental setting includes site topography, soil, surface water, geology and hydrogeology, groundwater, climate, ecology, and land use. Also provided are an overview of available information about the nature and extent of contamination at the TBP area and a discussion of exposure pathways. Contaminants of concern (COCs) are also identified in Section 2.

Section 3 describes the overall objectives of the interim remedial action. Other site aspects discussed include applicable or relevant and appropriate requirements (ARARs), to-be-considered requirements (TBCs), and the development of interim preliminary remediation goals (PRGs). In

Section 4, potentially applicable technologies are screened on the basis of effectiveness, implementability and cost, and site-specific conditions. Section 5 describes and screens the preliminary interim remedial action alternatives.

Section 6 provides a detailed description of the final interim remedial action alternatives. These alternatives are evaluated in Section 7 against nine general criteria: overall protection of human health and the environment; compliance with ARARS; long-term effectiveness and performance; reduction of contaminant toxicity, mobility, and volume through treatment; short-term effectiveness; implementability; and cost. The discussion of state and community acceptance has been deferred and will be addressed in the Responsiveness Summary prepared for the Record of Decision (ROD) for the site. Section 8 presents a comparative analysis.

All references cited in this report are listed in Section 9, and a list of preparers is given in Section 10. Appendix A evaluates potential metal transport via leaching through the unsaturated zone at the TBP area. Appendix B provides tables summarizing preliminary identification of ARARs and TBCs for J-Field. Appendix C consists of data summary tables for the TBP area. Appendix D describes the RI characterization activities and sampling results at the TBP area. Appendix E explains the uptake model and approach used to develop ecological-based interim PRGs. The methodology and assumptions used to determine the costs of the interim remedial action alternatives are presented in Appendix F. Appendix G gives conversion factors between metric and English units of measure.

#### 2 SITE BACKGROUND

#### 2.1 SITE HISTORY

Disposal operations at the TBP area began in the 1940s and continued until about 1980. The pits were used most extensively between the late 1940s and 1960s. Items disposed of included bulk chemical wastes, drummed chemical wastes, high explosives (HE) (by OB/OD), nerve agents, incapacitating agents (also known as riot-control agents), chlorinated solvents, and blister agents (Nemeth 1989).

Information from interviews, sampling, geophysical surveys, and analysis of historical aerial photographs indicates that five disposal pits were used at the TBP area (Figure 2.1). The two existing (or main) burning pits, each covering an area of about 4,500 ft<sup>2</sup>, were most actively used for the disposal of various chemical agents and explosives. These pits are referred to as the northern and southern main pits in this report and are the only visible open pits in this area. Two other burning pits (the VX and mustard pits, Figure 2.1) are buried. The liquid smoke disposal pit, a small pit measuring 4 × 6 ft, has also been reported (Nemeth 1989); however, its specific location is not clear. In the HE demolition ground, near the southern edge of the site (Figure 2.1), HE munitions were disposed of by detonation (Nemeth 1989). The TBP storage area, a fenced area near the southwestern end of the mustard pit, was used for storage, as evidenced by historical aerial photographs taken in the 1960s (U.S. Army 1965). A scrap metal mound has been observed in the field near the southwestern portion of the site, adjacent to the marsh. This area (the TBP southwestern suspect burning area, Figure 2.1) was probably active in the 1950s and 1960s, as indicated by historical aerial photographs, and may have been used for burning and/or demolition. A small square pit, measuring about  $3.5 \times 4.25$  ft, lies between the southwestern suspect burning area and monitoring well P9. This feature is believed to be the liquid smoke disposal pit.

Procedures for open burning in the TBP area involved placing 3-4 ft of wood dunnage in a pit, placing the materials to be burned on top of the dunnage, adding fuel oil, and igniting it. Scrap metal items were removed and reburned in the same manner. Large metal items were recovered and disposed of as scrap. The depths of the pits were maintained by pushing burned soil and ash out toward the adjacent marsh. The areas where this material now resides are referred to collectively as the "pushout" area (Figure 2.1). The pushout area associated with the four burning pits occupies about 100,000 ft<sup>2</sup> and extends more than 100 ft into the marsh (Sonntag 1991). Currently, the pushout area shows obvious signs of disturbance, including an uneven surface, areas of bare ground, and disposal debris (e.g., rebar, sheet metal, and rusted pipes).

An agent known as DANC (decontaminating agent, noncorrosive) was commonly used to decontaminate scrap metal contaminated with mustard, Lewisite, and VX (Nemeth 1989). DANC

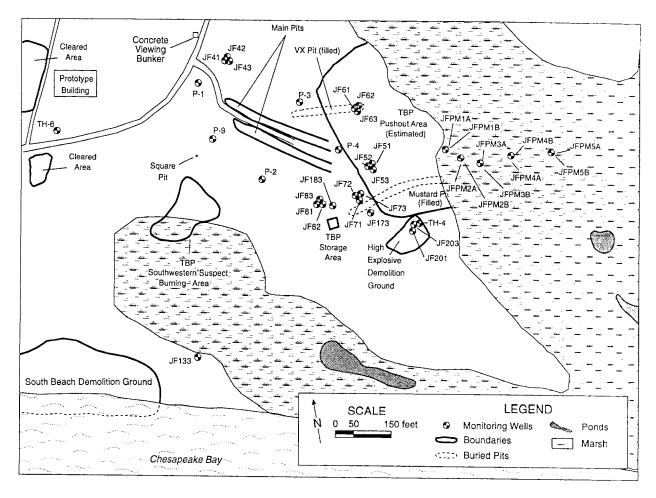


FIGURE 2.1 Locations of Main Features and Monitoring Wells at the TBP Area

is an organic N-chloroamide compound in solution with 1,1,2,2-tetrachloroethane (TCLEA); it typically contains 90-95% (by weight) TCLEA.

#### 2.2 SITE DESCRIPTION

## 2.2.1 Topography

J-Field is nearly flat, with a maximum relief of about 10 ft. The elevation in the TBP area ranges from about 14 ft above mean sea level (MSL) near the main pits to bay level at the shore. The adjacent marsh and freshwater pond are at an elevation of about 5 ft above MSL during wet periods.

#### 2.2.2 Soil

The APG was not mapped in the most recent Harford County soil survey (Smith and Matthews 1975). However, a previous survey by Perkins and Winant (1927) includes the APG. Their map indicates that the surface of J-Field comprises fairly equal areas of Elkton silt loam, Sassafras loam, and tidal marshes, with minor areas of Sassafras silt loam. Smith and Matthews (1975) describe the Elkton soil as slowly permeable and poorly drained. The Sassafras soils are moderately permeable and well drained. The tidal marsh soil is sandy to clayey, with peat or muck.

#### 2.2.3 Surface Water

Surface water features at J-Field include freshwater marshes, a marsh pond, and two unnamed streams that discharge into the Bush River (Figure 1.1). The freshwater marsh and pond along the southern shore near the TBP area are separated from Chesapeake Bay by sandy beaches. Because the beach acts as a dam, this marsh typically has water levels about 2 ft above high tide (Hughes 1993a). The large pond in this marsh has a maximum depth of about 5 ft. Other large marshes are present along J-Field's eastern and western shores. Water levels in these marshes are tidal influenced (Hughes 1993b). The streams along the eastern coast of J-Field are also affected by tides. Discharge is minimal except during heavy rainfalls (Hughes 1993a). The TBP area is located within the 100-year floodplain (Federal Emergency Management Agency [FEMA] 1986).

A tidal measuring station, located on Pooles Island (about 1 mi southeast of J-Field), shows that the difference between the mean high tide and the mean low tide is about 1.2 ft (National Oceanic and Atmospheric Administration [NOAA] 1993). The average level of the surrounding estuaries at Pooles Island is about 0.9 ft above MSL (Hughes 1993c). Vroblesky et al. (1989) determined an average bay level of 1.5 ft above MSL at an APG site north of J-Field. Hughes (1993b) suggested that the bay level measured by Vroblesky et al. was higher because the tide station for the study was located on an inland creek and not in the bay or estuary.

The depth of water in the Gunpowder River and Chesapeake Bay proper within 0.5 mi of J-Field's shores is generally shallower than 12 ft below mean lower low water (NOAA 1993). 1

# 2.2.4 Geology and Hydrogeology

The stratigraphy at J-Field consists of Pleistocene sediments of the Talbot Formation underlain by Cretaceous sediments of the Patapsco Formation (Potomac Group). The Pleistocene deposits are divided into three main units: a surficial, unconfined aquifer of interbedded sand, clay,

<sup>1 &</sup>quot;Mean lower low water" refers to the mean of the lower of the two daily low tides.

and silt; a confining unit of sandy and silty clay; and a confined sand and gravel aquifer (Hughes 1993a). On the basis of current knowledge of contaminant distribution, groundwater studies have focused on the surficial aquifer. The generalized stratigraphy in the TBP area is shown in Figure 2.2.

The deeper Talbot deposits fill a paleochannel eroded into the Patapsco sediments. Pleistocene paleochannels are common in the region (Kerhin et al. 1988), and similar features have been identified at other APG study areas (Lorah and Clark 1992; Vroblesky et al. 1989; Oliveros and Vroblesky 1989). Hughes (1991, 1992, 1993a) performed a marine seismic survey around the J-Field peninsula to determine the offshore extent of the geologic units. The seismic data suggest that the confining unit and the confined aquifer pinch out against the sides of a southwest-trending paleochannel. The width of this feature is approximately 1 mi.

Several monitoring wells were installed in the TBP area during the field investigations conducted at J-Field (Figure 2.1). One TH-series well (TH4) was installed near the TBP area by the U.S. Army Toxic and Hazardous Materials Agency (USATHAMA) during a 1977 environmental survey. Five additional wells (P1-4, 9) were installed by Princeton Aqua Science (1984). These wells

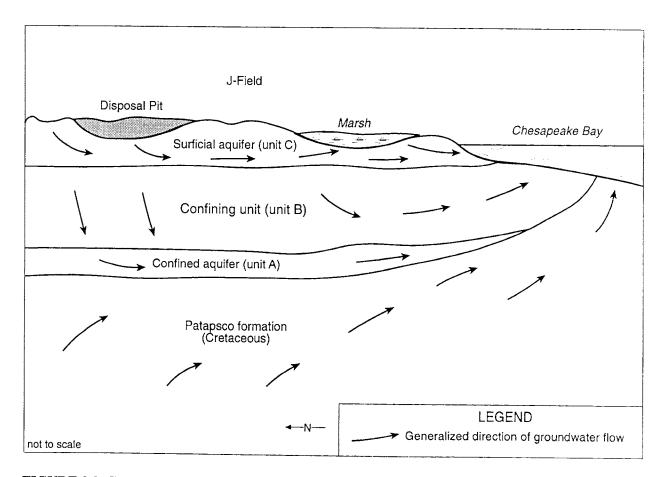


FIGURE 2.2 Generalized Hydrogeologic Cross Section and Direction of Groundwater Flow at J-Field TBP Area (Source: Modified from Hughes 1993a)

are all screened in the surficial aquifer to a maximum depth of 20 ft; they do not penetrate the confining unit.

The U.S. Geological Survey (USGS) installed five monitoring well nests (JF4-JF8), each consisting of three wells, in each of the three main Pleistocene units (Hughes 1993a). The well screens range in length from 3 to 5 ft. Hughes estimated hydraulic conductivities for these units based on slug tests performed in these and other J-Field monitoring wells.

In 1993, well JF173 was installed southeast of the main pits to monitor for dense, nonaqueous-phase liquids (DNAPLs) at the base of the surficial aquifer. The well is screened over a 5-ft interval, ranging from a depth of 26 to 31 ft (Patton 1994). Well JF183 was installed for use as a pumping well for the pumping test conducted in 1994; it is screened across the entire depth of the surficial aquifer.

In early 1996, wells JF201 and JF203 were installed adjacent to well TH4. Well JF203 is shallow, installed to replace the sediment-filled TH4. Well JF201 was installed to a depth of 165 ft to assess the confined aquifer.

## 2.2.4.1 Surficial Aquifer

The surficial aquifer is composed of medium- to fine-grained sand with interbedded clay (Hughes 1993a). The sand is generally red to gray in color; the clay is dark to light gray. The sandy facies are commonly silty or clayey. Hughes suggests that these sediments were deposited in an estuarine or marginal marine system during a period of higher sea level. A 10-ft vertical exposure of this unit can be viewed along the coast west of the TBP area. Here the stratigraphy is mainly reddish silt with fine sand. A 1-ft clay layer is also present. The depositional units appear to be continuous along the 50-ft length of the 10-ft-high cliff.

The total thickness of the surficial aquifer in the TBP area ranges from 30 to 40 ft. The USGS data suggest a basal contact elevation of 25-31 ft below MSL. Clay or silt units with thicknesses of up to 8 ft have been logged within this unit.

The hydraulic conductivity of the surficial aquifer has been estimated in several ways. Hughes (1993a) measured it to be in the range of  $1.0 \times 10^{-4}$  to  $3.7 \times 10^{-4}$  cm/s (0.29 to 1.04 ft/day), with a median of  $2.4 \times 10^{-4}$  cm/s (0.69 ft/day). These values are typical of silty sand or silt (Freeze and Cherry 1979). Because of the problems inherent in slug testing of wells, the results often underestimate the horizontal conductivity (Bradbury and Muldoon 1990; Hughes 1993b). During calibration of a draft modeling study, a value of  $2.8 \times 10^{-3}$  cm/s (8 ft/day) was assigned to this unit (Hughes 1993b). A pump test at well JF183, installed in 1994, indicated a hydraulic conductivity

of  $1.9 \times 10^{-3}$  cm/s (5.3 ft/day) (Quinn 1995). This value is consistent with a silty to clean sand (Freeze and Cherry 1979).

Despite the available data, the permeability of the surficial aquifer at the TBP area and throughout most of J-Field is uncertain. Because of the arrangement of interbedded facies, the permeability of the surficial aquifer varies over several orders of magnitude laterally and vertically. Any measurement of permeability is highly spatially dependent because of location-specific stratigraphy and well construction factors, including the depth interval of the well screen.

The flow rate produced by the pump test well was low. A yield of 2 gpm could not be sustained during a step drawdown test. The constant-rate pump test was completed with a flow rate of 1 gpm.

Continuous-head recordings available from the USGS for the 1993 water year indicate that the head in the surficial aquifer in the TBP area is more than 6 ft above MSL during a wet springtime and approximately 1 ft below MSL during August and September, when high evapotranspiration rates prevail.

The surficial aquifer receives most of its recharge by infiltration of precipitation; however, during dry periods, water levels in wells may be slightly lower than bay level (Hughes 1993a). A flow reversal may take place during these periods, with flow from the tidal estuaries providing recharge to the aquifer.

Under Maryland Code of Regulations 26.08.02.09, the surficial aquifer at J-Field is classified as a Type II(b) aquifer (Lemaster 1995). A Type II(b) aquifer has a transmissivity between 1,000 and 10,000 gal/day/ft, a permeability greater than 100 gal/day/ft<sup>2</sup>, and natural water with a total dissolved solids concentration of between 500 and 1,500 mg/L.

## 2.2.4.2 Confining Unit

The confining unit contains silty and sandy clay with shell fragments and varying amounts of organic matter. The clay is dark gray or olive gray in color. Hughes (1993a) interprets the sediment to be an estuarine deposit.

The basal contact elevation of the confining unit in the TBP area ranges from 64 ft to deeper than 145 ft below MSL (according to logs for well JF201). The thickness of this unit in the TBP area ranges from 35 ft in the northwestern portion of the area to greater than 125 ft in the southeastern portion. The thickness apparently depends on the depth of paleochannel erosion into the Potomac Group.

Slug tests performed by Hughes (1993a) on three wells in this unit yielded hydraulic conductivity values of less than  $3 \times 10^{-6}$  to  $7.0 \times 10^{-5}$  cm/s (0.01-0.20 ft/day), with a median value of  $1.8 \times 10^{-5}$  cm/s (0.05 ft/day). These values fall into the typical range of fine-grained materials (Freeze and Cherry 1979). Hughes (1993c) suggests that the slug test generally provides an estimate of horizontal conductivity, so the vertical conductivity may be much less. A value of  $1.8 \times 10^{-6}$  cm/s ( $5 \times 10^{-3}$  ft/day) used by Hughes (1993b) is comparable to that of Vroblesky et al. (1989), who modeled a nearby clay-filled estuarine paleochannel with a vertical hydraulic conductivity of  $6.8 \times 10^{-8}$  cm/s ( $2 \times 10^{-4}$  ft/day).

### 2.2.4.3 Confined Aquifer

The confined aquifer consists of gravelly sand with some clay and clayey sand. The confined aquifer is a fluvial deposit that blankets the bottom of a paleochannel identified by marine seismic and exploratory borings.

The thickness of the confined aquifer cannot be determined with USGS cluster well data because the confined aquifer monitoring wells do not fully penetrate the aquifer. Data from a deep USGS exploratory boring located near cluster JF8 (Figure 2.1) indicate that the base of the confined aquifer near the pits is about 120 ft below MSL (Hughes 1993a).

Slug tests were performed in four J-Field wells screened in the confined aquifer (Hughes 1993a). Hydraulic conductivity values determined for two wells completed in permeable portions of the aquifer were  $3.9 \times 10^{-2}$  and  $3.2 \times 10^{-1}$  cm/s (110 and 930 ft/day), with a median of  $1.3 \times 10^{-1}$  cm/s (390 ft/day). Hydraulic conductivity values determined for two other wells screened in fine-grained zones were  $1.1 \times 10^{-3}$  and  $1.8 \times 10^{-2}$ cm/s (3.2 and 52 ft/day), with a median of  $3.9 \times 10^{-3}$  cm/s (11 ft/day).

#### 2.2.4.4 Cretaceous Sediments

The Cretaceous Patapsco Formation sediments beneath J-Field are alluvial deposits of interbedded fine-grained sand and massive clay (Hughes 1993a). The Cretaceous clays are differentiated from Pleistocene clays by color in the USGS stratigraphic logs. Rather than the dark and olive gray of the confining unit, the Patapsco clays vary among reddish brown, light gray, white, light red, and olive yellow. The top of the Cretaceous sediments dips from 105 ft below MSL in the western portion of J-Field to 157 ft in the east (Hughes 1993a). Interpretation of gamma logs of the exploratory boreholes suggests that the sands range from 2 to 30 ft thick and that the clays range from 5 to 45 ft thick. Stratigraphic cross sections suggest that the various lenses are highly discontinuous laterally (Hughes 1993a). The Cretaceous deposits are underlain by metamorphic rock at a depth of about 800 ft.

#### 2.2.4.5 General Groundwater Flow

Field measurements of heads at J-Field indicate that groundwater flow is generally from topographically high areas to topographically low areas, such as marshes and estuary discharge areas (Hughes 1993a). In the TBP area, the water table has a local high in the area between the main pits and the Prototype Building. No pumping stresses are imposed on the Talbot aquifers at or in the vicinity of J-Field. The nearest pumping wells are 4 mi away, across the Gunpowder River (Hughes 1993a).

A comparison of the heads of wells screened in the three Pleistocene units indicates a downward gradient, suggesting leakage from the surficial aquifer into the confining unit and confined aquifer (Hughes 1993a). However, in wetlands and estuaries, flow is assumed to be vertically upward as groundwater flows from the deeper units to the discharge areas. The vertical head gradient within the surficial aquifer currently cannot be determined because only one well is screened in the surficial aquifer at each USGS well nest.

Hughes (1993a) describes the head distribution pattern in the confining unit and the confined aquifer as generally similar to that of the surficial aquifer. A low lateral head gradient was observed, so lateral flow is minimal. Discharge is believed to occur through the confining unit into the estuaries. During dry periods, flow reversals occur as the heads in the surficial aquifer fall slightly below those in the confined aquifer.

The tide in the bay influences water levels in some wells screened in the surficial aquifer at J-Field. The degree of influence depends on the distance from the shore and the presence of fine-grained confining or semiconfining units within the surficial aquifer. Continuous-head data are available from USGS for two wells (JF63 and JF43) in the TBP area. Stratigraphic logs for these wells indicate that they are screened in and overlain by fine sand (Hughes 1993a). The heads do not demonstrate the two highs and lows per day expected for tidally influenced wells. These data suggest that the distance from the shore to the TBP area is sufficient to negate tidal influence.

Hughes (1993a) did not observe tidal effects in wells screened in the confining unit. The confined aquifer, however, exhibited a strong tidal influence in continuous water-level data from six confined-aquifer monitoring wells. The heads fluctuate over a range of about 0.75 ft. Over a two-day period, the water levels in the wells show a high degree of similarity independent of distance from the shoreline. The head changes are therefore attributed to tidal loading on the confined aquifer.

#### 2.2.5 Climate

The Harford County area has a continental-type climate with significant temperature contrasts between winter and summer (Smith and Matthews 1975). The average daily high and low

are 65°F and 42°F, respectively. The average annual temperature is about 54°F. The average annual precipitation of 45 in./year is distributed fairly uniformly throughout the year.

### 2.2.6 Ecology

Ecological surveys were conducted at J-Field as part of the overall site ecological risk assessment (ERA), and the results of these surveys are presented in the draft ERA report (Hlohowskyj et al. 1996). The following subsections (2.2.6.1 through 2.2.6.4) summarize the ecological setting of the TBP area on the basis of the results presented in that report.

#### 2.2.6.1 Terrestrial Habitats

Terrestrial habitats at the TBP area include grassland and forested plant communities and represent less than 0.02% of the available terrestrial habitat at J-Field. The area immediately surrounding the main pits comprises disturbed habitats consisting exclusively of mowed and unmowed grassland with patches of bare ground. The areas of bare ground are located mainly in the pushout area, where a variety of disposal debris, such as rebar, sheet metal, and rusted pipes, is present. The grassland is bordered to the south, east, and north by marsh habitat, with a narrow band of upland forest occurring between the marsh and grassland habitat along the southern and southeastern boundary of the area. Dominant woody species in the forested habitats include black gum, sycamore, sweetgum, red and white oaks, and black tupelo.

## 2.2.6.2 Freshwater Habitats and Wetlands

The freshwater marsh southeast of the main pits and largely surrounding the pushout area is the major nontidal freshwater habitat in the southern portion of the Edgewood Peninsula. No other freshwater habitats are present at the main pits and pushout area. A large freshwater pond occurs near the center of the marsh. The pond is classified as an intertidal estuarine permanently flooded wetland (Hlohowskyj et al. 1996). The areal extent of the pond varies seasonally with changes in the water table and precipitation periods, averaging about 2.5 acres. No permanent surface water connections exist between the pond and Chesapeake Bay, although occasional connection is possible during very high tides and during extremely heavy rains. The marsh is classified as an intertidal estuarine wetland irregularly flooded by tides (Hlohowskyj et al. 1996). Common reed, the dominant plant species (Van Lonkhuyzen 1994), forms floating mats in some parts of the marsh. Palustrine forested and scrub/shrub seasonally flooded wetlands are also present along portions of the perimeter of the marsh, particularly along the southern and northern boundaries of the TBP area.

#### 2.2.6.3 Fish and Wildlife

More than 40 mammal, 22 reptile, and 15 amphibian species have been reported from the APG (USATHAMA 1993), and many of these species may occur at J-Field and use portions of the TBP main pits and pushout area.

Small mammals that have been collected from the TBP area include the white-footed mouse, meadow vole, and short-tailed shrew. Red fox and feral house cats have been observed at the TBP area, while bats have been observed flying over the site (Hlohowskyj et al. 1996). Although white-tailed deer, raccoon, muskrat, gray squirrel, and eastern chipmunk may occasionally forage at the site, the main pits and pushout area provide little or no habitat for these species.

The main pits and pushout area may provide seasonal or year-round habitat for a variety of reptile and amphibian species, primarily the American toad, northern spring peeper, spotted salamander, eastern box turtle, and a number of snake and lizard species; however, use of the main pits and pushout area by these species is most likely limited. In contrast, the marsh and pond represent good habitat for a variety of reptiles and amphibians, including bullfrog, leopard frog, eastern painted turtle, and common snapping turtle. The surrounding wooded areas also provide habitat for a variety of species.

Although more than 40 species of fish have been reported to occur in the APG area, most of these species are found in Chesapeake Bay and the Gunpowder and Bush Rivers. Fish collected from the marsh pond adjacent to the pushout areas include the bluespotted sunfish, banded killifish, spottail shiner, and golden shiner (Hlohowskyj et al. 1996).

The APG site is located along the Atlantic Flyway, a major migration corridor used by birds in spring and autumn. Because of the presence of the flyway and the diversity of habitat types in the vicinity, more than 140 species of birds have been reported from the APG (USATHAMA 1993), and many may use J-Field as foraging or nesting habitat. Qualitative auditory and visual surveys of birds were conducted in all seasons at the TBP area, and more than 110 species were reported from the site (Hlohowskyj et al. 1996). Birds reported for the area included mourning doves, American robins, eastern bluebirds, eastern kingbirds, and a variety of sparrows. Hawks that are known to nest in the J-Field area and may use the TBP area include the American kestrel and red-tailed hawk. Although osprey are known to nest at J-Field, it is unlikely that the osprey would feed on any of the fish species collected to date from the pond, primarily because of the small size of the fish.

A large variety of waterfowl occurs in the areas surrounding J-Field. The State of Maryland has designated Pooles Island, located about 1.5 km (1 mi) south of the Edgewood Peninsula and the TBP area, as a "Colonial Waterbird Nesting Site" (McKegg 1992). In addition, the open water areas north of the J-Field site have been identified by the state as a "Historic Waterfowl Staging and Concentration Area" (McKegg 1992). Waterfowl and shorebirds that have been reported from the

area and observed in the vicinity of the TBP area include mallard, tundra swan, wood duck, and great blue heron (Hlohowskyj et al. 1996). The small size of the pond, however, probably limits its use by large numbers of shorebirds or waterfowl.

# 2.2.6.4 Threatened and Endangered Species

No mammals, amphibians, or reptiles identified on the State of Maryland's rare, threatened, and endangered list have been reported for the J-Field area (Maryland Department of Natural Resources [MDNR] 1992). The bobcat, classified as in need of conservation by Maryland (MDNR 1992), has been reported for the general vicinity of APG (USATHAMA 1993) but not for the J-Field site. The eastern harvest mouse, listed by Maryland as endangered, historically has been reported for the area, but is now considered extirpated from the state (MDNR 1992). A number of bird species reported from the APG are state-listed species (MDNR 1992), including the barn owl (watchlist), sora (highly rare), dark-eyed junco (rare), and northern harrier (rare), and the grassland habitats at the pushout area and adjacent marsh provide suitable habitat for these species. The dark-eyed junco and the northern harrier were observed at the TBP area during the avian surveys conducted at the site. Table 2.1 identifies state and federally listed bird species that have been reported for APG.

TABLE 2.1 Bird Species of Special Federal and State Concern Reported at APG

Common Name	Scientific Name	State Status <sup>a</sup>	Federal Status
Bald eagle	Haliaeetus leucocephalus	Endangered	Threatened
Common barn owl	Tyto alba	Watchlist	Not listed
Common moorhen	Gallinula chloropus	Needs conservation	Not listed
Dark-eyed junco	Junco hyemalis	Rare	Not listed
Golden-crowned kinglet	Regulus satrapa	Rare	Not listed
Loggerhead shrike	Lanius ludovicianus	Endangered	C2 <sup>b</sup>
Northern harrier	Circus cyanus	Rare	Not listed
Peregrine falcon	Falco peregrinus	Endangered	Endangered
Sharp-shinned hawk	Aegolius acadicus	Highly rare	Not listed
Sora	Porzana carolina	Highly rare	Not listed
Yellow-bellied sapsucker	Sphyrapicus varius	Historical	Not listed
Yellow-rumped warbler	Dendroica coronata	Uncertain	Not listed

<sup>&</sup>lt;sup>a</sup> Source: MDNR (1992).

b C2 = Federal candidate for listing as a threatened or endangered species.

Several federally listed species are known to occur at the APG and may forage in suitable habitats near the TBP area (MDNR 1992; Wolflin 1992; USATHAMA 1993) (Table 2.1). The peregrine falcon is federally listed as endangered. No suitable habitat for the peregrine falcon occurs at the Edgewood Peninsula or the J-Field site; however, this species may be an occasional visitor to the area, especially during migration. The loggerhead shrike is a federal category 2 species that has been reported for APG and may forage in the TBP area. A federal category 2 species is a candidate for listing as either threatened or endangered. Neither the peregrine falcon nor the loggerhead shrike were observed at the TBP area during the avian surveys at the site.

APG supports the most significant concentration of bald eagles (federally listed as threatened) on the northern Chesapeake Bay (Wolflin 1992). A bald eagle nest site is located approximately 1.5 km (1 mi) north of the J-Field site. Occupation by the bald eagle is light to moderate on Pooles Island and the shoreline areas of the Edgewood Peninsula (Wolflin 1992). The bald eagle may forage on waterfowl at the marsh and pond adjacent to the pushout area but is unlikely to use the immediate TBP area because of the absence of suitable habitat and because of human activities at the site and surrounding areas.

#### **2.2.7** Land Use

The lower portion of the Edgewood Peninsula is generally dedicated to a military test range, with no resident population and only a limited number of site workers; however, the range fields offer restricted, seasonal accessibility to a limited number of hunters. Upland game hunting and bow hunting of deer and woodchuck are allowed along the northern border of (but not within) J-Field and farther north on the peninsula. Blinds for hunting deer with guns are scattered throughout the peninsula, including several locations in the northern portion of J-Field. However, the J-Field blinds are no longer available to hunters; hunting and trapping are no longer allowed within the J-Field boundary (Wrobel 1994). Seasonal hunting of migratory waterfowl is permitted along the entire shoreline of J-Field.

Although most of the AOCs are no longer used for OB/OD, a portion of the Robins Point Demolition Ground is currently active and is operating under interim status under RCRA. An open burning pan located 50 m west of the Prototype Building and an open detonation area at the White Phosphorus Pits area are also being used for emergency disposal operations.

Recent remedial activities at O-Field have required rerouting Robins Point Road through N-, D-, and I-Fields; the firing range; and H-Field; thus making J-Field less accessible. These changes in the main access route to J-Field will most likely preclude any future development of the J-Field site in the near term. In the long term, disturbances due to testing at H- and I-Fields make development of the J-Field site unlikely.

# 2.3 NATURE AND EXTENT OF CONTAMINATION

The following subsections summarize what is currently known about the nature and extent of contamination at the TBP area. Because the focus of this report is to address contaminated surface soil in the main pits and pushout area, the discussion of soil is most detailed. Site characterization data collected from 1993 to 1995 are presented in Appendix C. Appendix D, which is based on the draft RI report (Yuen et al. 1996), provides more details (including locations of samples) on the results of the RI at the TBP area.

#### 2.3.1 Soil

As part of the J-Field RI, soil samples have been collected from the TBP area and analyzed for various constituents. This discussion is divided into sections for the northern main pit, the southern main pit, and the pushout area east of the main pits to describe the contaminants found at each area. Two additional pits, the VX pit and the mustard pit, occur within the pushout area. These pits have been filled in and are partially covered by soil pushed out from the main pits. A brief discussion of these pits is included in Section 2.3.1.3.

#### 2.3.1.1 Northern Main Pit

Soil borings ranging in depth from 4 to 12 ft were drilled in six locations (JBP2-W, JBP2-C, JBP2-E, TBNPBOR1, TBNPBOR2, and TBNPBOR3; see Figure D.5) within the northern main pit as part of the RI. Samples taken from the borings indicate that contamination in the pit is not homogeneous. In general, contamination is higher in the western portion of the pit than in the eastern portion. The highest levels of volatile organic compound (VOC) contamination occur in sample TBNPBOR3 at the depth interval between 4 and 10 ft. Contaminants include acetone (up to 6,000  $\mu$ g/kg at 8-10 ft), tetrachloroethylene (TCLEE) (up to 750  $\mu$ g/kg at 6-8 ft), chlorobenzene (up to 23,000  $\mu$ g/kg at 4-6 ft), ethyl benzene (up to 6,600  $\mu$ g/kg at 6-8 ft), toluene (up to 4,200  $\mu$ g/kg at 6-8 ft), and total xylenes (up to 46,000  $\mu$ g/kg at 6-8 ft). Other VOCs, present at lower concentrations, include 1,1-dichloroethylene (11DCE), 12DCE, chloroform, TCLEA, and trichloroethylene (TRCLE). Semivolatile organic compounds (SVOCs) were also highest in sample TBNPBOR3 and include bis(2-ethylhexyl) phthalate (up to 1,900  $\mu$ g/kg at 6-8 ft), 2-methylnaphthalene (up to 2,100  $\mu$ g/kg at 6-8 ft), and naphthalene (up to 1,800  $\mu$ g/kg at 6-8 ft).

Metal contamination in soil underlying the northern main pit reflects a similar pattern as exhibited by the VOCs: contamination is highest in the western and central portions of the pit. The highest levels of heavy metal contamination occur in the upper 4 ft of soil. Metals include arsenic (up to 2,290 mg/kg), cadmium (up to 77 mg/kg), chromium (up to 240 mg/kg), copper (up to 7,120 mg/kg), lead (up to 4,790 mg/kg), and zinc (up to 17,800 mg/kg). In general, concentrations

of these metals decrease with depth, although elevated concentrations of metals were detected at depths of up to 12 ft. Toxicity characteristic leaching procedure (TCLP) analyses indicate high leachate levels of arsenic (155-216  $\mu$ g/L), lead (1,620-26,500  $\mu$ g/L), and cadmium (16.4-187  $\mu$ g/L). The soil is therefore considered hazardous (i.e., characteristically toxic) with respect to lead and cadmium.

Low levels of PCBs (<10 mg/kg at 0-2 ft) and dioxin/furan compounds (<  $5 \mu g/kg$  at 4-6 ft) were also detected in the western portion of the northern main pit. One sample (TBNPBOR3) had a total petroleum hydrocarbon (TPH) concentration of 19,000 mg/kg at 4-6 ft.

#### 2.3.1.2 Southern Main Pit

Soil borings ranging in depth from 4 to 10 ft were drilled in five locations (JBP1-W, JBP1-E, JHDP-C, TBSPBOR1, and TBSPBOR2; see Figure D.5) within the southern main pit as part of the RI. In general, the highest levels of VOC contamination are in the eastern portion of the pit. Although no VOC contamination was present at the surface, VOC concentrations were found to generally increase with depth. The highest concentrations of VOCs, found in the eastern portion of the pit, include 12DCE (8,400  $\mu$ g/kg at 10 ft), trans-12DCE (3,220  $\mu$ g/kg at 6 ft), 1,1,2-trichloroethane (112TCE) (8,540  $\mu$ g/kg at 6 ft), TCLEA (3,270,000  $\mu$ g/kg at 6 ft), TCLEE (25,700  $\mu$ g/kg at 6 ft), TRCLE (263,000  $\mu$ g/kg at 6 ft), and vinyl chloride (302  $\mu$ g/kg at 6 ft). SVOCs were highest in the upper 2 ft; these include N-nitrosodiphenylamine (up to 950  $\mu$ g/kg), hexachloroethane (up to 580  $\mu$ g/kg), and pyrene (up to 580  $\mu$ g/kg).

In general, heavy metals contamination is lower in the southern main pit than in the northern main pit; the highest levels occur in the upper 4 ft of soil. Metals include arsenic (up to 28.2 mg/kg), cadmium (up to 7 mg/kg), copper (up to 366 mg/kg), lead (up to 831 mg/kg), and zinc (up to 1,240 mg/kg). In general, metal concentrations decreased with depth in the southern main pit.

PCBs (e.g., Aroclor 1254) were found in the upper 4 ft in the eastern portion of the southern main pit (in sample JBP1-E, collected by Weston, Inc., in 1993). Sampling indicates a fairly localized area of PCB contamination in the surface soil.

#### 2.3.1.3 Pushout Area East of Main Pits

Surface soil samples collected by Weston, Inc., in 1992 indicate that VOC contamination is present in the upper 1 ft of the pushout area soil; these include acetone (up to 1,460  $\mu g/kg$ ), TCLEA (up to 13,200  $\mu g/kg$ ), and TRCLE (up to 26,000  $\mu g/kg$ ). SVOCs were also detected: hexachloroethane (up to 498  $\mu g/kg$ ), phenanthrene (up to 311  $\mu g/kg$ ), and pyrene (up to 215  $\mu g/kg$ ). Figure D.4 shows the locations of all samples collected within the pushout area.

Sampling in the pushout area also indicates that concentrations of heavy metals in the upper 4 ft exceed background concentrations (Table 2.2). The eastern portion of the pushout area exhibits the highest concentrations, although elevated concentrations were found throughout. In the upper 2 ft, metals include arsenic (up to 41 mg/kg), copper (up to 4,320 mg/kg), and lead (up to 94,000 mg/kg). TCLP analyses indicate that pushout area soil is hazardous with respect to lead. The vertical extent of metal contamination in soil is related to the thickness of the pushout material and increases toward the marsh. In general, metal concentrations decrease with depth.

Low levels of PCBs (Aroclor 1248, up to 3 mg/kg in sample JBPMA) were found in the upper 1 ft (Figure D.6). Low levels of pesticides were also found in the upper 1 ft (4,4'-dichloro-diphenyltrichloroethane [DDT], up to 177  $\mu$ g/kg in sample JBPMA). No explosives were detected in the surface soil samples.

Within the pushout area are two buried pits: the VX pit and the mustard pit (Figure 2.1). These pits were delineated during the RI by aerial photographic analysis (Yuen et al. 1996) and geophysical surveys (Daudt et al. 1994; Davies et al. 1995). Four surface soil samples (OT16A,

TABLE 2.2 Mean and Maximum Background Soil Concentrations of Metals Reported for Off-Site Areas Surrounding APG

	Concentration in Soil (mg/kg)		
Metal	Mean	Maximum	
Aluminum	7,940	17,300	
Antimony	3.14	<9.8	
Arsenic	2.57	5.29	
Barium	43.6	125	
Beryllium	0.44	1.42	
Calcium	534	1,980	
Chromium	16.8	68.9	
Copper	8.72	27.5	
Iron	12,300	23,500	
Lead	21.6	117	
Magnesium	1,010	3,920	
Manganese	276	1,140	
Mercury	0.04	<0.14	
Nickel	8.37	24.1	
Potassium	384	_ · · · -	
Selenium	0.21	1,700	
Zinc	37	0.497	
23110	31	242	

Source: ICF Kaiser Engineers (1995a).

OT16B, OT19A, and OT19B; see Figure D.4) were collected from two locations near the VX pit. Five borings (VXBOR1-VXBOR5; see Figure D.5), ranging in depth from 4 to 16 ft, were drilled into the VX pit as part of the RI. The nature of contamination at the VX pit was only partially characterized due to the potential presence of UXO (especially in the area of VXBOR3 and VXBOR4). On the basis of available data, the bottom of the VX pit is inferred to be about 4-6 ft below the ground surface.

Soil samples from the VX pit indicate that contamination is highest at the disposal center within the pit (near borings VXBOR3 and VXBOR4). In this area, surface soil is contaminated with moderate to high levels of heavy metals (including copper, up to 343 mg/kg; lead, up to 262 mg/kg; and zinc, up to 1,629 mg/kg), low levels of chlorinated ethenes and ethanes, petroleum-related compounds (including benzene, up to 42  $\mu$ g/kg; ethyl benzene, up to 2,900  $\mu$ g/kg; and xylenes, up to 2,300  $\mu$ g/kg), low levels of dioxins and furans, low levels of pesticides, 1,4-dithiane (a chemical warfare agent [CWA] degradation product), and phthalates. The vertical extent of contamination is estimated to be deeper than 6 ft. Near the disposal center, in areas near the western end of the pit, metal contamination is limited to the upper 2 ft, while organic contamination is minimal. The TPH content is high at depth.

Five surface soil samples (CLP6 [0-6 in.], CLP6 [6-24 in.], CLP7 [0-6 in.], CLP7 [6-24 in.], and CLP7 [24-48 in.]; see Figure D.4) were collected at two locations near the mustard pit. Three borings (HBOR1-HBOR3; see Figure D.5), ranging in depth from 10 to 16 ft, were drilled near the mustard pit (because of the potential presence of UXO, no borings were taken from within the pit). The bottom of the mustard pit is estimated to be about 4-6 ft below the ground surface.

Soil samples from the mustard pit indicate that heavy metals (including arsenic, up to 16 mg/kg; copper, up to 204 mg/kg; lead, up to 4,960 mg/kg; and zinc, up to 896 mg/kg) are present, especially in the upper 2 ft. Contamination decreases with depth and was not detected below 6 ft in the three borings. Very low levels of petroleum-related compounds and phthalates were detected in the upper 2 ft. Low levels of chlorinated ethanes and ethenes were detected in surface and subsurface soil. Low levels of CWA degradation products (diisopropylmethyl phosphonate and 1,4-dithiane) were also detected in samples taken at a depth of 6 ft.

#### **2.3.1.4 Summary**

Figures 2.3 and 2.4 illustrate the extent of metal contamination at two depth intervals (0-6 in. and 6-24 in., respectively), which exceeds the mean background values given in Table 2.2. The figures take into account the concentrations of several metals, including aluminum, arsenic, barium, cadmium, chromium, copper, lead, mercury, and zinc. The extent of contamination at these depth intervals is reduced when compared to maximum background values (Figures 2.5 and 2.6). Figure 2.7 shows an estimated area at 2-4 ft, which was inferred from limited soil sampling in the

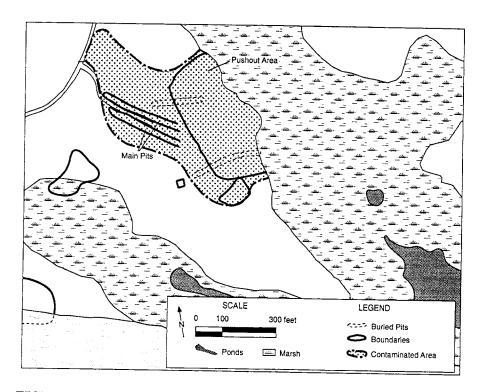


FIGURE 2.3 Estimated Contaminated Area at Depth Interval 0-6 in. with Heavy Metal Concentration Levels above Mean Background

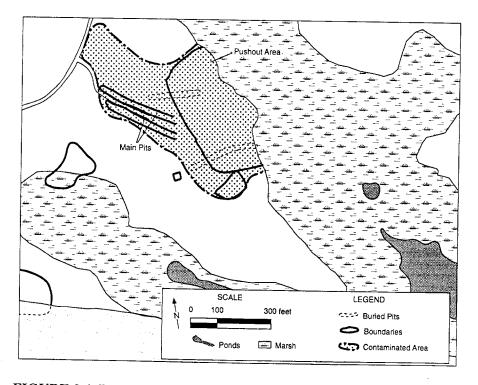


FIGURE 2.4 Estimated Contaminated Area at Depth Interval 6-24 in. with Heavy Metal Concentration Levels above Mean Background

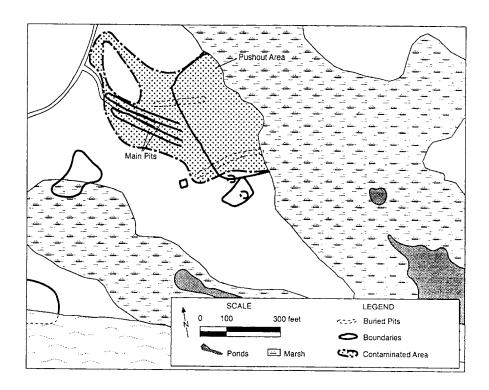


FIGURE 2.5 Estimated Contaminated Area at Depth Interval 0-6 in. with Heavy Metal Concentration Levels above Maximum Background

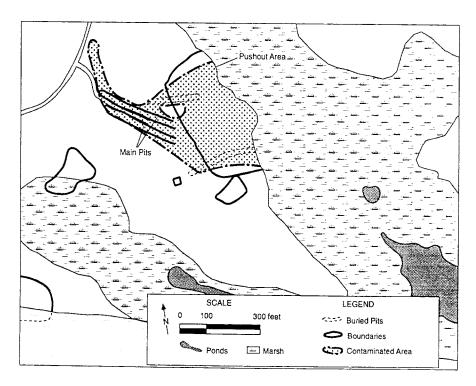


FIGURE 2.6 Estimated Contaminated Area at Depth Interval 6-24 in. with Heavy Metal Concentration Levels above Maximum Background

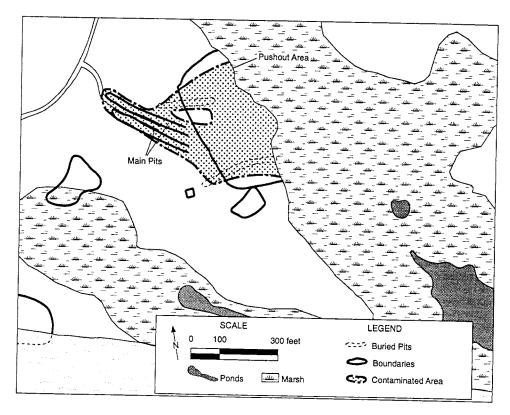


FIGURE 2.7 Estimated Contaminated Area at Depth Interval 2-4 ft with Heavy Metal Concentration Levels above Mean Background

interval of 2-4 ft. This estimate also considered the contamination present in the overlying soil and the likelihood for contaminant mobility. At the interval of 2-4 ft, VOC contamination is also present (and may also be present in low concentrations in the interval of 0-2 ft).

#### 2.3.2 Groundwater

Groundwater in the surficial aquifer in the TBP area is contaminated primarily with VOCs, particularly TCLEA (up to 260,000  $\mu$ g/L in well JF83), 112TCE (up to 2,000  $\mu$ g/L in well JF83), 12DCE (up to 37,000  $\mu$ g/L in piezometer JFPM1A), TRCLE (up to 41,000  $\mu$ g/L in well JF83), TCLEE (up to 3,400  $\mu$ g/L in well P-3), and vinyl chloride (up to 3,200  $\mu$ g/L in piezometer JFPM1A). Wells JF73 and JF83 and piezometer JFPM1A are located downgradient and about 200 to 400 ft from the two main pits.

#### 2.3.3 Sediment

Sediment samples were collected from the marsh as part of the RI (Yuen et al. 1996; see Figure D.7). Marsh sediments adjacent to the pushout area show significantly high levels of heavy metals in the upper 2 ft of the sediment column, including arsenic (up to 14 mg/kg), lead (up to 1,780 mg/kg), copper (up to 515 mg/kg), and zinc (up to 3,410 mg/kg). These contaminants correlate well with those found in the pushout area, suggesting that erosion via surface runoff is the mechanism of transport for these contaminants. Low levels of organic contaminants were also detected, including 4,4'-dichlorodiphenyldichloroethane (DDD) (up to 22  $\mu$ g/kg), 4,4'-dichlorodiphenyldichloroethylene (DDE) (up to 16  $\mu$ g/kg), 12DCE (45  $\mu$ g/kg), TRCLE (up to 29  $\mu$ g/kg), and vinyl chloride (up to 18  $\mu$ g/kg). In one surface sediment sample (TPSED1), 2,4-dinitrotoluene was found at a concentration of 2,110  $\mu$ g/kg. A variety of polyaromatic hydrocarbons (PAHs), including fluorene (410  $\mu$ g/kg), chrysene (1,200  $\mu$ g/kg), pyrene (up to 1,800  $\mu$ g/kg), anthracene (up to 280  $\mu$ g/kg), and the benzo- series (up to 2,300  $\mu$ g/kg), were found at a depth of 2-4 ft in sample SEDBOR7-1 (Figure D.7).

#### 2.3.4 Surface Water

Surface water samples from the adjacent marsh were also collected as part of the RI (Figure D.7). The samples were analyzed for organic compounds (VOCs, SVOCs, pesticides, and polychlorinated biphenyls [PCBs]) and inorganic materials (metals and cyanide). Several VOCs have been detected in the surface water, suggesting that contamination is migrating from the pit (source) areas via groundwater to the marsh. The organic contaminants detected include 12DCE (up to 1,700 µg/L), TRCLE (up to 3,615 µg/L), 112TCE (up to 138 µg/L), and TCLEA (up to 4,348 µg/L). SVOCs, pesticides, and PCBs were not detected. Lead (up to 1,590 µg/L), copper (up to 525 µg/L), and zinc (up to 4,040 µg/L) were also detected in several samples. The elevated concentrations of metals indicate that contamination has migrated to the marsh, most likely by surface runoff from the TBP and pushout areas.

# 2.4 CONTAMINATION SOURCES AND RELEASE MECHANISMS

Potential contamination sources identified in the RI include the two main pits, the mustard and VX pits, and their associated pushout areas; and a storage area, a disposal area, and an HE demolition area (Yuen et al. 1996; Figure 2.1). As described in the RI, surface and subsurface soil samples collected at the storage area did not indicate the presence of contamination. Soils from the HE demolition area showed concentrations of some metals above mean background levels but below maximum background levels (Table 2.2). The sources of contamination at the TBP area are the two main pits, the pushout area, the two buried pits within the pushout area (VX and mustard pits), and

the disposal area. The disposal area has not yet been fully characterized and is therefore not included in this FFS.

Release of contaminants from their primary source is most likely through lateral dispersion, surface runoff, leaching, and vertical infiltration. The lateral dispersion was enhanced by the pushout operations after each disposal/decontamination activity. Surface runoff is an important release mechanism for contaminants in the surface soil, especially in the pushout area. Leaching and infiltration are also important release mechanisms because fuel, liquid wastes (e.g., chlorinate solvents), and decontaminating agents (e.g., bleach and DANC) were used in the pits. These contaminants have been detected in groundwater downgradient of the site. A modeling study to evaluate the transport of metals through the vadose zone at the TBP area is presented in Appendix A.

## 2.5 POTENTIAL EXPOSURE PATHWAYS

#### 2.5.1 Human Exposure

Even under current conditions of restricted site access, a number of potential human exposure pathways have been identified at the J-Field study area (ICF Kaiser Engineers 1994b). These pathways, summarized in Table 2.3, include the exposure routes of dermal contact, ingestion of soil and water, and inhalation of vapors and particulate matter. The most likely human receptors include site workers, trespassers, or persons fishing along the J-Field shoreline. Because J-Field is situated in a restricted area with a wide range of physical security measures (including patrols by military police), the likelihood of trespassing at J-Field is considered low. However, the trespasser scenario is being retained in this study to comply with U.S. Environmental Protection Agency (EPA) guidance (1991a) requiring consideration of trespassers in the baseline risk assessment (BRA). Hunters are no longer considered likely human receptors because hunting at J-Field has been banned (Wrobel 1994).

Preliminary investigations indicate that the potential for most human exposure pathways at J-Field ranges from no (or negligible) potential for significant exposure to low potential (ICF Kaiser Engineers 1994b). Incidental soil ingestion and/or dermal contact with soil by on-site individuals may have a low to moderate potential for significant exposure because of the elevated chemical concentrations detected in surface soil in the vicinity of the TBP area. In addition, incidental ingestion and/or dermal contact with contaminated surface water or sediments has an unknown potential for significant exposure because of the limited data currently available. Infrequent contact with these media (surface soil, sediments, and surface water) and the use of protective clothing by on-site workers during remediation activities would be expected to prevent significant exposures.

TABLE 2.3 Potential Human Exposure Pathways at the TBP Area under Current Land-Use Conditions

Medium	Chemical Release Source	Exposure Pathway/Receptor <sup>a</sup>
Soil (surface, subsurface)	Previous dumping and disposal activities	Incidental ingestion and/or dermal contact with surface soil by personnel working at site, persons
	Aerial deposition from past explosions and fires	fishing, and trespassers
Groundwater	Leaching of chemicals from contaminated soil and buried wastes	Although groundwater is contaminated, no exposure pathway exists because there is currently no human uses of groundwater at this site or in downgradient areas
Surface water/ sediments	Groundwater discharge to Gunpowder River, Bush River, or Chesapeake Bay	Incidental ingestion and/or dermal contact by site workers, persons fishing, and trespassers
	Surficial runoff of chemicals from disposal areas	monning, and trespassers
Food chain (fish/shellfish)	Uptake from food sources that have bioaccumulated chemicals	Ingestion of contaminated fish and shellfish caught by local fishermen from the Gunpowder River, Bush
	Uptake of chemicals from exposure to contaminated media	River, or Chesapeake Bay near J-Field
Air (vapor-phase, particulate matter)	Volatilization from subsurface wastes or soil	Inhalation of contaminated dust and vapors by site workers, persons fishing along the J-Field shoreline,
,	Volatilization of chemicals from groundwater that has discharged to surface water	and trespassers
	Wind erosion of contaminated surface soil from disposal areas and spills	
	Dispersion and deposition of dust from destruction of explosive materials at Robins Point Demolition Ground	

<sup>&</sup>lt;sup>a</sup> Hunters are no longer considered likely human receptors because hunting at J-Field has been banned (Wrobel 1994).

Source: Adapted from ICF Kaiser Engineers (1994b).

For the purpose of an interim remedial action, it is assumed that the most likely human exposure pathway is incidental ingestion of contaminated surface soil by site workers at the TBP area.

#### 2.5.2 Ecological Exposure

Of the 35 species of aquatic and terrestrial biota identified as ecological receptors for the J-Field site (Hlohowskyj et al. 1995), 17 are appropriate for evaluation of ecological exposure and risk at the TBP area (Table 2.4). Exposure routes identified and evaluated in the J-Field ERA include exposure to and uptake from soil, surface water, and sediment.

For this FFS, contaminated surface soils represent the ecological media of concern at the TBP area and may affect 11 of the 17 receptors identified for this area (Table 2.4). The principal exposure pathways to the ecological receptors at the site are the incidental ingestion of contaminated soil by terrestrial biota, root uptake of contaminants by vegetation, and food-chain transfer of contaminants to higher trophic levels. This latter pathway represents the major contaminant route to the birds of prey that may use the site. For example, soil contaminants may be taken up by vegetation, which in turn is consumed by mice, which constitute a large portion of the diet of the red-tailed hawk.

#### 2.6 CONTAMINANTS OF CONCERN

# 2.6.1 Human Health Contaminants of Concern

COCs for soil were selected for the TBP area according to U.S. Environmental Protection Agency (EPA) Region III guidance for Superfund sites (EPA 1993a, 1995). Maximum chemical concentrations detected during the ongoing RI and earlier investigations of the TBP area were screened against EPA risk-based concentrations. EPA Region III guidance includes a database, updated semiannually, of approximately 600 contaminants in air, drinking water, fish tissue, and soil (EPA 1995). Risk-based concentrations for each contaminant were derived for the above exposure pathways corresponding to a hazard quotient of 1, or a lifetime cancer risk of 10<sup>-6</sup>, by using protective default exposure scenarios and the best available toxicity information (EPA 1993a).

Table 2.5 includes the maximum levels of all surface soil contaminants detected at least once in the TBP area. The maximum concentrations for surface soil (defined as 0-2 ft in depth) were screened against EPA's risk-based concentrations. Chemicals with maximum concentrations exceeding the risk-based concentrations were identified as COCs for surface-soil-related pathways

TABLE 2.4 Ecological Receptors for the TBP Area

Receptor	Habitat Type	Occurrence at the TBP Area	Exposure Point Media
Grasses (Andropogon spp.)	Upland	Pushout area, main pits area	Soil
Maple (Acer spp.)	Upland	Southern portion of area	Soil
Common reed	Wetlands	Marsh	Sediment
Phytoplankton and zooplankton	Quiet surface waters	Pond, marsh	Surface water
Golden shiner	Quiet surface waters	Pond, marsh, and marsh edges	Surface water
Leopard frog	Quiet surface waters	Marsh and pond margins	Surface water
Great blue heron	Surface waters and wetlands	Marsh and pond margins	Surface water
American kestrel	Semi-open grasslands	Pushout area, main pits area	Soil and surface water
Red-tailed hawk	Wetlands, grasslands, and forests	Entire area	Soil and surface water
American robin	Wetlands, grasslands, and forests	Entire area	Soil and surface water
Tree swallow	Grasslands with wooded edges or nearby woodlots	Entire area	Soil and surface water
Mallard duck	Wetlands with nearby grasslands	Pond, marsh, and pushout area	Soil, surface water, and sediment
White-tailed deer	Old fields, grasslands, and forests	Entire area	Soil and surface water
Muskrat	Surface waters and wetlands	Marsh, pond	Surface water and sediment
White-footed mouse	All upland habitats	Entire area	Soil and surface water
Eastern cottontail	Upland habitats and marsh edges	Entire area	Soil and surface water
Red fox	Wetlands, grasslands, and forests	Entire area	Soil and surface water

TABLE 2.5 Identification of Human Health–Based Contaminants of Concern for Soil (0-2 ft) in the TBP Area

Chemical	Maximum Concentration <sup>a</sup> (mg/kg)	Risk-Based Concentration <sup>b</sup> (mg/kg)	Contaminant of Concern
Volatile organic compounds			
Acetone	0.02	200,000	No
Benzene	0.1	200	No
2-Butanone	0.037	1,000,000	No
Carbon disulfide	0.038	200,000	No
Chlorobenzene	0.1	41,000	No
Chloroform	0.05	940	No
11DCE	0.10	9.5	No
trans-12DCE	0.17	18,000	No
Methylene chloride	0.10	760	No
TCLEA	2.5	29	No
TCLEE	1.0	110	No
Toluene	0.05	410,000	No
TRCLE	2.8	520	No
Xylenes	0.05	1,000,000	No
Semivolatile organic compounds			
Benzo(b)fluoranthene	1.25	7.8	No
Benzo(k)fluoranthene	1.25	78	No
Bis(2-chloroethyl)ether	1.25	5.2	No
Diethylphthalate	1.25	1,000,000	No
Fluorene	1.25	82,000	No
Hexachlorobenzene	3.1	3.6	No
Hexachloroethane	1.25	410	No
2-Methylnaphthalene	1.25	NA°	Yes
2-Methylphenol	1.25	100,000	No
4-Methylphenol	1.25	10,000	No
N-Nitrosodiphenylamine	0.95	1,200	No
Phenol	1.25	1,000,000	No
Pyrene	1.25	61,000	No
2,4,6-Trichloroaniline	7.9	170	No
Pesticide/PCB			
Aroclor 1248	0.57	0.74 <sup>d</sup>	No
Explosives			
Nitroglycerin	15.3	NA	Yes

TABLE 2.5 (Cont.)

Chemical	Maximum Concentration <sup>a</sup> (mg/kg)	Risk-Based Concentration <sup>b</sup> (mg/kg)	Contaminant of Concern
Inorganic materials			
Aluminum	22,600	1,000,000	No
Antimony	501	820	No
Arsenic	1,440	3.8	Yes
Barium	1,580	140,000	No
Beryllium	1.38	1.3	Yes
Cadmium	35.5	1,000	No
Calcium	36,000	_e	No
Chromium	878	10,000	No
Cobalt	108	120,000	No
Copper	4,320	82,000	No
Cyanide	120	41,000	No
Iron	154,000	610,000	No
Lead	94,200	400 <sup>f</sup>	Yes
Magnesium	3,880	_e	No
Manganese	633	10,000	No
Mercury	3.6	610	No
Molybdenum	6.9	10,000	No
Nickel	84.5	41,000	No
Potassium	1,460	_e	No
Selenium	7.12	10,000	No
Silver	41.9	10,000	No
Sodium	521	_e	No
Thallium	19.3	$160^{g}$	No
Vanadium	32.7	14,000	No
Zinc	17,800	610,000	No

Maximum detected concentrations for surface soil samples (0-2 ft in depth). These values are taken from currently available data, excluding non-Contract Laboratory Program data collected by Weston.

<sup>&</sup>lt;sup>b</sup> From EPA (1995), for "industrial" soil.

<sup>&</sup>lt;sup>c</sup> NA = not available.

<sup>&</sup>lt;sup>d</sup> Risk-based concentrations for PCB isomers are based on Aroclor mixtures.

e Risk-based concentrations for these compounds are not available due to low toxicity.

<sup>&</sup>lt;sup>f</sup> EPA currently recommends a soil lead cleanup value of 400 mg/kg for residential land use but site-specific pharmacokinetic modeling for nonresidential (adult) screening, when necessary (EPA 1994).

<sup>&</sup>lt;sup>g</sup> A risk-based concentration was not available for the class of thallium compounds, so the value for thallium sulfate was used instead.

of human exposure. In addition, except for several inorganic materials commonly found in soil (noted on Table 2.5), chemicals without a risk-based concentration were also included as COCs.

Although this approach is intended for use in BRAs (EPA 1993a), it is a conservative screening methodology that should be appropriate for identifying human health COCs for interim remedial action at the TBP area. The risk-based concentrations were derived by using conservative default exposure scenarios (EPA 1991a) and the most recent reference doses and carcinogenic potency slopes. Consequently, the risk-based concentrations "represent relatively protective environmental conditions at which EPA would typically not take action" (EPA 1993a). EPA Region III's risk-based screening approach was also used to select COCs at J-Field for determining soil concentrations that would be protective of human health, especially for site remediation workers (ICF Kaiser Engineers 1994a).

In addition, the COC screening methodology assumes a scenario of surface exposures to site personnel and remediation workers (i.e., the use of maximum surface soil concentrations and ingestion of industrial soils). This approach is not intended to take into account potential contaminant leaching to the groundwater. In that case, soil screening levels for groundwater use (recently developed by EPA) would be used to screen for COCs. Because the available soil screening levels for transfers from soil to groundwater are often orders of magnitude lower, there would be additional COCs to evaluate if the groundwater pathway were considered (e.g., VOCs). Similarly, human health—based interim PRGs would be correspondingly lower. However, the surface exposure scenario developed in this report is appropriate for a removal action, supplemented by a BRA and sitewide FS that considers potential contaminant leaching to groundwater.

## 2.6.2 Contaminants of Ecological Concern

The contaminants of ecological concern (COECs) in soils were identified for the TBP area by using data collected as part of the RI for J-Field (Yuen et al. 1996). These contaminants were identified by comparing measured soil concentrations with a number of chemical-specific factors, including background concentrations and screening benchmark values. The detection frequency, capacity to bioaccumulate or bioconcentrate, and importance as a micro- or macronutrient were also considered in the selection process. The procedure for selecting the COECs follows the general approach recommended by EPA in the Human Health Evaluation Manual (EPA 1989a) and EPA Region III guidance for identifying COCs (EPA 1993a) and is consistent with the screening approach used for the TBP area human health risk assessment (see Section 2.6.1 and ICF Kaiser Engineers 1995b).

Contaminant characterization data for soil were first evaluated with respect to analytical methods, detection limits, quality control (QC) samples, and blanks. Details regarding this portion

of the screening process are presented in the J-Field ERA report (Hlohowskyj et al. 1996). The following steps were then performed, in order, by using the remaining soil data for the TBP area:

- The detection frequency of each contaminant was evaluated, and all contaminants with detection frequencies of 0% were eliminated from further consideration in the ERA.
- Contaminant concentrations were compared to reported background concentrations (ICF Kaiser Engineers 1995a). The one-tailed nonparametric Mann-Whitney test (Zar 1984) was employed to test for similarity between background and site concentrations, and contaminants were retained for further screening if they were found to significantly (p < 0.05) exceed background concentrations. However, if the sample size for a contaminant from a particular site and medium was less than or equal to 3, a statistical comparison was not possible. In these instances, the maximum reported concentration was compared directly to the maximum reported background concentration, and a contaminant was retained for further screening if the reported maximum concentration exceeded the maximum background level.
- Chemical contaminants were compared to screening values. These values represent media concentrations considered to be protective of biota. Contaminants present at maximum concentrations exceeding screening concentrations were retained as the final COECs.
- Chemicals that were present at maximum concentrations within background levels but greater than the screening values were retained as final COECs.

Background concentrations used in the screening process are those identified for the region (ICF Kaiser Engineers 1995a). Soil screening values were obtained from a number of sources, including the EPA Region III screening levels, Opresko et al. (1994), Will and Suter (1994), NOAA/HAZMAT (undated), and the open scientific literature. Although soil benchmarks considered protective of human health are available (e.g., EPA 1993a), these were not used in the screening process.

The contaminants identified as the final COECs and carried through the remainder of the ERA were those with maximum concentrations that exceeded background and/or screening level concentrations. The screening process also considered essential plant and animal nutrients and bioconcentration and bioaccumulation potential. The final list of COECs for the TBP area soils is presented in Table 2.6.

TABLE 2.6 Contaminants of Ecological Concern for TBP Area Soils

Acetone	Cobalt	Nickel
Aluminum	Copper	Nitroglycerin
Antimony	Cyanide	N-Nitrosodiphenylamine
Aroclor 1248	1,1-Dichloroethene	Phenol
Arsenic	Diethyl phthalate	Potassium
Barium	Fluorene	Selenium
Benzene	Hexachlorobenzene	Silver
Benzo(b)fluoranthene	Hexachloroethane	1,1,2,2-Tetrachloroethane
Benzo(k)fluoranthene	Iron	2,4,6-Trichloroaniline
Beryllium	Lead	Trichloroethene
Bis(2-chloroethyl)ether	Magnesium	Vanadium
2-Butanone	Mercury	m & p Xylene
Cadmium	2-Methylnaphthalene	Zinc
Carbon disulfide	2-Methylphenol	
Chromium	4-Methylphenol	

## 2.7 CONCURRENT STUDIES

## 2.7.1 Remedial Investigation

The suspected contamination sources at the TBP area include the two main pits, the VX and the mustard pits, and their associated pushout areas; and a storage area, a disposal area, and an HE demolition ground. The specific location of the liquid smoke disposal pit is uncertain. Because the two main pits are visible at the surface today, their extent is easy to determine. Examination of historical aerial photographs indicates that their extent has not changed significantly over time. The remaining pits (VX, mustard, and liquid smoke) have been buried. The approximate locations of the VX and mustard pits have been identified with the aid of historical aerial photographs. The results of ground-penetrating radar (GPR) surveys conducted in the area of the VX pit correlate well with the location identified in the aerial photographs. Additional geophysical surveys and soil sampling will be conducted to verify the locations of all other pits and associated unloading zones, which are considered potential groundwater contamination sources.

Characterization of the TBP area is continuing so that long-term remedial alternatives may be developed for the entire site during the FS. The nature and extent of soil contamination in the remaining pits, the storage area, the disposal area, and demolition area are currently under investigation as part of the RI.

### 2.7.2 Human Health Risk Assessment

Human health risk assessments were recently conducted for J-Field (ICF Kaiser Engineers 1995a, 1996). The J-Field risk assessment focuses on potential exposure to chemical contaminants at hazardous waste disposal areas within J-Field, except for soil, surface water, and sediment at the TBP area, which were evaluated in a separate report (ICF Kaiser Engineers 1995a). Draft reports of these assessments have been released for public comment. The objective of the risk assessment is to determine the potential for and extent of human health risks resulting from contaminants at J-Field. Human exposures to chemical contaminants detected at J-Field or migrating from the waste disposal sites were evaluated specifically for the following environmental media: groundwater, soil, air, surface water, and sediment.

The approach for conducting the human health risk assessment at J-Field is outlined by ICF Kaiser Engineers (1994b). The preliminary list of potential receptors and human exposure pathways (which is based on a review of the site background and history) was reevaluated. Because the scope of the human health risk assessment is broad, additional field and laboratory data from the RI were needed to fully characterize the potential for human health risks. The data generated during the RI at J-Field were analyzed statistically to determine the significance of chemical concentration data. After the collection and statistical analysis of data, the BRA was performed to estimate human health impacts resulting from past activities at J-Field. Human health risks for each complete pathway were evaluated according to EPA guidance for BRAs (EPA 1989a,b). A similar approach was used for the TBP area baseline human health risk assessment (ICF Kaiser Engineers 1995a), which evaluated soil, surface water, and sediment data.

As stated in EPA guidance, risk-based remediation goals are initial guidelines that need to be reviewed and possibly modified upon completion of the BRA or when additional information is available (EPA 1991b). To reduce uncertainties and support the development of final remediation goals, the review should include exposure assumptions, future land use, and media and contaminants of potential concern.

## 2.7.3 Ecological Risk Assessment

An ERA (Hlohowskyj et al. 1996) was recently completed for the TBP area. This risk assessment included the following:

• Evaluations of species abundance and community composition that used quantitative and qualitative surveys of wetland and upland vegetation and terrestrial and aquatic invertebrate and vertebrate biota;

- Quantitative evaluations of physiological parameters of soil invertebrates, such as enzyme activity and respiration rates;
- Quantitative evaluations of processes mediated by soil invertebrates, such as litter decomposition and nitrogen mineralization; and
- Toxicity tests of site soils, sediments, and surface waters on a variety of invertebrates, vertebrates, and plants.

Surveys of the aquatic invertebrate community at the TBP area marsh and pond showed a diverse community representing a wide variety of taxa. Except in the site nearest the pushout area, the benthic fauna in the pond and marsh represented communities expected to occur in such habitats. The fish community was dominated by two species, one of which was present in very large numbers. No fish collected exhibited any external evidence (lesions, ulcers, fin rot, exopthalamus) of contaminant effects or other environmental stressors.

The vertebrate surveys showed a very diverse bird community inhabiting or using the site; more than 100 species were identified. No individuals exhibited any obvious external abnormalities. Four amphibian species and four native mammal species (white-footed mouse, meadow vole, short-tailed shrew, and red fox) were collected or observed, and none of the specimens exhibited any external abnormalities (lesions or tumors).

The survey results showed reduced invertebrate abundance in the disturbed soils of the area. Total macroinvertebrate numbers, bacterial and fungal biomass, and nematode numbers were significantly lower, particularly at the pushout area, than those measured at on-site and off-site reference areas. The trophic structure of the nematode community at the pushout area was also different from that observed at the reference area.

The activity of several bacterial and fungal nutrient-acquiring enzymes in the pushout area was significantly lower than that at the reference site. Enzyme activity was significantly and negatively correlated with the total metal content of the soil. Substrate-induced respiration and soil nitrogen dynamics were also lower in the pushout area than in the reference location.

Toxicity testing of aquatic media showed no acute toxicity of surface water from the pond or marsh, but chronic toxicity was indicated for surface water from the marsh. This toxicity was restricted to surface water collected from the marsh immediately next to the pushout area; it was manifested as growth inhibition of the floating vascular plant *Lemna* and reduced survival and growth in larval fish. No chronic toxicity was detected for surface water from the pond. Toxicity testing of sediments indicated bacterial (Microtox) inhibition, and some tests showed an increase in mortality of the amphipod *Hyalella*. The sediments exhibiting toxic effects were all collected from along the boundary between the pushout area and the marsh. No toxicity was detected in sediments

collected from the pond. The distribution of the measured surface water and sediment toxicity suggests that the toxicity of these media is a result of contaminated surface soil and precipitation running off from the pushout area to the marsh. Testing of groundwater from the surficial aquifer at the TBP area showed both acute and chronic toxicity to a variety of test organisms, including zooplankton, vascular plants, amphibians, and larval fish.

Soils from the TBP area had lethal and sublethal effects on earthworms and vegetation. Soils from the southern main pit and the pushout area resulted in nearly 100% mortality in earthworms, and significant weight loss in worms was detected in soil mixtures containing more than 25% of site soil. Toxicity testing evaluating seedling emergence, growth, and survival in lettuce showed seedling emergence rate  $\leq 2.5\%$  for soils from the pushout area and southern main pit and seedling emergence rate  $\leq 75\%$  from other areas of the TBP area.

Risk estimation based on uptake modeling (Appendix E) indicates that the soil contamination at the TBP poses a substantial risk to several receptors. Aluminum, antimony, arsenic, cadmium, chromium, cyanide, lead, mercury, selenium, and zinc were each found to pose an extreme risk (environmental effects quotient [EEQ] > 100) for at least one modeled ecological receptor. However, the exposure point concentrations used to model the uptake of particular contaminants were frequently based on the maximum detected concentrations for the site. In actuality, the concentrations to which receptors would likely be exposed at the site range from levels much lower than the exposure point concentrations (e.g., minimum detected values) to the maximum detected concentration. Thus, the results present a conservative view of risks likely to be incurred from contamination at the site (i.e., more likely to detect risks).

Extreme and moderate risks from soil contamination are indicated to terrestrial vegetation from 7 soil contaminants, and moderately high to extreme risk is indicated to a variety of terrestrial receptors for 13 COECs. Extreme risks were identified for 8 COECs (7 metals and cyanide) and 3 receptor species, while moderate to high risks were identified for 11 COECs (10 metals and cyanide). Low risks were identified for only two COECs, chromium and trichloroethene. The large number of receptors for which high or extreme risks were identified, together with the results of the effects assessment, supports the risk characterization that soil from the TBP area poses a high risk to ecological resources.

The EEQ and weight-of-evidence approaches for risk estimation indicate that the TBP area poses a high risk for adverse ecological impacts. This risk is associated primarily with contaminated soils in the pushout area and pits and is related primarily to heavy-metal concentrations rather than PCBs, pesticides, VOCs, or SVOCs. The contaminated soils of the pushout area are also the probable primary source of the contamination and high risk identified in surface water and sediment along the marsh–pushout area boundary (Hlohowskyj et al. 1996). The lack of suitable benchmark values for several of the receptor species precluded the calculation of EEQ risk values for a number

of the organic COECs. However, the absence of these risk estimates does not affect the conclusion that the TBP area soils pose a high risk for adverse ecological impacts.

The high risk identified for the TBP area may be significant at a local scale for most terrestrial wildlife that use the site. Adverse impacts were identified for several ecological variables and across multiple taxa and trophic levels. These impacts should largely be restricted to biota that occur within the TBP area boundaries and not extend to other areas of J-Field or APG. However, potential impacts on wide-ranging biota may affect wildlife populations that are not restricted to the J-Field boundary. The use of the site by migratory waterfowl and avian predators, such as the redtailed hawk and American kestrel, is of particular concern. Consequently, the TBP area poses risks that would be ecologically significant, given the potential importance of the surrounding marsh to waterfowl and the implications that impacts on raptor populations could have on control of prey populations in the J-Field area.

## 2.7.4 Sitewide Feasibility Study

A sitewide FS is being conducted for J-Field. The purpose of the FS is to gather sufficient information to develop and evaluate alternative remedial actions to address contamination at J-Field AOCs, including the TBP area groundwater.

# 3 INTERIM REMEDIAL ACTION OBJECTIVES

#### 3.1 OBJECTIVES AND SCOPE

The overall objectives of the proposed interim remedial action at the TBP area of J-Field are to:

- Reduce exposure of human and environmental receptors to surface contamination in three source areas: the two main pits and pushout area,
- Minimize the potential for contaminant migration via sediment transport from these areas,
- Minimize the potential for contaminant migration via downward leaching through these areas, and
- Support long-term site remediation.

The source areas of contamination at the TBP area are described in Section 2.4. Three of these source areas are the subject of this FFS: the northern main pit, the southern main pit, and the pushout area (Figure 2.1). Preliminary results of human health and ecological risk assessments (Section 2.7) indicate that contaminated surface soil in these areas should be addressed to protect human health and the environment. An interim remedial action could be implemented that would achieve the objectives stated above.

Technologies that could be applied to address contaminated surface soil in the main pits and pushout area are discussed in Section 4. Figures 3.1 and 3.2 show the areas of surface soil contamination to be addressed by the proposed action. Preliminary area and volume estimates for these areas are presented in Table 3.1. Volumes apply only to the excavation scenarios proposed under Alternatives 3, 4, and 5. Area of contamination applies to all the action alternatives, including

TABLE 3.1 Estimated Areas and Volumes of Contaminated Surface Soil at the Main Pits and Pushout Area

Contaminated Surface Soil Interval <sup>a</sup>	Contaminants of Concern	Area (ft²)	Volume (yd³)
0 to 6 in.	Metals, organics Metals, organics	218,300	4,043
6 to 24 in.		208,600	11,590

<sup>&</sup>lt;sup>a</sup> See Figure 3.1 for location of excavation area.

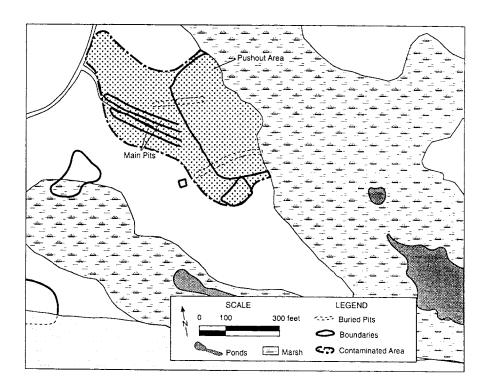


FIGURE 3.1 Contaminated Surface Soil, 0-6 in., to Be Addressed by the Proposed Action at the TBP Area

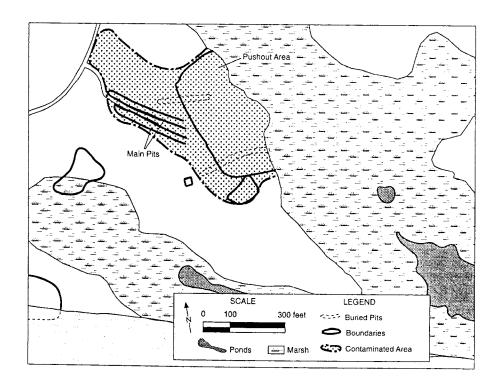


FIGURE 3.2 Contaminated Surface Soil, 6-24 in., to Be Addressed by the Proposed Action at the TBP Area

the covering of soil as proposed in Alternative 2 and the excavation scenarios proposed under Alternatives 3, 4, and 5. The areas depicted in Figures 3.1 and 3.2 represent the areas with metal concentrations exceeding the mean regional background as reported by ICF Kaiser Engineers (1995a). The methodology used to derive this estimate is presented in Section 2.3. The areas depicted in the figures also cover the areas with levels of contaminants exceeding the interim PRGs reported in Table 3.5. The interim PRGs developed for surface soils at the TBP area (Section 3.3) were used to determine which source areas at the TBP area should be addressed as part of the interim action and which could be appropriately left in their current condition until further action is taken at the site as part of the long-term remediation for J-Field. Other source areas and contaminated media (groundwater, surface water, sediment, and subsurface soils) at the TBP area are outside the scope of this interim remedial action and will be addressed as part of the long-term remediation for J-Field (Section 2.7.4).

# 3.2 COMPLIANCE WITH REGULATORY REQUIREMENTS

In accordance with Section 121 of CERCLA and the National Oil and Hazardous Substances Pollution Contingency Plan (NCP), the proposed action would be implemented in accordance with ARARs. Identification of ARARs is site-specific and is defined as follows:

- Applicable requirements are cleanup standards, standards of control, and other substantive environmental protection requirements, criteria, or limitations promulgated under federal or state environmental or facility siting laws that specifically address a hazardous substance, pollutant, contaminant, remedial action, location, or other circumstance at a CERCLA site. Only those state standards that are identified by a state in a timely manner and that are more stringent than federal requirements may be applicable.
- Relevant and appropriate requirements are cleanup standards, standards of control, and other substantive environmental protection requirements, criteria, or limitations promulgated under federal or state environmental or facility siting laws that, while not "applicable" to a hazardous substance, pollutant, contaminant, remedial action, location, or other circumstances at a CERCLA site, address problems or situations sufficiently similar to those encountered at the CERCLA site that their use is well suited to the particular site. Only those state standards that are identified by a state in a timely manner and that are more stringent than federal requirements may be relevant and appropriate.

As described in EPA guidance, ARARs can be divided into three categories: contaminant-specific, location-specific, and action-specific. Contaminant-specific ARARs address certain chemical species or classes of contaminants and relate to the allowable limits of contaminant

concentrations in various environmental media (soil, groundwater, surface water, air). Location-specific ARARs are based on the specific setting and nature of the site, such as location in a floodplain and proximity to wetlands. Action-specific ARARs relate to specific response actions (i.e., excavation or soil washing/leaching) that are proposed for implementation at the site.

In addition to ARARs, the NCP provides for the use of other advisories, criteria, or guidance TBCs. These are advisories, criteria, and standards that are issued by the federal or state regulatory body but not legally binding because they have not been properly promulgated. The identification of TBCs is not mandatory; however, they are to be used as appropriate to complement the ARARs.

Potential ARARs for the proposed action at the TBP area are identified on the basis of the nature of contamination, the site location, and the proposed activities (including both federal and State of Maryland requirements). A comprehensive list of potential ARARs is presented in Appendix B.

The NCP acknowledges the EPA's policy of responding by distinct operable unit at a site rather than waiting to take one consolidated response action (EPA 1990a). Therefore, this interim action has a limited scope, and requirements are ARARs only when they pertain to the issues to be addressed and the specific activities to be conducted under the interim action. For instance, this interim action addresses only the excavation and removal of contaminated soils and does not concern groundwater remediation, which is within the scope of a concurrent RI/FS. The interim action would also contribute to improving migration of contaminants to surface waters but does not address surface water remediation. Therefore, Appendix B does not address ARARs outside the scope of soil excavation, treatment, and disposal.

Appendix B identifies federal and state location-specific requirements for permitting and approvals for activities taking place in wetlands and floodplains (e.g., the movement of soils, grading, and fill discharge). Location-specific requirements also address impacts on wildlife and cultural, historic, or archeological sites.

Federal standards and guidelines are available for only a few of the soil contaminants of concern at the J-Field TBP area: metals (e.g., lead), dioxin, and PCBs. These soil contaminants are governed by RCRA and the Toxic Substances Control Act (TSCA), respectively. The management and disposal of soils contaminated with lead or arsenic are governed by the Maryland RCRA program. No Maryland State PCB standards or guidelines are established; therefore, federal standards for management and disposal of PCB-contaminated soils would be contaminant-specific ARARs.

Under RCRA regulations, any waste that fails the TCLP is a characteristic hazardous waste. The soil in the TBP area was analyzed and found to be characteristically hazardous for lead and

arsenic (i.e., the TCLP extract exceeded 5 mg/L). Because the operations at the TBP ceased before 1979, if the soil is left in place and no current actions are taken that constitute treatment, storage, or disposal as defined by RCRA, RCRA land disposal restriction (LDR) regulations would not be applicable to the TBP. However, if lead- or arsenic-contaminated characteristic hazardous waste is to be excavated and land-disposed, it must meet the RCRA LDR regulations and, thereunder, must be treated to meet an extraction procedure (EP) or TCLP concentration of less than 5 mg/L prior to disposal. Therefore, Maryland's EPA-authorized RCRA program regulations and standards will govern any activities concerning excavated contaminated soils, including any storage, treatment, or disposal activities.

The EPA recognized early on that RCRA constituent-contaminated soil is different than hazardous wastes generated in a process (as-generated wastes) and that treatment of soil should also be different. Therefore, the treatment standards for soil were delayed until May 8, 1992. An additional 1-year extension was granted and has since expired. Therefore, hazardous waste soil is now subject to the same treatment standards as as-generated wastes.

In the proposed Phase II LDR rule, the EPA included provisions to establish constituent-specific treatment standards for soils contaminated with hazardous wastes. However, after receiving a number of comments, the EPA issued a subsequent Federal Register Notice (58 FR 59,976, November 12, 1993) that announced its intention to not include soil LDRs in the Phase II LDR final rule. In the preamble of the Phase II LDR final rule, the EPA reiterated its position that "treatment standards for as-generated wastes are generally inappropriate or unachievable for soils contaminated with hazardous wastes" (59 FR 47,980, September 19, 1994). Now the EPA has issued its proposed Hazardous Waste Identification Rule — Media (61 FR 18,779, April 29, 1996).

This proposed rule would establish modified LDR treatment requirements and modified permitting procedures for higher-risk contaminated media that remain subject to hazardous waste regulations and give the EPA and any state authorized to implement the Hazardous Waste Identification Rule — Media the authority to remove certain lower-risk contaminated media from regulations as a hazardous waste, including minimum technological requirements. Once adopted, this rule would govern the storage, treatment, and disposal of hazardous waste—contaminated soils at remediation sites overseen by either the EPA or an authorized state. As proposed, this rule would also supersede the Corrective Action Management Unit (CAMU) regulations. Therefore, adoption of this rule would affect the management of any hazardous waste soils excavated after the rule's effective date.

However, until a new regulation is finalized, Maryland RCRA regulations, including LDRs, will apply to any hazardous waste soils excavated for storage, treatment, or disposal. Upon excavation, any soils that are hazardous wastes by definition (i.e., either characteristic waste due to lead contamination) must be stored pursuant to RCRA storage regulations and treated and disposed of pursuant to RCRA LDR regulations. Under the Phase II Universal Treatment Standards, hazardous

constituents other than those that cause wastes to exhibit a characteristic are "underlying hazardous constituents." These underlying hazardous constituents may also have to be treated to specified maximum concentrations prior to land disposal. Underlying hazardous constituents are defined as any constituent listed in 40 Code of Federal Regulations (CFR) 268.48, except zinc, that can reasonably be expected to be present at the point of generation of the hazardous waste at a concentration above the constituent-specific Universal Treatment Standard [40 CFR 268.2(i)]. Therefore, if any constituent listed in 40 CFR 268.48 is expected to be present in the waste soil in excess of the listed treatment standard concentration, it must also be treated before land disposal can take place.

In light of the uncertainty surrounding the treatment of hazardous waste soils, the EPA has presumed that a treatability variance will be used to comply with LDRs for the disposal of contaminated soil. EPA has published several guides for soils subject to treatability variances, including alternative treatment standards (EPA 1990b). The contaminated soil will need to be treated to meet either the concentration range or the percent reduction range for each hazardous constituent listed. A threshold concentration is used to determine which alternative standard to apply. The concentration range would be applied if the waste sample constituent concentration exceeded the threshold, and the percent reduction range would be applied if the threshold were not exceeded.

If an alternative cannot meet the identified ARARs, it may be assessed to determine whether the alternative meets the grounds for invoking a CERCLA waiver provision. The waiver provisions allow an alternative to not attain all ARARs where the alternative is an interim measure that will become part of a total remedial action that will attain ARARs [40 CFR 3000.430(f)(1)(ii)(C)].

TSCA regulations set forth standards for PCB-contaminated soils once they are excavated and there is a need to manage, store, treat, and dispose of them. TSCA also addresses cleanup standards and guidelines for contaminated soil left in place. Under TSCA regulations, materials contaminated with PCBs in concentrations exceeding 50 ppm (mg/kg) must be stored in specially designed facilities, and disposal must be either by incineration or in a chemical waste landfill at an EPA-approved facility. Under EPA's PCB spill cleanup policy (40 CFR Part 761 [Subpart G]) for spills of materials contaminated with greater than 50 ppm PCBs in restricted-access areas, soil within the spill area must be excavated and backfilled with soil containing PCBs in concentrations less than 25 ppm.

The PCB spill cleanup policy (40 CFR 761.120) applies only to spills taking place after the effective date of the regulation and is not a mandatory standard; therefore, the policy is not an ARAR but a TBC. However, according to EPA guidance, the policy sets forth the standards and guidelines that are used by EPA to determine PRGs for CERCLA sites (EPA 1990a). No state standards or guidelines exist for PCB-contaminated soils.

PCBs have been found to contaminate soils at the TBP area in concentrations exceeding 50 ppm; therefore, TSCA regulations for the management, storage, treatment, and disposal of PCBs will govern interim actions concerning such PCB-contaminated soil, and the PCB spill cleanup policy will be an important TBC in determining the cleanup standard.

# 3.3 INTERIM PRELIMINARY REMEDIATION GOALS

The development of risk-based PRGs early in the RI/FS process can aid in identifying appropriate remedial action alternatives at a site. The site-specific information required to develop risk-based PRGs usually consists of (1) the COCs, (2) the concentrations of the COCs by medium, (3) the human and ecological receptors, and (4) the probable future land use. For the TBP area, human health-based interim PRGs for soil were developed by using the EPA Region III derived risk-based soil concentrations (Table 3.2). EPA Region III derives the risk-based values following

TABLE 3.2 Human Health–Based Interim Preliminary Remediation Goals (PRGs) for Surface Soil at the TBP Area

Contaminant	Interim PRG (mg/kg)
Semivolatile organic compound	
2-Methylnaphthalene	NAª
Explosive	
Nitroglycerin	NA
Inorganic materials	
Arsenic	3.8 <sup>b</sup>
Beryllium	1.3
Lead	400°

<sup>&</sup>lt;sup>a</sup> NA = Not available.

b Human health-based interim PRG for arsenic is based on the EPA Region III risk-based concentration for arsenic as a carcinogen. The noncarcinogenic concentration for arsenic is 610 mg/kg.

<sup>&</sup>lt;sup>c</sup> EPA (1994) currently recommends a soil lead cleanup value of 400 mg/kg for residential land use but site-specific pharmacokinetic modeling for nonresidential adult screening, when necessary.

the Superfund risk assessment guidelines (EPA 1989a) and standard default assumptions (EPA 1991a); the values are based on carcinogenic and noncarcinogenic effects of ingestion of "industrial" soil (as opposed to residential soil). The interim PRGs for the TBP area are initial guidelines and are not intended to set final cleanup levels or establish that cleanup to meet these goals is warranted (EPA 1991b, 1994).

EPA has not developed risk-based soil concentrations for ecological resources similar to the human health-based PRGs. Two approaches were employed to develop ecological risk-based interim PRGs for the TBP area: (1) use of the contaminant uptake models developed for the ERA to back-calculate acceptable surface soil concentrations and (2) use of the human health-based interim PRGs for COCs for which uptake modeling was not performed. In this latter approach, risk reduction for ecological resources is considered to be directly correlated with the reduction of contaminant concentrations or with removal of contaminated media. Thus, reducing surface soil contaminant concentrations to the levels of the human health risk-based interim soil PRGs would also reduce risk for ecological resources. In both approaches, each derived interim soil PRG was screened against the regional background soil concentration of that contaminant, and the greater of the two values was selected as the interim PRG.

In the ERA for J-Field (Hlohowskyj et al. 1996), uptake modeling was used to estimate a daily uptake or dose (the applied daily dose [ADD]) for each COEC. Potential risk to an ecological receptor was estimated by comparing the modeled ADD value with a benchmark value representing a safe daily uptake. The ratio of the ADD to the benchmark value is termed the environmental effects quotient (EEQ) and is analogous to the hazard quotient used for estimating human health risks. For a particular contaminant, a potential risk is indicated for all values of the EEQ exceeding 1.00. Details regarding the uptake models and estimation of the EEQ are presented in Appendix E. It was possible to develop EEQ risk estimates only for those contaminants for which suitable benchmark values were available.

For the TBP area, the J-Field ERA modeled contaminant uptake and developed EEQ risk estimates for terrestrial vegetation and 11 vertebrate receptor species. Risk estimates exceeding values of 1.00 were identified for aluminum, antimony, arsenic, barium, cadmium, chromium, cobalt, copper, cyanide, lead, mercury, selenium, silver, zinc, and trichloroethene for terrestrial vegetation and six of the modeled terrestrial receptors (Table 3.3). To derive the ecological risk-based interim PRG values, the uptake models were used to identify contaminant surface soil concentrations that would limit the EEQ risk values for each receptor species to less than 1.00, and the lowest estimated soil concentration among all the receptors was selected as the interim PRG value. The species-specific interim surface soil PRGs derived for the TBP area are shown in Table 3.4.

TABLE 3.3 Ecological Effects Quotient Risk Estimates of Soil Contaminant Concentrations for Selected Ecological Receptors at the TBP Area<sup>a,b</sup>

			Ecologi	Ecological Effects Quotient	ent		
Contaminant	Vegetation	White-Footed Mouse	Eastern Cottontail	White-Tailed Deer	American Robin	American Kestrel	Red-Tailed Hawk
Aluminum	NAc	97.1	263	12.7	2.02	0.01	0.13
Antimony	100	50.7	138	5.98	NA	V V	CI.S
Arsenic	144	113	352	16.6	2.59	0.06	79.0
Barium	1.60	4.50	24.2	1.01	9.90	0.03	0.27
Cadmium	3.60	24.8	110	3.88	5.94	0.02	0.13
Chromium	251	1.21	3.61	0.15	7.30	0.03	0.13
Cobalt	4.30	NA	NA	NA	N A	3 Z	CC:0
Copper	43.2	3.79	19.1	0.80	9.40	0.03	0.22
Cyanide	NA	46.3	125	5.24	Z Z	CO.O	77.0 VA
Lead	942	103	507	21.3	1.527	5 20	40.4
Mercury	NA	16.2	117	5.53	68.2	0.20	10:1
Selenium	7.10	51.0	139	5.99	12.1	90:0	0.33
Silver	21.0	NA	NA	ŅĀ	Ϋ́	N N	S.S.
Trichloroethene	NA	1.37	3.46	1.02	NA	Y Z	C Z
Zinc	1,780	3.94	22.8	96.0	981	3.29	20.1

<sup>&</sup>lt;sup>a</sup> EEQ values exceeding 1.00 indicate potential risk to an ecological receptor. Values between 1.0 and 10 indicate low potential risk, values between 10 and 50 indicate moderate potential risk, values between 50 and 100 indicate high potential risk, and values exceeding 100 are indicative of extreme potential risk to ecological receptors.

<sup>&</sup>lt;sup>b</sup> EEQ values are shown only for those contaminants of ecological concern for which the EEQ values exceeded 1.00 for at least one receptor.

<sup>&</sup>lt;sup>c</sup> NA = A suitable benchmark value for this contaminant is not available for this receptor, and it is not possible to calculate an EEQ value at this time.

TABLE 3.4 Estimated Surface Soil Contaminant Concentrations Necessary to Achieve a No-Effects Level (100% Risk Reduction) for Selected Ecological Receptors that Use the TBP Area<sup>a</sup>

		Š	urface Soil Co	Surface Soil Contaminant Concentration (mg/kg)	ntration (mg/k;	g)	
		White-Footed	Eastern	White-Tailed	American	American	Red-Tailed
Contaminant	Vegetation	Mouse	Cottontail	Deer	Robin	Kestrel	Hawk
Aluminum	$NR^b$	$ m NP^c$	NP	NP	11,084	NR	NR
Antimony	5.00	9.55	3.36	80.1	$NA^d$	NA	NA
Arsenic	10.0	12.3	6.7	82.3	556	NR	NR
Barium	1,000	349.4	64.4	1,557	159	NR	NR
Cadmium	10.0	1.4	0.316	9.1	0.9	NR	NR
Chromium	3.50	727	242	NR	120	NR	NR
Cobalt	25.0	NA	NA	NA	NA	NA	NA
Copper	100	1,137	224	NR	459	NR	NR
Cyanide	NA	2.59	96.0	22.9	NA A	NA	NA
Lead	100	006	181	4,358	58.7	18,083	2,323
Mercury	NA	0.221	0.029	0.637	0.052	NR	3.0
Selenium	1.00	NR	NR	1.18	NR	NR	NR
Silver	2.00	NA	NA	NA	NA	NA	NA
Trichloroethene	NA	1.3	NP	1.9	NA	NA	NA
Zinc	10.0	4,516	776	NR	16.7	5,393	878

<sup>&</sup>lt;sup>a</sup> For each contaminant and receptor, the soil concentration identified will give a modeled applied daily dose to the ecological receptor that limits the ecological effects quotient (EEQ) to less than 1.00, resulting in a potential 100% reduction in risk.

NR = Current levels of soil contamination pose no risk to the receptor as determined by the modeled uptake and a resultant EEQ value less than 1.00.

NP = Because of contaminant uptake from exposure routes from media other than soil, a 100% risk reduction is not possible from remediation of soil alone.

<sup>&</sup>lt;sup>d</sup> NA = Suitable benchmarks for this contaminant are not available for this receptor, and it is not possible to calculate an EEQ value or an interim PRG at this time.

The ecological risk-based PRGs should be used with care. Because of the assumptions and uncertainties associated with the uptake models used to predict ADD values, the PRG values for some receptors may be greatly overprotective. The uptake models use a COEC exposure point concentration that is either the maximum concentration reported from the TBP area or the 95% upper confidence limit (UCL) of all the reported concentrations. For most of the contaminants, the maximum reported concentration was used as the exposure point concentration for the uptake modeling. The use of either value as the exposure point concentration is consistent with EPA Region III guidance (Davis 1994) and results in a very conservative modeled daily contaminant uptake. For the uptake modeling, the exposure point concentration is assumed to be homogeneously distributed across the entire site, and the resultant EEQ risk estimate represents the potential risk to ecological receptors from a uniform, sitewide contaminant concentration. In reality, the actual occurrence and distribution of the exposure point concentration (maximum or 95% UCL) is restricted to specific locations within the TBP area, such as the pits and trenches, and concentrations at most locations within the TBP area are much lower than the exposure point concentrations. Thus, the EEQ risk estimates are likely greatly overestimated, and the ecological risk-based PRGs are likely greatly overprotective for most if not all the contaminants and receptors.

Modeling contaminant uptake also requires exposure factor information on species-specific ecological and physiological parameters, such as body weight, ingestion rate, and diet. In contrast to the exposure factor data available for human health risk assessments, species-specific exposure factors for ecological receptors are largely unavailable. The exposure factors used to estimate contaminant uptake, and the benchmark values used to calculate the EEQs, were either (1) species-specific but not population specific, (2) not species-specific but from related taxa, or (3) developed with empirically derived allometric equations. Each of these sources adds to the uncertainty of the ADD and EEQ risk estimates and the interim PRGs.

Additional uncertainty with the uptake modeling that may affect the PRG values is related to the assumptions regarding contaminant transfer between trophic levels and contaminant assimilation. The ADD, EEQ, and PRG values were all derived assuming 100% contaminant transfer and assimilation; however, for most biota, it is unlikely that contaminant transfer or assimilation is 100% efficient. So, this assumption likely overestimates the true degree of contaminant uptake by the modeled receptors and, thus, the ADD and EEQ; therefore, the PRG values are probably overprotective.

The final uncertainty associated with the contaminant models that may result in overly protective PRGs is the nature of the benchmark values used for estimating the EEQ risk values. The benchmark values used were reported in the literature to produce no observable adverse effects in the exposed species. However, benchmark values specific to the species of concern at the TBP area were not available for all the COECs, and benchmark values were extrapolated between species as necessary. In some cases, uncertainty factors (Davis 1994) were used to minimize uncertainty

associated with benchmark extrapolations between species of the same class and between species of different classes (Appendix E).

For each COEC, the lowest PRG among those developed for all the receptors was chosen as the interim ecological risk-based PRG (Table 3.5). For some COECs, the interim PRG soil concentrations are less than the reported regional background concentrations, and attaining the PRGs for some COECs would require remediation to below background levels. For those COECs with interim PRGs below background levels, the background concentrations were selected as the ecological risk-based interim soil PRGs. However, remediation to background levels would produce risk reductions in excess of 97% for all the COECs except aluminum (Table 3.5). Regional background aluminum concentrations are indicated to pose a high risk to ecological resources. However, remediation to background levels of this COEC would result in a 64% reduction in the current risk level identified for the TBP area.

For those contaminants with insufficient available data to model an ADD, calculate an EEQ risk estimate, and derive an ecological risk-based interim soil PRG, the human health-based interim soil PRGs were used as the default ecological risk-based interim PRGs.

Table 3.6 integrates the ecological and human health risk-based PRGs and presents an overall list of preliminary interim soil PRGs for the TBP area. The identified interim soil PRG for each COC is the lower of the human health and ecological risk-based interim PRGs presented in Tables 3.2 and 3.5, respectively. These interim soil PRGs are initial guidelines and are not intended to set final cleanup levels or establish that cleanup to meet these goals is warranted (EPA 1991b, 1994). Rather, they provide a basis for the initial evaluation of remedial alternatives and for the development of remedial boundaries at the TBP area.

TABLE 3.5 Ecological Risk-Based Interim Preliminary Remediation Goals (PRGs) for Soils at the TBP Area

Current Risk Reduction by Using Background- Based PRG (%)	64.1	> 100°	> 100	> 100	0.66	98.1	> 100	> 100	> 100	> 100	9.86	97.0	> 100	V V	8.66
Background EEQ Risk Level	99.4	0.08	0.71	99.0	1.07	4.79	0.29	0.09	0.67	0.40	1.67	4.12	0.16	NA	3.70
Mean Background Soil Concentration (mg/kg)	7,940	3.14 <sup>d</sup>	2.57	43.6	0.34	16.7	7.25	8.72	$NA^f$	21.6	0.04	0.20	$0.31^{d}$	NA	37.0
Current EEQ Risk Level <sup>b</sup>	263	137	352	24.2	110	251	4.3	43.2	125	1,527	117	139	21	3.46	1,780
Ecological Risk-Based Interim PRG (mg/kg) for 100% Risk Reduction <sup>a</sup>	$NP^c$	3.7	6.7	64.4	0.32	3.5	25.0	100	1.0	58.7	0.03	0.05	2.0	1.3	10.0
Contaminant	Aluminum	Antimony	Arsenic	Barium	Cadmium	Chromium	Cobalt	Copper	Cyanide	Lead	Mercury	Selenium	Silver	Trichloroethene	Zinc

<sup>&</sup>lt;sup>a</sup> Predicted contaminant soil concentration necessary to limit the ecological effects quotient (EEQ) to less than 1.00, resulting in a potential 100% risk reduction.

<sup>&</sup>lt;sup>b</sup> Predicted EEQ risk values estimated by using current contaminant levels detected in soils at the TBP area.

<sup>&</sup>lt;sup>c</sup> Greatest risk was identified for the eastern cottontail (Table 3.3). Because of the contaminant uptake via ingestion of drinking water, a 100% risk reduction is not possible for this species from soil remediation alone. Remediation to background levels will result in a 64% risk reduction for this species. Similarly, 100% risk reduction from soil remediation is not possible for the white-tailed deer or the white-footed mouse. Remediation to background aluminum concentrations would result in a 100% risk reduction for the American robin but risk reductions of only 56% and 64% for the white-tailed deer and white-footed mouse, respectively.

Reported background concentrations were all below detection limits; mean background concentrations were estimated by using one-half the reported method detection limits (ICF Kaiser Engineers 1995a).

Risk reduction is greater than 100% when the background EEQ value is less than 1.00, while risk reduction equals 100% when the EEQ equals 1.00.

NA = Benchmark value is not available, so estimation of EEQ is not possible at this time.

TABLE 3.6 Risk-Based Interim Preliminary Remediation Goals (PRGs) for Surface Soil at the TBP Area

Contaminant	Interim PRG (mg/kg)
Volatile and semivolatile organic compounds	
Trichloroethene	1.3ª
2-Methylnaphthalene	NA <sup>b</sup>
Explosive	
Nitroglycerin	NA
Inorganic materials	
Aluminum	7,940°
Antimony	3.7 <sup>a</sup>
Arsenic	3.8 <sup>d</sup>
Barium	64.4 <sup>a</sup>
Beryllium	1.3 <sup>d</sup>
Cadmium	0.34 <sup>c</sup>
Chromium	16.7°
Cobalt	25.0 <sup>a</sup>
Copper	$100^{a}$
Cyanide	$1.0^{a}$
Lead	58.7/400 <sup>e</sup>
Mercury	0.04 <sup>c</sup>
Selenium	0.20 <sup>c</sup>
Silver	$2.0^{a}$
Zinc	37.0°

<sup>&</sup>lt;sup>a</sup> Interim PRG is ecological risk-based (see Table 3.5).

b NA = No interim PRG developed. Insufficient data to develop ecological risk-based interim PRG, and no EPA Region III human health-based PRG has been developed (see Table 3.2).

<sup>&</sup>lt;sup>c</sup> Ecological risk-based PRG value is below mean background concentration; interim PRG is mean background concentration.

<sup>&</sup>lt;sup>d</sup> Interim PRG is human health-based (see Table 3.2).

<sup>&</sup>lt;sup>e</sup> The ecological risk-based PRG is 58.7 mg/kg (see Table 3.5). EPA currently recommends a soil cleanup value of 400 mg/kg for residential land use (see Table 3.2). The cleanup goal for lead has been as high as 1,000 mg/kg at other APG sites (Wrobel 1995).

## 4 IDENTIFICATION AND SCREENING OF TECHNOLOGIES

Alternative interim remedial actions for the TBP area at J-Field were developed by identifying remedial technology types and process options that are potentially applicable to addressing contaminated soil. The technologies considered in selecting interim remedial action alternatives for contaminated soil in the TBP area include those identified in the NCP (EPA 1990a). These technology types and process options were screened for applicability to the site in accordance with EPA guidance (EPA 1988).

# 4.1 CRITERIA FOR IDENTIFYING AND SCREENING TECHNOLOGIES

The criteria for identifying potentially applicable technology types and process options are provided in EPA guidance (EPA 1988) and the NCP (EPA 1990a). Technologies identified in this section were screened on the basis of site-specific conditions and the current understanding of the TBP area at J-Field. Section 121 of the Superfund Amendments and Reauthorization Act of 1986 (SARA) identifies a strong statutory preference for remedial actions that are highly reliable and provide long-term protection. The primary requirements for a selected remedy are that it protects human health and the environment and meets the objectives of the proposed action in a cost-effective manner. Additional selection criteria include the following:

- In preferred remedies, the principal element is treatment to permanently or significantly reduce the toxicity, mobility, or volume of hazardous substances, pollutants, or contaminants.
- Where practical treatment technologies are available, off-site transport and disposal without treatment is the least preferred alternative.
- Permanent solutions and alternative treatment technologies or resource recovery technologies should be addressed and used to the maximum extent practicable.

These criteria have been considered in identifying and screening technologies to determine the appropriate components of remedial action alternatives for the TBP area.

The remedial action objectives for the TBP area are described in Section 3. The current understanding of contaminants and conditions at the site suggests the general response actions that could be implemented to achieve these objectives are institutional controls, in-situ containment, removal, treatment, short-term storage, and disposal. Specific application of these technologies to site conditions was evaluated to determine which would be most appropriate for interim remedial

action at the TBP area. These technologies were screened on the basis of effectiveness, implementability, and cost, as defined by the following factors:

- Effectiveness in terms of protecting human health and the environment in both the short term and the long term;
- Implementability in terms of technical and administrative feasibility and resource availability; and
- Cost in a comparative manner (i.e., low, moderate, or high) for technologies of similar performance and/or implementability.

These screening criteria were applied only to the technologies and general response actions being evaluated; combinations of technologies to address site-specific contamination problems were evaluated after the technologies were assembled into alternatives. This evaluation is presented in Section 5.

The no-action response was also included in this evaluation to provide a baseline for comparison; it is evaluated as an alternative in Sections 5 and 7. The technology types and process options identified for the other response actions were screened for applicability to address contaminated soils at the TBP area. Potentially applicable technologies are discussed in Section 4.3, and the results of the screening process are presented in Table 4.9.

### 4.2 TECHNOLOGY IDENTIFICATION AND SCREENING

#### **4.2.1** Institutional Controls

Institutional controls are measures that preclude or minimize public exposure by limiting access to or use of contaminated areas. Institutional controls include measures to restrict access, such as security guards or fencing, ownership and use or deed restrictions, and monitoring. Institutional controls do not reduce contaminant toxicity, mobility, or volume, but they can reduce the potential for exposure to contaminated material.

Access to the Edgewood down range area (including J-Field) is controlled by security guards and is limited to authorized personnel. Vehicular access to J-Field is controlled by a locked gate. These measures mitigate potential public exposure to contamination. However, these measures are not as effective at mitigating exposure of wildlife to contamination on-site. Fencing around the TBP area would not control exposures to wildlife, especially small and burrowing mammals and birds.

The U.S. Army has custody of the site and is expected to maintain this custody and accountability into the future. This measure controls public exposures to on-site contamination by restricting access and use.

An extensive environmental monitoring program is currently in place as part of the RI. Additional monitoring will be employed during future response actions. This measure can support the mitigation of potential exposures to humans and wildlife by providing data on the nature and extent of contamination and the effectiveness of primary control measures such as containment or removal. Future monitoring activities for the TBP area and other J-Field AOCs will be discussed in the FS report.

The screening analysis for institutional controls is summarized in Table 4.1. On the basis of effectiveness, implementability, and cost, all institutional controls currently in place have been retained. Fencing of the TBP area is not recommended. The most significant control at the site is land ownership. As long as the U.S. Army maintains custody of the APG site, the potential for significant public exposure will remain low because land use can be controlled and access to

**TABLE 4.1 Summary of Screening Analysis for Institutional Controls** 

Institutional Control Measure	Effectiveness	Implementability	Cost
Access restriction	Entry to the Edgewood down range area (which includes J-Field) is controlled by security guards. Vehicular access to the site is controlled by a locked fence. These measures are effective at mitigating potential public exposure to on-site contamination; however, they are less effective at mitigating potential exposures to wildlife. Fencing around the TBP area would not control exposures to wildlife and is not recommended.	Fences, guards, and other such measures are easy to implement, and resources are readily available.	Low
Ownership and use or deed restrictions	The U.S. Army has custody of APG and is expected to maintain this custody and accountability into the future.	Ownership and use or deed restrictions are easy to implement, and resources are readily available.	Low
Monitoring	An extensive monitoring program is currently in place as part of the RI. Additional monitoring will be employed during future response actions and will support mitigation of potential exposures to humans and wildlife.	Monitoring is easy to implement, and resources are readily available.	Moderate

contaminated areas can be restricted. Additional control measures would be needed to reduce the potential for significant exposure of wildlife.

#### 4.2.2 In-Situ Containment

In-situ containment involves technologies that confine contamination to its current locations in the soil. These technologies reduce contaminant mobility and the associated potential for exposure, but they do not reduce contaminant toxicity or volume. In-situ containment technologies that address contamination in surface soils include surface controls/diversions, vertical barriers, and caps/covers.

Surface controls/diversions are used to divert surface runoff around contaminated areas to minimize the potential for contaminant resuspension. Graded contours, swales, and berms can effectively control surface water runon and runoff and can limit the mobility of contaminants. These measures alone, however, may not be effective for surface water bodies (e.g., marsh) that are hydrologically connected to the surficial aquifer system or other surface water bodies. Therefore, vertical barriers may also be needed. In addition, erosion of contaminated soil from the portion of the pushout area in contact with the marsh could be mitigated by installing a vertical barrier.

A contaminated area can be encapsulated by placing barriers on top (caps/covers). Capping (e.g., a RCRA-engineered cap) of soil could effectively reduce airborne emissions (e.g., VOCs), precipitation-enhanced percolation and leaching, and contaminant resuspension via surface water runoff. This method would limit exposures of humans and wildlife to contamination in the surface soils. A stabilized surface would be required prior to cap placement and would most likely necessitate some grading and excavation. Therefore, a UXO survey would also be required over the area to be capped. In-situ capping of the main pits and pushout areas could prove difficult to implement, and the effectiveness of the cap would be constrained by the shallow water table and its proximity to the marsh. In addition, because the J-Field site is a designated floodplain, measures would be required to minimize the potential for damage from flood water. These measures would include shoreline stabilization and building the cap to elevations above flood level.

An alternative to an engineered cap is a protective soil cover, referred to here as a "risk-reduction cover." The risk-reduction cover also would be effective at minimizing contaminant mobility and limiting exposures at the surface. Under this option, a geotextile membrane would be laid over the contaminated area and covered with soil. This approach could be implemented with little or no UXO screening.

The screening analysis for in-situ containment is summarized in Table 4.2. On the basis of effectiveness, implementability, and cost, the in-situ containment technologies (surface controls/diversions, vertical barriers, and caps/covers) have been retained.

TABLE 4.2 Summary of Screening Analysis for In-Situ Containment

In-Situ Containment Measure	Effectiveness	Implementability	Cost
Surface controls/ diversions	Could effectively reduce contaminant mobility at the site. Such measures alone, however, might not be effective for the marsh because of its hydrological connection to the surficial aquifer.	Could be implemented with conventional equipment and procedures, and resources are readily available.	Low
Vertical barriers	Could effectively reduce contaminant mobility at the site, especially in groundwater and in places where the pushout area is in contact with the marsh.	Could be implemented with conventional equipment and procedures, and resources are readily available.	Low to moderate
Caps/covers	Could effectively limit airborne emissions (e.g., VOCs), precipitation-enhanced percolation and leaching, and contaminant resuspension via surface water runoff. Because J-Field is designated as a floodplain, measures would be required to minimize damage from flood waters.	Could be implemented with conventional equipment and procedures, and resources are readily available. UXO survey might be required.	Low to moderate

#### 4.2.3 Removal

Removal of contaminated material can limit contaminant mobility and volume at the affected source area and can facilitate treatment and disposal that could reduce contaminant toxicity, mobility, and volume.

Excavation with conventional earth-moving equipment (e.g., bulldozers, backhoes, and front-end loaders) could effectively remove soil in the main pits and pushout area at the TBP area. The area of excavation would have to undergo a UXO survey, and the soil would have to be monitored for CWAs during excavation.

Soil also could be removed by hand if the density of metal contacts is high or if only limited excavation is required. Human Factors Applications, Inc. (HFA) surveys have shown that conventional earth-moving equipment could be used in areas with metal contact densities of 1 to 28 contacts per 100-ft<sup>2</sup> area.

Contaminated vegetation can also be removed by standard clearing and grubbing methods to reduce exposures and limit biotic transport.

The screening analysis for removal is summarized in Table 4.3. On the basis of this evaluation, these technologies have been retained as potentially applicable to address surface soil contamination in the TBP area.

#### 4.2.4 Treatment

#### 4.2.4.1 In-Situ Physical/Chemical Treatment Technologies

Several in-situ physical/chemical treatment technologies are available for treating soil contaminated with metals, SVOCs, and PCBs (VOCs are present only at depth and will not be addressed by the proposed interim remedial action). These technologies include soil flushing, stabilization/solidification, vitrification, and electrokinetic processing. The main advantage of in-situ treatment is that it allows soil to be treated without being excavated and transported, resulting in potentially significant cost savings. However, in-situ treatment generally requires longer time periods to be effective, and there is less certainty about the uniformity of treatment because of the variability in soil characteristics and because the efficacy of the process is more difficult to confirm (EPA 1994). For the TBP area, UXO screening and removal would have to precede in-situ treatment.

The screening analysis of potential in-situ treatment technologies is discussed in the following subsections and is summarized in Table 4.4.

**Soil Flushing.** Soil flushing is the extraction of contaminants from the soil with water or other suitable aqueous solutions. This method is applicable to coarse-grained soils contaminated with

TABLE 4.3 Summary of Screening Analysis for Removal

Removal Measure	Effectiveness	Implementability	Cost
Excavation	Could remove the source of contamination to limit contaminant mobility and volume and to reduce related exposures. It could also facilitate treatment and disposal.	Could be implemented with conventional equipment and procedures, and resources are readily available. UXO surveys and CWA screening would be required.	Moderate to high
Clearing and grubbing	Could effectively remove vegetation from source areas to reduce related exposures and limit biotic transport.	Could be implemented with conventional equipment and procedures, and resources are readily available.	Low

TABLE 4.4 Summary of Screening Analysis for In-Situ Treatment Technologies

Treatment			
Technology	Effectiveness	Implementability	Cost
Soil flushing	Applicable to soils contaminated with SVOCs and metals; would not address soils with PCB contamination. Effectiveness is limited by variability in permeability across the TBP area. The potential for mobilizing contaminants beyond the capture zone is also a concern.	Implementation limited by variable soil conditions and low-permeability soil that could impede percolation and contact with reagents. Presence of UXO may also be a constraint.	Moderate to high
Stabilization/ solidification	Effective at limiting mobility of metals only; therefore, only applicable to soils in the pushout area. Contaminant mobility would be reduced, but contaminant toxicity would not. Volume of contaminated material would increase. Long-term effectiveness uncertain due to the unpredictable effects of weathering, groundwater infiltration, and physical disturbance (e.g., local detonations).	Implementation limited by scattered nature of contamination. Proximity to marsh and presence of UXO may also be constraints.	Moderate to high
In-situ vitrification	Applicable to soils contaminated with organics and metals. Toxicity and mobility of organics and metals would be reduced; volume would not.	Somewhat difficult to implement because of the innovative nature of this technology. Implementation limited by high organic content in soil. Presence of UXO may also be a constraint.	Moderate to high
Electrokinetic processing	Could reduce the area of contamination to be addressed by mobilizing metals and some organics with an electric current applied across the site. Studies indicate a period of 10 years may be required to concentrate contaminants. The potential for mobilizing contaminants beyond the capture zone is a concern. Would not be effective at mobilizing particulate lead from pushout area soils.	Somewhat difficult to implement because of the innovative nature of this technology. Presence of UXO may also be a constraint.	Moderate to high

SVOCs and metals. PCBs would not be removed by this process. Extraction fluids are passed through soil by using an injection or infiltration process, recovered, and recycled when possible. Recovered groundwater and fluids laden with desorbed contaminants may require treatment to meet appropriate discharge standards before release to wastewater treatment facilities or receiving streams. Treatment of the recovered fluids results in process sludges and residual solids, such as spent carbon and spent ion exchange resin, which must be appropriately treated before disposal. Air emissions of VOCs from recovered fluids would be collected and treated, as appropriate, to meet applicable regulatory standards.

The effectiveness of soil flushing is limited at sites, like the TBP area, with low-permeability soils. The permeability of soils in the TBP area varies laterally and vertically over several orders of magnitude because of the arrangement of interbedded facies, making any estimates of effectiveness difficult. In addition, the potential for immobilizing contaminants beyond the capture zone and introducing surfactants to the subsurface is of general concern. Soil flushing is considered a developing technology. For these reasons, soil flushing has been rejected from further consideration.

**Stabilization/Solidification.** In stabilization/solidification technologies, a fixing or stabilizing agent is mixed into the soil to create a product that is stable and resistant to leaching. In this process, additives are mixed directly into contaminated soil by conventional backhoes or equipment specifically designed for in-situ chemical injection and mixing. This technology effectively reduces the mobility of contaminants through both physical and chemical means. In-situ stabilization/solidification is considered most effective for immobilizing metals; therefore, it could only be applied in situ to fix contaminated soil at the pushout area. SVOC- and PCB-contaminated soil would have to be addressed by implementing an additional treatment technology.

The long-term effects of weathering (e.g., freeze-thaw cycles, acid precipitation, and wind erosion), groundwater infiltration, and physical disturbance (e.g., local detonations) could significantly affect the integrity of the stabilized mass, potentially reducing the long-term effectiveness of this technology. Contaminant mobility would be reduced, but contaminant toxicity would not. In addition, the volume of contaminated material would increase. For these reasons, stabilization/solidification has been rejected from further consideration.

In-Situ Vitrification. In-situ vitrification uses an electric current to melt soil or other earthen materials at extremely high temperatures (2,900 to 3,650°F) and thereby immobilize most inorganics (metals) and destroy organics (VOCs, SVOCs, and PCBs) by pyrolysis. To initiate the process, graphite electrodes are placed on the surface of the soil so that current can flow in the soil beyond the boiling temperature of water to the melting point of soil. The molten soil zone grows downward and outward. A vacuum-pressurized hood is placed over the vitrification zone to contain

and process any contaminants emanating from the soil during vitrification. The vitrification product is a chemically stable, leach-resistant glass and crystalline material similar to obsidian or basalt. Metals are incorporated within the vitrified product. Water vapor and organic pyrolysis combustion products are captured and drawn into an off-gas treatment system that removes particulates and other contaminants from the gas.

The in-situ application of this technology is considered innovative for waste treatment and has only been operated for test and demonstration purposes. For this reason, in-situ vitrification has been rejected from further consideration.

Electrokinetic Processing. Electrokinetic processing is an in-situ transport process that uses electric current to decontaminate soils contaminated with metals and some organics (polar species), such as phenol. The effectiveness of this technology is based on its ability to concentrate contaminants within a small area, thus reducing the area of contamination to be addressed. The application of electric current produces an acid in the anode compartment that sweeps across the soil, desorbing contaminants from the surface of soil particles and initiating electromigration of contaminants toward the respective electrodes. To initiate the process, anode and cathode series are inserted or laid on the ground, and a current is established across the electrodes. A conditioning fluid is introduced at the electrodes. A recent bench-scale study indicates that the time required to concentrate contamination in the TBP area could be as long as 10 years (Peters 1995). In addition, this process would not be effective at removing particulate lead, which makes up a significant proportion of contamination in the pushout area (U.S. Department of Energy [DOE] 1993/1994; EPA 1993b,d). For these reasons, electrokinetic processing has been rejected from further consideration.

# 4.2.4.2 Ex-Situ Physical/Chemical Treatment Technologies

Several ex-situ physical/chemical treatment technologies are available for treating soil contaminated with metals, SVOCs, and PCBs. These technologies include stabilization/fixation, vitrification, soil washing, soil leaching, oxidation/reduction, dehalogenation, incineration, and thermal desorption. The main advantages of ex-situ treatment are that it generally requires shorter time periods than in-situ treatment and that it affords more certainty about the uniformity of treatment because of the ability to homogenize, screen, and continuously mix the soil. However, ex-situ treatment requires excavation of soils, leading to increased costs and engineering for equipment, the necessity of UXO surveying, possible permitting, and material handling (EPA 1994).

The screening analysis of potential ex-situ treatment technologies is discussed in the following subsections and summarized in Table 4.5.

TABLE 4.5 Summary of Screening Analysis for Ex-Situ Treatment Technologies

Treatment			
Technology	Effectiveness	Implementability	Cost
Stabilization/ fixation	Effective at limiting mobility of metals and PCBs; applicable to soils from the pushout area and southern main pit. Contaminant mobility would be reduced, but contaminant toxicity would not. Final waste volume would increase.	Could be implemented with readily available equipment and materials.	Moderate
Vitrification	Effective at reducing toxicity of organics and metals. Volume would be reduced. Applicable to soils contaminated with organics and metals.	Could be implemented with readily available equipment and materials.	High
Soil washing	Effective at reducing toxicity, mobility, and volume of metals and SVOCs (and possibly PCBs); applicable to soils in the northern main pit and pushout area.	Could be implemented with readily available equipment and materials.	Low to moderate
Soil leaching	Effective at reducing toxicity, mobility, and volume of metals and SVOCs; applicable to soils in the northern main pit and pushout area.	Could be implemented with readily available equipment and materials.	Moderate
Reduction/oxidation	Effective at reducing contaminant toxicity, mobility, and volume; applicable to soils contaminated with metals only.	Could be implemented with readily available equipment and materials.	Moderate
Dehalogenation	Effective at reducing toxicity, mobility, and volume of PCBs and halogenated SVOCs; applicable only to soils in the main pits.	Could be implemented with readily available equipment and materials.	Moderate
Incineration	Effective at reducing contaminant volume; toxicity and mobility would not be reduced. Bottom ash would require further treatment by stabilization. May be required to treat PCB-contaminated soil (depending on concentrations).	Could be implemented with readily available equipment and materials.	Moderate to high
High-temperature thermal desorption	Effective at reducing toxicity, mobility, and volume of organics only; applicable to soils in the main pits.	Could be implemented with readily available equipment and materials.	Moderate

Stabilization/Fixation. In stabilization/fixation technologies, a fixing or stabilizing agent is mixed into the soil to create a product that is stable and resistant to leaching. This process involves mixing reagents with contaminated soil in a mixing vessel, such as a pug mill, to immobilize the contaminants and solidify the waste. This technology effectively reduces the mobility of contaminants through both physical and chemical means. Stabilization/fixation is considered effective for immobilizing metals and PCBs; therefore, it could only be applied to soils excavated from the pushout area and southern main pit. SVOC-contaminated soil may have to be addressed by implementing an additional treatment technology (e.g., if treatment codes are exceeded).

The predominant fixing agents currently in use are Portland cement, lime/fly ash, Portland cement/fly ash, and Portland cement/sodium silicate. Gypsum, bentonite, and zeolites could also be used, as could a number of proprietary agents. Chemical stabilization with cement and fly ash is an established practice for treating hazardous waste and has been retained as potentially applicable to address surface soil contamination in the TBP area.

Ex-Situ Vitrification. Ex-situ vitrification is designed to encapsulate inorganic contaminants rather than reduce inorganic contaminant concentration. This process involves electrically heating contaminated material to temperatures high enough to cause it to melt. Vitrification is applicable to a full range of contaminants. Inorganic contaminants are encapsulated in the vitrified mass; organic contaminants are destroyed because of the high temperatures achieved during processing. Organic off-gases and some volatile metal contaminants would have to be controlled and treated during process operation. The vitrified product is a chemically stable, leach-resistant glass or crystalline material similar to obsidian or basalt, ready for disposal. This technology is not considered cost-effective for the small volume of soils to be treated at the TBP area; on this basis, ex-situ vitrification has been rejected from further consideration.

Soil Washing. Soil washing is an aqueous-based process that reduces soil contamination through the use of particle size separation. This technology is applicable to soils contaminated with metals, SVOCs, and possibly PCBs. The effectiveness of soil washing is based on the finding that most organic and inorganic contaminants tend to bind, either chemically or physically, to the fine-grained fraction of the soil matrix (clay, silt, and organic soil particles). Washing processes that separate the fine particles (silt and clay) from the coarser material (sand and gravel) separate and concentrate the contaminants into a smaller volume of soil that can be further treated or disposed of.

Gravity separation is an effective method for removing high or low specific gravity particles, such as lead. Attrition scrubbing removes adherent contaminant films from coarser particles. The clean, larger fraction can be returned to the site for use as clean fill. The washing agent and soil fines are residuals from this process that would require further treatment (e.g., through soil

leaching). A bench-scale study has indicated that density separation is effective at removing metals from the sand fraction of TBP area pushout soils (Alternative Remedial Technologies, Inc. [ART] 1995). An additional study is needed to evaluate the efficacy of treating the fine-grained fraction through chemical extraction (soil leaching). On the basis of this preliminary study, it is estimated that a volume reduction of 94% may be achievable. For this reason, this technology type has been retained.

Soil Leaching. Soil leaching is a chemical process used to remove metals that remain in the fine sand, silt, and clay soil particles after the soil washing and separation process. Sand and fines are leached in separate parallel circuits with an acid solution. The effectiveness of soil leaching is achieved through dissolving smaller metallic particles and ionic metals remaining in the soil. A final step may include precipitation of metals from the leaching solution for recovery and recycling. Soil leaching is applicable to soils contaminated with metals and possibly SVOCs. PCB-contaminated soil would have to be addressed by implementing an additional treatment technology; however, these soils constitute a small fraction of the TBP area soils to be treated. On this basis, soil leaching has been retained.

Reduction/Oxidation. Reduction/oxidation (redox) reactions chemically convert contaminants to nonhazardous or less toxic compounds that are more stable, less mobile, and/or inert (EPA 1994). The oxidizing agents most commonly used for treatment are ozone, hydrogen peroxide, hypochlorites, chlorine, and chlorine dioxide. Chemical redox is applicable to soils contaminated with metals. SVOC- and PCB-contaminated soil would have to be addressed by implementing an additional treatment technology. This technology is not considered cost-effective for soils with high contaminant concentrations (because of the large amounts of oxidizing agent required); on this basis, chemical redox has been rejected from further consideration.

Dehalogenation. Dehalogenation is a decomposition process (e.g., base-catalyzed decomposition or glycolate) designed to remediate soils contaminated with chlorinated organic compounds (especially PCBs and halogenated SVOCs). During the dehalogenation process, contaminated soil is screened, processed with a crusher and pug mill, and mixed with sodium bicarbonate. The mixture is heated in a rotary reactor to decompose and partially volatilize the contaminants. The effectiveness of this technology is limited in soils with high clay and moisture content. In addition, it is only applicable to a small portion of the soils to be excavated at the TBP area. On this basis, dehalogenation has been rejected from further consideration.

**Incineration.** Waste material can be incinerated by rotary kiln, fluidized bed, slagging, and liquid injection incinerators. Rotary kiln and fluidized bed incineration are typically used to destroy

organic contaminants. Rotary kilns are refractory-lined, slightly inclined, rotating cylinders that function as a combustion chamber and operate at temperatures up to 1,800°F. Waste is introduced at the high end, and ash is collected from the bottom end. Flue gases pass through a secondary chamber and control equipment before exiting to the atmosphere. Fluidized-bed incinerators contain a bed of sized granular refractory material in a refractory-lined vessel. Waste is injected onto the bed and incinerated as air is forced up through the bed at a velocity sufficient to fluidize the burning material.

Incineration technologies do not reduce the toxicity or mobility of inorganic constituents; they do reduce the total waste volume but produce an ash residue. Heavy metals can produce a bottom ash that requires further treatment by stabilization, although volatile metals would leave the combustion unit with the flue gases and would require the installation of gas cleaning systems for removal. In addition, given the small volume of soil to be treated, this option would only be applied at an off-site facility. Waste transport would result in an increased risk of transportation accidents and related impacts, the magnitude of which would depend on the waste volume and type, mode of transportation, and facility location. Given the high costs associated with transportation and treatment, incineration is not considered cost-effective. However, this option has been retained because it may be required to treat PCB-contaminated soil (depending on concentrations, incineration may be the required treatment).

High-Temperature Thermal Desorption. High-temperature desorption involves heating wastes to temperatures of 600 to 1,000°F. It is applicable to soils contaminated with SVOCs and PCBs. In the treatment process, a carrier gas or vacuum system transports volatilized water and organics to the gas treatment system. Some volatile metals may also be removed. Contaminants are removed from the gas stream and disposed of. Bench-scale studies would be required to determine the effectiveness of organics removal. Thermal desorption is only applicable to a small portion of the soils to be excavated at the TBP area, but it is considered a cost-effective treatment for PCB-contaminated soil. On this basis, this option has been retained.

# 4.2.4.3 Biological Treatment Technologies

Several biological technologies are available for treating contaminated soil. Biological treatment technologies use living organisms, such as bacteria or fungi, to detoxify or immobilize contaminants in waste. These technologies are applied primarily for converting organic contaminants into nontoxic products. Bioremediation has also been used to degrade inorganic contaminants, such as nitrates, and it can be used to detoxify or immobilize certain metals by changing their oxidation state. The SVOC- and PCB-contaminated soils in the main pits are candidates for bioremediation.

Biological treatment technologies can be implemented as either in-situ or ex-situ processes. The advantage of in-situ treatment is that it allows soil to be treated without being excavated and transported, resulting in a significant cost savings. However, in-situ treatment generally requires longer time periods, and there is less certainty about the uniformity of treatment because of the variability in soil characteristics and because the efficacy of the process is more difficult to confirm (EPA 1994). Ex-situ treatment generally requires shorter time periods; however, it requires excavation of soils, leading to increased costs and engineering for equipment, the necessity of UXO surveying, possible permitting, and material handling/worker exposure considerations (EPA 1994).

The screening analysis of potential biological treatment technologies is discussed in the following subsections and is summarized in Table 4.6.

In-Situ Bioremediation. In-situ bioremediation is a destructive technique aimed toward stimulating microorganisms in the soil to grow and use the contaminants as a food and energy source by creating a favorable environment for the microorganisms. Generally, this means providing some combination of oxygen, nutrients, and moisture and controlling the temperature and pH. Sometimes, microorganisms adapted for degradation of specific contaminants can be applied to enhance the process. This technology is applicable to soils contaminated with organics (e.g., PCBs); it is not applicable to soils contaminated with metals.

A recent treatability study has shown that sufficient numbers of microorganisms suitable for removal of the organic COCs are present; however, the low permeability of the soil would limit the effectiveness of in-situ bioremediation (Huang 1994).

Because in-situ bioremediation is only applicable to a small portion of the soils to be excavated at the TBP area, it has been rejected from further consideration.

TABLE 4.6 Summary of Screening Analysis for Biological Treatment

Biological Treatment Technology	Effectiveness	Implementability	Cost
In-situ bioremediation	Applicable to soils contaminated with organics; would not address metals contamination in the pushout area.	Implementation limited by low-permeability soil that could impede aeration of soils.	Moderate
Ex-situ bioremediation	Applicable to soils contaminated with organics; would not address metals contamination in the pushout area.	Can be implemented with conventional equipment and procedures.	Moderate

**Ex-Situ Bioremediation.** Ex-situ bioremediation technologies include slurry-phase and solid-phase processes applicable to soils contaminated with organics (e.g., PCBs). Slurry-phase methods, such as batch treatment in a reactor, might be effective because operating parameters could be closely controlled. Solid-phase methods, such as placing in piles, have the advantage of low setup costs. Metals would not be addressed through either treatment.

A recent treatability study reported that significant numbers of microorganisms, including methanotrophs and biphenyl degraders (that are appropriate for degrading organics), are present and that ex-situ bioremediation by using bulking agents might be effective. Another treatability study has indicated that significant numbers of heterotrophs suitable for degradation of chlorinated organics may also be present (Huang 1994).

Because ex-situ bioremediation is only applicable to a small portion of the soils to be excavated at the TBP area, it is not considered a cost-effective treatment. Therefore, ex-situ bioremediation has been rejected from further consideration.

### 4.2.5 Short-Term Storage

Short-term storage involves isolating contaminated material to protect human health and the environment until the material can be treated and/or permanently disposed of. This technology typically involves constructing an engineered facility to minimize the potential for contaminant migration. Short-term storage would not reduce contaminant toxicity or volume, but it could reduce contaminant mobility and potential exposures. Such facilities also could be developed to support cleanup activities. The storage facility could be an enclosed structure or an outdoor area with a gravel pad or other base and covers, such as tarpaulins. These options could be applied either on-site or at an off-site facility.

The screening analysis of short-term storage is summarized in Table 4.7. Use of an off-site facility would require that the material be transported off-site and that an appropriate facility be available. Waste transport would result in an increased risk of transportation accidents and related impacts, the magnitude of which would depend on the waste volume and type, mode of transportation, and facility location. No suitable off-site facility exists for short-term storage of site waste, and none is likely to become available in the near future. Therefore, only the on-site application has been retained.

## 4.2.6 Disposal

Disposal options for contaminated soil involve confinement for permanent disposition. The only disposal option considered for the waste resulting from site remediation at the TBP area is

TABLE 4.7 Summary of Screening Analysis for Short-Term Storage

Short-Term Storage Facility	Effectiveness	Implementability	Cost
Open and enclosed structures	Could effectively protect human health and the environment in the short term by reducing contaminant mobility and limiting exposures. The off-site application of this option has been rejected on the basis of potential increased risk to workers and the general public due to increased likelihood of exposures associated with waste transportation.	The on-site application could be easily implemented with readily available resources. Such facilities also could be developed to support cleanup activities. An appropriate storage facility is not currently available off-site, and it is unlikely that such a facility would become available in the near future.	Low to moderate

off-site land-based facilities. Material determined to be hazardous under RCRA could be shipped to a permitted facility; nonhazardous material could be shipped to a sanitary or demolition landfill. Related requirements are presented in Appendix B (Table B.4).

Under RCRA regulations (40 CFR Part 261), a solid waste is considered to be a regulated hazardous waste if it is not otherwise excluded from regulation as a hazardous waste and either exhibits any of the characteristics identified in Subpart C of this regulation (a "characteristic hazardous waste") or is listed in Subpart D of the regulation (a "listed hazardous waste"). Waste determined to be hazardous as defined by RCRA must be disposed of according to the requirements of 40 CFR Parts 264 and 268 (see Appendix B). In addition, the LDRs promulgated by EPA in 40 CFR Part 268 preclude the disposal of certain contaminants without prior treatment. The LDRs would apply to any TBP area soils regulated under 40 CFR Part 268, and land disposal of such waste would require that appropriate treatment standards be met. The treatment standards are based on contaminant concentrations in the waste, the treatment technology used, or waste characteristics after treatment. Untreated soils from the TBP area pushout area may be categorized as hazardous waste code D008 under 40 CFR Part 261 (Subpart C) because of the high lead concentrations. These soils would require treatment by stabilization to satisfy LDRs prior to disposal. PCB-contaminated soils would also require treatment prior to disposal. If TBP area soils are found to be regulated under TSCA (e.g., if concentrations of PCBs exceed 50 ppm in unrestricted-access areas), these soils may require incineration prior to disposal. If PCB-contaminated soils are not regulated under TSCA (i.e., if concentrations are less than 50 ppm), they would undergo stabilization prior to disposal.

The implementability of land disposal at an off-site facility is affected by the availability of suitable sites for disposal of APG waste. Special pricing through an on-site contractor at APG could keep disposal costs down relative to off-site contractors, thereby making disposal a cost-effective option.

The screening analysis of the disposal option is summarized in Table 4.8. On the basis of this evaluation, the disposal option has been retained.

# 4.3 POTENTIALLY APPLICABLE TECHNOLOGIES

Potentially applicable technologies for remediation of the TBP area are summarized in Table 4.9. This summary is based on the screening analysis presented in Section 4.2. The technology types that have been retained through this analysis were used to develop preliminary interim remedial action alternatives for the TBP area. These alternatives are identified in Section 5.

TABLE 4.8 Summary of Screening Analysis for Disposal

Disposal Option	Effectiveness	Implementability	Cost
Off-site, land- based disposal	Provides protection of environment and reduces exposures. Treatment of contaminated soils would be required prior to disposal.	Subtitle C and D landfills are available in the eastern region of the United States (e.g., Virginia and Pennsylvania).	Moderate

TABLE 4.9 Summary of Potentially Applicable Technologies

General Response Action	Technology Type	Evaluation Result	Comments
No action	Not applicable	Retained	Would provide a baseline for comparison with action alternatives.
Institutional control	Access restriction	Retained	Could effectively limit entry to contaminated areas and could be used to support other response actions.
	Ownership and use or deed restrictions	Retained	Could minimize exposures to site contaminants by limiting use of contaminated areas and could be used to support other response actions.
	Monitoring	Retained	Could provide data useful for minimizing exposures and could be used to support other response actions.
In-situ containment	Surface control diversions	Retained	Could limit contaminant mobility by directing surface runoff from contaminated areas and could be used to support other response actions.
	Vertical barriers	Retained	Could limit contaminant mobility by impeding groundwater discharge to the marsh and by minimizing erosion of contam- inated soil from the pushout area.
·	Caps/covers	Retained	Could limit airborne emissions, precipitation-enhanced percolation and leaching, and contaminant resuspension via surface water runoff.
Removal	Excavation	Retained	Could effectively remove the source of contamination and could be readily implemented. UXO survey would be required.
	Clearing and grubbing	Retained	Could effectively remove vegetation from the site to support overall cleanup activities.

TABLE 4.9 (Cont.)

General Response Action	Technology Type	Evaluation Result	Comments
In-situ treatment	Soil flushing	Rejected	Difficult to implement because of limitations due to site conditions. Ineffective for removing PCBs.
	Stabilization/solidification	Rejected	Difficult to implement because of limitations due to site conditions. Ineffective for limiting mobility of PCBs.
	Vitrification	Rejected	Difficult to implement because of innovative nature of technology and site conditions.
	Electrokinetic separation	Rejected	Difficult to implement because of innovative nature of technology and site conditions. Ineffective at mobilizing particulate lead.
Ex-situ treatment	Stabilization/fixation	Retained	Could reduce contaminant mobility but increases volume. Would require an engineered treatment facility.
	Vitrification	Rejected	Could reduce contaminant toxicity and mobility; volume would not be reduced. Would require an engineered treatment facility; not considered cost-effective.
	Soil washing	Retained	Could reduce contaminant toxicity, mobility, and volume. Could be implemented on-site.
	Soil leaching	Retained	Could reduce contaminant toxicity, mobility, and volume. Would facilitate recovery of lead for recycling. Could be implemented on-site.
	Reduction/oxidation	Rejected	Could reduce toxicity, mobility, and volume (for metals only). Not considered cost-effective.
	Dehalogenation	Rejected	Could reduce toxicity, mobility, and volume (for chlorinated organic compounds only). Effectiveness limited by site conditions.

TABLE 4.9 (Cont.)

General Response Action	Taabnala ay Tura	Evaluation Result	
Action	Technology Type	Result	Comments
Ex-situ treatment (cont.)	Incineration	Retained	Could reduce contaminant toxicity, mobility, and volume (for organics). Would require an engineered treatment facility.
	Thermal desorption	Retained	Could reduce contaminant toxicity, mobility, and volume (for organics).
Biological treatment	In-situ bioremediation	Rejected	Could reduce contaminant toxicity, mobility, and volume (for organics). Effectiveness limited by site conditions.
	Ex-situ bioremediation	Rejected	Could reduce contaminant toxicity, mobility, and volume (for organics). Not considered cost-ffective.
Short-term storage	Open and closed structures	Retained	Could effectively reduce contaminant mobility in the short term and support cleanup activities.
Disposal	Land-based facility	Retained	Could effectively provide protection of environment and reduce exposures. Treatment required prior to disposal.

# 5 DEVELOPMENT AND SCREENING OF PRELIMINARY ALTERNATIVES

Preliminary alternatives for remediating surface soil in the J-Field TBP area were developed and screened in accordance with CERCLA, EPA guidance (EPA 1988), and the NCP (EPA 1990a). Five preliminary alternatives, including no action, were developed on the basis of the criteria presented in Section 5.1. These alternatives, identified and described in Section 5.2, were then screened on the basis of the criteria defined in Section 5.3. Each of the five preliminary alternatives was passed through the screening analysis presented in Section 5.4; they are identified as final alternatives and listed in Section 5.5. These final alternatives are further developed and described in Section 6 and evaluated in detail in Section 7. The preferred alternative will be identified in the proposed plan for this interim remedial action.

# 5.1 CRITERIA FOR DEVELOPING ALTERNATIVES

EPA has established an approach for developing remedial action alternatives that are appropriate to the site-specific conditions (EPA 1988, 1990a). In this approach, the scope, characteristics, and complexity of the site are considered in developing a range of alternatives that would be protective of human health and the environment. This protection can be achieved by eliminating, reducing, and/or controlling risks posed by each pathway at a site. Two major categories of response, containment and treatment, are considered in developing remedial action alternatives:

- Containment involves little or no treatment but protects human health and the
  environment by preventing or controlling exposures to contaminants through
  engineering measures and by using institutional controls as necessary to
  ensure the continued effectiveness of a response; and
- Treatment ranges from alternatives that use treatment as the primary element
  of the response to address the principal threat(s) posed by a site to alternatives
  that use treatment to reduce the toxicity, mobility, or volume of contaminated
  material to the maximum extent feasible, minimizing the need for long-term
  management.

As stated in Section 121(b) of CERCLA, as amended, the most preferred alternatives (1) represent permanent and cost-effective solutions for protecting human health and the environment; (2) permanently and significantly reduce the toxicity, mobility, or volume of contaminated material; and (3) apply alternative treatment or resource recovery technologies to the extent possible. Least preferred alternatives involve the transport and disposal of waste off-site without treatment.

A no-action alternative is also included to provide a baseline for comparison with other alternatives. For the analysis in this FFS, the baseline condition of the J-Field TBP area would include only minor maintenance activities, such as mowing and monitoring.

#### 5.2 IDENTIFICATION OF PRELIMINARY ALTERNATIVES

Technologies potentially applicable to managing contaminated surface soil at the J-Field TBP area are identified and screened in Section 4 (Table 4.9). On the basis of this screening, various technologies were identified as potential components of remedial action alternatives for the site. These technologies have been incorporated into five preliminary alternatives:

- Alternative 1: No Action;
- Alternative 2: Limited Removal and Disposal, and In-Situ Containment;
- Alternative 3: Removal and Short-Term Storage;
- Alternative 4: Removal, On-Site Treatment, and Limited Disposal; and
- Alternative 5: Removal, Off-Site Treatment, and Disposal.

In-situ containment is the primary emphasis of Alternative 2, and source control through removal is the primary emphasis of Alternatives 3, 4, and 5.

The J-Field site would remain secured, and existing institutional controls, such as ownership and use or deed restrictions, are implicitly included in each action alternative during the cleanup period. These controls are also included for the no-action alternative. Controls, such as monitoring, would be increased as needed. Each alternative would require various support activities before implementation, including the design and construction of staging areas, procurement of appropriate equipment, and development of contingency plans and operational controls to minimize contaminant releases. Site preparation activities would include clearing and grubbing contaminated areas, conducting UXO surveys, constructing access roads, and emplacing site perimeter dikes for surface water control. Shoreline stabilization has been included only for Alternatives 2 and 3. Because Alternatives 4 and 5 involve total removal of contaminated surface soil, shoreline stabilization was not included as a component.

### 5.2.1 Alternative 1: No Action

Alternative 1 is included to provide a baseline for comparison with the other action alternatives. Under this alternative, the site would continue to operate under restricted access. Current conditions of the contaminated surface soil would continue.

# 5.2.2 Alternative 2: Limited Removal and Disposal, and In-Situ Containment

Under Alternative 2, contamination in the two main pits and the pushout area would be contained in place with a protective cover. Limited excavation and disposal would be implemented to address contaminated soil in the main pits.

The "risk-reduction cover" would involve placing a geotextile fabric over the vegetation-cleared surface (about 5 acres [218,000 ft²]), which would then be covered by a layer of soil sufficient to raise the site topography above the 100-year floodplain. The soil layer would be stabilized with a vegetative cover. Runoff from the area would be diverted by surface grading controls. Erosion of contaminated soil from the portion of the pushout area in contact with the marsh would be mitigated by installing a vertical barrier. Because the soil layer would be emplaced incrementally in 2-ft layers, UXO screening most likely would not be required for this component. A UXO visual survey would be required for the limited excavation component. Because the area of excavation is limited, hand removal methods may be used (depending on the density of metal contacts present). The limited disposal component under Alternative 2 would involve shipping about 400 yd³ of soil to a permitted off-site facility for treatment (stabilization and possibly incineration) and disposal.

Stabilization of the shoreline about 2,000 ft to the south of the TBP area would also be implemented as part of this alternative. A UXO survey would be conducted by a two-man clearance team. Installation of a boulder riprap berm would reduce the potential for long-term erosion along the shoreline.

# 5.2.3 Alternative 3: Removal and Short-Term Storage

Under Alternative 3, contaminated surface soil in the two main pits and pushout area that exceed soil cleanup criteria would be removed and transported to an on-site, short-term storage facility, where it would be stored until final remediation is addressed at J-Field.

Soil would be excavated to a depth of 2 ft from the source areas with standard construction equipment, such as front-end loaders, scrapers, and backhoes. The excavation area would be surveyed for UXO in the upper 2 to 4 ft. Approximately 16,000 yd<sup>3</sup> of soil would be removed.

Runoff from the area would be diverted by surface grading controls. Following excavation, the area would be backfilled with clean soil. The short-term storage facility would be constructed adjacent to the Prototype Building to minimize the distance over which the material would have to be transferred. This location is the highest topographic area at J-Field. The short-term facility would be a modular building consisting of fabric walls. The base of the structure would require the construction of a bermed and drained concrete or asphalt pad equipped with a leachate collection system. An air infiltration system may also be required.

Stabilization of the shoreline about 2,000 ft to the south of the TBP area would also be implemented as part of this alternative. A UXO survey would be conducted by a two-man clearance team. Installation of a boulder riprap berm would reduce the potential for long-term erosion along the shoreline.

# 5.2.4 Alternative 4: Removal, On-Site Treatment, and Limited Disposal

Under Alternative 4, contaminated soil in the two main pits and pushout area that exceeds soil cleanup criteria would be removed and treated by soil washing/leaching to reduce the volume of waste for off-site disposal.

The removal component of Alternative 4 would be similar to that of Alternative 3; however, the excavated soil would be transported to an on-site soil washing/leaching treatment facility, which would be constructed adjacent to the Prototype Building (though I- and H-Fields are other potential sites). For the conceptual design, it was estimated that 16,000 yd³ of soil would be treated. A portion of the soil from the southern main pit, contaminated with PCBs, would be sent off-site for treatment and disposal. After removal and treatment, the cleaned material would either remain on-site to be used as fill or be shipped to an off-site solid waste disposal facility, in accordance with the State of Maryland's preference. Limited disposal under Alternative 4 would involve shipping recovered lead, scrap metal, and other (nonhazardous) refuse, as necessary, to an off-site facility (smelter and/or municipal landfill).

# 5.2.5 Alternative 5: Removal, Off-Site Treatment, and Disposal

Under Alternative 5, contaminated soil in the two main pits and pushout area that exceeds soil cleanup criteria would be removed and transported to an off-site treatment and disposal facility.

The removal component of Alternative 5 would be similar to that of Alternative 3; however, the excavated soil would be temporarily staged on-site following excavation, then transported to an off-site treatment and disposal facility.

# 5.3 CRITERIA FOR SCREENING ALTERNATIVES

The five preliminary alternatives were evaluated for applicability to remediating the J-Field TBP area on the basis of three general criteria: effectiveness, implementability, and cost. The effectiveness of an alternative is defined by its overall ability to protect human health and the environment in both the short term and long term. Measures of effectiveness include (1) reduction of potential long-term impacts to human health and the environment; (2) reduction of contaminant toxicity, mobility, or volume through treatment; (3) control of potential impacts to human health and the environment during the action period; (4) timeliness; and (5) consistency with regulatory requirements.

The implementability of an alternative is defined by its technical and administrative feasibility and availability of resources. Technical feasibility refers to the ability to construct, reliably operate, and meet technology-specific regulations for process options until the remedial action is complete. It also addresses the operation, maintenance, replacement, and monitoring of the technical components of an alternative, as appropriate; potential constraints associated with the site environment are also considered. Administrative feasibility addresses the acceptability of an alternative by other agencies and groups and pertinent environmental requirements, such as permits, as appropriate. Resource availability addresses the resources required to implement specific components of an alternative and the ability to obtain them.

The cost of an alternative is considered only in a comparative manner at the screening stage by comparing general estimates for each alternative to evaluate relative cost. This comparison helps decision makers to determine whether the cost of one alternative is much greater than that of another alternative of similar effectiveness and implementability. If the cost of an alternative is inordinately excessive compared to the effectiveness it provides, that alternative can be screened from further consideration.

# 5.4 SCREENING OF PRELIMINARY ALTERNATIVES

### 5.4.1 Alternative 1: No Action

Under Alternative 1, the TBP area would remain unchanged. Activities that would continue under the no-action alternative include maintenance and periodic monitoring.

#### 5.4.1.1 Effectiveness

Alternative 1 would not involve any treatment to reduce the toxicity, mobility, or volume of contaminated surface soil at the TBP area, and it would not provide for a timely or permanent response to the contamination problem. In addition, certain regulatory requirements would not be met. The potential for exposures of wildlife, trespassers, and site workers would continue in the short term and could increase over time if contaminants were released to other media (groundwater, surface water, sediment, or air). Potential long-term health impacts to site workers and the general public would be low as a result of institutional controls; however, adverse impacts to wildlife would be expected due to continued exposure.

#### 5.4.1.2 Implementability

Minimum site operations, including maintenance activities (e.g., mowing) and monitoring, would continue with readily available resources.

#### 5.4.1.3 Cost

Costs associated with Alternative 1 (baseline conditions) include those for continuing the general maintenance and security of the site. Monitoring activities would also be included. Annual costs are estimated to be about \$300,000.

#### 5.4.2 Alternative 2: Limited Removal and Disposal, and In-Situ Containment

#### 5.4.2.1 Effectiveness

The effectiveness of the soil cover could be limited by the shallow water table and its proximity to the marsh and by the potential for flooding, because portions of the TBP area are located within the 100-year floodplain (FEMA 1986). Other factors that could limit the effectiveness of the soil cover include the activities of burrowing animals present at the site and growth of deeprooted vegetation on the soil cover. Engineering controls (e.g., contouring and landscaping) and maintenance could effectively alleviate these limitations. Regular maintenance and monitoring would be necessary to maintain long-term effectiveness. Some repair might be necessary if the soil cover were to deteriorate.

#### 5.4.2.2 Implementability

Alternative 2 could be implemented with readily available resources. Technical feasibility of the in-situ containment component and shoreline stabilization would be high. Timeliness would be affected by the potential presence of UXO, which would require labor-intensive removal methods for the localized areas. However, localized removal would take considerably less time than full excavation and removal or excavation and on-site treatment.

Approximately 60% of the site is above the 100-year floodplain; another 20% is above the 5-ft contour and would only require 2 to 3 ft of cover to exceed the 100-year floodplain line. The remaining 20% would have to be raised 6 to 7 ft to exceed the 100-year floodplain line. The edge of the cover along the marsh would be engineered as a berm that would be covered by geotextile fabric and rock riprap to protect against erosion from the 100-year flooding.

The disposal component of Alternative 2 would be relatively straightforward. About 400 yd<sup>3</sup> of contaminated soil would be sent off-site for treatment and disposal.

#### 5.4.2.3 Cost

The estimated cost of in-situ containment would be about \$2 million. This cost includes site preparation activities (clearing and grubbing, UXO surveying, constructing and improving roads, and limited excavating), construction of the soil cover, limited disposal, direct and indirect costs, remedial design costs, and project management. The cost of Alternative 2 would be comparable to Alternative 3 and lower than Alternatives 4 and 5.

# 5.4.3 Alternative 3: Removal and Short-Term Storage

#### 5.4.3.1 Effectiveness

Alternative 3 would not include a treatment component; therefore, contaminant toxicity, mobility, and volume would not be reduced through treatment. However, potential exposures to wildlife, trespassers, and site workers would be reduced through removal of surface soil. The storage component of this alternative would provide for short-term protection, but it would not be protective in the long term.

Short-term risks to on-site workers and the general public would be significant for Alternative 3 because of the removal component. Risks include those related to the potential for encountering UXO and airborne emissions. Potential short-term environmental impacts, mainly a

result of excavation and grading activities, include disturbance to the soil, temporary increases in fugitive dust emissions and ambient noise level, increased sediment transport to the adjacent marsh, and displacement or loss of vegetation and wildlife habitat. Mitigative measures to reduce these impacts are discussed in Section 7.4.

#### 5.4.3.2 Implementability

Implementing the removal component of Alternative 3 would be relatively straightforward with regard to the availability of resources. Technical feasibility of the on-site storage component and shoreline stabilization would be high. Contaminated surface soil from the main pits and pushout area could be excavated with standard equipment and readily available materials. Timeliness would be affected by the potential presence of UXO, which will require time-consuming surveys in the upper 2 to 4 ft. Implementing the on-site, short-term storage component would also be straightforward. The storage facility could be built by on-site workers with assistance from the manufacturer's technical representative.

#### 5.4.3.3 Cost

This cost includes site preparation activities (clearing and grubbing, UXO surveying, extending electric service, and constructing and improving roads), excavation, construction of the storage facility, transporting material to a storage facility, landscaping the excavation area, direct and indirect costs, remedial design costs, and project management. For this estimate, operation of the facility was assumed to be 5 years. The cost of Alternative 3 would be comparable to Alternative 2 and lower than Alternatives 4 and 5. Although the costs associated with Alternative 3 are lower than those for Alternatives 4 and 5, they do not represent the total cost of long-term remediation. Short-term storage following excavation is only a temporary measure to limit exposures to contaminated surface soil; this action would have to be followed by treatment and disposal at some point in the future.

### 5.4.4 Alternative 4: Removal, On-Site Treatment, and Limited Disposal

#### 5.4.4.1 Effectiveness

Excavating contaminated surface soil at the TBP area would reduce potential impacts to human health and the environment in a timely manner and would reduce the potential for contaminant migration. Subsequent treatment of the waste (via soil washing/leaching) would reduce

contaminant toxicity, mobility, and volume over the short and long term. Alternative 4 satisfies the statutory preference for treatment as a principal element of the remediation.

Recent treatability studies (ART 1995) have shown that through a combination of physical separation treatment and chemical extraction (leaching) of sand and fines, a volume reduction of about 94% may be achieved. The remaining 6% would consist of oversize material (>2 mm) that may require disposal as a hazardous waste. These studies were based on achieving interim PRGs for metals. Final remediation goals would have to be determined before further studies are conducted so that total effectiveness could be accurately assessed.

Short-term risks to site workers and the general public would be significant for Alternative 4 because of the removal component. Risks include those related to the potential for encountering UXO and airborne emissions. Potential short-term environmental impacts, mainly a result of excavation and grading activities, include disturbance to the soil, temporary increases in fugitive dust emissions and ambient noise level, increased sediment transport to the adjacent marsh, and the displacement or loss of vegetation and wildlife habitat. Mitigative measures to reduce these impacts are discussed in Section 7.4.

### 5.4.4.2 Implementability

Implementing the removal component of Alternative 4 would be relatively straightforward with regard to the availability of resources. Contaminated surface soil from the main pits and pushout area could be excavated with standard equipment and readily available materials. Timeliness would be affected by the potential presence of UXO, which will require time-consuming surveys in the upper 2 to 4 ft. Implementing the on-site treatment component (soil washing/leaching) and limited disposal (recovered lead and scrap metal) would also be straightforward.

#### 5.4.4.3 Cost

The estimated cost of removal, on-site treatment, and limited disposal would be about \$13 million. This cost includes site preparation activities (clearing and grubbing, UXO surveying, extending electric service, and constructing and improving roads), excavation (to 2 ft), construction of the soil washing/leaching facility, transporting and loading material into the facility, landscaping the excavation area, direct and indirect costs, remedial design costs, and project management. The estimate given here is based on a 94% reduction in soil volume; that is, 6% of washed soil volume and all PCB-contaminated soil would be sent off-site. The cost of Alternative 4 is the highest of all the action alternatives.

#### 5.4.5 Alternative 5: Removal, Off-Site Treatment, and Disposal

#### 5.4.5.1 Effectiveness

Excavating contaminated surface soil at the TBP area would reduce potential impacts to human health and the environment in a timely manner and would reduce the potential for contaminant migration. Subsequent treatment of the waste (off-site) would reduce contaminant toxicity, mobility, and volume over the short and long term. Alternative 5 satisfies the statutory preference for treatment as a principal element of the remediation but does not satisfy the statutory preference for on-site treatment; instead, it involves the transport of a large volume of soil to an off-site facility for treatment.

Short-term risks to site workers and the general public would be significant for Alternative 5 because of the removal component. Potential short-term risks to the general public would also exist because of the off-site transport of untreated hazardous material. Risks include those related to the potential for encountering UXO and airborne emissions. Potential short-term environmental impacts, mainly a result of excavation and grading activities, include disturbance to the soil, temporary increases in fugitive dust emissions and ambient noise level, increased sediment transport to the adjacent marsh, and the displacement or loss of vegetation and wildlife habitat. Mitigative measures to reduce these impacts are discussed in Section 7.4.

Leachability tests have been conducted on surface soils (0 to 4 ft) in the pushout area (Peters 1995). Test results indicate that soil fails the TCLP requirements for hazardous waste with respect to lead. Waste determined to be hazardous as defined by RCRA (40 CFR Part 261) must be disposed of according to the requirements of 40 CFR Parts 264 and 268. Because of the high lead concentrations, untreated soils from the TBP pushout area would be categorized as waste code D008 under 40 CFR Part 268 (Subpart D). These soils would require treatment by stabilization before disposal. The semivolatile components most likely will not present a disposal problem for the soils excavated from the main pits (and volatiles are present mainly at depths greater than 2 ft). However, PCB-contaminated soils (found in the southern main pit) would also require treatment before disposal. Depending on concentrations, the PCB-contaminated soils would undergo either stabilization (if less than 50 ppm) or incineration (if greater than 50 ppm). These technologies have been demonstrated to be effective and acceptable for treating hazardous soils before disposal.

#### 5.4.5.2 Implementability

Implementing the removal component of Alternative 5 would be relatively straightforward with regard to the availability of resources. Contaminated surface soil from the main pits and pushout area could be excavated with standard equipment and readily available materials. Timeliness

would be affected by the potential presence of UXO, which will require time-consuming surveys in the upper 2 to 4 ft. Implementing the off-site treatment and disposal components would also be straightforward. Off-site facilities are available to handle the type and volume of waste that would be generated.

#### 5.4.5.3 Cost

The estimated cost of removal, off-site treatment, and disposal would be about \$11 million. This cost includes site preparation activities (clearing and grubbing, UXO surveying, and constructing and improving roads), excavation (to 2 ft), construction of a temporary staging area, landscaping the excavation area, direct and indirect costs, remedial design costs, and project management. The cost of Alternative 5 is higher than Alternatives 1 through 3 but less than Alternative 4.

# 5.5 SCREENING SUMMARY AND IDENTIFICATION OF FINAL ALTERNATIVES

The results of the screening analysis for the preliminary alternatives are summarized in Table 5.1. Information for each alternative was evaluated relative to EPA's screening criteria of effectiveness, implementability, and cost.

Although it would not be protective of human health and the environment in the long term, the no-action alternative (Alternative 1) was retained through this screening to provide a basis for comparison with the remaining action alternatives during the subsequent detailed analysis. Alternative 3 was retained because it offers a low-cost option for reducing risk; however, it is not considered effective in the long term and would thus require an additional action at a future date.

On the basis of the screening analysis, all five preliminary alternatives were retained.

TABLE 5.1 Screening Results for Preliminary Alternatives

Cost	hent Annual costs estimated to be strail \$300,000. The baseline costs would intain probably increase with inflation. uch as ng.	ward More expensive than Alternative 1 h but less than other action alternatives e (4 and 5). Comparable in cost to the Alternative 3. For the and
Implementability	Standard practices and equipment would be used to conduct general maintenance activities and maintain current institutional controls, such as access restriction and monitoring.	Installation of the protective cover would be relatively straightforward and could be implemented with readily available resources. The presence of UXO could affect the timeliness of implementation. For the limited excavation and (off-site) disposal component, transport and disposal of waste material generated from grading activities would be relatively straightforward. Installation of riprap berm at the shoreline would also be straightforward.
Effectiveness	Exposures to humans and wildlife at the contaminated areas would continue, and migration could result in increased exposures over time. No treatment would be implemented to reduce toxicity, mobility, or volume of contaminated material; no permanent solution would be achieved; and certain regulatory requirements would not be met.	More protective in the long term than Alternative 1. Short-term impacts to workers would be higher than for Alternative 1 because of the potential for exposures during construction and grading activities. Exposures would be reduced in the short term, and this alternative could be implemented in a timely manner. Factors limiting the effectiveness of the cover include the activities of burrowing animals and growth of deep-rooted vegetation. Regular maintenance and monitoring would be required.
Alternative	Alternative 1: No Action	Alternative 2: Limited Removal and Disposal, and In-Situ Containment

TABLE 5.1 (Cont.)

Alternative	Effectiveness	Implementability	Cost
Alternative 3: Removal and Short- Term Storage	More protective in the short term than Alternative I because it would reduce potential exposures to humans and wildlife through removal. No treatment would be implemented to reduce contaminant toxicity, mobility, or volume. Alternative 3 would not be protective in the long term and would require an additional action at a future date. Potential risks to human health and the environment would be significant in the short term because of the removal component but would be reduced in the long term. Good engineering practices and mitigative measures would be implemented to minimize impacts.	Removal and storage would be relatively straightforward and could be implemented with readily available resources and standard procedures. The presence of UXO could affect the timeliness of implementation. Installation of riprap berm at the shoreline would also be straightforward.	Total estimated cost is \$5 million but does not include costs of long-term remediation. More expensive than Alternative 1 but less than other action alternatives (4 and 5).  Comparable in cost to Alternative 2.
Alternative 4: Removal, On-Site Treatment, and Limited Disposal	More protective than Alternatives 1 through 3 in the long term because contaminated soil would be removed from the source areas and treated. The statutory preference for treatment as a principal element of remediation would be met. Contaminant toxicity, mobility, and volume would be reduced. Potential risks to human health and environment would be significant in the short term because of the removal component but would be reduced in the long term. Good engineering practices and mitigative measures would be impacts.	Removal and treatment would be relatively straightforward and could be implemented with readily available resources and standard procedures. The presence of UXO could affect the timeliness of implementation.	Total estimated cost is \$13 million. Alternative 4 is the highest of all the action alternatives.

TABLE 5.1 (Cont.)

Cost	Total estimated cost is \$11 million.  More expensive than Alternatives 1 through 3 but less than Alternative 5.  f
Implementability	Removal, treatment, and disposal would be relatively straightforward and could be implemented with readily available resources and standard procedures. The presence of UXO could affect the timeliness of implementation. Off-site treatment and disposal facilities are available.
Effectiveness	More protective than Alternatives 1 through 3 in the long term because contaminated soil would be removed from the source areas and treated. The statutory preference for treatment as a principal element of remediation would be met; however, it would not satisfy the statutory preference for on-site treatment. Contaminant toxicity, mobility, and volume would be reduced. Potential risks to human health and the environment would be significant in the short term because of the removal component but would be reduced in the long term. Potential short-term risks to the general public would also exist because of off-site transport of untreated hazardous material. Good engineering practices and mitigative measures would be implemented to minimize impacts.
Alternative	Alternative 5: Removal, Off-Site Treatment, and Disposal

### 6 DESCRIPTION OF FINAL ALTERNATIVES

Five interim remedial action alternatives for the J-Field TBP area main pits and pushout area were retained through the screening process:

- Alternative 1: No Action;
- Alternative 2: Limited Removal and Disposal, and In-Situ Containment;
- Alternative 3: Removal and Short-Term Storage;
- Alternative 4: Removal, On-Site Treatment, and Limited Disposal; and
- Alternative 5: Removal, Off-Site Treatment, and Disposal.

The components of these alternatives are described in Sections 6.1 through 6.5. Under Alternative 2, contaminated soil would be contained in place with a protective cover. Under Alternative 3, contaminated surface soil would be removed through excavation and placed in a short-term storage facility. Under Alternatives 4 and 5, contaminated surface soil would be removed through excavation and treated (either on-site or off-site). Treated soil under Alternative 4 would be returned to the site to be used as fill (pending a treatability variance or CERCLA waiver as described in Section 3.2) or disposed of as nonhazardous waste. Metal scrap and metals recovered from the soil washing and leaching processes would be sent to an off-site smelter, and PCB-contaminated soil would be sent to an off-site treatment and disposal facility. Under Alternative 5, all contaminated soil would be sent off-site for disposal. Because contaminated soil would remain on-site under Alternatives 2 and 3, shoreline stabilization would be an additional component as a measure to control erosion of the peninsula south of the TBP area. Engineering procedures and equipment presented in the following sections are provided to compare the feasibility of the alternatives. Final design components, procedures, and equipment selection will be developed during the remedial design phase and will incorporate information developed during the course of detailed design.

#### 6.1 ALTERNATIVE 1: NO ACTION

The no-action alternative is included as a final alternative to provide a baseline for comparison with the action alternatives. Under this alternative, no action would be taken at the TBP area, and contaminated surface soil would remain in place. Institutional control measures (described in Section 4.2.1), including access restriction, ownership and use or deed restrictions, and monitoring, would remain in effect at the area.

# 6.2 ALTERNATIVE 2: LIMITED REMOVAL AND DISPOSAL, AND IN-SITU CONTAINMENT

#### 6.2.1 Limited Removal

Under Alternative 2, approximately 400 yd<sup>3</sup> of contaminated surface soil would be removed from the two main pits, and the 5-acre area encompassing the main pits and pushout areas would be contained in place (Figure 6.1). An estimated 350 yd<sup>3</sup> of arsenic-contaminated soil would be removed from the northern main pit. The removal would include excavation of soil containing arsenic levels that exceed the EPA Region III noncarcinogenic, industrial risk-based concentration of 610 mg/kg. Confirmation samples would be collected in unexcavated soil to determine whether arsenic has been removed to levels below this criterion.

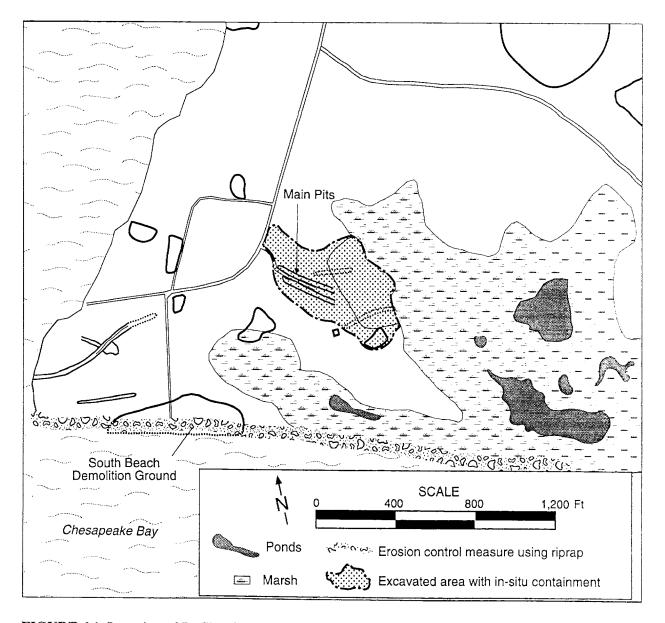


FIGURE 6.1 Location of In-Situ Containment Area and Shoreline Stabilization, Alternative 2

At the southern main pit, arsenic concentrations were below the risk-based concentration of 610 mg/kg. However, PCBs (Aroclor 1248) were detected at a concentration of 143 mg/kg in a sample collected at a depth of 2 ft from boring JBP1-E, which is at the western section of the pit (see Figure D.6). Soil would be excavated at this boring. Confirmatory analyses would be performed on the unexcavated soil to ensure that PCB concentrations were below the ARAR value of 50 mg/kg. It is estimated that about 17 yd³ of PCB-contaminated soil would be removed from the southern main pit. The site would be prepared by personnel performing a visual UXO inspection of the excavation area, grubbing and clearing vegetation in the excavation area, constructing a temporary staging area, and improving existing access roads. A work zone would be designated with temporary fencing. Access routes would be established by relying on the U.S. Army's ordnance avoidance procedures. After excavation activities were completed, the areas would be backfilled with clean soil and prepared for the emplacement of the protective cover.

#### 6.2.1.1 UXO Screening

The UXO screening effort for Alternative 2 would consist of visually inspecting the two main pits and conducting a UXO survey along the southern shoreline of the Gunpowder Neck Peninsula where erosion control measures would be implemented. The UXO survey for the main pits does not include the use of metal detection devices because the amount of buried metal debris is expected to be high. A UXO supervisor, however, would be on-site throughout the remediation effort. The survey of the shoreline would be conducted by a two-man UXO clearance team with handheld excavation tools, active electromagnetic induction detectors, and passive ferromagnetic detectors. The total area to be surveyed would be 24,000 ft<sup>2</sup> (8 ft wide by 3,000 ft long), to a depth of 2 ft below the surface.

As metal contacts were identified, the following activities would be carried out:

- Explosive-, chemical-, propellant-, or pyrotechnic-loaded UXO would be marked with a yellow survey marker; work would cease; and Emergency Notification Procedures would be implemented immediately; and
- Nonexplosive-loaded ordnance components, including armor-piercing projectiles, empty ejection munitions, and spent rocket motors (separated from warheads), and nonexplosive-loaded training munitions would be segregated for pickup by the U.S. Army Technical Escort personnel.

Other metal debris that is not ordnance-related would not be removed.

#### 6.2.1.2 Site Preparation

Following the initial UXO survey, site preparation activities would begin, including grubbing and clearing vegetation, constructing a temporary staging area, and improving existing access roads. Vegetation, which consists mainly of grass and *Phragmites*, would be removed from the entire 5-acre area. The access road would be repaired and resurfaced. The final surfacing of the road would occur after the remediation of the main pit areas is completed and all equipment is demobilized. Construction and improvement of roads would require the placement of 525 yd<sup>3</sup> of fill and gravel obtained from an off-site location.

#### 6.2.1.3 Excavation

At the northern main pit, 2 ft of soil would be removed from the entire pit (approximately 350 yd³). An additional 10 yd³ from boring TBPNPBOR1 would be removed because the arsenic concentration (2,290 mg/kg) in a sample collected from a depth of 4 ft exceeds the industrial risk-based concentration. At the southern main pit, 17 yd³ would be excavated from boring JBP1-E because Aroclor 1248 was detected at a concentration of 143 mg/kg.

#### 6.2.1.4 Mitigation and Monitoring

During remediation activities, good engineering practices and mitigative measures would be implemented to control both contaminant releases and potential exposures to workers and the general public. All workers engaged in soil removal activities would be required to wear an appropriate level of personal protective equipment. Work zones would be clearly delineated and monitored to ensure worker safety. Equipment and personnel would be decontaminated before leaving the area. Monitoring and mitigative measures for Alternative 2 are summarized in more detail in Section 7.6. Erosion control measures to be implemented include engineering the soil cover as a berm along the pushout-marsh boundary.

#### 6.2.2 Disposal

Soil from the northern main pit would be placed directly into 20-yd<sup>3</sup> rolloffs. Approximately 20 rolloffs would be required for disposal of arsenic-contaminated soil. PCB-contaminated soil would likely be containerized in 55-gal U.S. Department of Transportation—(DOT-) and TSCA-approved drums. Approximately 68 drums would be needed for containerizing the excavated soil. One trailer would be required for transportation off-site. If PCB-contaminated soil is required to be incinerated and containerized, a temporary staging area would be constructed in the work zone. Because the PCB-contaminated soil meets the definition of PCBs and PCB items

that may be stored temporarily for up to 30 days [40 CFR 761.65 (c)(I)(iii)], design specifications of PCB storage areas are not required [40 CFR 761.65(c)(I)].

Soil in the northern main pit has been characterized as Toxicity Characteristic for Lead (D008) under RCRA. The soil would require stabilization prior to land disposal. In addition, because the soil is from a National Priorities List (NPL) site, the soil must be disposed of at a facility that can accept CERCLA wastes. Soil from the northern main pit would be transported to a RCRA Subtitle C Landfill for stabilization and final disposal.

Soil in the southern main pit would be containerized in 55-gal drums and transported to a TSCA-approved incinerator. If the moisture content and PCB concentrations are low (i.e., below 500 ppm), the PCB-contaminated soil can be disposed of in a chemical waste landfill that meets the requirements of 40 CFR 761.75.

Remediation-derived wastes, such as personal protection equipment, decontamination water, debris, and scrap metal, would be characterized and disposed of accordingly (e.g., incineration, stabilization). Because remediation-derived wastes that are RCRA-hazardous would not be stored on-site for more than 90 days, the generator is not required to construct a storage facility that meets the standards of 40 CFR Parts 264 or 265, and the generator is not required to meet the permit requirements of 40 CFR Part 270. Likewise, because remediation-derived wastes that are regulated under TSCA because of PCB concentrations are not expected to be stored on-site for more than 30 days, the generator is not required to construct a PCB storage facility that meets the standards of 40 CFR 761.65(c)(I). The number of 55-gal drums for disposal of remediation-derived waste is not expected to exceed 20.

### 6.2.3 In-Situ Containment

Once the excavation was completed, the excavated areas would be backfilled with clean soil. The backfill soil would be tested to ensure that it is "clean" prior to use. The 5-acre (218,000-ft<sup>2</sup>) area would be covered with geotextile fabric and then clean soil to an elevation of just about the 100-year floodplain (7.5 ft), compacted (by standard construction equipment), regraded, covered with topsoil, and seeded (the end result would be a natural grass habitat).

### **6.2.4** Shoreline Stabilization

Shoreline stabilization would include removing debris from the shoreline and placing a fabric liner and riprap over an area 8 ft wide by 3,000 ft long along the southern shoreline (Figure 6.1). After the area has been surveyed for UXO, riprap would be transported by truck to the shore via the unnamed road that extends south from Robins Point Road to the South Beach

Demolition Ground. Front-end loaders (or comparable equipment) would be used to place riprap along the shoreline. The trucks would drive over the riprap to compact it into the sand as the wall is built along the shoreline.

#### 6.3 ALTERNATIVE 3: REMOVAL AND SHORT-TERM STORAGE

#### 6.3.1 Removal

Under Alternative 3, approximately 16,000 yd<sup>3</sup> of contaminated surface soil would be removed from the two main pits and pushout area to achieve permanent source control and long-term protection of human health and the environment (Figures 3.1 and 3.2). Removal activities would be carried out with standard construction equipment and procedures. Removal activities at the TBP area would begin with a visual surface UXO survey; subsurface UXO surveys would be conducted as required. The site would be prepared by grubbing and clearing vegetation, constructing and/or improving access roads, and extending electric service to the storage facility location. A work zone would be designated with temporary fencing. Access routes would be established by relying on the U.S. Army's ordnance avoidance procedures.

### 6.3.1.1 UXO Screening

The work zone would be surveyed for UXO in the upper 2 to 4 ft. The total area to be surveyed would be 218,000 ft<sup>2</sup>. The survey would be conducted by two-man UXO clearance teams with handheld and mechanical excavation tools, active electromagnetic induction detectors, and passive ferromagnetic detectors.

As metal contacts were identified and removed from the excavation area, the following activities would be carried out:

- Explosive-, chemical-, propellant-, or pyrotechnic-loaded UXO would be marked with a yellow survey marker; work would cease; and Emergency Notification Procedures would be implemented immediately;
- Nonexplosive-loaded ordnance components, including armor-piercing projectiles, empty ejection munitions, and spent rocket motors (separated from warheads), and nonexplosive-loaded training munitions would be segregated for pickup by the U.S. Army Technical Escort personnel; and

• Miscellaneous metallic debris would be collected, drummed, and stored at the Prototype Building until it was shipped to an off-site smelter.

Metal contacts would be marked and removed at each depth interval. Removal of metal contacts might proceed during soil excavation to limit "double handling" of the soil.

#### **6.3.1.2** Site Preparation

Following the initial UXO survey, site preparation activities would begin, including clearing and grubbing vegetated areas, constructing access roads and staging areas, and extending electric service to the site. Site preparation would require clearing and grubbing an estimated 5 acres (218,000 ft<sup>2</sup>) of vegetation. Construction and improvement of access roads connecting the excavation area with the storage facility would require clearing an area of about 15,000 ft<sup>2</sup> (15 ft by 1,000 ft). Construction of roads would require placing approximately 525 yd<sup>3</sup> of fill and gravel base obtained from an off-site location. Vegetation, which consists mainly of grass and *Phragmites*, would be removed from these areas, drummed, and characterized to determine appropriate handling. Electric service would be extended to J-Field; power needs for the storage facility would be 440 V at 580-600 kW.

#### 6.3.1.3 Excavation

Excavation and transport of the soil from the TBP area would be accomplished by using standard earthmoving equipment. After excavation, soil would be placed in trucks and transported to the storage facility. A fleet of three trucks, each with a 32-yd³ capacity, would be required. The soil would most likely require dewatering before being loaded onto the trucks. If needed, belt filter presses would be used to dewater soils through a combination of gravity draining and compression between two filter cloth belts. The water resulting from this process would be collected and characterized to determine the appropriate method of disposal. Once the excavation activities were completed, the excavated pit would be backfilled with clean soil. Backfilling activities might be necessary intermittently as the excavation proceeds to prevent sidewall collapse and water infiltration. Additional clean soil would be brought in from off-site to be used as backfill as necessary. The excavated area would be regraded, covered with topsoil, and seeded.

# 6.3.1.4 Mitigation and Monitoring

During remediation activities, good engineering practices and mitigative measures would be implemented to control both contaminant releases and potential exposures to workers and the general public. All workers engaged in soil removal activities would be required to wear an appropriate level of personal protective equipment. Work zones would be clearly delineated and monitored to ensure worker safety. Equipment and personnel would be decontaminated before leaving the area. Monitoring and mitigative measures for Alternative 3 are summarized in more detail in Section 7.6. Dust control measures would include spraying water, covering stockpiles, and covering loads during transport. Erosion control measures would be used to mitigate impacts to both the air and surface water. For air, these measures would include wetting loose material and minimizing construction stockpiles; for surface water, these measures would include isolating work areas with berms, covering stockpiles, and possibly using temporary vegetative covers. An air monitoring program would be implemented for the period of the remediation activities. Biomonitoring would also be employed to assess the impact of excavation activities on biota in the adjacent marsh.

#### **6.3.2** Storage Facility Construction

The excavated soil would be stored in an area adjacent to the Prototype Building (Figure 6.2) in a building consisting of fabric-covered steel arch trusses. Buildings of this type are relatively inexpensive, are durable enough for extended time periods, and can be erected quickly. In addition, they can cover a large area free of interior supports, are weather tight, and can be easily dismantled. A liner and leachate collection system would probably be needed to meet the RCRA waste pile regulations. The design of this type of structure could be patterned after a design for a similar soil storage project performed by the U.S. Army Corps of Engineers in Winfield, West Virginia. The foundation of the storage building would consist of a bottom layer of 3 ft of impermeable clay, a leachate collection system, and a type K concrete pad 1 ft thick with an area of about 48,000 ft<sup>2</sup> for 16,000 yd<sup>3</sup> (25,000 ft<sup>2</sup> for 4,000 yd<sup>3</sup>). Concrete barriers (like those used for highway median dividers) would be placed around the inside perimeter to protect the walls from the trucks and bulldozers piling the soil, and an air scrubber system would be installed to filter any VOCs emitted during the storage period. An excavation of about 5 ft would be needed to accommodate the foundation required for the storage facility. The soil removed from this excavation could then be used to backfill the excavated contaminated area.

#### 6.3.3 Shoreline Stabilization

Shoreline stabilization would include removing debris from the shoreline and placing a fabric liner and riprap over an area 8 ft wide by 3,000 ft long along the southern shoreline (Figure 6.1). After the area has been surveyed for UXO, riprap would be transported by truck to the shore via the unnamed road that extends south from Robins Point Road to the South Beach Demolition Ground. Front-end loaders (or comparable equipment) would be used to place riprap along the shoreline. The trucks would drive over the riprap to compact it into the sand as the wall is built along the shoreline.

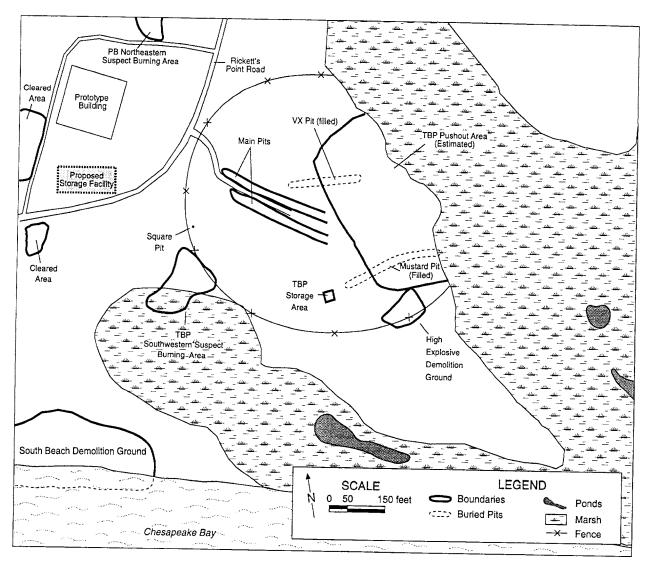


FIGURE 6.2 Location of Fenced Work Zone and Proposed Storage Facility, Alternative 3

# 6.4 ALTERNATIVE 4: REMOVAL, ON-SITE TREATMENT, AND LIMITED DISPOSAL

### 6.4.1 Removal

Under Alternative 4, contaminated surface soil would be removed to a depth of 2 ft from the two main pits and pushout area in the same manner as described for Alternative 3 (Section 6.2.1). The area to be excavated is shown in Figures 3.1 and 3.2. Electric service would be extended to J-Field; power needs for a 15 ton/h soil washing/leaching facility would be 440 V at 580-600 kW. A temporary staging area for excavated soil would be constructed in the work zone. Contaminated soil awaiting treatment would be stockpiled on a bermed and drained concrete or asphalt pad (about

48,000 ft<sup>2</sup>) and covered with plastic laminate to control dust and vapor emissions. Treated soil would be staged at a separate location (in the clean zone) until analytical testing was performed to verify that treatment goals had been achieved. The pad would be decontaminated following remedial activities and would remain in place at the site. Once the excavation and treatment activities were completed, the excavated pit would be backfilled with clean soil. The excavated area would be regraded, covered with topsoil, and seeded.

#### 6.4.2 On-Site Treatment

#### 6.4.2.1 Metals- and Semivolatile-Contaminated Soil

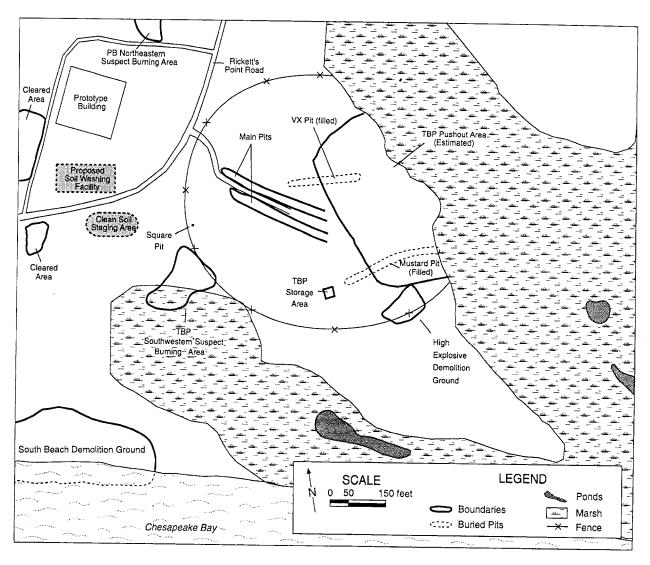
Under Alternative 4, a soil washing/leaching facility would be constructed on-site to treat all contaminated soils except those contaminated with PCBs. The approximate location of the facility is shown in Figure 6.3. The facility would be constructed on a concrete pad, about 100 by 200 ft in area. This area would require clearing and grading before the concrete foundation and necessary utilities (i.e., electricity) are installed. Water used in the process would be trucked to the site.

The soil washing/leaching facility would be used to treat metals- and semivolatile-contaminated soils (approximately 16,000 yd³) by (1) physically separating metallic fragments; (2) physically separating soil into various size fractions (coarse sand and gravel, fine sand, and silt and clay); (3) chemically removing metals bound to sand, silt, and clay; and (4) precipitating metals out of the wash for recovery. PCB-contaminated soil, if found, would be excavated and sent off-site for treatment and disposal (see Section 6.4.2.2). The preliminary conceptual design of the soil washing/leaching facility (Figure 6.4) is based on a bench-scale study that used J-Field TBP area soil (ART 1995).

The treated soil product would be placed in a pile located adjacent to the treatment plant. Treated soil would be sampled and analyzed to ensure that treatment goals had been met. Soil failing analysis might be reprocessed. Soil meeting treatment goals would be either returned to the excavated area once all excavation activities were completed (pending a treatability variance or CERCLA waiver) or disposed of as nonhazardous waste.

Delivery trucks would travel on clean access roads to eliminate the need for vehicle decontamination. Vehicles would be decontaminated, as required, before leaving the work zone.

Operation of the soil washing/leaching facility would require supervision, laborers, and laboratory and maintenance personnel. The soil washing/leaching plant would operate for 8 months, assuming a throughput of 15 tons per hour and 6-day work weeks (10-hour shifts).



 $FIGURE \ 6.3 \ Location \ of \ Fenced \ Work \ Zone \ and \ Proposed \ Treatment \ Facility \ and \ Staging \ Area, \ Alternative \ 4$ 

All processing equipment would be contained within the facility area (built on a concrete pad). Air emissions from the facility would be minimized through the use of emission control equipment, such as carbon absorption units. Because of these engineering controls, airborne emissions from the plant would be expected to be very low.

Support facilities would also be maintained on-site to provide potable water, portable sanitary facilities, and offices for the construction management staff. The soil washing/leaching plant would be dismantled at the end of the interim remedial action period and decontaminated for reuse. Process wash water would be sent to a sanitary wastewater treatment facility.

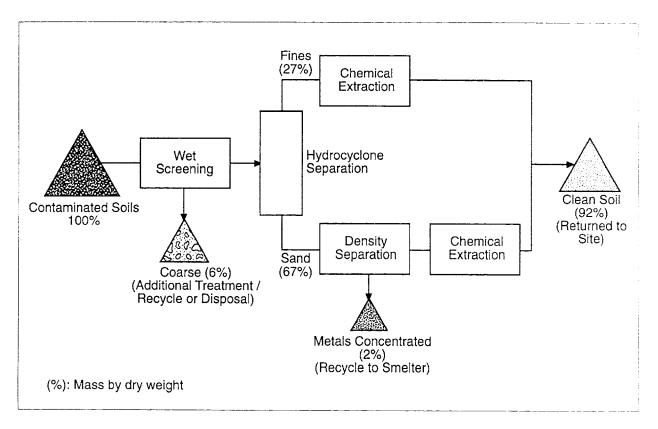


FIGURE 6.4 Process Flowchart for the Soil Washing/Leaching Facility (Source: Modified from ART 1995)

#### 6.4.2.2 PCB-Contaminated Soil

PCB-contaminated soil, if present, would likely be containerized in 55-gal DOT- and TSCA-approved drums. If PCB-contaminated soil were required to be incinerated and containerized, a temporary staging area would be constructed in the work zone. Because the PCB-contaminated soil meets the definition of PCBs and PCB items that may be stored temporarily for up to 30 days [40 CFR 761.65(c)(I)(iii)], design specifications of PCB storage areas are not required [40 CFR 761.65(c)(I)].

#### 6.4.3 Limited Disposal

All disposal activities would take place at an off-site facility. Under Alternative 4, PCB-contaminated soil would be sent off-site for treatment and disposal immediately following excavation and characterization.

Other disposal activities would include shipping metal scrap and recovered metals (from the treatment process) to an off-site smelter to be recycled.

# 6.5 ALTERNATIVE 5: REMOVAL, OFF-SITE TREATMENT, AND DISPOSAL

#### 6.5.1 Removal

Under Alternative 5, contaminated surface soil would be removed from the two main pits and pushout area in the same manner as described for Alternative 3 (Section 6.2.1). The area to be excavated is shown in Figures 3.1 and 3.2. A temporary staging area for excavated soil would be constructed in the work zone.

### 6.5.2 Off-Site Treatment and Disposal

Under Alternative 5, all excavated soil from the TBP area would be sent off-site for treatment and disposal. Following excavation, soil would be stockpiled at a temporary staging area to await loading. Hauling trucks would be decontaminated before leaving the work zone. Soils would be treated by stabilization, then disposed of in a Subtitle C landfill. PCB-contaminated soil would require stabilization or incineration, depending on the concentrations.

### 7 DETAILED EVALUATION OF ALTERNATIVES

The EPA has identified nine criteria in the NCP that must be evaluated for each alternative retained through the screening stage [Section 300.430(e)(9)(iii)]. The criteria are explained as follows.

- 1. Overall protection of human health and the environment. Addresses protection from unacceptable risks in both the short term and the long term by minimizing exposures, in accordance with the purpose and objectives of the proposed action described in Section 3. Because of its broad scope, this criterion also reflects the focus of criteria 2 through 5.
- 2. Compliance with ARARs. Addresses the attainment of federal and state environmental requirements determined to be either applicable or relevant and appropriate to the alternative on the basis of site-specific considerations. Potential ARARs are listed in Appendix B.
- 3. Long-term effectiveness and permanence. Addresses residual risks (i.e., those risks remaining after completion of a remedial action). EPA guidance states that it is usually sufficient to indicate whether an alternative has the potential to achieve the PRGs and not necessary to quantify the risk that would remain after the alternative was implemented (EPA 1991c).
- 4. Reduction of contaminant toxicity, mobility, and volume. Addresses the degree to which treatment is used to address the principal threat(s) at the TBP area; the amount of material treated; the magnitude, significance, and irreversibility of the given reduction; and the nature and quantity of treatment residuals.
- 5. Short-term effectiveness. Addresses the potential impacts of implementing the alternative to site workers, the general public, and the environment during the action period; the effectiveness and reliability of mitigative measures; and the time required to achieve protectiveness.
- 6. *Implementability*. Addresses technical feasibility, including the availability and reliability of required resources (such as specific technologies, materials and equipment, facility capacities, and skilled workers), the ease of implementation, and the ability to monitor effectiveness. This criterion also addresses administrative feasibility.

- 7. Cost. Addresses both capital costs and annual operation and maintenance costs. Costs for the individual components of the alternatives are also considered. Costs presented for Alternative 1, 3, 4, and 5 were estimated by using the Remedial Action Cost Engineering and Requirements System (RACER). Costs for Alternative 2 were developed by ICF Kaiser Engineers, Inc. Details concerning the structure of the cost estimates and assumptions used are discussed in Appendix F.
- 8. State acceptance. Addresses the comments made by the State of Maryland on the alternatives being considered for site remediation. Because the state's comments will not be received until after this report has been issued for public review, the state acceptance criterion will be addressed in the ROD.
- 9. Community acceptance. Addresses the comments made by the community on the alternatives being considered. Because public comments will not be received until after this report has been issued for public review, the community acceptance criterion will be addressed in the ROD.

Five alternatives (Alternatives 1, 2, 3, 4, and 5) were retained through the screening process presented in Section 5. Section 6 describes these final alternatives. Each alternative is evaluated in detail in the following subsections on the basis of criteria 1 through 7. In each subsection, the evaluation of an alternative against these seven criteria is organized to follow the order in which the criteria are listed in the NCP (EPA 1990a). A comparative analysis is presented in Section 8.

### 7.1 ALTERNATIVE 1: NO ACTION

# 7.1.1 Overall Protection of Human Health and the Environment

Alternative 1 would not be protective of human health and the environment over the long term. Under this alternative, the site would continue to be managed as it has been in the past. Current institutional controls, such as restricted access and monitoring, would continue to be implemented. Exposure of biota to surface soil contaminants would continue, as would the potential for exposure of site workers. Under the no-action alternative, contaminant source areas would not be removed or treated and exposures could increase over time, leading to potential adverse environmental impacts. Results of the J-Field ERA (Hlohowskyj et al. 1996) indicate that ecological resources at the TBP area are heavily impacted by past activities and current levels of contamination. The current levels of contamination also pose a risk to wildlife species at J-Field. These impacts and risks would continue under the no-action alternative.

### 7.1.2 Compliance with ARARs

Potential regulatory requirements that might be applicable or relevant and appropriate to the interim remedial action alternatives are identified and evaluated in Appendix B. Under the no-action alternative, some ARARs would not be met. For example, although RCRA disposal site requirements and LDRs are not applicable under the no-action alternative (because no movement or placement of soil would occur and no waste was placed in J-Field after 1979) (Office of Solid Waste and Emergency Response [OSWER] Directive 9234.1-01, August 8, 1988), they may be relevant and appropriate if the problem to be addressed (i.e., migration of hazardous waste constituents into the environment) is sufficiently similar. In addition, because the remediation at J-Field is being performed under the joint authority of CERCLA and RCRA (EPA and U.S. Department of the Army 1990), under Alternative 1, TBP area soil contaminated with RCRA characteristic hazardous wastes (e.g., lead), and possibly RCRA-listed hazardous wastes, would not meet the corrective action requirement under RCRA. Also, the TSCA cleanup policy (a TBC for J-Field), which requires that PCB-contaminated soil in restricted access areas be cleaned to less than 25 ppm, would not be met.

The no-action alternative also fails to meet the CERCLA-mandated preference for remedies that reduce contaminant toxicity, mobility, or volume through treatment [CERCLA Section 121(b)(1)].

# 7.1.3 Long-Term Effectiveness and Permanence

Alternative 1 would not be protective of human health and the environment over the long term. Under this alternative, maintenance and monitoring activities would be carried out for an indefinite period. Existing institutional controls would continue to limit access to site workers and other authorized personnel, thus reducing risk to the general public. However, exposures to the general public could occur through unauthorized entry. Exposures to biota and risks associated with these exposures would continue at current levels; adverse ecological impacts would be expected with long-term exposure to surface soil contamination. In addition, contaminant levels in the marsh adjacent to the pushout area could increase because of transport of contaminated material via surface water runoff, resulting in increased risks to biota in the marsh over time.

### 7.1.3.1 Protection of Human Health

Workers would be on-site periodically to carry out monitoring (e.g., toxicity testing in the marsh) and maintenance activities (e.g., mowing); the risks to workers involved in these activities are considered to be low. Because the site is restricted, potential impacts to members of the general public over the long term are also considered to be low (ICF Kaiser Engineers 1994b, 1995a).

#### 7.1.3.2 Environmental Protection

Under Alternative 1, the level of contamination and risks to biota in terrestrial and aquatic habitats in the long term would be similar to current levels; however, these levels might increase over time as a result of contaminant transport via surface water runoff.

#### 7.1.4 Reduction of Toxicity, Mobility, or Volume

Reduction of toxicity, mobility, or volume through treatment is not applicable to Alternative 1 because it does not involve treatment.

#### 7.1.5 Short-Term Effectiveness

#### 7.1.5.1 Protection of Human Health

Under Alternative 1, existing institutional controls would continue to limit access to site workers and other authorized personnel, thus reducing risks to the general public. However, exposures to the general public could occur through unauthorized entry. Short-term risks to workers involved in monitoring and maintenance activities are considered to be low (ICF Kaiser Engineers 1994b).

#### 7.1.5.2 Environmental Protection

Under Alternative 1, protection of the environment would be minimal. The level of contamination and risks to biota in terrestrial and aquatic habitats in the short term are considered to be significant as indicated by the results of the J-Field ERA (Hlohowskyj et al. 1996). Short-term risks to the environment as a result of monitoring and maintenance activities are considered to be low.

#### 7.1.6 Implementability

Minimum site operations would continue with use of readily available resources for monitoring and maintenance activities.

#### 7.1.7 Cost

Costs for Alternative 1 are associated with continuing the environmental monitoring program and maintenance activities (e.g., mowing). Estimated total annual costs for Alternative 1 would be about \$300,000.

# 7.2 ALTERNATIVE 2: LIMITED REMOVAL AND DISPOSAL, AND IN-SITU CONTAINMENT

# 7.2.1 Overall Protection of Human Health and the Environment

Alternative 2 would be protective of human health and the environment over the short term. Under this alternative, local "hot spots" of contamination (in particular, arsenic and PCBs) would be removed from the two main pits and the excavation areas would be backfilled. Following the limited removal and backfilling, surface soils in the main pits and pushout areas would be contained in place with a protective cover of soil (see Section 6.2). Exposures of biota to surface contaminants would be reduced, as would the potential for exposure of site workers and the general public.

# 7.2.2 Compliance with ARARs

Alternative 2 would comply with pertinent ARARs and TBCs, with waivers as appropriate. Location-specific ARARs and TBCs address the protection of historic sites, archeological and cultural resources, endangered species and habitats, floodplains, and wetlands. No impacts to archeological and cultural resources are expected because the area to be remediated has been subject to disturbances on a massive scale, including open burning/open detonation of high explosives and wastes, plowing, excavation of soil from large areas, and disposal of a variety of hazardous wastes. Because the pushout area, in effect, filled in existing wetlands, removal of soil from the pushout area as part of this alternative could be viewed as a wetland mitigative measure.

No critical habitats have been identified at the TBP area. Most of the area to be addressed by Alternative 2 is located within the 100-year floodplain (FEMA 1986) and borders wetlands. To address the contamination present at the site, wetlands and a portion of the 100-year floodplain would have to be disturbed, and adjacent wetlands could be affected. Mitigative measures may be required, such as wetland replacement.

Action-specific ARARs address the protection of water, sediment, soil, watersheds, and air during implementation of the remedial action. Alternative 2 would involve limited soil movement and covering; state regulations related to sediment and erosion control would apply (Water

Management 4[1] and 4[2]). Installation of the soil protective cover may require the preparation of a sediment and erosion control plan (Water Management 25[09:1] and [09:2]).

Some of the soil excavated at the site may meet the regulatory definition of characteristic waste (e.g., toxic for lead) as determined by the TCLP test. In addition, Maryland lists the following as acute hazardous wastes: CWA, waste CWA, mixtures of any of these substances and any characteristic or listed hazardous waste, and residues from treatment of CWA listed wastes. Because the TBP area was used for the disposal of CWA and CWA treatment residues, soil excavated from the site may be considered a listed acute hazardous waste (i.e., if it contains CWAs listed by Maryland). However, no CWAs have been detected in the TBP area soils to date.

Soil excavated from the southern main pit would likely contain detectable concentrations of PCBs. Because a number of samples from the southern main pit contained soil with PCB concentrations greater than 50 mg/kg (the regulatory "trigger" requiring implementation of the TSCA PCB handling regulations), the soil generated during the excavation of the southern main pit may have to be handled according to TSCA regulations. Areas contaminated with PCBs may have to be remediated to satisfy the PCB spill cleanup policy for a contaminant-specific TBC. However, as stated in the policy, PCB contamination resulting from historical spills is to be decontaminated to levels established at the discretion of EPA regional offices. Because EPA Region III is involved in the regulatory review and public participation process for this FFS, this PCB TBC would be satisfied.

State regulations related to air releases would also be action-specific ARARs. Particulate matter and VOCs would have to be controlled during earthmoving activities in accordance with Maryland requirements (Air Quality 26[11:1], 26[11:3], and 26[11:6]).

Appropriate permits would have to be obtained from the U.S. Army Corps of Engineers before the riprap berm could be constructed along the southern shoreline of the Gunpowder Neck Peninsula.

# 7.2.3 Long-Term Effectiveness and Permanence

Alternative 2 would be protective of human health and the environment over the long term. Under this alternative, contaminated surface soil from the main pits and pushout area would be contained in place with a "risk-reduction cover," and maintenance activities would continue. Monitoring activities would be conducted to assess the effectiveness and potential impacts of the protective cover. In-situ containment would reduce contaminant mobility (especially via surface water runoff); therefore, potential exposures due to contaminant releases from these source areas into the marsh would be low. Although the potential for downward leaching of contaminants into the

surficial aquifer would still exist, the flux rate of leaching contaminants is estimated to be fairly slow (see Appendix A).

Shoreline stabilization would be part of this alternative as an erosion control for the southern shoreline of the Gunpowder Neck Peninsula. However, long-term effectiveness may be affected by the potential for flooding because the TBP area is located within the 100-year floodplain (FEMA 1986).

#### 7.2.3.1 Protection of Human Health

Workers would be on-site periodically to carry out monitoring and maintenance activities. Residual risks would be reduced because contaminated soil exceeding cleanup criteria would be contained in place. Long-term exposures of workers and the general public to contaminants would be negligible because contaminated soil would be covered.

#### 7.2.3.2 Environmental Protection

Alternative 2 would result in a significant reduction in contaminant levels and exposures to biota. The in-situ containment of contaminated soil would reduce the potential for transport of contaminants to the nearby marsh and pond via surface water runoff, thereby producing a positive benefit to surface water and wetland resources in the long term. In addition, direct exposures to terrestrial wildlife and threatened and endangered species would also be reduced.

# 7.2.4 Reduction of Toxicity, Mobility, or Volume

Under Alternative 2, about 400 yd<sup>3</sup> of soil would be removed from the main pits and treated, thereby reducing contaminant toxicity, mobility, and volume in these areas. The in-situ containment component would reduce contaminant mobility via leaching and surface runoff but would not reduce contaminant toxicity or volume.

#### 7.2.5 Short-Term Effectiveness

#### 7.2.5.1 Duration of Remedial Activities

Remedial action activities (including limited soil removal and installation of a 2-ft soil cover) under Alternative 2 would be expected to be completed in about 30 days. Monitoring

activities would take place during and after implementation of the action; maintenance activities would begin after the action was implemented.

#### 7.2.5.2 Protection of Human Health

Short-term risks to site workers and the general public would be significant for Alternative 2, and mitigative measures would need to be employed (see Section 7.6). Excavation of contaminated soil at the TBP area, though not as extensive as that proposed under Alternatives 3 through 5, is complicated by the possible presence of buried UXO, some of which may contain chemical agents. The health and safety implications of encountering UXOs and CWAs or other chemical contaminants would be addressed in the health and safety plan for the selected alternative. Soil would be excavated using conventional equipment following U.S. Army standard operating procedures.

The release of particulate emissions during excavation would need to be minimized. Although workers could wear respiratory protection, off-site dispersion of contaminated dust could present a health risk to other APG personnel and the general public. Engineering controls would be needed to reduce the air concentrations of CWAs and toxic chemicals released during excavation. The use of enclosures to contain airborne emissions may be required.

Monitoring for chemical agents and hazardous chemicals would also be required during excavation. Continuous personal and work-site monitoring for airborne vapors and particulates is a standard industrial hygiene practice. Even if the highest levels of protective clothing and equipment were used, chemical monitoring would be necessary in the event that personal protective gear malfunctioned or was improperly operated. Personal protective clothing and equipment would also be required to prevent dermal exposure through direct contact with contaminated soil.

Alternative 2 also involves the placement of a 2-ft protective layer of soil over geotextile fabric. Because no ground disturbance would occur as part of this component, a UXO survey would not be required for the in-situ containment component. However, respiratory protection would still be needed to protect workers against particulate inhalation.

#### 7.2.5.3 Environmental Protection

Limited excavation and in-situ containment would disturb approximately 5 acres of soil, vegetation, and wildlife habitat. Soil excavation would disturb less than 0.3 acres, in-situ containment would disturb about 5 acres, and construction and improvement of access roads would disturb an additional 0.3 acres of the TBP area. Adverse impacts associated with these activities include increases in fugitive dust emissions and ambient noise levels and loss of vegetation and wildlife

habitat at the main pits and pushout areas. The construction of a berm along the pushout-marsh boundary would reduce surface water drainage from the pushout area into the marsh. However, the pushout area accounts for only a small portion of the area draining into the marsh, and impacts from the resulting changes in surface water runoff would be localized and minor. Although the reduction of surface water input into the marsh could result in localized adverse impacts to marsh habitat, the berm would benefit the surrounding marsh ecosystem by reducing the erosion and transport of contaminated soils into the marsh. Air quality and noise impacts could disturb ecological resources near the construction areas, but these impacts would be minor and temporary. Although some loss of vegetation and wildlife habitat is expected, the amount of vegetation and habitat that would be affected is small, none of the areas that would be disturbed contained high-quality habitats or support listed biota, and most of the disturbed areas (77%) are contaminated and have been identified to pose risk to ecological receptors at the TBP area and at J-Field as a whole.

Placement of riprap along the southern shoreline of J-Field also has potential impacts to ecological resources, although the degree of impacts would depend upon the final design of the riprap structure. Marsh vegetation that is inundated during high tides often serves as refuge and foraging habitat for many fish species, especially smaller species and juveniles of larger species. Construction of a riprap barrier can curtail or prevent access to such areas. Fishery surveys conducted by the U.S. Fish and Wildlife Service off the southern end of J-Field (Robins Point) determined that nursery habitat for striped bass and marginal nursery habitat for Atlantic croaker and white perch exist in this area (Swihart et al. 1994). The impact to fish populations due to loss of access to this area of marsh habitat is unknown, although it would probably be minor. The presence of riprap could benefit some fish species by providing artificial habitat for a number of invertebrate species that could serve as food. There is also a potential benefit from reduction of erosion along the southern shoreline of J-Field because future erosion of contaminated soils in the TBP area could pose an ecological risk to aquatic organisms.

Potential impacts to air quality and noise levels during implementation of Alternative 2 would be temporary and would cease upon completion of all excavation, construction, and restoration activities. In addition, the magnitude of the potential impacts could be mitigated during all phases of Alternative 2 through the use of good engineering practices. For example, siltation fences, berms, and a runoff collection system could be used to control runoff from excavation and construction areas, mufflers could be used to limit noise levels, and water spraying could be employed to minimize fugitive dust emissions (see Section 7.6). No state or federally listed species would be expected to be disturbed during the implementation of Alternative 2.

#### 7.2.6 Implementability

Construction and operation of the components of Alternative 2 would be straightforward. Resources are readily available for UXO screening (and safety measures), soil removal, in-situ

containment, and shoreline stabilization. Standard hand excavation tools and UXO screening methods would be used to remove contaminated soil. Soil would be sampled during excavation activities to verify that all contaminated surface soil within the excavation area was removed. The sampling procedures for soil at APG are well established. Standard earthmoving equipment would be used to install the soil cover.

Installation of the riprap berm along the southern shoreline of the Gunpowder Neck Peninsula would be relatively straightforward. Because of the shallow depths offshore, riprap would most likely have to be trucked in via Rickett's Point Road and the unnamed road that runs south toward the South Beach Demolition Ground. The berm would be built across the shoreline from west to east so that trucks could drive over the top of it.

The implementation of Alternative 2 would not adversely affect the performance of additional remedial actions that might be required at the TBP area. The ability to implement future groundwater remediation at the site (if needed) would not be affected by the excavation operations or construction of the soil cover (e.g., the soil cover would be built around the trees currently in place as part of the groundwater phytoremediation pilot study). Any additional wells (e.g., for pumping) or excavation could still be implemented, if needed.

#### 7.2.7 Cost

The components that make up Alternative 2 are shown in Table 7.1. The costs provided are for limited excavation to 2 ft, with some localized excavation to 4 ft if needed (for a total of about 400 yd³). Because this alternative would involve only limited excavation, estimates for excavation are not broken down for two depths (6 in. and 2 ft) as they are for Alternatives 3 through 5. Cost components of Alternative 2 include obtaining all necessary federal, state, and local permits; repairing and resurfacing the existing access road (final course will be applied when remediation work is complete); UXO screening; clearing and grubbing; constructing a staging area; excavating pits; disposing of materials off-site; and landscaping all excavated areas. Excavations to depths greater than 2 ft may occur if localized contamination is detected at deeper levels. This alternative also would include emplacing a 2-ft protective soil cover with a dike along the pushout-marsh boundary to minimize the impacts of flooding. Shoreline stabilization would also be included. All excavated material would be disposed of in an off-site hazardous waste landfill and TSCA-approved incinerator, as required.

The estimated cost for the UXO screening effort is \$25,000, which would consist of providing UXO supervision of the remediation effort at the main pit area and surveying up to 24,000 ft<sup>2</sup> along the shoreline. The existing access road would be repaired and resurfaced in conjunction with the construction of the staging area. The final surface course would be applied when remediation is complete and all equipment is demobilized. The estimated cost for this work is \$40,000.

TABLE 7.1 Cost Estimate for Alternative 2 (April 1996 \$)

	Cost (\$1,0	000)
Activity	Subcontract and Other Direct Costs	Contractor <sup>a</sup> Unloaded
Direct costs		
Repair/resurface existing road		40
UXO screening	25	
Clear and grub		5
Excavate/drum/stage soil (PCB)	1.4	12.6
Excavate/drum/stage soil (metals)	5.5	49.5
Off-site disposal, soil (PCB)	30	,,,,,
Off-site disposal, soil (metals)	195	
Off-site disposal, remediation-derived waste	8.5	
Sampling and analysis	9	
Landscape excavated area		12
Install 2-ft earth cover	194.5	17.5
Site stakeout and control		16
Sediment control plans and specs		5
Perimeter dike with flood protection	46	17
Remedial action professional labor		17
Remedial design		45
Shoreline stabilization	73.4	45.6
Contractor general conditions <sup>b</sup>		125
30-year operation and maintenance	50	
Contractor unloaded cost <sup>a</sup>		407.2
Contractor loaded cost <sup>a</sup>	638.3	
Indirect costs		
Contractor unloaded and overhead		736
Contractor subtotal cost		1,375
Fee (8%)		110
Total contractor cost		1,485
Other		
Contingencies (10%)		148
Project management (8%)		119
Total project cost		1,752

<sup>&</sup>lt;sup>a</sup> Loaded costs = subcontract and other direct costs; unloaded = contractor costs, less overhead.

For this estimate, ICF Kaiser Engineers, Inc., included contractor general conditions under direct costs.

The cost of metals-contaminated soil disposal in a hazardous waste disposal facility would be about \$286/ton. The cost includes transportation in roll-offs to the facility, stabilization of the material, and all taxes and fees for proper disposal. The cost for PCB-contaminated soil in 55-gal drums at a TSCA-approved incinerator is approximately \$425 per drum, which includes transportation and incineration. The cost of remediation-derived wastes such as personal protective equipment, if it requires incineration, is approximately \$425 per drum, which includes transportation.

The sampling and analysis effort would consist of performing analysis to fully characterize the excavated soil for acceptance into disposal facilities and to confirm that the excavations removed arsenic and PCB to levels at or below the noncarcinogenic industrial risk-based concentration and ARAR, respectively. The estimated cost for sampling and analysis is \$9,000.

The final component of this alternative would be placing erosion control measures along 3,000 ft of beachfront to the south of the TBP area. This component would involve clearing debris, lining the area with geotextile fabric, and then overlaying the fabric with riprap (at a cost of about \$16 per linear foot) over an area of about 8 ft by 3,000 ft. An alternative to riprap is geotube (at a cost of about \$135 per linear foot). To minimize costs, the estimate assumes that riprap would be used. The estimated beachfront stabilization cost is \$119,000.

The final two direct-cost items for this alternative are professional labor and remedial design. Professional labor amounts to \$17,000. Remedial design costs for the 2-ft excavation, soil cover, and shoreline erosion control are estimated at \$45,000 on the basis of a low level of complexity for this alternative.

The total direct and contractor overhead costs for this alternative are estimated to be \$736,000. Loaded costs, which include subcontractor and other direct costs (ODCs), added \$638,300. An 8% fee of the sum of the loaded cost, total direct cost, and overhead cost is included in this project. The sum of the fee, loaded cost, contractor direct cost, and overhead cost is the total contractor cost, which is estimated to be \$1,485,000. A contingency factor of 10% of the total contractor cost adds \$148,000 to the cost of the project. Project management adds an additional 8% of the total contractor cost (\$119,000) to the cost of the alternative. The total project cost of this alternative, which is the sum of the total contractor cost, contingency factor, and project management, is estimated to be \$1,752,000.

# 7.3 ALTERNATIVE 3: REMOVAL AND SHORT-TERM STORAGE

# 7.3.1 Overall Protection of Human Health and the Environment

Alternative 3 would be protective of human health and the environment over the short term. Under this alternative, contaminated surface soil in the main pits and pushout area would be removed and stored in a containment building located north of the TBP area for an estimated five years (see Section 6.3). Exposures of biota to surface contaminants would be reduced, as would the potential for exposure of site workers and the general public.

# 7.3.2 Compliance with ARARs

Compliance with location-specific ARARs under Alternative 3 would be the same as for Alternative 2.

Action-specific ARARs address the protection of water, sediment, soil, watersheds, and air during implementation of the remedial action. Alternative 3 would involve moving, grading, transporting, or otherwise disturbing soil; state regulations related to sediment and erosion control would apply (Water Management 4[1] and 4[2]). Implementing the soil excavation component of Alternative 3 would require the preparation of a sediment and erosion control plan (Water Management 25[09:1] and [09:2]).

Some of the soil excavated at the site might meet the regulatory definition of characteristic waste (e.g., toxic for lead) as determined by the TCLP test. In addition, Maryland lists the following as acute hazardous wastes: CWAs, waste CWAs, mixtures of any of these substances and any characteristic or listed hazardous waste, and residues from treatment of CWA wastes listed as acute hazardous waste. Because the TBP area was used for the disposal of CWAs and CWA treatment residues, soil excavated from the site might be considered a listed acute hazardous waste (i.e., if it contains CWAs listed by Maryland). However, no CWAs have been detected in the TBP area soils to date.

If hazardous waste is present, RCRA siting requirements for new treatment, storage, and disposal (TSD) facilities would apply for certain storage unit operations constructed to store waste under Alternative 3, that is, the temporary storage facility. The RCRA requirements and similar state requirements specify that any facility located in a 100-year floodplain should be constructed, operated, and maintained to prevent washout of any waste by a 100-year flood. These requirements might apply to locating and operating the short-term storage facility.

General TSD facility standards, including waste analysis, security, inspections, training, preparedness and prevention, and contingency planning/emergency procedures, would be applicable to such waste management activities. Alternatively, site managers could rely on 40 CFR 264.553 (Temporary Units) to satisfy RCRA ARARs for the storage of wastes. Under this regulation, the EPA regional administrator can determine that temporary tanks and container storage areas used for the treatment or storage of hazardous remediation wastes may meet an alternate design, operating, or closure standard.

Soil excavated from the southern main pit would likely contain detectable concentrations of PCBs. Because a number of samples from the southern main pit contained soil with PCB concentrations greater than 50 mg/kg (the regulatory "trigger" requiring implementation of the TSCA PCB handling regulations), the soil generated during the excavation of the southern main pit might have to be handled according to TSCA regulations. Areas contaminated with PCBs might have to be remediated to satisfy the PCB spill cleanup policy for a contaminant-specific TBC. However, as stated in the policy, PCB contamination resulting from historical spills is to be decontaminated to levels established at the discretion of EPA regional offices. Because EPA Region III is involved in the regulatory review and public participation process for this FFS, this PCB TBC would be satisfied.

State regulations related to air releases would also be action-specific ARARs. Particulate matter and VOCs would have to be controlled during earthmoving activities in accordance with Maryland requirements (Air Quality 26[11:1], 26[11:3], and 26[11:6]).

Appropriate permits would have to be obtained from the U.S. Army Corps of Engineers before the riprap berm could be constructed along the southern shoreline of the Gunpowder Neck Peninsula.

# 7.3.3 Long-Term Effectiveness and Permanence

#### 7.3.3.1 Protection of Human Health

Under Alternative 3, workers would be on-site periodically to carry out monitoring and maintenance activities. Residual risks would be reduced because contaminated soil exceeding cleanup criteria would be removed and contained. Long-term exposures of workers and the general public to contaminants would be negligible because contaminated soil would be removed and transported to a storage building and the excavated areas backfilled. Alternative 3 is not considered a long-term or permanent remedy; therefore, another component to this alternative would also involve (at some future date) off-site transport, treatment, and disposal of contaminated soil. Because this component would involve the off-site transport of contaminated materials, the eventual overall

risk of exposure to the general public would be comparable to that expected for Alternative 5. Site workers and off-site contractors responsible for the transport of soil would be equipped with and trained in the use of personal protective equipment. Mitigative measures are further addressed in Section 7.6.

#### 7.3.3.2 Environmental Protection

Alternative 3 would result in a significant reduction in environmental contaminant concentrations in site soils and thus reduce direct exposure of vegetation and wildlife to surface soil contaminants as well as reduce or eliminate food chain transport of contaminants to ecological receptors in higher trophic levels. Excavation and storage of contaminated surface soils would largely eliminate the transport of soil-bound contaminants by surface runoff to the nearby marsh and pond, thereby producing a long-term benefit to surface water and wetland resources at the site. Restoration of the excavation areas would also establish better-quality wildlife habitat than is currently present at the site.

# 7.3.4 Reduction of Toxicity, Mobility, or Volume

Under Alternative 3, about 16,000 yd<sup>3</sup> (27,200 tons) of soil would be removed from the main pits and pushout areas and stored temporarily on-site. Storage would take place over a period of 5 years until soil was sent off-site for permanent treatment and disposal; therefore, the reduction of contaminant toxicity, mobility, and volume through treatment would be deferred.

#### 7.3.5 Short-Term Effectiveness

# 7.3.5.1 Duration of Remedial Activities

Remedial action activities (including construction of storage facility and removal of soil) under Alternative 3 should be completed in about 4 months; the storage facility would operate for about 5 years. Monitoring activities would take place during and after implementation of the action.

#### 7.3.5.2 Protection of Human Health

The short-term risks to site workers and the general public would be significant for Alternative 3, and mitigative measures would need to be employed (see Section 7.6). Excavation of contaminated soil at the TBP area is complicated by the possible presence of buried UXO, some of

which may contain chemical agents. The health and safety implications of encountering UXO and CWAs or other chemical contaminants would be addressed in the health and safety plan for the selected alternative. Soil would be excavated by using conventional equipment following U.S. Army standard operating procedures.

The release of particulate emissions during excavation would need to be minimized. Although workers could wear respiratory protection, off-site dispersion of contaminated dust could present a health risk to other APG personnel and the general public. Engineering controls would be needed to reduce the air concentrations of CWAs and toxic chemicals released during excavation. The use of enclosures to contain airborne emissions might be required.

Monitoring for chemical agents and hazardous chemicals would also be required during excavation. Continuous personal and work-site monitoring for airborne vapors and particulates is a standard industrial hygiene practice. Even if the highest levels of protective clothing and equipment were used, chemical monitoring would be necessary in the event that personal protective gear malfunctioned or was improperly operated. Personal protective clothing and equipment would also be required to prevent dermal exposure through direct contact with contaminated soil.

Alternative 3 would also involve short-term, on-site storage of excavated soil from the TBP area. Construction of a storage facility would need to incorporate a vapor-phase emission control system to reduce risks to site workers and the general public associated with the inhalation of VOCs. Site workers at the storage facility might need to be equipped with and trained in the use of personal protective equipment, which would reduce the potential for exposure during the storage period. Mitigative measures are further addressed in Section 7.6.

#### 7.3.5.3 Environmental Protection

Excavation of contaminated soil and construction of the interim storage facility would disturb approximately 6.5 acres of soil, vegetation, and wildlife habitat. Soil excavation would disturb about 5 acres, construction of the storage facility would disturb up to 1.1 acres, and construction and improvement of access roads would disturb an additional 0.3 acres of the TBP area. Adverse impacts associated with these activities include increases in fugitive dust emissions and ambient noise levels; potential increased surface soil runoff to the marsh and pond; and loss of vegetation and wildlife habitat at the main pits, pushout area, and interim storage facility areas. Air quality and noise impacts could disturb ecological resources near the construction areas, but these impacts would be minor and temporary. Although some loss of vegetation and wildlife habitat would be expected, the amount of vegetation and habitat that would be affected is small, none of the areas that would be disturbed contain high-quality habitats or support listed biota, and most of the disturbed areas (77%) are contaminated and have been identified to pose a risk to ecological receptors at the TBP

area and at J-Field as a whole. When all excavation activities were completed, the excavated areas would be backfilled with clean fill and revegetated to restore wildlife habitat.

Placement of riprap along the southern shoreline of J-Field also has potential impacts to ecological resources, although the degree of impacts would depend upon the final design of the riprap structure. Marsh vegetation that is inundated during high tides often serves as refuge and foraging habitat for many fish species, especially smaller species and juveniles of larger species. Construction of a riprap barrier can curtail or prevent access to such areas. Fishery surveys conducted by the U.S. Fish and Wildlife Service off the southern end of J-Field (Robins Point) determined that nursery habitat for striped bass and marginal nursery habitat for Atlantic croaker and white perch exist in this area (Swihart et al. 1994). The impact to fish populations due to loss of access to this area of marsh habitat is unknown, although it would probably be minor. The presence of riprap could benefit some fish species by providing artificial habitat for a number of invertebrate species that could serve as food. There is also a potential benefit from reduction of erosion along the southern shoreline of J-Field because future erosion of contaminated soils in the TBP area could pose an ecological risk to aquatic organisms.

Potential impacts to surface water, air quality, and noise levels during implementation of Alternative 3 would be temporary and would cease upon completion of all excavation, construction, and restoration activities. In addition, the magnitude of the potential impacts could be mitigated during all phases of Alternative 3 through the use of good engineering practices. For example, siltation fences, berms, and a runoff collection system could control runoff from excavation and construction areas; mufflers could limit noise levels; and water spraying could minimize fugitive dust emissions (see Section 7.6). No state or federally listed species would be expected to be disturbed during the implementation of Alternative 3.

#### 7.3.6 Implementability

Construction and operation of most components of Alternative 3 would be straightforward. Resources are readily available for UXO screening (and safety measures), soil removal, and off-site treatment and disposal. Standard excavation and construction equipment would be used to remove contaminated soil. Soil would be sampled during excavation activities to verify that all contaminated surface soil within the excavation area was removed. The sampling procedures for soil at APG are well established.

A recent HFA survey of 10- by 10-ft areas in the pushout area showed a range of 1 to 28 contacts per 100-ft<sup>2</sup> area. Twenty-five contacts were found in a burning pit (an area of 100 by 6 ft). It is estimated that about 100 contacts per day would be identified and excavated per two-man team, provided none of the contacts were UXO. If UXO were encountered, work would cease until the UXO was located and destroyed. Downtime for this operation would be from one to four hours.

Once all contacts were excavated or destroyed, conventional earthmoving equipment could be used to excavate the cleared area to a depth of 2 ft.

The storage facility would be relatively straightforward to construct and operate. All necessary equipment is readily available; temporary structures are available through a number of companies and can be purchased as kits, which can be customized to fit project specifications. The facility would require a supervisor and a small inspection/maintenance crew.

Installation of the riprap berm along the southern shoreline of the Gunpowder Neck Peninsula would be relatively straightforward. Because of the shallow depths offshore, riprap would most likely have to be trucked in via Rickett's Point Road and the unnamed road that runs south toward the South Beach Demolition Ground. The berm would be built across the shoreline from west to east such that trucks could drive over the top of it.

Implementation of Alternative 3 would not adversely affect the performance of additional remedial actions that might be required at the TBP area. The ability to implement future groundwater remediation at the site (if needed) would not be affected by the excavation and backfilling operations; however, in the case of phytoremediation, some coordination would be needed in the timing of excavation and planting of vegetation.

#### 7.3.7 Cost

The components that make up the treatment train for Alternative 3 are shown in Table 7.2. Costs are given for both the 6-in. and 2-ft excavations. Before the interim remedial action could be performed, several site improvements must be made. The first site improvement is improvement/construction of an access road, which was deemed necessary to handle the heavy equipment that would be required to excavate and haul the soil to the temporary storage building. Construction/improvement of a one-lane gravel road was estimated to cost \$26,000. The second site improvement is extension of electric service to the storage site to provide power to the storage building for the lighting, ventilation system, and air scrubber system. The cost of this site improvement was estimated at \$82,000.

Before the site is excavated, it must be screened for UXO and then cleared of vegetation (clear and grub). Because excavation would not be greater than 2 ft, only one UXO screening is required for either excavation depth. The area screened is about 5 acres (218,000 ft<sup>2</sup>). Because of the potential for CWA, it is assumed that the UXO screening teams would use a MINICAMS to detect the presence of CWA. The total cost of UXO screening is \$156,000, for both the 6-in. and the 2-ft excavations. The cost to clear and grub the site was estimated at \$20,000.

TABLE 7.2 Cost Estimate for Alternative 3 (April 1996 \$)<sup>a</sup>

	Cost (\$	1,000)
Activity	6-in. Excavation	2-ft Excavation
Direct costs		
Construct/improve access road	26	26
Extend electric service	82	82
UXO screening	156	156
Clear and grub	20	20
Excavate pushout and pits	81	197
Construct storage building	1,024	1,966
Transport to storage building	7	23
Landscape excavated area	4	4
Remedial action professional labor	15	16
Remedial design	47	87
Beach erosion control	119	119
5-year operation and maintenance (O&M) <sup>b</sup>	66	145
Total direct costs	1,647	2,841
Indirect costs		
Contractor general conditions	501	797
Contractor overhead and profit	410	639
Total contract indirect costs	911	1,436
Total contract cost (direct and indirect)	2,558	4,277
Other		
Contingencies (10%)	256	428
Project management (8%)	205	342
Total project cost	3,019	5,047
Unit cost (\$/yd³)	749	313

<sup>&</sup>lt;sup>a</sup> Costs for the 6-in. excavation are included for comparative purposes only.

b Annual O&M cost would be about \$13,000 for the 6-in. excavation and \$29,000 for the 2-ft excavation.

Construction of the storage building would begin before excavation of the contaminated area so that the building would be ready once the excavation took place. To accommodate the different layers used in the special foundation, which include the impermeable clay layer, the leachate collection system, and the concrete pad, the area where the building will be constructed would need to be excavated to a depth of about 5 ft. However, the soil excavated could be stockpiled and used as backfill in the contaminated area. In addition to the foundation excavation and the special foundation, other cost components associated with the storage building are the building fabric, support trusses, and building erection; protective concrete barriers around the inside perimeter; inside ventilation system; air scrubber system; and site work needed for proper drainage around the buildings. The total estimated cost for these activities is about \$1,024,000 for the 6-in. excavation and \$1,966,000 for the 2-ft excavation. These costs were estimated from cost data obtained from the Army Corps of Engineers for a similar soil storage building that was constructed in Winfield, West Virginia. The storage building accounts for 67% of the total direct cost for the 6-in. excavation and 72% of that for the 2-ft excavation.

Also, while the soil is stored, annual costs would be associated with operation and maintenance (O&M) of the scrubber system. These costs were estimated to be about \$13,000/yr for the 6-in. excavation and \$29,000/yr for the 2-ft excavation. The O&M cost shown in Table 7.2 assumes that the soil is stored for 5 years. Consequently, the estimated 5-year O&M cost is \$66,000 for the 6-in. excavation and \$145,000 for the 2-ft excavation.

Excavation is the next component of the treatment train. This cost estimate is for conventional, and not remote, excavation. Because part of the excavation area is adjacent to the marsh, precautions must be taken to prevent infiltration of marsh water into the excavated area. Sheet piling would be used as a mitigative measure. In addition, the area near the marsh would be excavated and backfilled before the rest of the area was excavated. It is also assumed that all areas would be backfilled with clean soil so that potentially contaminated soil would not be exposed. The estimated cost to excavate the area to a depth of 6 in. is \$81,000, while the cost to excavate to a depth of 2 ft is \$197,000. Excavated soil would also need to be transported to the storage building by truck and piled with earthmoving equipment, such as dozers. The cost to transport and pile soil from the 6-in. and 2-ft excavations would be \$7,000 and \$23,000, respectively. After the excavation was complete, the site would be landscaped by grading and seeding, for a cost of \$4,000.

The final component of this alternative is the placement of erosion control measures along 3,000 ft of beachfront to the south of the TBP area. This component would involve clearing debris, lining the area with geotextile fabric, and then overlaying the fabric with riprap (at a cost of about \$16 per linear foot) over an area of about 8 ft by 5,300 ft. An alternative to riprap is geotube (at a cost of about \$135 per linear foot). To minimize costs, the estimate assumes that riprap would be used. The estimated beachfront stabilization cost is \$119,000.

The final two direct-cost items for this alternative are professional labor and remedial design. Professional labor accounts for the professional labor tasks required in all components of the treatment train. This cost was estimated to be \$15,000 and \$16,000 for the 6-in. and 2-ft excavations, respectively. The estimated cost of the remedial design for the 6-in. excavation is \$47,000, and the cost for the 2-ft excavation is \$87,000, assuming a moderate level of complexity for this alternative.

The total direct cost of this alternative is estimated to be \$1,647,000 for the 6-in. excavation and \$2,841,000 for the 2-ft excavation. Indirect costs, such as contractor general conditions and contractor overhead and profit, added \$501,000 and \$410,000, respectively, to the total cost for the 6-in. excavation and \$797,000 and \$639,000, respectively, for the 2-ft excavation. A contingency factor of 10% was assumed for this remedial action to account for unforeseen conditions at the site or changes in the project scope based on more detailed site information. This added \$256,000 to the 6-in. excavation and \$428,000 to the 2-ft excavation. The contingency factor used here is less than in previous draft reports because the site is now better characterized. Because more data have been obtained and analyzed, fewer unknowns and changes are expected. Finally, it is assumed that project management adds an additional 8% or \$205,000 for the 6-in. excavation and \$342,000 for the 2-ft excavation. Consequently, the total estimated cost of this alternative is \$3,019,000 for the 6-in. excavation and \$5,047,000 for the 2-ft excavation. This translates into a unit cost of \$749/yd³ for the 6-in. excavation and \$313/yd³ for the 2-ft excavation.

# 7.4 ALTERNATIVE 4: REMOVAL, ON-SITE TREATMENT, AND LIMITED DISPOSAL

# 7.4.1 Overall Protection of Human Health and the Environment

Alternative 4 would be protective of human health and the environment over the long term. Under this alternative, contaminated surface soil in the main pits and pushout area would be removed and treated. Exposures of biota to surface contaminants would be reduced, as would the potential for exposure of site workers and the general public.

# 7.4.2 Compliance with ARARs

Compliance with location-specific and contaminant-specific ARARs under Alternative 4 would be the same as for Alternatives 2 and 3. General TSD facility standards, including waste analysis, security, inspections, training, preparedness and prevention, and contingency planning/emergency procedures, would be applicable to such waste management activities. Alternatively, site managers could rely on 40 CFR 264.553 (Temporary Units) to satisfy RCRA ARARs for the treatment of wastes. Under this regulation, the EPA regional administrator can determine that

temporary tanks and container storage areas used for the treatment or storage of hazardous remediation wastes may meet an alternate design, operating, or closure standard. In addition, should soil be excavated, treated, and replaced, the RCRA LDR would apply.

Action-specific ARARs address the protection of water, sediment, soil, watersheds, and air during implementation of the remedial action. Alternative 4 would involve moving, grading, transporting, or otherwise disturbing soil; state regulations related to sediment and erosion control would apply (Water Management 4[1] and 4[2]). Implementing the soil excavation component of Alternative 4 would require the preparation of a sediment and erosion control plan (Water Management 25[09:1] and [09:2]).

State regulations related to air releases would also be action-specific ARARs. Particulate matter and VOCs would have to be controlled during earthmoving activities in accordance with Maryland requirements (Air Quality 26[11:1], 26[11:3], and 26[11:6]). Emissions resulting from remedial actions, such as soil washing/leaching, would also have to be controlled in accordance with Maryland requirements (Air Quality 26[11:15]).

#### 7.4.3 Long-Term Effectiveness and Permanence

Alternative 4 would be protective of human health and the environment over the long term. Under this alternative, contaminated soil from the main pits and pushout area would be removed and treated, and maintenance activities would continue. Monitoring activities would be conducted to assess the effectiveness and potential impacts of the remedial action. Excavation and treatment by soil washing/leaching would greatly reduce contaminant mobility and volume; therefore, potential for exposures due to contaminant releases from these source areas would be low.

#### 7.4.3.1 Protection of Human Health

Workers would be on-site periodically to carry out monitoring and maintenance activities (e.g., mowing). Residual risks would be reduced because contaminated soil exceeding cleanup criteria would be removed and treated. Long-term exposures of workers and the general public to contaminants would be negligible because contaminated soil would be removed from the site and the excavated areas backfilled.

#### 7.4.3.2 Environmental Protection

Alternative 4 would result in a significant reduction in contaminant levels and exposures to biota. The removal and treatment of contaminated soil would reduce the potential for transport

of contaminants to the nearby marsh and pond via surface water runoff, thereby producing a positive benefit to surface water and wetland resources in the long term. In addition, direct exposures to terrestrial wildlife and threatened and endangered species would also be reduced. Restoration of the excavation areas would create wildlife habitat of better quality than is currently present at these locations.

# 7.4.4 Reduction of Toxicity, Mobility, or Volume

The treatment technology implemented as part of Alternative 4 would be soil washing/leaching. The physical and chemical treatment of TBP area soils with soil washing/leaching technologies would significantly reduce contaminant mobility and volume.

It is estimated that about 16,000 yd³ (27,200 tons) of soil would be treated on-site by soil washing/leaching. The soil washing process would separate and treat the oversize and sand fractions from the soil, while concentrating the contaminants in the fines. Treating TBP area soils by soil washing/leaching could reduce the original contaminated soil volume by 94%; the remaining 6% would be sent off-site for disposal. Cleaned soil would be returned to the site, and recovered metals would be recycled. A treatability study would be needed to confirm the effectiveness of the process in achieving PRGs. Toxicity of lead would not be reduced; however, the metal would be removed and recycled.

#### 7.4.5 Short-Term Effectiveness

# 7.4.5.1 Duration of Remedial Activities

Remedial action activities (including construction of the treatment facility and soil removal, treatment, and disposal) under Alternative 4 should be completed in about 8 months; monitoring activities would take place during and after implementation of the action.

# 7.4.5.2 Protection of Human Health

The short-term risks to site workers and the general public would be significant for Alternative 4, and mitigative measures would need to be employed (see Section 7.5). Excavation of contaminated soil at the TBP area is complicated by the possible presence of buried UXO, some of which may contain chemical agents. The health and safety implications of encountering UXO and CWAs or other chemical contaminants would be addressed in the health and safety plan for the

selected alternative. Soil would be excavated with conventional equipment following U.S. Army standard operating procedures.

The release of particulate emissions during excavation would need to be minimized. Although workers could wear respiratory protection, off-site dispersion of contaminated dust could present a health risk to other APG personnel and the general public. Engineering controls would be needed to reduce the air concentrations of CWAs and toxic chemicals released during excavation. The use of enclosures to contain airborne emissions might be required.

Monitoring for chemical agents and hazardous chemicals would also be required during excavation. Continuous personal and work-site monitoring for airborne vapors and particulates is a standard industrial hygiene practice. Even if the highest levels of protective clothing and equipment were used, chemical monitoring would be necessary in the event that personal protective gear malfunctioned or was improperly operated. Personal protective clothing and equipment would also be required to prevent dermal exposure through direct contact with contaminated soil.

Another component of Alternative 4 would involve the on-site treatment of excavated soil via soil washing/leaching. The soil washing/leaching facility might need to incorporate a vaporphase emission control system to reduce risks to site workers and the general public associated with the inhalation of VOCs. Site workers operating the treatment facility would be equipped with and trained in the use of personal protective equipment, which would reduce the potential for exposure during the pretreatment and washing activities. Mitigative measures are further addressed in Section 7.6.

#### 7.4.5.3 Environmental Protection

Soil, vegetation, and wildlife habitat disturbance would occur as a result of the soil excavation activities, which would cover an area of about 5 acres. An estimated 1.5 acres of land would be disturbed to provide areas for staging soil and operating the soil washing/leaching facility. Adverse impacts associated with these activities include increases in fugitive dust emissions and ambient noise levels; potential increased soil transport via surface runoff to the marsh and pond; and loss of vegetation and wildlife habitat at the main pits, pushout area, and soil staging areas. Air quality and noise impacts could disturb ecological resources near construction areas, but these impacts would be minor and temporary. Although some permanent loss of vegetation and wildlife habitats would be expected, the amount of vegetation and habitat eliminated would be small; moreover, because of the existing contaminant levels and past activities, none of the areas to be excavated represent high-quality habitat or support listed biota. These potential risks are considered acceptable when weighed against the current adverse risk associated with contaminated surface soil (Sections 2.7.3 and 3.3). Upon completion of all treatment activities, the soil washing/leaching facility would be removed and the disturbed land would be backfilled and revegetated to restore

wildlife habitat of better quality than is currently present. Potential impacts to surface water, air quality, and noise levels during implementation of Alternative 4 could be mitigated through the use of good engineering practices. For example, siltation fences could control runoff from the excavation and construction sites, berms and a runoff collection system could control runoff from the staging areas, mufflers could limit noise levels, and water spraying could minimize dust emissions (Section 7.6). No state or federally listed species would be expected to be disturbed during the implementation of Alternative 4.

#### 7.4.6 Implementability

Construction and operation of most components of Alternative 4 would be straightforward. Resources are readily available for UXO screening (and safety measures), soil removal, and on-site treatment by soil washing/leaching. Standard excavation and construction equipment would be used to remove contaminated soil. Soil would be sampled during excavation activities to verify that all contaminated surface soil within the excavation area was removed. The sampling procedures for soil at APG are well established.

The soil washing/leaching facility would be relatively straightforward to construct and operate. All necessary equipment is readily available because the process has been used frequently in the mining industry and hazardous waste treatment applications. The treatment system would consist of a relatively standard configuration of industrial equipment. The facility would require a supervisor and general laborers with industrial work experience, as well as maintenance personnel and laboratory and administrative employees. Further bench-scale testing is required to define and optimize the design of the system.

The soil washing/leaching process would require delivery of contaminated soil by truck to the facility each day. Most likely, contaminated soil would be stockpiled at a temporary staging area until being transported to the facility for treatment. Clean soil would be stockpiled in a clean zone south of the treatment facility until all excavation activities had been completed; once excavation was complete, the cleaned soil would be used to fill the excavated area. Alternative 4 would use established technologies. The technology for soil washing/leaching has been used at several hazardous waste sites (currently, it is being demonstrated at TCAAP, New Brighton, Minnesota) and probably would not require further development before implementation at the TBP area. From 1986 to 1989, soil washing was one of the selected source control remedies at eight Superfund sites (EPA 1994). Several vendors would be available to submit competitive bids.

Disposal of the waste generated from the soil washing/treatment process is considered to be minimal. Contaminated fines and process residuals generated from soil washing would be further treated through soil leaching. The leachate solution would be cleaned and recycled continuously throughout the process. At the end of treatment, process fluids would be cleaned and put through the

sanitary wastewater system. Metal scrap and recovered metals would be shipped to an off-site smelter.

Implementation of Alternative 4 would not adversely affect the performance of additional remedial actions that might be required at the TBP area. The ability to implement future groundwater remediation at the site (if needed) would not be affected by the excavation and backfilling operations; however, in the case of phytoremediation, some coordination would be needed in the timing of excavation and planting of vegetation.

#### 7.4.7 Cost

The components that make up the treatment train for Alternative 4 are shown in Table 7.3. As for the previous alternative, costs are given for both the 6-in. and 2-ft excavations. Many of the components that make up the soil remediation treatment train for Alternative 4 are identical to those in Alternative 3: UXO and CWA screening, constructing an access road, extending electric service (to provide electric power to the soil washing/leaching facility), clearing and grubbing, excavating all areas, transporting soil to a treatment/storage area, and landscaping all excavated areas. The new components in the treatment train of this alternative are construction of a soil washing/leaching facility and staging area instead of temporary storage.

As soil was excavated, it would be transported by truck to a staging area. A staging area would be necessary because the soil would be excavated faster than it could be washed. With a soil washing facility throughput of 15 tons/h and a 10-hour daily shift working six days a week, 8 weeks would be required to wash the soil from the 6-in. excavation and 31 weeks for the 2-ft excavation. It is assumed that the soil would be stored in a bermed and drained concrete pad and covered with plastic laminate. A pad of about 25,000 ft<sup>2</sup> would be needed to store soil from the 6-in. excavation and a pad of about 48,000 ft<sup>2</sup> for the 2-ft excavation. The pads would cost \$126,000 and \$268,000, respectively. It is assumed that excavated soil would be loaded directly onto a truck and hauled to the staging area. A fleet of about three trucks with a capacity of 32 yd<sup>3</sup> each would be needed to continuously haul the excavated soil to the staging area. The estimated cost to haul the soil is about \$7,000 for the 6-in. excavation and \$23,000 for the 2-ft excavation.

The cost of the soil washing/leaching component was based on a treatability study on the TBP area soil performed by ART (1995). The cost includes both treatment of residual water from the process and disposal of the 6% solid waste fraction resulting from the process. This waste would be disposed of in an off-site hazardous waste landfill and would cost about \$312/ton (Whorton 1996). This cost includes loading onto trucks, transporting to the landfill, stabilizing the waste, and all taxes and fees for landfill disposal. Consequently, the total estimated direct cost of the soil washing/leaching component is \$2,361,000 for the 6-in. excavation and \$6,723,000 for the 2-ft excavation. This accounts for about 79% of the direct cost for the 6-in. excavation and 86% for the 2-ft excavation.

TABLE 7.3 Cost Estimate for Alternative 4 (April 1996 \$)<sup>a</sup>

	Cost (S	\$1,000)
Activity	6-in. Excavation	2-ft Excavation
Direct costs		
Construct/improve access road	26	26
Extend electric service	82	82
UXO screening	156	156
Clear and grub	20	20
Excavate pushout and pits	81	197
Construct and operate soil wash/leach facility	2,361	6,723
Construct staging area	126	268
Transport to staging area	7	23
Sampling and analysis	6	11
Landscape excavated area	4	4
Remedial action professional labor	23	23
Remedial design	101	275
Total direct costs	2,994	7,810
Indirect costs		
Contractor general conditions	752	1,562
Contractor overhead and profit	728	1,717
Total contract indirect costs	1,480	3,279
Total contract cost (direct and indirect)	4,473	11,089
Other		
Contingencies (10%)	447	1,109
Project management (8%)	358	887
Total project cost	5,279	13,085
Unit cost (\$/yd³)	1,309	811

<sup>&</sup>lt;sup>a</sup> Costs for the 6-in. excavation are included for comparative purposes only.

Before the solid waste fraction from the soil washing process was shipped to the off-site hazardous waste landfill, sampling and analysis would be required. Random samples of waste would be taken and TCLP analyses performed to characterize the waste, as required by the landfill operator. Sampling and analysis costs \$6,000 for the 6-in. excavation and \$11,000 for the 2-ft excavation.

The final two direct-cost items for this alternative are professional labor and remedial design. Professional labor amounts to \$23,000 for both excavations. Remedial design costs \$101,000 for the 6-in. excavation and \$275,000 for the 2-ft excavation, assuming a moderate level of complexity for this alternative.

The total direct cost of this alternative is estimated to be \$2,994,000 for the 6-in. excavation and \$7,810,000 for the 2-ft excavation. Indirect costs, such as contractor general conditions and contractor overhead and profit, add \$752,000 and \$728,000, respectively, to the total cost for the 6-in. excavation and \$1,562,000 and \$1,717,000, respectively, for the 2-ft excavation. A contingency factor of 10% adds \$447,000 to the cost of the 6-in. excavation and \$1,109,000 to the 2-ft excavation. Finally, it is assumed that project management adds an additional 8%, or \$358,000 for the 6-in. excavation and \$887,000 for the 2-ft excavation. Consequently, the total estimated cost of this alternative is \$5,279,000 for the 6-in. excavation and \$13,085,000 for the 2-ft excavation. This translates into a unit cost of \$1,309/yd³ for the 6-in. excavation and \$811/yd³ for the 2-ft excavation.

# 7.5 ALTERNATIVE 5: REMOVAL, OFF-SITE TREATMENT, AND DISPOSAL

# 7.5.1 Overall Protection of Human Health and the Environment

Alternative 5 would be protective of human health and the environment over the long term. Under this alternative, contaminated surface soil in the main pits and pushout area would be removed and treated. Exposures of biota to surface contaminants would be reduced, as would the potential for exposure of site workers and the general public.

#### 7.5.2 Compliance with ARARs

Compliance with location-specific and contaminant-specific ARARs under Alternative 5 would be the same as for Alternatives 2, 3, and 4. Alternative 5 would comply with the same action-specific ARARs and TBCs as Alternative 4, except for on-site disposal requirements. The key ARARs related to the off-site disposal component of this alternative would be the RCRA hazardous waste generator and record-keeping regulations and the TSCA PCB generator and record-keeping regulations. In general, the site operator would be required to ensure proper characterization of excavated soil (as soil containing listed hazardous waste, characteristic hazardous waste, and/or

PCBs), proper completion of hazardous waste and PCB manifests, receipt of "come-back" copies of manifests from destination disposal sites, and proper retention of waste characterization and manifest records for at least five years. In addition, under the CERCLA procedures for planning and implementing off-site response action (40 CFR 300.440), any facility chosen to receive CERCLA wastes for treatment, storage, or ultimate disposal must be deemed in compliance with applicable RCRA regulations and have no unpermitted releases of hazardous waste, constituents, or substances into the groundwater, surface water, soil, or air.

# 7.5.3 Long-Term Effectiveness and Permanence

Alternative 5 would be protective of human health and the environment over the long term. Under this alternative, (1) contaminated soil from the main pits and pushout area would be removed and treated and (2) maintenance activities would continue. Monitoring activities might be conducted to assess the effectiveness and potential impacts of the remedial action. Excavation and off-site treatment by chemical stabilization/solidification would greatly reduce contaminant mobility; therefore, risk of potential exposures due to contaminant releases from these source areas would be low. Restoration of the excavated areas would also create wildlife habitat of better quality than is currently present in these areas.

#### 7.5.3.1 Protection of Human Health

Workers would be on-site periodically to carry out monitoring and maintenance activities (e.g., mowing). Residual risks would be reduced because contaminated soil exceeding cleanup criteria would be removed and treated. Exposures of workers and the general public to contaminants would be negligible because contaminated soil would be removed from the site and the excavated areas backfilled.

#### 7.5.3.2 Environmental Protection

Alternative 5 would significantly reduce contaminant levels and exposures to biota. The removal and treatment of contaminated soil would reduce the potential for transport of contaminants to the nearby marsh and pond via surface water runoff, thereby benefiting surface water and wetland resources in the long term. In addition, direct exposures to terrestrial wildlife and threatened or endangered species would be greatly reduced, and restoration activities would create higher-quality habitat at the site.

#### 7.5.4 Reduction of Toxicity, Mobility, or Volume

The treatment technology implemented as part of Alternative 5 would be chemical stabilization/solidification, and any soils regulated under TSCA (i.e., PCB concentrations that exceed 50 ppm) would require incineration prior to stabilization. The chemical treatment of TBP area soils with stabilization technology would significantly reduce contaminant mobility. Incineration, if needed, would significantly reduce the toxicity of PCB contamination.

It is estimated that about  $16,000 \text{ yd}^3$  (27,200 tons) of soil would be excavated and shipped off-site for treatment by chemical stabilization/solidification. The chemical stabilization/solidification process could increase the original soil volume and weight by as much as 30% and 60%, respectively, with the addition of cement and fly ash. This would result in a total treated volume of about  $20,800 \text{ yd}^3$  and a total weight of about 35,360 tons, which would be placed in a landfill. Contaminant toxicity would not be reduced.

Limited information is available to quantify the durability of chemically stabilized/ solidified waste upon exposure to the environment, that is, if the disposal facility failed and no corrective actions were taken. The durability of the product after failure would depend on the degree of failure and the quantity and quality of infiltrating water. Contaminants could leach from the cement slowly over time, and the leach rate would increase with an increase in dissolution or fracturing of the treated waste. Because the chemically treated product could eventually be leached and degraded if the waste were continuously exposed to the environment over a long period, the chemical stabilization/solidification process could be considered not entirely irreversible. However, appropriate design and good engineering practices (e.g., monitoring and maintenance activities) over the long term would reduce the likelihood of waste exposure and degradation.

#### 7.5.5 Short-Term Effectiveness

#### 7.5.5.1 Duration of Remedial Activities

Remedial action activities (including removal of soil, treatment, and disposal) under Alternative 5 should be completed in about 4 months; monitoring activities would take place during and after implementation of the action.

#### 7.5.5.2 Protection of Human Health

The short-term risks to site workers and the general public would be significant for Alternative 5, and mitigative measures would need to be employed (Section 7.6). Excavation of

contaminated soil at the TBP area is complicated by the possible presence of buried UXO, some of which may contain chemical agents. The health and safety implications of possibly encountering UXO and CWAs or other chemical contaminants would be addressed in the health and safety plan for the selected alternative. Soil would be excavated with conventional equipment following U.S. Army standard operating procedures.

The release of particulate emissions during excavation would need to be minimized. Although workers could wear respiratory protection, off-site dispersion of contaminated dust could present a health risk to other APG personnel and the general public. Engineering controls would be needed to reduce the air concentrations of CWAs and toxic chemicals released during excavation. The use of enclosures to contain airborne emissions might be required.

Monitoring for chemical agents and hazardous chemicals would also be required during excavation. Continuous personal and work-site monitoring for airborne vapors and particulates is a standard industrial hygiene practice. Even if the highest levels of protective clothing and equipment were used, chemical monitoring would be necessary in the event that personal protective gear malfunctioned or was improperly operated. Personal protective clothing and equipment would also be required to prevent dermal exposure through direct contact with contaminated soil.

Another component of Alternative 5 would involve the off-site transport, treatment, and disposal of contaminated soil. Because this component would involve the off-site transport of contaminated materials, the overall risk of exposure to the general public is higher than for Alternative 4. Site workers and off-site contractors responsible for transporting soil would be equipped with and trained in the use of personal protective equipment. Mitigative measures are further addressed in Section 7.6.

#### 7.5.5.3 Environmental Protection

Soil, vegetation, and wildlife habitat disturbance would occur as a result of the soil excavation activities that would cover an area of about 5 acres. An estimated 0.5 acre of land would be disturbed to provide staging areas for soil awaiting transport to an off-site facility. Adverse impacts associated with these activities would include increases in fugitive dust emissions and ambient noise levels; potential increased sediment transport to the marsh and pond; and loss of vegetation and wildlife habitat at the main pits, pushout area, and soil staging areas. Air quality and noise impacts could disturb ecological resources near construction areas, but these impacts would be minor and temporary. Although some permanent loss of vegetation and wildlife habitat would be expected, the amount of vegetation and habitat eliminated would be small; moreover, none of the areas to be excavated represent a quality habitat, contain important or critical plant species, or support listed biota. These potential risks are considered acceptable when weighed against the current adverse risk associated with contaminated surface soil (Sections 2.7.3 and 3.3). When all

excavation activities were complete, the disturbed land would be backfilled and revegetated to create wildlife habitat of better quality than is currently present at the site.

Potential impacts to surface water, air quality, and noise levels during implementation of Alternative 5 could be mitigated through the use of good engineering practices. For example, siltation fences could control sediment runoff from the excavation and construction sites, berms and a runoff collection system could control runoff from the staging areas, mufflers could limit noise levels, and water spraying could minimize dust emissions (Section 7.6). No state or federally listed species would be expected to be disturbed during the implementation of Alternative 5.

#### 7.5.6 Implementability

Construction and operation of most components of Alternative 5 would be straightforward. Resources are readily available for UXO screening (and safety measures), soil removal, and off-site treatment and disposal. Standard excavation and construction equipment would be used to remove contaminated soil. Soil would be sampled during excavation activities to verify that all contaminated surface soil within the excavation area was removed. The sampling procedures for soil at APG are well established.

Alternative 5 would use established treatment technologies. The stabilization/solidification technology is considered to be a reliable process and would not require further development before implementation. Several off-site treatment/disposal vendors would be available to submit competitive bids.

The implementation of Alternative 5 would not adversely affect the performance of additional remedial actions that might be required at the TBP area. The ability to implement future groundwater remediation at the site (if needed) would not be affected by the excavation and backfilling operations; however, in the case of phytoremediation, some coordination would be needed in the timing of excavation and planting of vegetation.

#### 7.5.7 Cost

The components that make up the treatment train for Alternative 5 are shown in Table 7.4. As for Alternatives 3 and 4, costs are given for both the 6-in. and 2-ft excavations. Many of the components that make up the soil remediation treatment train for Alternative 5 are identical to those in Alternatives 2 through 4: UXO and CWA screening, constructing an access road, clearing and grubbing, excavating all areas, constructing a staging area, transporting soil to the staging area, and landscaping all excavated areas. Extension of electric service is not needed in this alternative. The

TABLE 7.4 Cost Estimate for Alternative 5 (April 1996 \$)<sup>a</sup>

	Cost (S	\$1,000)
Activity	6-in. Excavation	2-ft Excavation
Direct costs		
Construct/improve access road	26	26
UXO screening	156	156
Clear and grub	20	20
Excavate pushout and pits	81	197
Construct staging area	126	268
Transport to staging area	7	23
Off-site land disposal	2,119	5,660
Sampling and analysis	23	82
Landscape excavated area	4	4
Remedial action professional labor	16	19
Remedial design	93	238
Total direct costs	2,672	6,695
Indirect costs		
Contractor general conditions	444	824
Contractor overhead and profit	618	1,415
Total contract indirect costs	1,062	2,239
Total contract cost (direct and indirect)	3,734	8,934
Other		
Contingencies (10%)	373	893
Project management (8%)	299	715
Total project cost	4,406	10,542
Unit cost (\$/yd³)	1,093	653

<sup>&</sup>lt;sup>a</sup> Costs for the 6-in. excavation are included for comparative purposes only.

new component in the treatment train of this alternative is disposal of all excavated material in an off-site hazardous waste landfill.

As for Alternative 4, a staging area would be needed to stockpile the soil because it would probably be excavated faster than it could be transported to the off-site landfill. The staging area in this alternative was assumed to be the same size and construction as in Alternative 4. Soil would be trucked from the excavation site to the staging area and covered with plastic laminate.

As noted for Alternative 4, the contract cost of soil disposal in a hazardous waste landfill would be about \$208/ton (\$312/ton for the 6-in. excavation, because of the smaller quantities). The cost includes loading onto trucks, transporting to the landfill, stabilizing the waste, and all taxes and fees for landfill disposal. The cost of disposal for the 6-in. excavation would be \$2,119,000, whereas the cost of disposal for the 2-ft excavation would be \$5,660,000. This would amount to nearly 79% of the direct cost for the 6-in. excavation and nearly 85% for the 2-ft excavation.

To characterize the excavated soil before off-site shipment, screening for CWAs and TCLP analyses would be performed. If CWAs were present, additional processing (on-site) would be required. Additional processing costs are not included in this estimate. The estimated cost for sampling and analysis is \$23,000 for the 6-in. excavation and \$82,000 for the 2-ft excavation.

The final two direct-cost items for this alternative are professional labor and remedial design. Professional labor amounts to \$16,000 for the 6-in. excavation and \$19,000 for the 2-ft excavation. Remedial design costs \$93,000 for the 6-in. excavation and \$238,000 for the 2-ft excavation, assuming a low level of complexity for this alternative.

The total direct cost of this alternative is estimated to be \$2,672,000 for the 6-in. excavation and \$6,695,000 for the 2-ft excavation. Indirect costs, such as contractor general conditions and contractor overhead and profit, add \$444,000 and \$618,000, respectively, to the total cost for the 6-in. excavation and \$824,000 and \$1,415,000, respectively, to the total cost for the 2-ft excavation. A contingency factor of 10% adds \$373,000 to the cost of the 6-in. excavation and \$893,000 to the 2-ft excavation. Finally, it is assumed that project management adds an additional 8%, or \$299,000 for the 6-in. excavation and \$715,000 for the 2-ft excavation. Consequently, the total estimated cost of this alternative is \$4,406,000 for the 6-in. excavation and \$10,542,000 for the 2-ft excavation. This translates into a unit cost of about \$1,093/yd³ for the 6-in. excavation and \$653/yd³ for the 2-ft excavation.

# 7.6 MONITORING AND MITIGATIVE MEASURES

The primary monitoring and mitigative measures that would be used at the TBP area during remediation are summarized in Table 7.5. These measures would provide a high degree of effectiveness in minimizing the potential for adverse effects associated with remediation activities.

TABLE 7.5 Major Monitoring and Mitigative Measures for Action Alternatives

Factor	Potential Impact	Mitigative Measure
Construction and excavation activities	Transport of contaminated and uncontaminated soil to adjacent marsh by surface water runoff	Good engineering practices would be implemented, including sediment barriers, dikes, siltation ponds, and drainage channels to direct runoff away from marsh; surface would be backfilled upon completion of excavation.
		In addition, surface water and sediment would continue to be monitored for related contaminants so contaminated media could be addressed appropriately (through removal or treatment), as needed.
	Loss of terrestrial habitats	Habitats would be restored, as appropriate. The final form of mitigation would be determined in consultation with appropriate state and federal agencies.
	Disturbance of local biota and site workers by noise	Vehicle and equipment mufflers would be checked periodically and maintained in good condition.
	Disturbance of local biota, site workers, and general public and impacts to local air quality as a result of fugitive dust emissions	Dust would be controlled by using wet methods and covers at the excavation site, along access roads, and at staging areas. Work areas would be covered, as needed (e.g., at night and during high winds).
Transport of contaminated soil from TBP area to on-site storage or treatment facility	Accidental spill (release) of contaminated material as a result of equipment failure or vehicular accident	Soil would be transported in covered trucks traveling at low speeds. Soil would be dewatered before transport.  Contingency plans would be in place to address any spills that might occur during transport.
	Inadvertent transport of contaminated material on haul vehicle surfaces or tires leaving work zone	Haul vehicles would be decontaminated and inspected before leaving the work zone.

TABLE 7.5 (Cont.)

Mitigative Measure	Soil would be transported in closed containers. Contingency plans would be in place to address any spills that might occur during transport.	Haul vehicles and containers would be decontaminated and inspected before leaving the work zone.	All activities would be conducted in accordance with project health and safety plans and would include continuous monitoring of the work environment, UXO screening, and the use of protective equipment, as needed.	Air and water would be monitored at the site and vicinity, and appropriate responses would be implemented if measured contaminant levels increased significantly above background. Access to construction and excavation areas would be limited; public vehicle access would also be limited along some of the off-site haul routes. Engineering controls would be applied to minimize dust and erosion during remedial action activities. Decontamination methods would be employed to minimize vehicle tracking of contaminants to surrounding uncontaminated areas. All traffic associated with the remedial action would be coordinated to minimize impacts on nearby facilities.
Potential Impact	Accidental spill (release) of contaminated material as a result of equipment failure or vehicular accident	Inadvertent transport of contaminated material on haul vehicle surfaces or tires leaving work zone	Protection of site workers	Protection of the general public
Factor	Transport of contaminated soil to an off-site treatment and disposal facility		All phases of remedial activities	

TABLE 7.5 (Cont.)

Mitigative Measure	Air quality would be monitored for contaminated particulates at the site perimeter. Surface water downgradient of excavation and construction areas would be monitored for contaminants. Appropriate responses would be implemented as indicated by monitoring results.	Disturbed areas would be restored by backfilling with clean fill, regrading, and revegetating with native and/or forage species. If necessary, wetlands would be constructed, as indicated, on the basis of consultation with the appropriate state and federal agencies.
Potential Impact	Environmental monitoring	Environmental restoration
Factor	All phases of remedial activities (cont.)	

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# 8 COMPARATIVE ANALYSIS OF ALTERNATIVES

The comparative analysis of final interim remedial action alternatives for the J-Field TBP area compares the alternatives according to the nine evaluation criteria described in Section 7. This analysis is the second stage of the detailed evaluation process and provides information for making a balanced decision for site cleanup. For this analysis, the nine criteria are grouped into three general categories that make up the tiered evaluation system identified in the NCP (EPA 1990a): threshold criteria, primary balancing criteria, and modifying criteria.

The alternatives are compared in Sections 8.1 and 8.2 according to the threshold and primary balancing criteria, and Table 8.1 presents the results of this analysis. Section 8.3 introduces the modifying criteria. The comparative analysis is summarized in Section 8.4.

#### 8.1 THRESHOLD CRITERIA

The threshold criteria category contains the two criteria that must be satisfied by the selected alternative:

- Overall protection of human health and the environment and
- Compliance with ARARs, unless a waiver condition applies.

These criteria are of greatest importance in the comparative analysis because they reflect the key statutory mandates of CERCLA. If an alternative does not satisfy both criteria, it cannot be selected as the cleanup remedy.

# 8.1.1 Overall Protection of Human Health and the Environment

Alternatives 2, 4, and 5 would provide long-term protection of human health and the environment. This protection could not be ensured by the no-action alternative (Alternative 1) because only general baseline maintenance and monitoring activities would continue, and contaminants could migrate over time (e.g., from surface runoff into the marsh) and result in possible future adverse impacts. Alternative 3, intended to be an interim measure only, would also not provide long-term protection of human health and the environment, although it would meet project objectives by reducing exposures and minimizing contaminant migration by removing the sources of contamination. Alternative 2 would limit exposures to contaminants by removing "hot spots" in the main pits and covering remaining contaminated surface soil. Arsenic- and PCB-contaminated soil would be sent off-site for treatment and disposal. Shoreline stabilization would be an adjunct to this

# TABLE 8.1 Comparative Analysis of Alternatives

Alternative I: No Action	Alternative 2: Limited Removal and Disposal, and In- Situ Containment	Alternative 3: Removal and Short-Term Storage	Alternative 4: Removal, On-Site Treatment, and Limited Disposal	Alternative 5: Removal, Off-Site Treatment, and Disposal
Overall Protection of Human Health and the Environment	ilth and the Environment			
Would not ensure protection of human health and the environment in the long term. Source areas would not be removed or treated, and exposures could increase over time.	Engineering and mitigative measures would be employed during the remedial action period so that no significant adverse impacts would occur to the general public or environment. Worker exposures would be similarly controlled to levels within health-protective limits. Long-term exposures would be minimized by removing "hot spots" (arsenic and PCBs) from the main pits and by covering remaining contaminated soil with a protective soil cover.	Somewhat greater than Alternative 2. Long-term exposures would be minimized by removing and storing contaminated soil from the main pits and pushout area for about 5 years. At the end of that time, an additional component (e.g., off-site treatment and disposal) would be needed to ensure long-term protection.	Generally similar to Alternative 3.  Long-term exposures would be minimized by removing and treating contaminated soil from the main pits and pushout area.	Generally similar to Alternatives 3 and 4. Treatment and disposal of a large volume of contaminated soil would be conducted off-site; therefore, the overall risk of exposure to the general public would be higher than that for Alternatives 2, 3, and 4.
Compliance with ARARs				
Would not meet all ARARs, including the corrective action requirements of RCRA and TSCA. In addition, would not satisfy the CERCLA-mandated preference for remedies that reduce contaminant toxicity, or volume mobility, or volume through treatment.	Would meet all pertinent ARARs (with waivers as appropriate), including those that address protection of endangered species and habitats, floodplains, and wetlands. Implementing Alternative 2 might require preparation of a sediment and erosion control plan.	Same as Alternative 2. The on-site storage facility would be operated to ensure compliance with RCRA.	Same as Alternatives 2 and 3. The on-site treatment facility would be operated to ensure compliance with RCRA.	Same as Alternatives 2, 3, and 4, with additional disposal requirements that would be met, including RCRA hazardous waste generator.

# TABLE 8.1 (Cont.)

Alter Alter Situ Costinct of Situ Cost and Situ Contain what reduce and disposition and the situ Cost and Situ Cos	mative 2:  Alternative 4:  Alternative 5:  Alternative 5:  Alternative 5:  Removal, On-Site Treatment, Alternative 5:  Removal, On-Site Treatment, and Disposal  and Alternative 1, Similar to Alternative 4, except that ated soil would be because contaminated soil would be because contaminated soil would be surface soil would be removed and stored in an enclosed surface soil would be removed and to the environment could result in	facility to provide an interim solution freated (soil washing/leaching) to for risk posed by contaminated surface soil. Soil cleanup criteria would be applied as appropriate to the removal of soils, so the action would reduce contaminant concentrations remaining in soil to the most protective concentrations remaining in soil to the most protective concentrations remaining in soil to the most protective levels practicable.	ment component Short-term storage of soil in an on-site reduce contaminant facility would significantly reduce contaminant contaminant mobility. An estimated contaminant mobility. An estimated 16,000 yd³ (27,200 tons) of soil would not be stored on-site for about 5 years. Would not be reduced.  Short-term storage of soil in an on-site washing/leaching would significantly reduce contaminant mobility. An estimated icantly reduce contaminant mobility would be treated on-site by soil would be treated off-site by chemical stabilization/ washing/leaching. This process could about then be sent off-site for disposal.  Reduced. Soil would be treated on-site by soil with the stanted incantly reduce contaminant mobility. An estimated incanted in
Alternative I: Limitee No Action  Long-Term Effectiveness and Permanence Current exposures and impacts would continue and could increase over time because of continued contaminant migra- tion. Existing institutional controls would continue to limit access to site workers and other cleanup authorized personnel, thereby removal reduction of Toxicity, Mobility, or Volume Toxicity, mobility, and volume Toxicity, mobility, and volume Toxicity, mobility, and volume removal Contaminated soil would mobility be some removal Contaminated contaminated soil would si mobility	Alternative 2: Situ Containment Situ Containment  protective than Alternative 1  c contaminated soil would be becty removed; the remaining soil ren	on. Existing institutional exposure of human and environmental for risontrols would continue to limit receptors to surface contamination. Soil soil. Scess to site workers and other cleanup criteria would be applied to the applied to the applied to the applied to the contamination of soil ducing risk to the general of the area to be covered.	ifu containment component ignificantly reduce contaminant. Contaminant volume would what reduced by the limited and disposal component. Inant toxicity would not be

# TABLE 8.1 (Cont.)

Alternative 1: No Action	Alternative 2: Limited Removal and Disposal, and In- Situ Containment	Alternative 3: Removal and Short-Term Storage	Alternative 4: Removal, On-Site Treatment, and Limited Disposal	Alternative 5: Removal, Off-Site Treatment, and Disposal
Short-Term Effectiveness				
Current exposures and adverse impacts would continue. Existing institutional controls would continue to limit access to site workers and other authorized personnel, thus reducing risks to the general public. Short-term risks to site workers and the environment as a result of monitoring and maintenance activities are considered low.	Current exposures and adverse impacts would continue. Existing institutional controls would continue. Existing institutional controls would continue. Existing institutional controls would continue to limit access to site workers and other authorized personnel, thus reducing risks to the general public. Short-term risks to site workers and the environment as a result of monitoring and maintenance activities are considered low.  No adverse impacts to the general public are expected from contaminant releases during implementation of this alternative. Soil, vegetation, and wildlife disturbances would be significant during soil excavation activities. An estimated 5 acres would be affected. Most impacts to biota would be temporary. Surface water imput would be reduced, but reduction in erosion and transport of soils into the marsh ecosystem. Mitigative measures would be implemented to minimize impacts to air quality and those associated with noise levels.	Exposures could be higher than Alternatives I and 2 in the short term because of particulate and VOC emissions associated with removal, transportation, and storage activities. Mitigative measures would be implemented to minimize potential human health and environmental impacts. Risks to a site worker would increase compared to Alternative 1; however, worker health and safety precautions would be used to control exposures. No adverse impacts to the general public are expected from contaminant releases during implementation of this alternative. Soil, vegetation, and wildlife disturbances would be significant during soil excavation activities. An estimated 5 acres would be affected. Most impacts to biota would be temporary. Activities are not expected to affect threatened or endangered species. Mitigative measures would be employed to minimize impacts to surface water resources (including the adjacent marsh), air quality, and those associated with noise levels.	Similar to Alternative 3.	nilar to Alternative 3. Similar to Alternatives 3 and 4, except that overall risk to the general public would be higher because this alternative would involve the off-site transport of a large volume of contaminated soil. Additional mitigative measures would be implemented to reduce related impacts.

TABLE 8.1 (Cont.)

Alternative 5: Removal, Off-Site Treatment, and Disposal	Similar to Alternatives 3 and 4. Chemical stabilization/solidification is an established technology and would not require further development before it could be implemented.	The total cost would be about \$10.5 million (for 16,000 yd³ of soil), which is lower than Alternative 4 for the same overall level of effectiveness. Off-site treatment and disposal would account for 85% of the total direct cost of this alternative.
Alternative 4: Removal, On-Site Treatment, and Limited Disposal	Similar to Alternative 3. Soil washing/leaching has been used at several hazardous waste sites and probably would not require further not development before it could be implemented at the TBP area. Further bench-scale testing would be required to refine and optimize	The total cost would be about \$10.5 million (for 16,000 yd³ of soil), of soil), which is the highest of the action alternatives. The soil account for 86% of the total direct cost of this alternative.  The total cost would be about \$10.5 million (for 16,000 yd³ of soil), which is lower than Alternative 4 for the action alternative. Off-site treatment and disposal would account for 86% of the total direct cost of this alternative.
Altemative 3: Removal and Short-Term Storage	Fairly straightforward to implement. Resources are readily available for UXO screening, soil removal, and on-site storage. Short-term storage facilities are available as kits and can be customized to meet project specifications.	The total cost would be about \$5.0 million (for 16,000 yd³ of soil), the second lowest of all the action alternatives. This cost, however, does not include the cost of an additional component (off-sie treatment and disposal) that would be needed to ensure long-term protectiveness and permanence. Therefore, while this alternative would achieve project objectives of reducing exposures and minimizing contaminant migration at a low cost, it is considered the least costeffective because additional components would be required when the 5-year storage period ended.
Alternative 2: Limited Removal and Disposal, and In- Situ Containment	Fairly straightforward to implement. Resources are readily available for UXO screening, soil removal, in-situ containment, and shoreline stabilization.	t in the short term (about to in the short term (about to in the short term (about to in the short term (about a 10-year action alternatives. This cost, however, reflects only O&M costs for the next of effectiveness would be addition, the cost could an addition, the cost could entially higher than the alternatives over the long ceause conditions could are also not reflected in this estimate.
Alternative 1: No Action	Implementability  Minimum site operations (monitoring and maintenance) would continue with the use of readily available resources.	The total cost would be the lowest in the short term (about \$3 million over a 10-year period), but the comparative level of effectiveness would be low. In addition, the cost could be potentially higher than the action alternatives over the long term because conditions could worsen over time, necessitating an expensive expanded response in the future.

activity to prevent erosion of the shoreline to the south of the TBP area. Alternative 4 would limit exposures to contaminants by removing the sources of contamination, treating the contaminated soil on-site (via soil washing/leaching), then returning the cleaned soil to the site to be used as fill. In addition, PCB-contaminated soil would be sent off-site for treatment (incineration or stabilization) and disposal. Alternative 5 is similar to Alternative 4, except that the soil would be sent off-site for treatment (via stabilization/solidification) and disposal. Clean soil would be brought in to fill the excavated areas.

The two main differences between Alternatives 4 and 5 are the treatment method and disposal location, which includes a transportation component for off-site disposal. Under both Alternatives 4 and 5, PCB-contaminated soil would be sent off-site for treatment and disposal. Under Alternative 4, contaminated soil (an estimated 16,000 yd³ for an excavation to a depth of 2 ft) would be treated on-site with a small volume of material sent off-site for disposal. Under Alternative 5, all of the 16,000 yd³ of contaminated soil would be sent off-site for treatment and disposal.

Although potential health impacts during cleanup activities could be significant for all the alternatives, mitigative measures would be employed to control exposures to levels within health-protective limits. Impacts to the general public would be relatively higher for Alternative 5 than for Alternatives 2, 3, and 4 because of the increased likelihood of exposures and accidents during the waste handling and transportation activities associated with off-site disposal. Impacts to site workers would be highest for Alternatives 3, 4, and 5 because these alternatives involve large-scale excavation components.

Environmental impacts for all the alternatives could be significant. Each alternative involves excavating soil from source areas and constructing temporary staging areas, although excavation for Alternative 2 is fairly limited. Potential environmental impacts include increased fugitive dust and VOC emissions, increased ambient noise levels, increased sediment transport to the marsh and pond, and loss of vegetation and wildlife habitat. Most of these impacts are considered short term, although some loss of habitat may result. This impact would be offset by backfilling and revegetating the site, creating a wildlife habitat of better quality than is currently present at the site. Mitigative measures, such as those described in Section 7.6, would be employed to minimize these impacts.

#### 8.1.2 Compliance with ARARs

Except for Alternative 1, the attainment of ARARs under each final alternative would be comparable; applicable requirements would be met both during and following cleanup unless a waiver condition is applied. A comprehensive list of potential ARARs for this remedial action is presented in Appendix B; key requirements are discussed in Section 7 within the evaluation of each alternative against this criterion.

Alternative 1 would not attain certain applicable requirements, including RCRA corrective action requirements and TSCA. Alternatives 2, 3, 4, and 5 would meet applicable standards.

# 8.2 PRIMARY BALANCING CRITERIA

The primary balancing criteria category contains the five criteria used to assess the relative advantages and disadvantages of the alternatives to determine the most appropriate solution for a given site:

- Long-term effectiveness and permanence;
- Reduction of toxicity, mobility, or volume through treatment;
- Short-term effectiveness:
- Implementability; and
- Cost.

The first and second criteria address the statutory preference for treatment as a principal element of the remedy and the bias against off-site disposal of untreated waste. Together with the third and fourth criteria, they form the basis for determining the general feasibility of the remedy and whether costs are proportional to the overall effectiveness, considering both the cleanup period and the time following cleanup. By this means, it can be determined whether the remedy is cost-effective.

# 8.2.1 Long-Term Effectiveness and Permanence

Alternative 1 would not ensure long-term protection of human health or the environment because contaminants would be left in place, resulting in continued exposures to site workers and biota. In addition, contaminant levels in the marsh adjacent to the pushout area could increase due to contaminant migration from the source areas. Alternative 3, an interim measure, also would not ensure long-term protection, although it would accomplish project objectives. A long-term and more permanent component (such as off-site disposal) would be required to ensure long-term effectiveness. In contrast, Alternative 2 would provide long-term protection by limited removal and treatment and in-situ containment; however, long-term effectiveness could be affected by the potential for flooding. Alternatives 4 and 5 also would provide a permanent solution and long-term protection because they both involve removing and treating contaminated soil. The cleanup activities under all alternatives would reduce risks to levels considered protective of both human health and the environment.

# 8.2.2 Reduction of Toxicity, Mobility, or Volume

The toxicity, mobility, or volume of contaminated material at the TBP area would not change under Alternative 1. In contrast, Alternative 2 would reduce contaminant mobility; however, because Alternative 2 involves only limited removal and treatment, contaminant toxicity and volume would be only somewhat reduced. Alternatives 3, 4, and 5 would remove contaminated soil, and the overall reduction in contaminant mobility would be higher than Alternative 2 and generally similar for each alternative. The toxicity and volume of contaminated soils would not be affected by Alternative 3. The volume of contaminated soils would be reduced by about 94% under Alternative 4; cleaned soil would be returned to the site, and recovered metals would be recycled. Under Alternative 5, soil volume would increase because of the addition of cement and fly ash to the soil matrix to stabilize it before disposal. Soil volume could increase by as much as 30% under this alternative. The toxicity of PCB-contaminated soil would be reduced by incineration under Alternatives 2, 4, and 5. Toxicity of lead would not be reduced under any alternative; however, the metal would be removed and recycled under Alternative 4.

#### 8.2.3 Short-Term Effectiveness

For Alternative 1, conditions would remain essentially the same in the short term, and no significant changes in potential exposures would be expected. Estimated risks associated with these exposures have been reported in the BRA (ICF Kaiser Engineers 1995b) and ERA (Hlohowskyj et al. 1996). For Alternatives 2, 3, 4, and 5, the various removal, treatment, and disposal activities would result in increased short-term exposures compared with Alternative 1. The short-term impacts associated with excavating and constructing temporary staging areas would be similar for all alternatives because the same procedures would be used. Potential impacts to the public would be minimized through the use of protective mitigative measures (Section 7.5). The risk of transportation accidents and related exposures would be highest for Alternative 5 because the greatest volume of soil would be sent off-site for disposal (resulting in a larger number of truck trips to the treatment and disposal facilities). The berm to be built as part of the in-situ containment component under Alternative 2 would result in a reduction of surface water input into the adjacent marsh but would produce the overall benefit of reducing erosion transport of contaminated soils into the marsh.

Potential short-term environmental impacts for all alternatives include increased fugitive dust and VOC emissions, increased ambient noise levels, increased sediment transport to the marsh and pond, and temporary loss of vegetation and wildlife habitat. Mitigative measures, such as those described in Section 7.6, would be employed to minimize these impacts.

# 8.2.4 Implementability

All the action alternatives would be fairly straightforward to implement. The UXO screening, removal, treatment, disposal, and in-situ containment activities could be carried out with standard equipment and procedures and readily available resources. The storage facility under Alternative 3 can be purchased as a kit and customized according to project specifications. The soil washing/leaching technology under Alternative 4 has been used at several hazardous waste sites and probably would not require further development for implementation at the TBP area, although further pilot-scale testing would be required. The stabilization/solidification treatment technology under Alternative 5 is well established and reliable.

The implementation of Alternative 2, 3, 4, or 5 would not adversely affect the performance of any future remedial actions that might be required at the TBP area. For example, the ability to implement future groundwater remediation (if needed) at the site would not be affected by excavation and backfilling operations or by emplacement of the "risk-reduction cover."

#### 8.2.5 Cost

Alternative 1 would include monitoring and maintenance costs and would be the least expensive of all the alternatives in the short term. However, total costs are expected to be highest in the long term because site conditions could worsen over time in the absence of cleanup, such that the potential impacts and the magnitude of the cleanup effort could increase in the future. Therefore, the cost-effectiveness of the no-action alternative is low.

Preliminary costs estimated for Alternatives 2, 3, 4, and 5 allow a balanced comparison for considering overall effectiveness. Final costs will be developed during the detailed design stage after the remedy for site cleanup is selected. The costs presented in this report were estimated by using the RACER model (Appendix F) and information supplied by vendors. Table 8.2 lists comparative costs for the removal, storage, treatment, and disposal components of Alternatives 2, 3, 4, and 5. Alternative 2 has the lowest total costs. Alternative 4 costs more than Alternative 5 because higher costs are associated with on-site treatment (soil washing/leaching) and disposal. Although Alternative 4 offers no increased benefit for overall protectiveness, it does satisfy EPA's preference for on-site treatment. Of concern for Alternative 5 is the continued, long-term monitoring and maintenance of the off-site disposal facility. Because it would be a commercial facility, the maintenance of institutional controls at the site would be the responsibility of a private company instead of the federal government. Breakdowns in institutional controls resulting in contaminant releases from U.S. Army waste could lead to future liability issues. This concern does not exist for the on-site treatment of contaminated soil under Alternative 4 because soil would be cleaned and returned to the site (pending a treatability variance), and metals would be recovered and recycled.

TABLE 8.2 Comparative Costs for Cleanup Activities under Alternatives 2, 3, 4, and 5

		Estimated	Cost (\$ million)	
Activity	Alternative 2	Alternative 3	Alternative 4	Alternative 5
Removal	0.1	$0.4 (0.3)^a$	0.4 (0.3)	0.4 (0.3)
In-situ containment	0.4	$NA^b$	NA <sup>b</sup>	NAb
Storage	NA <sup>c</sup>	2.2 (1.2)	NA <sup>c</sup>	NA <sup>c</sup>
Treatment	d	NA <sup>e</sup>	6.4 (2.5)	_d
Disposal	0.3	NA <sup>e</sup>	0.3 (0.1)	6.1 (2.3)
Other <sup>f</sup>	1.0	2.4 (1.5)	6.0 (2.4)	4.0 (1.8)
Total	1.8	5.0 (3.0)	13.1 (5.3)	10.5 (4.4)

<sup>&</sup>lt;sup>a</sup> Cleanup costs listed are for excavation to 2 ft (total soil volume of 16,000 yd<sup>3</sup>); numbers in parentheses represent cleanup costs for excavation to 6 in. (total soil volume of 4,000 yd<sup>3</sup>) for Alternatives 3, 4, and 5.

Cost estimates for Alternative 4 were based on the assumption that soil volume would be reduced by 94%, resulting in 6% of the soil being sent off-site for further treatment and disposal.

Alternative 2 is considered the most cost-effective alternative because it provides overall protection for human health and the environment and has the lowest total cost as compared to Alternatives 3, 4, and 5. However, because this alternative has only been costed to 30 years, it is not directly comparable to the costs for Alternatives 3, 4, and 5.

b Not applicable; in-situ containment is not a component of Alternative 3, 4, or 5.

<sup>&</sup>lt;sup>c</sup> Not applicable; storage facility is not a component of Alternative 4 or 5.

<sup>&</sup>lt;sup>d</sup> Treatment and disposal costs are estimated together for Alternatives 3 and 5.

<sup>&</sup>lt;sup>e</sup> Not applicable; treatment and disposal is not a component of Alternative 3.

Other costs include indirect costs, such as contractor overhead and profit, remedial design, labor and project management, and contingencies.

#### 8.3 MODIFYING CRITERIA

The modifying criteria include:

- State acceptance and
- Community acceptance.

As discussed in Section 7, this category can be fully considered only after this FFS has been issued to the state and the public for formal comment. Therefore, these modifying criteria are not addressed in this comparative analysis. They will be addressed in detail in the Responsiveness Summary for the ROD for this remedial action.

#### 8.4 SUMMARY

In summary, all of the final remedial action alternatives for the TBP area, except for the no-action alternative (Alternative 1), satisfy the threshold criteria for protecting human health and the environment and complying with regulatory requirements, with waivers as appropriate. Under each alternative, exposures and risks would be minimized by removing the sources of contamination and treating the contaminated soil. Alternative 2 would also involve covering contaminated soil in place to reduce surface exposures. Overall protectiveness under Alternatives 4 and 5 would be comparable to and somewhat greater than that under Alternative 2 because contaminant removal and treatment would be the major components of these alternatives. Alternative 3, while protective in the short term, would require an additional component (e.g., off-site treatment and disposal) to be as protective as Alternatives 4 and 5.

With regard to the primary balancing criteria, only Alternatives 2, 4, and 5 are expected to provide a permanent solution that would ensure protection for a very long time, although long-term effectiveness under Alternative 2 could be affected by the potential for flooding. It is possible that the soil washing/leaching treatment under Alternative 4 would be more protective than Alternative 5 if, at some future date, the stabilized/solidified waste were to be exposed to the environment and contaminants leached. However, appropriate design and good engineering practices would minimize the likelihood of such an occurrence.

Each action alternative would reduce contaminant mobility. Waste toxicity and volume would be only somewhat reduced under Alternative 2. Waste volume would not be affected by Alternative 3. Waste volume would decrease under Alternative 4. Waste volume would increase under Alternative 5 because of the addition of cement and fly ash to stabilize the waste. Treatment methods (under Alternatives 4 and 5) would reduce contaminant toxicity; under Alternative 4, metals would be recovered and recycled.

The short-term effectiveness of Alternatives 2, 3, 4, and 5 is comparable, except that the overall risk to the general public would be higher for Alternative 5 because it would involve off-site transport of contaminated soil. Environmental impacts at the TBP area from excavation and construction activities would be common for all action alternatives, and comparable impacts would be expected. Mitigative measures would be used to minimize potential short-term impacts.

Emplacement of the "risk-reduction cover" would be fairly straightforward. Construction of the storage facility under Alternative 3 also would be fairly straightforward. The implementation of Alternative 4 would be fairly straightforward, although additional studies would be required to refine the soil washing/leaching treatment system design and cost estimates. The chemical stabilization/solidification treatment that would be performed off-site under Alternative 5 is fairly well established and would not require further development before implementation.

Alternative 2 has the lowest overall cost (\$1.8 million) of the action alternatives; however, this cost is not directly comparable to Alternatives 3, 4, and 5. Alternative 5 is considered more cost-effective than Alternative 4 for site cleanup. The estimated total cost of Alternative 5 is \$10.5 million (for excavation to 2 ft), and it would provide a similar level of overall effectiveness as Alternative 4, with an estimated cost of \$13.1 million.

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# 10 LIST OF PREPARERS

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Terri Patton	M.S., geology; 13 years experience in environmental research and assessment.	FFS task leader; development of remedial action objectives; description of alternatives; engineering evaluation; comparative summary.
Paul Benioff	Ph.D., nuclear chemistry; 17 years experience in theoretical chemistry; 16 years experience in environmental assessment.	Technology screening; assessment of soil impacts.
Carole Biang	B.S., chemical engineering; 16 years experience in management and remediation of hazardous waste sites, environmental assessment.	Description of alternatives; engineering evaluation; comparative summary.
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# APPENDIX A:

# VADOSE ZONE MODEL OF METAL TRANSPORT FOR THE TOXIC BURNING PITS AREA OF J-FIELD

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# VADOSE ZONE MODEL OF METAL TRANSPORT FOR THE TOXIC BURNING PITS AREA OF J-FIELD

# A.1 INTRODUCTION

A model of the vadose zone of the Toxic Burning Pits (TBP) area of J-Field was developed to evaluate transport of metal contamination from the surface to the water table (i.e., vadose zone) by using the TRACR3D code (Travis and Birdsell 1991). The vadose zone model assumed a depth to water table of 2 m. Simulations were performed for a 50-year period. The model calculates saturation and concentration values at 2-cm vertical intervals through the vadose zone.

The objective of this modeling study was to evaluate the various processes that may affect the transport of metals from the near-surface to the water table and estimate approximate travel times for selected contaminants of concern (e.g., arsenic, cadmium, lead, zinc).

Surface contamination can reach groundwater via pathways other than those modeled by vadose zone transport models. In the vadose zone, numerous preferential pathways exist for contaminant and fluid transport, especially in the near-surface; these include such common features as desiccation cracks, holes that result from the decay of a former plant root, and animal burrows. These features can provide a means for surface contamination to reach the groundwater much faster than predicted by using vadose zone matrix flow and the transport mechanisms. In general, these types of features are difficult to model because it is difficult to characterize their location and properties in the field.

#### A.2 CONCEPTUAL MODEL

A conceptual model of the vadose zone was developed on the basis of the results of field investigations performed at the TBP area (Yuen et al. 1996). Metal transport within the vadose zone to the water table was assumed to occur because of near-surface metal contamination in the top 10 cm of the soil. The metal was assumed to undergo linear, instantaneous, equilibrium sorption-desorption between the aqueous and solid phases as it was transported through the vadose zone. The area contaminated with metals was assumed to be large, so that transport within the vadose zone could be assumed to be vertically downward with no lateral dispersion (i.e., one-dimensional). This assumption is conservative, because lateral dispersion and diffusion would tend to lower the concentration and delay the arrival time of a contaminant at the water table given a finite duration source, such as that assumed for this model.

The vadose zone comprised a homogeneous silty loam material 2 m thick (Perkins and Winant 1927), with a hydraulic conductivity of 2.0 Darcies<sup>1</sup> (Quinn 1995), a porosity of 0.45 (Guymon 1994), and an irreducible water saturation of 0.067 (Guymon 1994).

This model assumed a monthly rainfall of 10.0 cm (Smith and Matthews 1975), which occurred in a single event at the start of each month. This assumption is conservative because a single wetting event will transport material faster and further than constant infiltration, given the same total volume over the entire month.

# A.3 MODEL DESIGN AND INPUT PARAMETERS

A finite-difference grid was designed to model a depth to groundwater of 2 m, which is representative of the TBP area. The grid developed for this model contains 10 divisions in the horizontal direction and 100 in the vertical. The cell size is 2.0 cm square. The results from these simulations can be used to evaluate the transport of metals in areas where the depth to groundwater is less than 2 m by using the results for the 1-m depth. These estimates will not be conservative because of faster transport times in more saturated soils. As the depth to water table decreases, the saturation of the model increases due to capillary effects. However, these estimates should be relatively accurate. When depth to the water table becomes shallower (less than 1 m), these assumptions become worse and the model simulations reported here cannot be used to estimate contaminant transport. At shallow depths, a conservative estimate would be to assume that concentrations in the vadose zone water at the groundwater table are equal to the initial vadose zone concentrations. Dilution would occur as the vadose zone fluids mix with the saturated zone fluids, and concentrations would be lowered further as contaminants are transported within the saturated zone because of dilution, sorption onto the solid phases, and other geochemical and physical processes.

The model grid is two-dimensional; however, initial and boundary conditions are such that the flow regime modeled is one-dimensional. The left and right sides of the model are no-flow boundaries, creating a one-dimensional flow regime. No lateral dispersion is modeled. The bottom boundary represents the water table and is modeled by using a constant-pressure/constant-saturation condition for fluid flux and a zero-gradient condition for contaminant transport. This boundary condition allows continuous contaminant outflow when a contaminant plume reaches it. The surface boundary (top of the model) models the total monthly precipitation event as a single ponded water condition. This boundary condition applies a fluid saturation and pressure equal to a given depth of water. The ponded water depth automatically decreases as fluid leaves the surface by infiltration and enters the subsurface. When all of the water has infiltrated the subsurface, air pressure at the upper

<sup>&</sup>lt;sup>1</sup> 2.0 Darcies is equivalent to a saturated hydraulic conductivity of  $1.9 \times 10^{-3}$  cm/s.

boundary is set to 0 gage pressure, and air saturation is set to 1.0 (completely air-saturated). Surface infiltration occurred once per month for the entire 50-year run.

The material distribution in the vadose zone is assumed to be homogeneous, with a permeability of 2.0 Darcies in the horizontal and vertical directions. This aspect of the model setup adds a degree of conservatism, because vertical saturated permeability is generally less than horizontal saturated permeability (typically 1/10) (Domenico and Schwartz 1990).

Determining the saturation-permeability relationship for soils is a difficult task that requires special laboratories and considerable time and effort. In addition, the results of these tests can have large uncertainty (Guymon 1994). For this study, the relationship between saturation and permeability developed by van Genuchten (1980) was used. Values for porosity, irreducible soil moisture, and the van Genuchten parameters from Guymon (1994) were used because no site-specific data were available. The porosity is 0.45, and the irreducible water content is 0.067. The van Genuchten parameters are 0.2908 and 50 cm for  $(1-1/\beta)$  and  $1/\alpha$ , respectively. These parameters match those of a silty loam (Guymon 1994).

Initial water saturations approximate equilibrium values, with the soils near the water table more water-saturated than soils near the surface. The values used for initial water saturations approximate a drained condition for the soil.

#### A.4 TRANSPORT STUDIES

For all computations, the fluid in the top 10 cm of the vadose zone was assumed to be a tracer at a concentration ratio,  $\mathrm{C/C_0}$  equal to 1, in equilibrium with sorbed tracer on the solid matrix. In one-dimensional studies, the  $\mathrm{C/C_0}$  parameter is a way to nondimensionalize concentration measurements, providing general information that is invariant with respect to the initial concentration. Model-computed values of  $\mathrm{C/C_0}$  can be multiplied by a chosen initial concentration to arrive at model-predicted concentrations. As a conservative estimate, the solubility limit of a compound is typically used as an initial concentration. Multiplying the computed  $\mathrm{C/C_0}$  value by a compound's solubility gives a conservative estimate of subsurface contamination values.

A set of simulations was completed by using a range of  $K_d$  values: 0, 5, 15, and 150 mL/g (Table A.1). This range covers the estimated  $K_d$  values for the metal contaminants present at the TBP area. In the following sections, these simulations are discussed. The metal that could be modeled by a given  $K_d$  is identified in each section along with reported  $K_d$  ranges from the literature. No site-specific values exist for  $K_d$  at the TBP area.

 $<sup>^{2}</sup>$  C is the current concentration of a tracer, and  $C_{0}$  is the initial concentration.

TABLE A.1 K<sub>d</sub> Values and Results

Description	K <sub>d</sub> (mL/g)	Depth after 50 years (cm)	Notes
Conservative tracer	0.0	-	Tracer was flushed from the vadose zone after approximately 4 months
Lead	150.0	~ 45	Very immobile
Cadmium	15.0	~ 85	Relatively immobile, after 50 years the contamination has moved less than 1 m
Zinc	5.0	-	Tracer was flushed from the vadose zone after approximately 30 years

#### **A.4.1** Nonreactive Tracers

An initial set of simulations was run as a base-case assuming a conservative tracer (i.e., nondegrading, nonreactive, and nonsorbing  $[K_d=0]$ ). Nonreactive tracer transport was rapid. The tracer reached the groundwater table in approximately two months. After four months, all the tracer had been flushed from the vadose zone. The model was configured to produce output at the end of each 30-day period. At the end of the third period, tracer concentrations were still measurable in the vadose zone. By the end of the fourth period, no tracers were present in the model — they had all been transported across the lower boundary.

#### A.4.2 Lead

 $K_d$  values for lead range from  $1.8 \times 10^2$  to  $6.3 \times 10^4$  mL/g (Sheppard et al. 1984), with the higher values generally occurring in soils rich in organic matter. In sand, the highest reported value is  $3.5 \times 10^3$  mL/g. For this study, a conservative value<sup>3</sup> of 150 mL/g was used to model lead transport in the vadose zone. Under these conditions, the lead was not mobile.

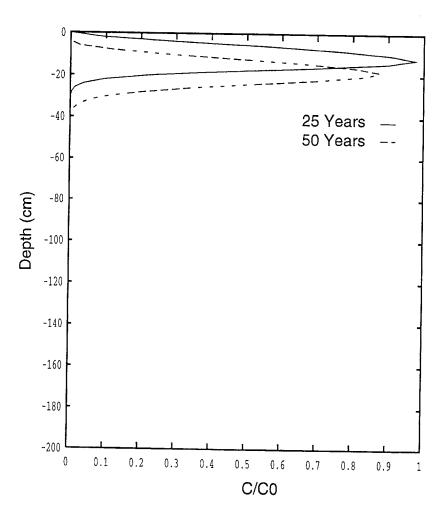
Pure lead is insoluble (Chemical Rubber Company [CRC] 1968); however, various lead compounds are soluble to varying degrees. The solubility of lead-oxide (PbO) is 0.0017~g/100~cc, while red lead-oxide (Pb<sub>3</sub>O<sub>4</sub>) is insoluble (CRC 1968). Lead is mostly present in the lead-oxide form (Peters 1995).

 $<sup>^{3}</sup>$  Lower values of  $K_{d}$  allow the contaminant to be more mobile in the subsurface.

Figure A.1 shows the results of a calculation spanning 50 years. The initial depth of contamination was 10 cm with a concentration ratio,  $C/C_0$ , of 1. After 50 years, the center of mass of the lead plume is at a depth of approximately 20 cm, while the leading edge is at a depth of less than 40 cm.

#### A.4.3 Cadmium

Reported  $K_d$  values for cadmium range from 47.6 to 500 mL/g in sands, 9.8 to 76 mL/g in silty soils, and 23 to  $1.7 \times 10^4$  mL/g in soils rich in organic matter (Sheppard et al. 1984; Thibault et al. 1990). For this study, a value of 15 mL/g was used to model cadmium transport, which is near the low end of the range for silty soils. Cadmium is slightly soluble, and the compound cadmium-sulfide has a solubility of 0.00013 g/100 cc (CRC 1968).

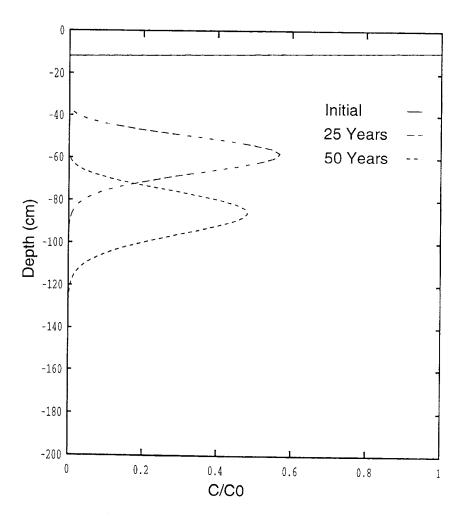


**FIGURE A.1** Concentration Profiles for Lead Using a  $K_d$  Value of 150 mL/g for a 50-Year Period

Figure A.2 shows the results of a calculation spanning 50 years. The initial depth of contamination was 10 cm with a concentration ratio,  $C/C_0$ , of 1. After 50 years, the center of mass of the cadmium plume is at a depth of approximately 85 cm, while the leading edge is at a depth of less than 130 cm.

#### A.4.4 Zinc

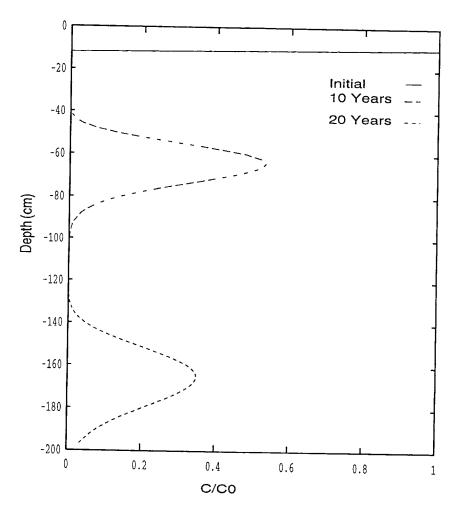
Reported  $K_d$  values for zinc range from 0.1 to 2,120 mL/g in sands, 3.6 to 100 mL/g in silty soils, and 70 to  $1.3 \times 10^4$  mL/g in soils rich in organic matter (Sheppard et al. 1984; Thibault et al. 1990). For this study, a value of 5 mL/g was used, which is near the low end of the range for silty soils. The results of the calculations for lead and cadmium, which used  $K_d$  values of 150 and 15 mL/g, respectively, can be used to estimate the effects of a higher  $K_d$  on zinc transport.



**FIGURE A.2** Concentration Profiles for Cadmium Using a  $K_d$  Value of 15 mL/g (Shown are  $C/C_0$  values versus depth for 0, 25, and 50 years)

Figure A.3 shows the results of a calculation spanning 20 years. The initial depth of contamination was 10 cm with a concentration ratio,  $C/C_0$ , of 1. After 30 years, the zinc plume has been flushed from the vadose zone; no zinc occurs in solution in any of the vadose zone water. During this 30-year period, the maximum  $C/C_0$  ratio at the water table was approximately 0.38.

Assuming the zinc is present as zinc-oxide, with a maximum solubility of 0.00016 g/100 cc (CRC 1968), the maximum concentration at the water table would be below  $6.0 \times 10^{-7}$  g/mL. This is calculated by assuming that the initial concentration of zinc in the vadose zone fluid was equal to the maximum solubility limit of zinc-oxide and multiplying this value by the computed ratio of the current concentration to the initial concentration, C/C<sub>0</sub>, which in this case was 0.38.



**FIGURE A.3** Concentration Profiles for Zinc Using a  $K_d$  Value of 5.0 mL/g (Shown are  $C/C_0$  versus depth profiles for 0, 10, and 20 years)

#### A.5 OTHER CONTAMINANTS

Table A.2 lists some inorganic materials of concern at J-Field. Estimates of how these materials would behave during vadose zone transport can be made by determining which of the modeled  $K_d$  values most closely approximates that of the material of interest. The reported values of  $K_d$  span a large range because of the sensitivity of  $K_d$  values to soil types and other factors. Generally, the lowest reported values are for clean sands (no organics, clays, or silts) or clean gravels. The highest values reported are generally for clay and organic-rich soils.

#### A.6 CONCLUSIONS

The model assumptions made in this study were conservative (i.e., tended to encourage transport at higher rates); however, even using these assumptions, lead contamination of the groundwater due to vadose zone transport from surface soil contamination was unlikely to be above levels of concern. Because of the high affinity of lead to sorb onto solids (especially clays and organic particles), lead is only slightly mobile in the subsurface. If lead is transported through the vadose zone to the water table, it would take several hundreds of years and be at low concentrations because of dispersion and dilution. Because of the estimated time required for lead to reach the groundwater, simulations were not performed to predict concentrations at the groundwater table.

TABLE A.2 Inorganic Materials of Concern K<sub>d</sub> Values<sup>a</sup>

Material	K <sub>d</sub> Range (mL/g)	Material	K <sub>d</sub> Range (mL/g)
Antimony	180 - 4,000	Copper	76 - 266
Arsenic	1.0 - 18	Lead	180 - 63,000
Barium	0.4 - 6.0	Selenium	36 - 310
Cadmium	9.0 - 17,000	Silver	75 - 1,000
Chromium	50 - 1,000	Zinc	0.1 - 13,000
Cobalt	10 - 2,500		

 <sup>&</sup>lt;sup>a</sup> K<sub>d</sub> values cover a large range. Generally, the lowest values reported here are for either a laboratory clean sand or gravel, while the highest values are for a clay or organic-rich soil. Typical field values are generally toward the higher portion of the range.

Source: Thibault et al. (1990).

These low concentrations would be further reduced by dilution that would occur as the vadose zone fluids mixed with the fluids below the water table. This dilution would be significant during any lateral transport, such as moving from the TBP area to the nearby marsh. Given the results of these calculations and the low concentrations of lead in groundwater samples from the TBP area,  $^4$  the transport of lead through the vadose zone to the groundwater table presents a low risk of contamination at levels of concern (50  $\mu$ g/L).

Cadmium, or other metals with a  $K_d$  of 15 mL/g, is only slightly mobile, with the center of the contamination plume moving less than 1.5 m during the 50-year period modeled.

Zinc, or metals with small  $K_d$  values (5 mL/g), reached the groundwater in approximately 20 years. Under the conditions modeled, the maximum concentration was estimated to be  $6.0 \times 10^{-7}$  g/mL.

This model simulates only transport of metals via matrix flow in the vadose zone. If preferential pathways for flow exist, highly retarded metals, such as lead, could reach the groundwater in a much shorter time. This will, however, be a local event, and the groundwater should dilute the contaminant to below levels of concern. This is the most likely pathway for the lead contamination measured in the groundwater at the TBP area.

This study does not look at the effects of saturated zone transport of metal contaminants. Metals undergoing transport in the saturated zone would undergo a number of processes that would lower the initial concentrations at the source, if the source existed for a finite time (was not continuous). Thus, concentrations in the groundwater at points of concern may be lower than those estimated at the groundwater table by this model.

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The maximum concentration found was 92  $\mu$ g/L (in only one sample); most samples have lead values below the MCL of 50  $\mu$ g/L (5 × 10<sup>-3</sup> mg/L).

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# APPENDIX B:

APPLICABLE OR RELEVANT AND APPROPRIATE REQUIREMENTS

TABLE B.1 State Regulations and To-Be-Considered Requirements for J-Field

Remarks	Applies to Chesapeake Bay only. The bay borders the site to the south and may receive discharge from groundwater aquifers. Potential impacts will depend on the remedial actions.	Any excavation of soils may require compliance with this law.	Any soil movement may require compliance with these requirements, including mitigation.	Because the definition of "person" does not include the federal government, these regulations would not be applicable; however, any clearing, filling, excavation, or dredging within wetland areas may require consultation with the Department.
Preliminary Determination	Potentially applicable	Potentially applicable	Potentially applicable	Potentially applicable
Requirements	Protection of the bay, and criteria for any development in the bay area. Hazardous waste collection or disposal facilities are not permitted in the Critical Area unless no environmentally acceptable alternative exists.	Construction, including soil movement, grading, transporting, or otherwise disturbing land, requires a grading and building permit, issued after review and approval of the Department of the Environment.	A federal agency may not clear land, move soil, or engage in construction without submitting a stormwater management plan for approval to the Department of the Environment.	Any person proposing to conduct any activity not authorized in the wetland regulations must apply for a permit from the Department of Natural Resources.
Medium	Water, sediment, and soil	Soil and sediment	Stormwater	Wetlands
Potential ARAR	Chesapeake Bay Critical Area Protection Program, Annotated Code of Maryland, NR § 8-1801 et seq.; COMAR 27	Maryland Nonpoint Source Pollution Control Laws, Annotated Code of Maryland EN § 4-101 et seq.	Maryland Nonpoint Source Pollution Control Laws, Stormwater Management, Annotated Code of Maryland, EN § 4-201 et seq.; COMAR 26.09.05	Maryland Wetland Law, Annotated Code of Maryland, EN § 9-301 et seq.; COMAR 08.07

TABLE B.1 (Cont.)

Potential ARAR	Medium	Requirements	Preliminary Determination	Remarks
Nontidal Wetlands Protection Act, Annotated Code of Maryland, NR § 8-1201 et seq.; COMAR 08.05.04	Nontidal wetlands	No person may conduct a regulated activity, including removal, excavation, or dredging of soil; grading; or destroying or removing plant life, without first obtaining a permit from the Department of Natural Resources. No activity may (1) degrade the aquatic ecosystem, diversity, productivity and stability, wildlife, recreational and economic value, and public welfare; (2) violate any applicable State water quality standard or the Clean Water Act; or (3) degrade surface or groundwater quality. An activity must be consistent with any applicable comprehensive watershed management plan.	Potentially applicable	Any excavation, dredging, or dumping of soil in a nontidal wetland area must comply with these regulations.
COMAR 26.08.04	Water	No person may construct a wastewater treatment plant that discharges to the waters of the State without first obtaining an NPDES permit from the Department of the Environment.	Potentially applicable	If construction of a new wastewater treatment plant at J-Field is part of the proposed action, permit discharge limits would be applicable. If wastewater from a J-Field soil washing/leaching facility is to be treated at the O-Field treatment plant or discharged through the POTW, a permit modification may be necessary.

TABLE B.1 (Cont.)

Remarks	These requirements do not apply directly to sources of emissions; rather, they are limitations on ambient concentrations. The site is an unconfined source in what is termed an Area III (per 26.11.01).  Particulate matter must be controlled during demolition or other earth-moving activities.	Potentially applicable to any emissions resulting from remedial actions, such as in-situ treatment and pump-and-treat actions. In addition, VX, GB, and mustard gas are Class I toxic air pollutants (COMAR 26.11.15.11). Potential emissions may occur in the marshes where chemical agents and UXO may exist and during emergency use of the TBP and WPP.
Preliminary Determination	Potentially relevant and appropriate	Potentially applicable
Requirements	State-adopted national ambient air quality standards and guidelines must be met for the following: PM10 – 50 µg/m³ annual mean, 150 µg/m³ 24-hour average; sulfur dioxide – 80 µg/m³ annual arithmetic mean, 365 µg/m³ maximum 24-hour concentration; nitrogen dioxide – 100 µg/m³ annual arithmetic mean; lead – 1.5 µg/m³ maximum mean over calendar quarter; carbon monoxide – 10 µg/m³ maximum 8-hour concentration, 40 µg/m³ maximum 1-hour concentration, reasonable precautions must be taken to prevent matter from mecoming airborne; i.e., use of water or chemicals for dust control and covering open-bodied vehicles transporting soils.	Any installation or source that discharges Class I or II toxic air pollutants into the ambient air and must obtain a permit under this subtitle must meet the requirements of these regulations. Even if the installation need not obtain a permit, the Department of the Environment may require any source that is not subject to permit or approval to meet these requirements if the Department determines that the installation has the potential to discharge toxic air pollutants in quantities that may unreasonably endanger the public health (COMAR 26.11.15.03(A)(4)).
Medium	Air	Air
Potential ARAR	Maryland Air Quality Control Act, Amotated Code of Maryland, EN § 2-101 et seq.; COMAR 26.11.01, 26.11.03, and 26.11.06	COMAR 26.11.15

TABLE B.1 (Cont.)

Potential ARAR	Medium	Requirements	Preliminary Determination	Remarks
Maryland Air Quality Control Act, Annotated Code of Maryland, EN § 2-101 et seq.; COMAR 26.11.01, 26.11.03	Air	No person may construct a new source protection source, including a sewage treatment plant, without a permit to construct from the Department of Environment.	Potentially applicable	If construction of a new wastewater treatment plant at J-Field is part of the proposed action, permit discharge limits would be applicable.
COMAR 26.11.04(A)(44)	Air	A person may not operate any petroleum-contaminated soil treatment facility without first obtaining a permit to operate from the Department of the Environment.	Potentially applicable	If treatment of petroleum-contaminated soil will be part of the proposed action at J-Field, permit requirements would be applicable.
COMAR 26.11.06.06	Air	A person may not treat or dispose of waste containing VOCs in a manner that results in evaporation of greater than 20 lb/day into the atmosphere.	Potentially applicable	If treatment of contaminated soil will be part of the proposed action at J-Field, VOC emissions must meet these requirements.
Maryland Landfill Siting Law, Annotated Code of Maryland, EN § 9-209 et seq.; COMAR 26.04.07	Solid waste	General regulations for disposing of solid waste in a land-based unit, including procedures for closure and postclosure. Also, sites that act as a transfer station or processing facility for on-site disposal must follow these regulations.	Potentially applicable	Should the remedial action involve landfilling, transferring, and/or processing the waste (as defined by COMAR 26.04.07.02), these regulations must be followed.

TABLE B.1 (Cont.)

Remarks	These requirements are applicable to any removal actions if the waste at the site meets the prerequisites for definition as characteristic or listed hazardous waste.  These requirements would apply to any interim waste storage or treatment facilities or to groundwater monitoring for such storage or treatment facilities, such as a contaminated soil waste pile.
Preliminary Determination	Potentially applicable
Requirements	These regulations identify and list hazardous waste and define hazardous waste.  Maryland-listed acute hazardous wastes include the chemicals GA (tabun), GB (sarin), GD (soman), VX, L (lewisite), DM (adamsite), HD (sulfur mustard), and T, waste military chemical warfare agents (Hazardous waste codes K991 through K999) having any of these substances as their active or principal ingredient, or mixtures of any of these substances and any characteristic or listed hazardous waste or residues from the treatment of wastes K991 through K999. Maryland-listed hazardous waste from specific sources includes residues from the treatment of wastes residues from the treatment of the above-listed wastes, except when treated at the following:
Medium	Hazardous waste
Potential ARAR	COMAR 26.13.02

- U.S. Army Aberdeen Proving Ground/U.S. Army Chemical Research Development and Engineering Center;
- U.S. Army Product Assurance Directorate, Aberdeen Proving Ground; or
- U.S. Army Aberdeen Proving Ground/Medical Research Institute of Chemical Defense

TABLE B.1 (Cont.)

Potential ARAR	Medium	Requirements	Preliminary Determination	Remarks
		under the following protocol: "To Support the Delisting of Decontaminated Liquid Chemical Surety Materials as Hazardous Waste from Specific Sources COMAR 26.13.02.167 and 26.13.02.18 K991-K999 and Residues from K991-K999" (U.S. Army Chemical Research Development and Engineering Center, Aberdeen Proving Ground, Md., 1988).		
COMAR 26.13.05	Waste treatment, storage, and disposal	General requirements are established for storage, treatment, and disposal facility location, design, and inspection; waste compatibility determination; emergency contingency plans; preparedness plans; recordkeeping; reporting; and worker training.	Potentially applicable	These requirements are potentially applicable because storage, treatment, or disposal, as defined under RCRA, may be included in the scope of a remedial action.
COMAR 26.13.05	Waste treatment, storage, and disposal	Location requirements include that (1) facilities must not be located within 61 m (200 ft) of an active fault and (2) facilities located in a 100-year floodplain must be constructed, operated, and maintained to prevent washout of any hazardous waste by a 100-year flood.	Potentially applicable	These requirements would be applicable to the siting of any waste storage or treatment facility constructed for the remedial action.
COMAR 26.13.05.06	Waste disposal	A groundwater monitoring system must be maintained for a SWMU. The concentration of a hazardous constituent in the uppermost aquifer beneath a regulated unit is not to exceed the existing background concentration or the following maximum concentration if higher than the background level, or an	Potentially applicable	These requirements are potentially applicable if the scope of a remedial action includes maintaining a surface impoundment, waste pile, land treatment, or landfill SWMU.

TABLE B.1 (Cont.)

Potential ARAR	Medium	Requirements	Preliminary Determination	Remarks
		alternative concentration limit, unless an exemption is granted:		
		$\label{eq:maximum} {\sf Maximum Concentration} \\ {\sf for Groundwater Protection} \\ {\sf Constituent} \\ ({\it \mu g/L})$		
		Arsenic 0.05 Lead 0.05		
COMAR 26.13.05.09	Waste storage	Containers used to store hazardous waste must be closed and in good condition.	Potentially applicable	These requirements may be applicable to the remedial action, container storage, and transportation of site waste that meets the prerequisites for definition as characteristic hazardous waste.
COMAR 26.13.05.12	Waste storage	A waste pile that is inside or under a structure that provides protection from precipitation so that neither runoff nor leachate is generated need not meet the design and operating requirements of this Subpart if certain controls are established.	Potentially applicable	If contaminated soils are to be excavated and stored in a waste pile, these regulations would apply.
COMAR 26.13.05.10	Waste storage	Wastes stored in tanks must meet specific integrity and design requirements such that leaks can be detected and spills responded to.	Potentially applicable	If contaminated wastewaters meet the definition of hazardous waste and are to be stored in tanks before transfer to the O-Field treatment plant, these regulations would apply.
COMAR 26.13.03	Waste management	Standards for accumulating labeling, marking, and shipping hazardous waste. Also, requirements for record keeping and reporting hazardous waste activities.	Potentially applicable	If hazardous wastes are generated due to remedial actions, these wastes must be managed pursuant to these requirements.

TABLE B.1 (Cont.)

Potential ARAR	Medium	Requirements	Prcliminary Determination	Remarks
COMAR 26.13.05.07	Waste management	All treatment, storage, and disposal facilities must have a closure plan; all waste must be removed and equipment decontaminated. For certain facilities, postclosure requirements for monitoring and notification are necessary.	Potentially applicable	If storage or treatment facilities are a part of the remedial action, closure requirements must be met.
Maryland Hazardous Waste Facilities Siting Law, Annotated Code of Maryland, NR § 3-701 et seq.	Waste management	The Hazardous Waste Facilities Siting Board must issue a certificate of public necessity for siting HW facilities.	Potentially applicable	If an HW storage or treatment facility is to be constructed at J-Field, the requirements for these would be applicable.

TABLE B.2 Federal Regulations and To-Be-Considered Requirements for J-Field — Potential Contaminant-Specific Requirements

Remarks	These requirements do not apply directly to source-specific emissions; rather, they are national limitations on ambient concentrations. However, they will be addressed in controlling particulate emissions and ozone (VOC) emissions that could result from implementing a remedial action.	These requirements do not apply directly to source-specific emissions; rather, they are national limitations on ambient concentrations. However, they will be addressed in controlling lead emissions that could result from implementing a remedial action.
Preliminary Determination	Potentially relevant and appropriate	Potentially relevant and appropriate
Requirements	Federal standards for ambient air quality include the following: sulfur dioxide – 80 µg/m³ annual arithmetic mean, 365 µg/m³ maximum 24-hour concentration; particulate matter – 150 µg/m³ 24-hour average concentration, 50 µg/m³ annual arithmetic mean; carbon monoxide – 10 mg/m³ 8-hour average concentration; ozone – 235 µg/m³ maximum hourly average concentration; ozone – 235 µg/m³ maximum hourly average concentration; introgen dioxide – 100 µg/m³ annual arithmetic mean; lead – 1.5 µg/m³ annual arithmetic mean averaged over calendar quarter.	As for the above conditions, the standard for lead and its compounds, as elemental lead, is 1.5 µg/m³ maximum arithmetic mean averaged over one calendar quarter.
Medium	Air	Air
Contaminant	Particulate matter	Lead
Potential ARARs	Clean Air Act, as amended (42 USC 7401-7642); National Primary and Secondary Ambient Air Quality Standards (40 CFR Part 50)	

TABLE B.2 (Cont.)

Remarks	Any such spills at the site would have preceded the effective date of this requirement; hence, it is not applicable. Because the EPA policy on PCBs is not an enforceable requirement, it is not considered relevant and appropriate. However, these regulations are the guidelines used by the EPA for setting preliminary remediation goals for PCBs in soil at a remediation site.
Preliminary Determination	To be considered
Requirements	For spills of materials contaminated with >50 ppm PCBs in unrestricted access areas (e.g., residential areas), soil within the spill area must be excavated and backfilled with soil containing <1 ppm PCBs. Contaminated soil may be decontaminated to 10 ppm by weight by excavating a minimum of 10 in, and backfilling with soil containing <1 ppm PCBs. For spills at outdoor electrical substations, the soil must be cleaned to 25 ppm by weight (as for other restricted access areas) or to 50 ppm by weight with posting of a visible notice.
Medium	Soil
Contaminant	PCBs
Potential ARARs	TSCA, as amended (15 USC 2607-2629, PL 94-469 et seq.); Polychlorinated Biphenyls Manufacturing, Processing, Distribution in Commerce, and Use Prohibitions [40 CFR Part 761 (Subpart G)]

TABLE B.3 Federal Regulations and To-Be-Considered Requirements for J-Field — Potential Location-Specific Requirements

Potential ARARs	Location	Requirements	Preliminary Determination	Remarks
Antiquity Act, Historic Sites Act [16 USC 431-433, 16 USC 461-467, 40 CFR 6.301(a)]	Land	Cultural resources, such as historic buildings and sites and natural landmarks, must be preserved on federal land to avoid adverse impacts.	Potentially applicable	Should this site be identified as a cultural resource, the requirement could be potentially applicable during a remedial action.
National Historic Preservation Act, as amended [16 USC 470 et seq., 40 CFR 6.301(b), 36 CFR Part 800]	Land	The effect of any federally assisted undertaking must be taken into account for any district, site, building, structure, or object included in or eligible for the <i>National Register of Historic Places</i> .	Potentially applicable	Should such a property be identified, the requirement would be potentially applicable during a remedial action.
Archeological and Historic Preservation Act [16 USC 469, 40 CFR 6.301(c), PL 93-291, 88 Stat. 174]	Land	Prehistoric, historic, and archeological data that might be destroyed as a result of a federal, federally assisted, or federally licensed activity or program must be preserved.	Potentially applicable	Should such data be identified, the requirement would be potentially applicable during a remedial action.
Archeological Resources Protection Act [16 USC 470(a)]	Land	A permit must be obtained if an action on public or Native American lands could affect archeological resources.	Potentially applicable	Should such resources be identified, the requirement would be potentially applicable during a remedial action.
Protection and Enhancement of the Cultural Environment (EO 11593, 40 CFR 6.301)	Land	Historic, architectural, archeological, and cultural resources must be preserved, restored, and maintained and must be evaluated for inclusion in the <i>National Register</i> .	Potentially applicable	Should such resources be identified, the requirement would be potentially applicable during a remedial action.
Endangered Species Act, as amended [16 USC 1531-1543, 50 CFR 17.402, 40 CFR 6.302(h)]	Any	Federal agencies must ensure that any action authorized, funded, or carried out by the agency is not likely to jeopardize the continued existence of any threatened or endangered species or destroy or adversely modify any critical habitat.	Potentially applicable	Should such species or habitat exist, the requirement would be potentially applicable during a remedial action.

TABLE B.3 (Cont.)

Remarks	Should modification of any stream or any other water bodies or areas affecting water bodies be required during a remedial action, the requirement would be potentially applicable.	If any remedial activities would disturb the habitat of the bald eagle, consultation with the U.S. Department of the Interior would be necessary.	The site is within the 100-year floodplain.	Wetlands exist in the affected area.
Preliminary Determination	Potentially applicable	Potentially applicable	Potentially applicable	Potentially applicable
Requirements	Adequate protection of fish and wildlife resources is required when any federal department or agency proposes or authorizes any modification (e.g., diversion or channeling) of any stream or other water body or any modification of areas affecting any stream or other water body.	It is unlawful to take, pursuc, molest, or disturb bald eagles, their nests, or eggs.	Federal agencies must avoid, to the maximum extent possible, any adverse impacts associated with direct and indirect development of a floodplain.	Federal agencies must avoid, to the extent possible, any adverse impacts associated with the destruction or loss of wetlands and the support of new construction in wetlands if a practicable alternative exists.
Location	Any	Any	Floodplain	Wetlands
Potential ARARs	Fish and Wildlife Coordination Act [14 USC 441-444, 40 CFR 4.302(a)]	Bald and Golden Eagle Protection Act (16 USC 668-668d)	Floodplain Management (EO 11988, 40 CFR Part 6)	Protection of Wetlands (EO 11990, 40 CFR Part 6)

requirements are not subject to evaluation for attainment or waiver as part of the ARAR process. However, they constitute requirements for worker protection with which any remedial action will comply.

TABLE B.4 Federal Regulations and To-Be-Considered Requirements for J-Field — Potential Action-Specific Requirements

Potential ARARs	Action	Requirements	Preliminary Determination	Remarks
Hazardous Material Transportation Act, as amended (49 USC 1801-1812); DOT HAZMAT Transportation Regulations (49 CFR Parts 172-179)	Transportation	Generic requirements are established for minimizing the environmental impacts of spills or releases of hazardous materials, as are procedures for transporting hazardous waste.	Potentially applicable	These requirements are not part of an environmental law and hence are not subject to evaluation for attainment or waiver as part of the ARAR process. However, they could be pertinent to any remedial action in which hazardous waste is transported off-site. In this case, the pertinent requirements (e.g., for spill response) would be addressed during implementation.
Noise Control Act, as amended; Noise Pollution and Abatement Act (42 USC 4901 et seq.)	Detonation, heavy equipment usage	The public must be protected from noises (e.g., that could result from remedial action activities) that jeopardize health or welfare.	Potentially applicable	This requirement is potentially applicable if, during a remedial action, any heavy equipment is used.
OSHA Standards, Occupational Health and Environmental Control [29 CFR 1910.95(g)]	Noise, air	The permissible occupational exposure level for noise is 90 dBA (slow response) for an 8-hour day; with decreasing times of exposure, the levels increase to 115 dBA per 15-minute day.	Potentially applicable	These requirements are part of an employee protection law (rather than an environmental law) with which all CERCLA response actions should comply. Because this compliance is directly required, as promulgated in the NCP, these

TABLE B.4 (Cont.)

Potential ARARs OSHA Standards for Hazardous	Action	Requirements General worker protection requirements are	Preliminary Determination Potentially	Remarks These requirements are parts of an
Waste Operations and Emergency Response (29 CFR Part 1910)	management	established, as are requirements for worker training and the development of an emergency response plan and a safety and health program for employees. In addition, procedures are established for hazardous waste operations—including decontamination and drum/container handling.	applicable	cmployce protection law (rather than an environmental law) with which all CERCLA response actions should comply. Because this compliance is directly required, as promulgated in the NCP, these requirements are not subject to evaluation for attainment or waiver as part of the ARAR process. However, they constitute requirements for worker protection with which a remedial action will comply.
TSCA, as amended (15 USC 2607-2629, PL 94-469 et seq.); Polychlorinated Biphenyls Manufacturing, Processing, Distribution in Commerce, and Use Prohibitions [40 CFR Part 761 (Subpart A)]	PCB testing	Inspection and testing are required for material contaminated with PCBs.	Potentially applicable	This requirement may be applicable to characterization of site waste for PCBs.
TSCA, as amended (15 USC 2607-2629, PL 94-499 et seq.) Polychlorinated Biphenyls Manufacturing, Processing, Distribution in Commerce, and Use Prohibitions [40 CFR Part 761 (Subpart D)]	PCB storage and disposal	When material contaminated with PCBs >50 ppm is stored, the facility used for storage must be marked as such and must not be located in a 100-year floodplain. Disposal of material contaminated with PCBs >50 ppm requires either incineration or chemical waste landfilling at EPA-approved facilities.	Potentially applicable	These requirements are potentially applicable because disposal of material from the site that may be contaminated with PCBs may be included in the scope of the remedial action.

TABLE B.4 (Cont.)

Potential ARARs	Action	Requirements	Preliminary Determination	Remarks
Solid Waste Disposal Act, as amended (42 USC 6901 et seq.); Waste Characterization [40 CFR Part 261 (Subparts C and D)]	Solid waste	A waste must be evaluated to determine whether it is a hazardous waste, i.e., either a listed waste or a characteristic waste. A characteristic waste is determined by its (1) ignitability, (2) corrosivity, (3) reactivity, or (4) toxicity, as defined by TCLP for several contaminants. See 40 CFR Part 261.	Potentially applicable	This requirement is applicable to the characterization and management of site waste.
Standards for Owners and Operators [40 CFR Part 264 (Subparts B through E)]	Waste treatment, storage, and disposal	General requirements are established for storage, treatment, and disposal facility location, design, and inspection; waste compatibility determination; emergency contingency plans; preparedness plans; record keeping; reporting; and worker training. Location requirements include (1) facilities must not be located within 61 m (200 ft) of an active fault and (2) facilities located in a 100-year floodplain must be constructed, operated, and maintained to prevent washout of any hazardous waste by a 100-year flood.	Potentially applicable	These requirements are potentially applicable because storage, treatment, or disposal, as defined under RCRA, may be included in the scope of a remedial action.
Releases from Solid Waste Management Units [40 CFR Part 264 (Subpart F)]	Waste disposal	A groundwater monitoring system must be maintained for a SWMU. The concentration of a hazardous constituent in the uppermost aquifer beneath a regulated unit is not to exceed the existing background concentration or a listed maximum concentration if higher than the background level, or an alternative concentration limit, unless an exemption is granted.	Potentially applicable	These requirements are potentially applicable if the scope of a remedial action includes maintaining a surface impoundment, waste pile, land treatment, or landfill SWMU.
Use and Management of Containers [40 CFR Part 264 (Subpart I)]	Waste storage	Containers used to store hazardous waste must be closed and in good condition.	Potentially applicable	These requirements may be applicable to the remedial action, container storage, and transportation of site waste that meets the prerequisites for definition as characteristic hazardous waste.

TABLE B.4 (Cont.)

Potential ARARs	Action	Requirements	Preliminary Determination	Remarks
Prohibitions on Land Disposal [40 CFR Part 268 (Subpart C)]	Waste disposal	The land disposal of certain hazardous waste (e.g., contaminated soil and debris) without proper treatment is restricted, unless a treatability variance is appropriate. Such treatment must attain levels achievable by the "best demonstrated available technologies" for each hazardous constituent in the listed waste.	Potentially applicable	These requirements are potentially applicable for any remedial action at J-Field that includes disposal or any consolidation of site waste into a new disposal or storage area.
Waste Piles [40 CFR Part 264 (Subpart L)]	Waste storage	A waste pile that is inside or under a structure that provides protection from precipitation so that neither runoff nor leachate is generated need not meet the design and operating requirements of this Subpart if certain controls are established.	Potentially applicable	If contaminated soils are to be excavated and stored in a waste pile, these regulations would apply.
Containment Buildings [40 CFR Part 264 (Subpart DD)]	Waste storage	Wastes stored within a completely enclosed, self-supporting structure that has a primary barrier to contain all wastes must conform to these regulations.	Potentially applicable	If contaminated soils are to be excavated and stored in a containment building, these regulations would apply.
Tanks [40 CFR Part 264 (Subpart J)]	Waste storage	Wastes stored in tanks must meet specific integrity and design requirements such that leaks can be detected and spills responded to.	Potentially applicable	If contaminated wastewaters meet the definition of hazardous waste and are to be stored in tanks before transfer to the O-Field treatment plant, these regulations would apply.
Corrective Action for Solid Waste Management Units [40 CFR Part 264 (Subpart S)]	Waste storage	Temporary tanks and container storage areas for treatment or storage of hazardous remediation wastes generated during RCRA corrective actions may meet alternative requirements contained in this Subpart.	Potentially applicable	These regulations would apply if tank or container storage is a part of the proposed action at J-Field.
Closure and Post-Closure [40 CFR Part 264 (Subpart G)]	Waste	All treatment, storage, and disposal facilities must have a closure plan; all waste must be removed and equipment decontaminated. For certain facilities, postclosure requirements for monitoring and notification are necessary.	Potentially applicable	If storage or treatment facilities are a part of the remedial action, closure requirements must be met.

## APPENDIX C:

SAMPLING DATA FOR THE TOXIC BURNING PITS AREA JANUARY 1993–JUNE 1995

#### APPENDIX C:

# SAMPLING DATA FOR THE TOXIC BURNING PITS AREA JANUARY 1993-JUNE 1995

This appendix presents the sampling data for surface water, sediment, soil, and groundwater collected at the J-Field Toxic Burning Pits (TBP) area between January 1993 and June 1995. Soilgas measurements are not presented here. A complete data set for all J-Field areas of concern, including the TBP area, is summarized and discussed in the *Remedial Investigation Report for J-Field, Aberdeen Proving Ground, Maryland, Volume 1: Remedial Investigation Results* (Yuen, R., et al., 1996, prepared by Argonne National Laboratory, Argonne, Ill., for U.S. Army, Aberdeen Proving Ground, Md., Directorate of Safety, Health, and Environment), which is currently in draft form.

Table C.1 provides an index to the data tables.

TABLE C.1 Index to Data Tables for the J-Field TBP Area

Table No.	Sample Medium	Analyses	Date	Page Number
C.2a	Surface water	Volatile organics	Apr 1993	C-6
C.2b	Surface water	Volatile organics	Sept 1993	C-8
C.2c	Surface water	Volatile organics	Feb 1994	C-10
C.2d	Surface water	Volatile organics	May 1994	C-13
C.3a	Surface water	Semivolatile organics	Apr 1993	C-14
C.3b	Surface water	Semivolatile organics	Sept 1993	C-14 C-17
C.3c	Surface water	Semivolatile organics	May 1994	C-17
C.4a	Surface water	Inorganics	Sept 1993	C-17
C.4b	Surface water	Inorganics	May 1994	C-21 C-23
C.5a	Surface water	Pesticides and PCBs	Apr 1993	C-23
C.5b	Surface water	Pesticides and PCBs	Sept 1993	C-24
C.5c	Surface water	Pesticides and PCBs	May 1994	C-28
C.6a	Surface water	General chemistry	May 1994	C-29
C.6b	Surface water	Explosive compounds	May 1994	C-30
C.6c	Surface water	CWA degradation products Organosulfur compounds	Apr 1993	C-31
C.7a	Sediment	Volatile organics	Apr 1994	C-32
C.7b	Sediment	Volatile organics	June 1995	C-38
C.8a	Sediment	Semivolatile organics	Apr 1994	C-40
C.8b	Sediment	Semivolatile organics	June 1995	C-53
C.9a	Sediment	Inorganics	Apr 1994	C-55
C.9b	Sediment	Inorganics	June 1995	C-60
C.10	Sediment	Pesticides and PCBs	June 1995	C-62
C.11	Sediment	Explosive compounds	June 1995	C-63
C.12	Sediment	CWA degradation products Organosulfur compounds	June 1995	C-64
C.13a	Soil	Volatile organics	Jan 1993	C-65
C.13b	Soil	Volatile organics	Jan 1994	C-71
C.13c	Soil	Volatile organics	May 1994	C-77
C.13d	Soil	Volatile organics	May 1995	C-91
C.14a	Soil	Semivolatile organics	Jan 1993	C-102
C.14b	Soil	Semivolatile organics	Jan 1994	C-116
C.14c	Soil	Semivolatile organics	May 1994	C-126
C.14d	Soil	Semivolatile organics	May 1995	C-142
C.15a	Soil	Inorganics	Jan 1993	C-160
C.15b	Soil	Inorganics	Jan 1994	C-165
C.15c	Soil	Inorganics	May 1994	C-169
C.15d	Soil	Inorganics	May 1995	C-176

TABLE C.1 (Cont.)

Table No.	Sample Medium	Analytes	Date	Page Number
C.16a	Soil	Pesticides and PCBs	Jan 1993	C-188
C.16b	Soil	Pesticides and PCBs	May 1995	C-193
C.17	Soil	On-site analytical suite	Dec 1993- May 1994	C-195
C.18	Soil	Explosives	May 1995	C-198
C.19	Soil	CWA degradation products Organosulfur compounds	May 1995	C-200
C.20	Soil	Dioxins Total petroleum hydrocarbons	May 1995	C-202
C.21a	Groundwater	Volatile organics	May 1994	C-203
C.21b	Groundwater	Volatile organics	Sept 1994	C-208
C.22	Groundwater	Semivolatile organics	May 1994	C-211
C.23	Groundwater	Inorganics	May 1994	C-215
C.24	Groundwater	General chemistry	May 1994	C-220
C.25	Groundwater	Pesticides and PCBs	May 1994	C-223
C.26	Groundwater	Explosive compounds	June 1994	C-224
C.27	Groundwater	Radiochemistry	June 1994	C-225

TABLE C.2a Volatile Organics Analyses for Surface Water, April 1993<sup>a</sup>

							Surface	Water Co	Surface Water Concentrations (µg/L)	ns (µg/L)						
Compound	S	M-6	S	SW-7	S	SW-8	SW	SW-10	SW-1	SW-10-Dup	SW-11	11	SW	SW-12	SW-13	.13
Acetone	6	-	10		œ	_	32	_	110	D	∞	<b>-</b>	∞	-	=	
Benzene	10	U	10	Ω	10	Ω	62	Ω	250	Ω	10	Ω	10	Ω	10	n
Bromodichloromethane	10	n	10	D	10	Ω	62	Ω	250	ם	10	D	10	Ω	10	n
Bromoform	10	n	10	Ŋ	10	Ω	62	Ω	250	Ω	10	n	10	n	10	Þ
Bromomethane	10	n	10	Ω	10	Ω	62	D	250	n	10	Ŋ	10	Ω	10	Ω
2-Butanone	2	_	2	'n	10	Ω	62	Ω	250	Ω	10	Ω	10	Ω	5	-
Carbon disulfide	10	n	10	Ω	10	Ω	62	Ω	250	Ω	10	Ω	10	U	01	n
Carbon tetrachloride	10	Ω	10	D	10	n	62	Þ	250	U	10	D	10	Ω	10	n
Chlorobenzene	10	Ω	10	Ω	10	D	62	D	250	Ω	10	Ω	10	Ω	10	Ω
Chloroethane	10	Ω	10	n	10	Ω	62	Ω	250	U	10	Ω	10	Ω	10	n
Chloroform	10	n	10	Ω	10	Ω	62	n	250	Ω	10	Ω	10	Ω	10	D
Chloromethane	10	Ω	10	Ω	10	D	62	Ω	250	Ω	10	Ω	10	Ω	10	Ω
Dibromochloromethane	10	Ω	10	ב	10	Þ	62	ח	250	Ω	10	Ω	10	Ω	10	D
1,1-Dichloroethane	10	O	10	Þ	10	n	62	Þ	250	Ω	10	n	10	Ω	10	D
1,2-Dichloroethane	10	Ω	10	n	10	Ω	62	Ω	250	Ω	10	n	10	U	10	n
I, I-Dichloroethene	10	ח	10	Ω	10	n	62	Ω	250	Ω	10	Ω	10	Ω	10	Ω
1,2-Dichloroethene (total)	10	Ω	19		10	n	1,400		1,400	Q	16		10	Ŋ	10	Ω
1,2-Dichloropropane	10	n	10	Ω	01	Ω	62	Ω	250	n	01	Ω	10	n	10	Ω
cis-1,3-Dichloropropene	10	D	10	Ω	10	n	62	Ω	250	n	10	n	10	Ω	10	n
trans-1,3-Dichloropropene	10	D	10	ם	10	Ω	62	n	250	Ω	10	Ω	10	Ω	01	Ω
Ethylbenzene	10	ņ	10	Ω	10	n	62	Ω	250	n	10	Ω	10	Ω	10	Ω
2-Hexanone	10	ב	01	ב	10	n	62	Ω	250	Ω	10	n	10	Ω	10	Ω
4-Methyl-2-pentanone	10	Ω	10	n	10	Ω	62	Ω	250	Ω	10	n	10	D	01	D
Methylene chloride	10	n	-	BJ	10	Ω	62	D	27	BDJ	10	n	10	Þ	10	D
Styrene	10	ם	10	ח	10	Ω	62	Ω	250	U	10	n	10	D	10	Ω
1,1,2,2-Tetrachloroethane	10	D	190		<b>C</b> I	_	2,200		2,300	D	œ	~	10	n	10	Ω
Tetrachloroethene	10	D	-	<u>,,</u>	10	n	77	-	40	DJ	10	n	10	n n	10	Ω
Toluene	10	Ξ	10	n	01	Ω	62	D	250	n	01	Ω	10	n	01	Ω
I, I, I-Trichloroethane	01	n	9	Þ	01	Ω	62	ח	250	Ω	9	Ω	10	a	01	n
1,1,2-Trichloroethane	10	D	<u></u>		10	n	66		250	Ω	_	_	10	n	01	n

TABLE C.2a (Cont.)

				Surface Water Co	Surface Water Concentrations (µg/L)			
Compound	SW-6	L-MS	SW-8	SW-10	SW-10-Dup	SW-11	SW-12	SW-13
Trichloroethene	10 U	59	10 U	2.100	250 11	1	1	;
17						•	0	) (1
vinyi chionde	D 01	10 U	10 U	62 U	250 U	2	10 11	11 01
Xylenes (total)	10 U	10 U	10 U	11 69	11 050	. :		o ;
					0 000	0	0 01	10 O

<sup>a</sup> Sampled by ANL and USGS, analyzed by Enesco, CLP/HSL volatile organics. Provisional data, validation in progress.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

B = analyte was found in the associated blank.

D = sample was diluted for analysis.

TABLE C.2b Volatile Organics Analyses for Surface Water, September 1993<sup>a</sup>

	Surface Water Concentrations (µg/L)
Compound	SW-7
Acetone	17
Benzene	10 U
Bromodichloromethane	10 U
Bromoform	10 U
Bromomethane	10 U
2-Butanone	10 U
Carbon disulfide	10 U
Carbon tetrachloride	10 U
Chlorobenzene	10 U
Chloroethane	10 U
Chloroform	10 U
Chloromethane	10 U
Dibromochloromethane	10 U
1,1-Dichloroethane	10 U
1,2-Dichloroethane	10 U
1,1-Dichloroethene	10 U
1,2-Dichloroethene (total)	10 U
1,2-Dichloropropane	10 U
cis-1,3-Dichloropropene	10 U
trans-1,3,-Dichloropropene	10 U
Ethylbenzene	10 U
2-Hexanone	10 U
4-Methyl-2-pentanone	10 U
Methylene chloride	1 BJ
Styrene	10 U
1,1,2,2-Tetrachloroethane	10 U
Tetrachloroethene	10 U
Toluene	1 J
1,1,1-Trichloroethane	10 U
1,1,2-Trichloroethane	10 U

## TABLE C.2b (Cont.)

	Surface Water Concentrations (µg/L)
Compound	SW-7
Trichloroethene	10 U
Vinyl chloride	10 U
Xylenes (total)	10 U

<sup>&</sup>lt;sup>a</sup> Sampled by USGS, analyzed by Enseco, CLP/HSL volatile organics. Provisional data, validation in progress.

### Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

B = analyte was found in the associated blank.

J = estimated value.

TABLE C.2c Volatile Organics Analyses for Surface Water, February 1994<sup>a</sup>

1							Surfac	e Water	Conce	Surface Water Concentrations (µg/L.)	(µg/L)							
Compound	05;	QSSSW	Q55SW-Dup	-Dup	Q56SW	3W	Q58SW	3W	Q59SW	SW	WS09Q	SW	Q62SW	SW	W859Q	SW	WSE60	W.S
Acetone	17	В	18	В	14	ĸ	15	<b>,</b>	12	В	10	Ω	14		2	α	~	l a
Benzene	10	n	10	Ω	10	ם	10	Ω	10	Ω	10	'n	10	n	: 2	i D	2	3 =
Bromodichloromethane	10	Ω	10	n	10	Ω	10	Ω	10	n	10	Ω	10	Ω	01	n	10	n
Вготобогт	10	Ω	10	Ω	10	U	10	n	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω
Bromomethane	10	Ω	10	n	10	U	10	Ω	10	Ω	10	Ω	10	n	10	Ω	10	n
2-Butanone	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω
Carbon disulfide	10	n	10	Ω	7	ſ	10	Ω	10	Ω	10	Ω	10	n	10	n	10	n
Carbon tetrachloride	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	01	D	10	Ω	10	Ω
Chlorobenzene	10	D	10	n	10	Ω	10	Ω	10	Ω	10	n	10	Þ	10	n	10	Ω
Chloroethane	10	Ω	10	Ω	10	Ω	10	n	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω
Chloroform	10	Þ	10	n	10	n	10	Þ	10	n	10	n	10	n	10	n	10	n
Chloromethane	10	Ω	10	Ω	10	D	10	n	10	Ω	10	n	10	Þ	10	n	10	Ω
Dibromochloromethane	10	n	10	Ω	10	Ω	10	Ω	10	n	10	D	10	Ω	10	n	10	Ω
1,1-Dichloroethane	10	Ω	10	n	10	Ω	10	n	10	n	10	Ω	10	Ω	10	Þ	10	n
1,2-Dichloroethane	10	n	10	n	01	Ω	10	Ω	10	n	10	Ω	10	Ω	10	n	01	n
1,1-Dichloroethene	10	⊃	10	n	10	Ω	10	Ω	10	n	10	n	10	Ω	10	n	10	Ω
cis-1,2-Dichloroethene	10	Ω	10	n	01	Ω	10	Ω	10	n	9	_	S	ſ	10	n	9	<b>–</b>
trans-1,2-Dichloroethene	10	Ω	10	n	01	Ω	10	n	10	Ω	10	Ω	10	n	10	Ω	10	n
1,2-Dichloropropane	10	Ω	10	Ω	10	Ω	10	₽	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω
cis-1,3-Dichloropropene	10	D	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	n	10	D	10	Ω
trans-1,3-Dichloropropene	10	n	10	ก	10	Ω	10	Ω	10	Ω	10	n	10	Ω	10	ם	10	D
Ethyl benzene	10	ם	10	D	10	D	10	n	10	Ω	10	Ω	10	Ω	10	Ω	01	Ω
2-Hexanone	10	Ω	10	Ω	10	Ω	10	Ω	10	n	10	D	01	D	10	n	01	Ω
4-Methyl-2-pentanone	10	ם	01	n	10	n	10	n	10	n	10	Ω	10	n	10	D	10	n
Methylene chloride	10	n	10	Ω	10	נ	01	n	01	n	10	Ω	10	Ω	10	n	10	n
Styrene	10	n	01	ם	10	D.	10	Ω	01	Ð	01	Ω	10	Ω	10	Ω	10	n
1,1,2,2-Tetrachloroethane	01	⊃	v	1	10	Ω	10	Ω	10	n	10	n	10	Ω	S	-	9	_
Toluene	01	$\Box$	10	Ω	10	D	01	ם	10	ت ت	10	U	10	n	10	Ω	10	)
1,1,1-Trichloroethane	10	Ω	01	ם	10	n	01	ם	10	Ω	0.	n	01	n	10	Ω	01	n
1,1,2. Trichloroethane	01	Ξ	10	Ω	10	Ω	10	n	01	ם	01	<u>ت</u>	10	ב	10	n	01	n

TABLE C.2c (Cont.)

							Surfac	e Water	Surface Water Concentrations (µg/L)	trations	(µg/L)		<u> </u> 					
Compound	95	QSSSW	Q55SV	Q55SW-Dup	05	Q56SW	Q58SW	M.S.	Q59SW	W	WS09Q	>	Q62SW	≥	WSS9Q	MS.	WSE6O	<b> </b> ≱
Trichloroethene	10	n	10	n	10	n	10	D	10	Ω	01		1	_	٤	=		
Vinyl acetate	10	n	10	n	10	ם	5	Ξ	10	. =		-	` {	, :	2 5	<b>)</b> ;	٠ ;	-, ;
Vinyl chloride	10	Ω	10	ב	0	- =	2	> =	2 2	> <b>=</b>		> <b>=</b>	2 2	<b>)</b>	2 :	<b>&gt;</b> ;	≘ :	<b>D</b> :
m- and p-Xylene	10	ב	10	Ξ	: 01	) <u>=</u>	2 2	) <u>=</u>	2 2	o :		o :		<b>)</b>	2	<b>-</b>	01	D
. Xvlana	2 -	) ;	2 :	);	2	>	01	<b>-</b>	2	<b>-</b>		D	10	n	10	n	10	Ω
0-Aylelle	01	o    -	0		2	ח	0	ח	2	n	01	n	10	U	10	U	10	n
				Surface V	Vater Con	centratio	Surface Water Concentrations (μg/L)	į								i       		i i i
Compound	60	Q95SW	SW-7	1-7	SW	SW-10	SW-10-Dup	-Dup	S	SW-11	i 1							
Acetone	22	В	10	D	10	ח	30		10	=								
Benzene	10	Ω	10	Ω	10	n	10	Ξ	2 2	) <u>=</u>								
Bromodichloromethane	10	ם	10	Ω	10	D	10	) ⊃	2	) =								
Bromoform	10	Ω	10	n	10	Ω	10	D	10	) D								
Bromomethane	10	n	10	U	10	D	10	ב	10	Ω								
2-Butanone	10	Ω	10	n	10	Ω	10	Ω	10	D								
Carbon disulfide	10	D	10	Ω	10	Ω	10	Ω	10	: =								
Carbon tetrachloride	10	Ω	10	U	10	n	10	ח	10	n								
Chlorobenzene	10	Ω	10	Ω	10	n	10	Ω	10	n								
Chloroethane	10	n	10	Ω	10	n	10	Þ	10	n								
Chloroform	10	Ω	10	n	ъ	ſ	3	~	10	Ω								
Chloromethane	10	Ω	10	U	10	n	10	n	10	· 12								
Dibromochloromethane	10	Ω	10	n	10	Ω	10	ב	10	n								
1,1-Dichloroethane	10	Ω	10	Ω	10	Ω	10	n	10	· D								
1,2-Dichloroethane	10	D	10	Ω	10	n	10	Ω	10	n								
1,1-Dichloroethene	10	Ω	10	Ω	S	_	4	~	10	n								
cis-1,2-Dichloroethene	==		7	J	1,724		1,809		7	_								
trans-1,2-Dichloroethene	10	Ω	2	ſ	173		239		10	Ω								
1,2-Dichloropropane	10	n	10	Ω	10	Ω	10	Ω	01	n								

TABLE C.2c (Cont.)

Compound	60	WS269	SW	SW-7	SW	SW-10	SW-10-Dup	-Dup	S	SW-11
cis-1,3-Dichloropropene	10	Ω	10	Ω	10	n	10	D	10	n
trans-1,3-Dichloropropene	10	Ω	10	Ω	10	Ω	10	Ω	10	n
Ethyl benzene	10	Ω	10	Ω	10	n	10	D	10	Þ
2-Hexanone	10	Ω	10	n	26		59		10	D
4-Methyl-2-pentanone	10	Ω	10	n	10	Ω	10	n	10	Ω
Methylene chloride	10	Ω	10	n	10	n	10	n	10	n
Styrene	10	Ω	10	n	10	D	10	Ω	10	D
1,1,2,2-Tetrachloroethane	15		211		3,220		4,348		10	ח
Toluene	10	Ω	10	Ω	10	Ω	10	Ω	10	n
1,1,1-Trichloroethane	10	n	10	Ω	10	n	10	n	10	ח
1,1,2-Trichloroethane	-	_	10	n	124		138		10	n
Trichloroethene	15		37		3,214		3,615		2	_
Vinyl acetate	10	n	10	D	10	Ω	10	b	01	D
Vinyl chloride	10	D	10	D	26		29		10	Þ
m- and p-Xylene	10	n	10	n	01	n	10	Ω	10	Þ
o-Xylene	10	Ω	10		10	Ξ	01	=	2	=

a Sampled by ANL, analyzed by ANL-ACL, CLP/HSL volatile organics.

U = analyte was analyzed for but not detected.

B = analyte was found in the associated blank.

J = estimated value.

TABLE C.2d Volatile Organics Analyses for Surface Water, May 1994<sup>a</sup>

		<del> </del>	S	urface '	Water Co	ncentrat	tions (µg/	L)		
Compound	S	W-7	sv	V-10	s	W-11	sw	'-11 Dup	S\	W-12
Acetone	10	U	10	U	10	U	10	U	10	U
Benzene	10	U	10	U	10	U	10	U	10	U
Bromodichloromethane	10	U	10	U	10	U	10	U	10	U
Bromoform	10	U	10	U	10	U	10	U	10	U
Bromomethane	10	U	10	U	10	U	10	U	10	U
2-Butanone	10	U	10	U	10	U	10	U	10	U
Carbon disulfide	10	U	10	U	10	U	10	U	10	U
Carbon tetrachloride	10	U	10	U	10	U	10	U	10	บ
Chlorobenzene	10	U	10	U	10	U	10	U	10	U
Chloroethane	10	U	10	U	18	Ü	12	O	10	U
Chloroform	10	U	10	U	10	U	10	U	10	U
Chloromethane	10	U	10	U	10	U	10	Ŭ	10	U
Dibromochloromethane	10	U	10	U	10	U	10	U	10	U
1,1-Dichloroethane	10	U	10	U	10	U	10	U	10	U
1,2-Dichloroethane	10	U	10	U	10	U	10	U	10	U
1,1-Dichloroethene	10	U	3	J	10	U	10	U	10	U
1,2-Dichloroethene (total)	10	U	1,700	Е	86	Ŭ	56	O	10	U
1,2-Dichloropropane	10	U	10	U	10	U	10	U	10	U
cis-1,3-Dichloropropene	10	U	10	U	10	U	10	U	10	บ
trans-1,3-Dichloropropene	10	U	10	U	10	U	10	U	10	U
Ethylbenzene	10	U	10	U	10	U	10	U	10	U
2-Hexanone	10	U	10	บ	10	U	10	U	10	U
4-Methyl-2-pentanone	10	U	10	U	10	U	10	U	10	IJ
Methylene chloride	10	U	10	U	10	U	10	U	10	U
Styrene	10	U	10	U	10	U	10	U	10	U
1,1,2,2-Tetrachloroethane	10	U	910	Е	3	J	10	U	10	U
Tetrachloroethene	10	U	10	U	10	U	10	U	10	U
Toluene	10	U	10	U	10	Ü	10	U	10	U
1,1,1-Trichloroethane	10	U	10	U	10	U	10	U	10	U
1,1,2-Trichloroethane	10	U	95		3	J	10	Ü	10	U
Trichloroethene	3	J	94		13	-	9	J	10	U
Vinyl chloride	10	U	26		22		14	•	10	U
Xylenes (total)	10	U	10	U	10	U	10	U	10	U

<sup>&</sup>lt;sup>a</sup> Sampled by ANL, analyzed by EA Laboratories, CLP/HSL volatile organics.

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

E = analyte concentration exceeded calibration range of instrument; concentration is greater than value given.

TABLE C.3a Semivolatile Organics Analyses for Surface Water, April 1993<sup>a</sup>

							Surface	. Water	Concer	Surface Water Concentrations (µg/L)	(µg/L)					
Compound	SV	SW-6	SV	SW-7	SV	SW-8	SW	6-MS	NS.	SW-10	SV	SW-11	SV	SW-12	SV	SW-13
Acenaphthene	10	Ω	10	ח	10	Þ	10	D	10	Þ	10	Ω	10	n	10	Ω
Acenaphthylene	10	Ω	10	Ω	10	Ω	10	Ω	10	D	10	ח	10	Ω	01	D
Anthracene	10	n	10	Ω	10	Ω	10	n	10	Ω	10	Ω	10	Ω	10	n
Benz(a)anthracene	10	D	10	Ω	10	n	10	Ω	10	Ω	01	Ω	10	n	10	ח
Benzo[a]pyrene	10	D	10	Ω	10	D	10	n	10	Ω	10	Ω	10	Ω	10	n
Benzo[b]fluoranthene	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	n	01	n
Benzo[g,h,i]perylene	10	n	10	D	10	n	10	Ω	10	n	10	Ω	10	D	10	Þ
Benzo[k]fluoranthene	10	Ω	10	Ω	10	Ω	10	Ω	10	n	10	Ω	01	n	10	U
Bis(2-chloroethyl)ether	10	D	10	Ω	10	Ω	10	n	10	Ω	10	Ω	10	n	10	n
Bis(2-chloroisopropyl)ether	10	n	10	Ω	10	Ω	10	Ω	10	n	01	n	10	Ω	10	ח
Bis(2-ethylhexyl)phthalate	2	_	2	_	10	Ω	10	Ω	10	ם	10	n	10	Ω		_
4-Bromophenyl phenyl ether	10	n	10	n	10	Ω	10	Ω	10	n	10	n	10	Ω	10	D
Butylbenzylphthalate	10	Ω	10	Ω	10	Ω	10	Ω	10	D	10	Ω	10	Ω	10	ח
Carbazole	10	Þ	10	Ω	10	n	10	Ω	10	Ω	10	D	10	n	10	D
4-Chloro-3-methylphenol	10	n	10	n	10	₽	01	n	10	n	10	n	10	n	10	Þ
4-Chloroaniline	10	Ω	10	n	10	n	01	n	10	Ω	10	Ω	10	n	10	Þ
cis(2-Chloroethoxy) methane	10	Ω	01	ב	01	n	10	Ω	10	D	10	Ω	10	Ω	10	n
2-Chloronaphthalene	10	Ω	10	n	10	Ω	01	Ω	10	n	10	Ω	10	Ω	10	D
2-Chlorophenol	10	Ω	01	n	10	Ω	10	Ω	10	Ω	10	n	10	D	10	Ω
4-Chlorophenyl phenyl ether	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	01	Ω	10	ם	10	Ŋ
Chrysene	10	n	10	Ω	10	Ω	10	Ω	01	Ω	10	Ω	10	n	10	D
Di-n-buty1phthalate	10	Ω	10	Ω	10	n	10	ח	10	D	10	n	10	n	10	D
Di-n-octyl phthalate	10	n	01	Ω	10	n	10	Ω	10	n	10	n	10	Þ	10	D
Dibenz[a,h]anthracene	10	n	10	Ω	10	n	10	Ω	10	D	10	D	10	Ω	10	n
Dibenzofuran	10	n	10	n	10	D	01	n	10	Ω	01	Þ	10	Ω	10	D
1,2-Dichlorobenzene	10	Ω	01	n	10	Ω	01	Ω	01	Ω	10	n	10	Ω	10	n
1,3-Dichlorobenzene	10	n	0	n	10	n	01	Ω	01	Þ	10	Ω	10	Ω	01	ח
1,4-Dichlorobenzene	10	Ω	01	n	01	n	10	n	10	Ω	10	Ω	10	D	01	Ω
3,3'-Dichlorobenzidine	10	Ω	01	n	0	ח	01	Ω	10	Ω	10	n	10	ם	<u>0</u>	Þ
2,4-Dichlorophenol	10	n	9	Ω	01	⊃	01	n	01	Ω	9	Ω	10	D	10	ב

TABLE C.3a (Cont.)

							Surfac	. Water	Concer	ıtration	Surface Water Concentrations (μg/L)					
Compound	S	9-MS	S	SW-7	SV	8w-8	NS.	6-MS	AS.	SW-10	SV	SW-11	S	SW-12	NS.	SW-13
Diethylphthalate	10	n	10	D	10	D	10	Þ	10	=	0	=	2	=	٤	=
2,4-Dimethylphenol	10	n	10	Ω	10	n	10	) D	10	) D	2 9	) <u>=</u>	2 2	) <b>=</b>	2 2	> =
DimethyIphthalate	10	Ω	10	Ω	10	Ω	10	ם	10	Ω	2	> =	2 =	> =	2 2	) <u>=</u>
4,6-Dinitro-2-methylphenol	25	Ω	25	n	25	n	25	D	25	n	25	) =	25	> =	3 %	> =
2,4-Dinitrophenol	25	Ω	25	Ω	25	Ω	25	D	25	n	25	ם כ	25	) D	3 %	> =
2,4-Dinitrotoluene	10	n	10	Ω	10	Ω	10	n	10	D	10	ר	10	) D	01	> <b>=</b>
2,6-Dinitrotoluene	10	n	10	Ω	10	n	10	Ω	10	Ω	10	D	10	) )	2 01	· ⊃
Fluoranthene	10	n	10	Ω	10	Ω	10	Ω	10	D	10	D	10	n	10	'n
Fluorene	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω
Hexachlorobenzene	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	n	10	D
Hexachlorobutadiene	10	n	10	n	10	Ω	10	Ω	10	Ω	10	כ	10	Ω	10	ם ב
Hexachlorocyclopentadiene	10	n	10	Ω	10	n	10	Ω	10	Ω	10	Ω	10	Ω	01	ı D
Hexachloroethane	10	n	10	n	10	Ω	10	n	10	Ω	10	Ω	10	n	10	=
Indeno[1,2,3-c,d]pyrene	10	n	10	n	10	n	10	Ω	10	n	10	n	10	Ω	10	) <u> </u>
Isophorone	10	n	10	n	01	Ω	10	Ω	10	n	10	n	10	n	10	n
2-Methyinaphthalene	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	01	Ω	10	Ω	10	n
2-Methylphenol	10	Ω	01	Ω	01	ח	10	n	10	n	10	Ω	10	n	10	n
4-Methylphenol	10	Ω	01	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	n
Naphthalene	10	Ω	10	Ω	10	n	10	U	10	Ω	10	D	10	D	10	n
2-Nitroaniline	25	ח	25	Ω	25	n	25	Ω	25	Ω	25	Ω	25	D	25	, ,
3-Nitroaniline	25	D	25	Ω	25	n	25	n	25	Ω	25	D	25		25	· =
4-Nitroaniline	25	n	25	Ω	25	Ω	25	Ω	25	Ω	25	Ω	25	n	25	n
Nitrobenzene	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	n	10	Ω
2-Nitrophenol	10	Ω	10	Ω	10	Ω	10	n	10	D	10	Ω	10	n	: 0	) D
4-Nitrophenol	25	Ω	25	Ω	25	Ω	25	n	25	Ω	25	Ω	25	n	25	o D
N- Nitrosodiphenylamine (1)	2	ſ	-	ſ	_	-	7	ſ	2	-	2	'n	2	_	_	
N-Nitroso-di-n-propylamine	10	n	01	Ω	01	Ω	10	n	10	Ω	10	D	10	Ω	01	. n
Pentachlorophenol	25	n	25	n	25	Ω	25	D	25	Ω	25	n	25	Ω	25	Ω
Phenanthrene	01	n	10	Ω	10	Ω	10	n	10	Ω	10	Ω	10	Ω	10	Ω
Phenol	10	Ω	10	Ω	10	Ω	10	n	10	Ω	10	Ω	10	n	10	n

TABLE C.3a (Cont.)

						Surface Water Concentrations	Water	100000	, 000:101	( )					
						Suriace	v alter	Ollicelli	allons	hg/L)					
Compound	9-MS		SW-7	SV	SW-8	6-MS	6-	SW-	SW-10	SW-11	Ξ	SW-12	.12	SW-13	.13
Pyrene	10 L	10	n	10	n	10	n	10	n	10	D	10	n	10	Ω
1,2,4-Trichlorobenzene	10 U	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	ם
2,4,5-Trichlorophenol	25 U	25	n	25	n	25	n	25	Ω	25	Ω	25	Ω	25	D
2,4,6-Trichlorophenol	10 U	10	Ω	10	n	10	n	10	Ω	10	n	10	n	10	Ω

<sup>a</sup> Sampled by ANL and USGS, analyzed by Enseco, CLP/HSL semivolatile organics. Provisional data, validation in progress.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

TABLE C.3b Semivolatile Organics Analyses for Surface Water, September 1993<sup>a</sup>

	Surface Water Concentrations (µg/L)
Compound	SW-7
Acenaphthene	10 U
Acenaphthylene	10 U
Anthracene	10 U
Bis(2-chloroethoxy)methane	10 U
Bis(2-chloroisopropyl)ether	10 U
Bis(2-chloroethyl)ether	10 U
4-Bromophenyl phenyl ether	10 U
4-Chloro-3-methylphenol	10 U
4-Chloroaniline	10 U
2-Chloronaphthalene	10 U
2-Chlorophenol	10 U
4-Chlorophenyl phenyl ether	10 U
Dibenzofuran	10 U
1,2-Dichlorobenzene	10 U
1,3-Dichlorobenzene	10 U
1,4-Dichlorobenzene	10 U
2,4-Dichlorophenol	10 U
Diethylphthalate	10 U
Dimethyl phthalate	10 U
2,4-Dimethylphenol	10 U
4,6-Dinitro-2-methylphenol	5 Ј
2,4-Dinitrophenol	10 U
2,4-Dinitrotoluene	10 U
2,6-Dinitrotoluene	10 U
Fluorene	10 U
Hexachlorobenzene	10 U
Hexachlorobutadiene	10 U
Hexachlorocyclopentadiene	10 U
Hexachloroethane	10 U
Isophorone	<sup>-</sup> 10 U
2-Methylnaphthalene	10 U
2-Methylphenol	3 Ј

TABLE C.3b (Cont.)

	Surface Concent (µg/	rations
Compound	SW	-7
4-Methylphenol	9	J
Naphthalene	10	U
2-Nitroaniline	10	U
3-Nitroaniline	10	U
4-Nitroaniline	10	U
Nitrobenzene	10	U
2-Nitrophenol	10	U
4-Nitrophenol	10	U
N-Nitroso-di-n-propylamine	10	U
N-Nitrosodiphenylamine	10	U
Pentachlorophenol	10	U
Phenanthrene	10	U
Phenol	6	Ĵ
1,2,4-Trichlorobenzene	10	U
2,4,5-Trichlorophenol	10	U
2,4,6-Trichlorophenol	10	U

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and USGS, analyzed by Enseco, CLP/HSL volatile organics. Provisional data, validation in progress.

J = estimated value.

U = analyte was analyzed for but not detected; detection limit given.

TABLE C.3c Semivolatile Organics Analyses for Surface Water, May 1994<sup>a</sup>

			<u> </u>	Surface	Water (	Concer	ntrations (µ	ıg/L)		
Compound	S	W-7	S	W-10	S	W-11	SW-1	l 1-Dup	sv	V-12
Acenaphthene	10	U	10	U	10	U	10	U+	10	U
Acenaphthylene	10	U	10	U	10	U	10	U+	10	Ŭ
Anthracene	10	U	10	U	10	U	10	U+	10	U
Benz[a]anthracene	10	U	10	U	10	Ŭ	10	U+	10	U
Benzo[a]pyrene	10	U	10	U	10	U	10	U+	10	U
Benzo[b]fluoranthene	10	U	10	U	10	U	10	U+	10	U
Benzo[g,h,i]perylene	10	U	10	U	10	U	10	U+	10	U
Benzo[k]fluoranthene	10	U	10	U	10	U	10	U+	10	U
Bis(2-chloroethoxy)methane	10	U	10	U	10	U	10	U+	10	U
Bis(2-chloroethyl)ether	10	U	10	U	10	U	10	U+	10	U
Bis(2-ethylhexyl)phthalate	10	U	10	U	10	U	10	U+	10	U
4-Bromophenyl phenyl ether	10	U	10	U	10	U	10	U+	10	U
Butylbenzylphthalate	10	U	10	U	10	U	10	U+	10	Ŭ
Carbazole	10	U	10	U	10	U	10	U+	10	U
4-Chloro-3-methylphenol	10	U	10	U	10	U	10	U+	10	U
4-Chloroaniline	10	U	10	U	10	U	10	U+	10	U
2-Chloronaphthalene	10	U	10	U	10	U	10	U+	10	U
2-Chlorophenol	10	U	10	U	10	U	10	U+	10	U
4-Chlorophenyl phenyl ether	10	U	10	U	10	U	10	U+	10	U
Chrysene	10	U	10	U	10	U	10	U+	10	U
Di-n-butylphthalate	10	U	10	U	10	U	10	U+	10	U
Di-n-octyl phthalate	10	U	10	U	10	U	10	U+	10	U
Dibenz[a,h]anthracene	10	U	10	U	10	U	10	U+	10	Ŭ
Dibenzofuran	10	U	10	U	10	U	10	U+	10	Ŭ
1,2-Dichlorobenzene	10	U	10	U	10	U	10	U+	10	U
1,3-Dichlorobenzene	10	U	10	U	10	U	10	U+	10	U
1,4-Dichlorobenzene	10	U	10	U	10	U	10	U+	10	U
3,3'-Dichlorobenzidine	10	U	10	U	10	U	10	U+	10	U
2,4-Dichlorophenol	10	U	10	U	10	U	10	U+	10	U
Diethylphthalate	10	U	10	U	10	U	10	U+	10	U
2,4-Dimethylphenol	10	U	10	U	10	U	10	U+	10	U
Dimethylphthalate	10	U	10	U	10	U	10	U+	10	U
2,4-Dinitrophenol	25	U	25	U	25	U	25	U+	25	U

TABLE C.3c (Cont.)

		· · · · · ·	S	urface	Water C	Concent	rations (μ	g/L)		
Compound	SV	V-7	SV	V-10	SV	V-11	SW-1	1-Dup	SW	V-12
2,4-Dinitrotoluene	10	U	10	U	10	U	10	U+	10	U
2,6-Dinitrotoluene	10	U	10	U	10	U	10	U+	10	U
Fluoranthene	10	U	10	U	10	U U	10	U+	10	U
Fluorene	10	U	10	U	10		10	U+	10	U
Hexachlorobenzene	10	U	10	U	10	U	10	U+	10	U
Hexachlorobutadiene	10	U	10	U	10	U	10	U+	10	U
Hexachlorocyclopentadiene	10	U	10	U	10	U	10	U+	10	U
Hexachloroethane	10	U	10	U	10	U	10	U+	10	U
Indeno[1,2,3-c,d]pyrene	10	U	10	U	10	U	10	U+	10	U
Isophorone	10	U	10	U	10	U	10	U+	10	U
2,Methyl-4,6-Dinitrophenol	25	U	25	U	25	U	25	U+	25	U
2-Methylnaphthalene	10	U	10	U	10	U	10	U+	10	U
2-Methylphenol	10	U	10	U	10	Ū	10	U+	10	U
4-Methylphenol	10	U	10	U	10	U	10	U+	10	U
Naphthalene	10	U	10	U	10	U	10	U+	10	U
2-Nitroaniline	25	U	25	U	25	U	25	U+	25	U
3-Nitroaniline	25	U	25	U	25	Ü	25	U+	25	U
4-Nitroaniline	25	U	25	U	25	U	25	U+	25	U
Nitrobenzene	10	U	10	U	10	U	10	U+	10	U
2-Nitrophenol	10	U	10	U	10	U	10	U+	10	U
4-Nitrophenol	25	U	25	U	25	U	25	U+	25	U
N-Nitroso-di-n-propylamine	10	U	10	U	10	Ü	10	U+	10	U
N-Nitrosodiphenylamine	10	U	10	U	10	U	10	U+	10	U
2,2'-Oxybis(1-Chloropropane)	10	U	10	U	10	U	10	U+	10	U
Pentachlorophenol	25	U	25	U	25	U	25	U+	25	U
Phenanthrene	10	U	10	U	10	U	10	U+	10	U
Phenol	10	U	10	Ŭ	10	U	10	U+	10	U
Pyrene	10	U	10	U	10	U	10	U+	10	U
1,2,4-Trichlorobenzene	10	U	10	U	10	U	10	U+	10	U
2,4,5-Trichlorophenol	25	U	25	U	25	U	25	U+	25	
2,4,6-Trichlorophenol	10	U	10	U	10	U	10	U+	10	U U

<sup>&</sup>lt;sup>a</sup> Sampled by ANL, analyzed by EA Laboratories, CLP/HSL semivolatile organics.

U = analyte was analyzed for but not detected; detection limit given.

<sup>+ =</sup> surrogate recoveries were outside QC limit(s).

TABLE C.4a Inorganic Analyses for Surface Water, September 1993<sup>a</sup>

ć															***	
Parameter	AS	SW-6	1S	SW-7	SV	SW-8	S	6-WS	S	SW-10	SW	SW-11	S	SW-12	S	SW-13
Metals																
Aluminum	698		261		230		64.9	В	47	Ξ	582	Œ	7.1	=	001	
Antimony	49	D	49	n	49	Ω	49		49	· =	1.00	a =	Ì Ç			
Arsenic	2.3	В	2	Ω	2	1	, ,	> =	7 6	2 0	÷ ;	ם כ	<del>,</del>		49	
Barium	149	В	61.2	· œ	79.1	) <u>c</u>	4 01	ם	0.0	α ε	7:7	<b>2</b>	7			
Beryllium	_	n	_		-	> =	-	<u> </u>	20.0	ο :	0¢ •	n :	133		58.8	
Cadmium	4	Ω	4	D	. 4	) =	• 7	) <u>=</u>	→ ▼	> =	- T	) :	- •	⊃ ;	_	<b>D</b> :
Calcium	27,700		33,500		123,000	,	63.500	)	78 500	)	43 800	>	4	)	4 00	) 
Chromium	9	ם	9	Þ	9	Ξ	9	Ξ	ocato.	Ξ	000,01	:	000,62		19,400	
Cobalt	9	n	9.1	В	9	· =	o vo	=	<b>&gt;</b>	> =	0 7	) <u>:</u>	۰ ۵	o ;	9 '	
Copper	15.3	В	11.1	В	6.9	В	7.5	) <u>m</u>	o v	> =		۵ د	0 3		•	
Iron	181,100		3,740		2,980		458	}	1.170	)	1.050	a	3 190	n	0.1.3	n 
Lead	6.1		1	D	-	n	•	⊐	-	Ξ	8 7		0,1,0		1,0%	
Magnesium	19,500		25,100		52,000		24,900	•	15.800		58.000		40 900		4.2	r
Manganese	352		385		295		70.4		93 \$		106		007,74		000,01	_
Mercury	0.1	Ω	0.1	D	0 1	=	-	Ī	10	Į	3 -	:	751	;	0.76	
Nickel	13	Ω	2	=		· =	7	) <u>=</u>	1.0	o :	1.0	<b>&gt;</b> ;	0.1	<b>&gt;</b>	0.1	
Potassium	3.410	ď	4 970		7 430	•		ء د	C1 .	) ,	51	<b>D</b>	13	D	13	⊃
Selenium		ì	2//:		7,430		4,020	n	3,500	m	8,290		5,870		4,040	В
Silver	۲.۶	;	7	<b>-</b>	3.4		2	n	2	D	2	Ω	2	Ω	2	D
Silver	7	D	7.6	ш	7	n	7	n	7	Ω	7	Ω	7	Ω	7	n
Sodium	99,200		156,000		416,000		110,000		97,500		58,600		55,500		74.800	
Thallium	20	Þ	20	n	20	Ω	2	Ω	2	Ω	20	n	2	Ξ	C	Ξ
Vanadium	6.1	В	6.7	В	5	Ω	S	Ω	5	D	٧.	- 1	۱ ۲۰	> =	1 v	
Zinc	73.4		43		18	В	32.8		21.5		262	ı	782	)	36	)
Other parameter																
Cvanide	=	11	•	;												

TABLE C.4a (Cont.)

Parameter	SW-14	-14	SW	SW-15
Metals				
Aluminum	1,090		237	
Antimony	54	n	54	Ω
Arsenic	3.4	В	2.9	В
Barium	173	В	173	В
Beryllium	3.4	Ω	3.4	Ω
Cadmium	2.5	Ω	2.5	Ω
Calcium	101,000		99,500	
Chromium	5.2	Ω	5.2	Ω
Cobalt	3.6	Ω	3.6	n
Copper	4.0	Ω	4.7	В
Iron	2,900		1,310	
Lead	2.4	В	1.0	n
Magnesium	218,000		224,000	
Manganese	2,530		2,620	
Mercury	0.05	D	0.05	D
Nickel	8.0	D	8.0	n
Potassium	53,900		61,300	
Selenium	1.0	Ω	1.0	D
Silver	5.6	Þ	5.6	D
Sodium	1,530,000		1,620,000	
Thallium	5.0	Ω	5.0	n
Vanadium	9.1	8	6.9	В
Zinc	57.1		30	

Sampled by USGS, analyzed by Enseco, CLP/TAL metals and cyanide. Provisional data, validation in progress.

D 01

10 U

Cyanide

U = analyte was analyzed for but not detected; detection limit given.

 $B_{\parallel}$  reported value is less than the contract-required detection limit but greater than the instrument detection limit.

TABLE C.4b Inorganic Analyses for Surface Water, May 1994a

	·		Sı	ırface Wa	iter Concenti	ations	(μg/L)			
Metal	SV	V-7	SV	V-10	SW-	11	SW-11	Dup	SW	-12
Aluminum	429		226		18,000		2,000		731	
Antimony	20	U	20	U	32.9	В	20	U	20	U
Arsenic	2.4	В	6.4	В	36.3		9.4	В	18.2	O
Barium	69.5	В	43.4	В	559		167	В	234	
Beryllium	1	U	1	U	1	U	1	U	1	U
Cadmium	3	U	3	U	4	U	4	U	4	U
Calcium	35,800		106,000		98,700		90,700	_	33,800	Ü
Chromium	5	U	5.6	В	64.8		7.6	В	8	В
Cobalt	8.4	В	9.7	В	7	U	7.0	U	7	U
Copper	3	U	3	U	525		57.6	_	105	O
Iron	21,100		3,470		128,000		27,800		13,000	
Lead	2.1	В	6.5		1,590		210	S	169	
Magnesium	44,500		28,500		110,000		97,300	-	68,900	
Manganese	664		239		1,080		759		496	
Mercury	0.2	U	0.2	U	1.7		0.31		0.2	U
Nickel	12	U	12	U	22.4	В	12	U	12.1	В
Potassium	12,800		4,430	В	17,000		15,300	Ü	9,000	Ъ
Selenium	1	UW	1	UW	1	U	1	UW	2,000 1	U
Silver	4	U	5.3	В	4	U	4	U	4	U
Sodium	347,000		127,000		228,000		197,000	Ü	129,000	U
Thallium	2.00	UNW	2	UNW	2	U	2	UW	2	U
Vanadium	6.3	В	5	U	42.7	В	5	U	6.8	В
Zinc	11.8	B*@	18.4	B*@	4,040		1,670	E	968	D

<sup>&</sup>lt;sup>a</sup> Sampled by ANL, analyzed by EA Laboratories, CLP/TAL metals.

U = analyte was analyzed for but not detected; detection limit given.

B = reported value is less than contract-required detection limit but greater than the instrument detection limit.

W = postdigestion spike out of control limits.

N =spiked sample recovery is not within control limits.

<sup>\* =</sup> duplicate analyses are not within control limits.

<sup>@ =</sup> percent RPD of sample duplicate was outside of control limits.

S = reported value is determined by method of standard additions.

E = reported value is estimated because of the presence of interference.

TABLE C.5a Pesticide and Polychlorinated Biphenyl (PCB) Analyses for Surface Water, April 1993<sup>a</sup>

					Sur	face W	Surface Water Concentrations (μg/L)	entra	tions (µg	(T)				
Compound	SW	9-MS	SW-7	7-	SI	6-MS	SW-10	-10	SW-11	-11	SW-12	-12	SW-13	[3
Aldrin	0.05	Ω	0.05	Ω	0.05	Ω	0.05	n	0.05	D	0.05	Ω	0.05	1
Aroclor 1016	1.00	n	1.00	Ω	1.00	D	1.00	Ω	1.00	n	1.00	'n	1.00	n
Aroclor 1221	2.00	U	2.00	n	2.00	Ω	2.00	n	2.00	Ω	2.00	Ω	2.00	'n
Aroclor 1232	1.00	Ω	1.00	Ω	1.00	Ω	1.00	D	1.00	Ω	1.00	Ω	1.00	n
Aroclor 1242	1.00	Ω	1.00	D	1.00	n	1.00	n	1.00	Ω	1.00	Ω	1.00	n
Aroclor 1248	1.00	D	1.00	D	1.00	Ω	1.00	Ω	1.00	D	1.00	n	1.00	n
Aroclor 1254	1.00	n	1.00	Ŋ	1.00	Ω	1.00	D	1.00	n	1.00	Ω	1.00	Ω
Aroclor 1260	1.00	n	1.00	Ω	1.00	Ω	1.00	n	1.00	Ω	1.00	Ω	1.00	D
$\alpha ext{-BHC}$	0.05	D	0.05	Ω	0.05	n	0.05	Ω	0.05	Ŋ	0.05	Ω	0.05	Ω
<b>β-</b> ВНС	0.05	$\Box$	0.05	Ω	0.05	D	0.05	Ω	0.05	Ω	0.05	Ω	0.05	Ω
ô-BHC	0.05	Ω	0.05	n	0.05	n	0.05	Ω	0.05	n	0.05	n	0.05	Ω
γ-BHC (Lindane)	0.05	n	0.05	Ω	0.05	Ω	0.05	n	0.05	Ω	0.05	U	0.05	n
$\alpha$ -Chlordane	0.05	n	0.05	Ω	0.05	n	0.05	D	0.05	Ω	0.05	Ω	0.05	Ω
$\gamma$ -Chlordane	0.05	ר	0.05	D	0.05	n	0.05	D	0.05	D	0.05	Ω	0.05	Ω
4,4'-DDD	0.10	Ω	0.10	n	0.10	n	0.10	n	0.10	Ω	0.10	D	0.10	Ω
4,4'-DDE	0.10	n	0.10	Ω	0.10	Ω	0.10	Ω	0.10	n	0.10	Ω	0.10	Ω
4,4'-DDT	0.10	Ω	0.10	Ω	0.10	Ω	0.10	Ω	0.10	D	0.10	Ω	0.10	n
Dieldrin	0.10	n	0.10	Ω	0.10	Ω	0.10	Ω	0.10	D	0.10	Ω	0.10	Ω
Endosulfan I	0.05	$\Box$	0.05	Ω	0.05	Ω	0.05	D	0.05	n	0.05	Ü	0.05	n
Endosulfan sulfate	0.10	D	0.10	n	0.10	Ω	0.10	Ω	0.10	n	0.10	D	0.10	U
Endosulfan II	0.10	D	0.10	n	0.10	n	0.10	n	0.10	Ω	0.10	n	0.10	n

TABLE C.5a (Cont.)

					Sur	face W	Surface Water Concentrations (µg/L)	entrat	ions (µg/	(L)	;			
Compound	9-MS	9-/	SW-7	7-	SV	6-MS	SW-10	.10	SW-11	11	SW-12	.12	SW-13	13
Endrin	0.10	n	0.10	Ω	0.10	Ω	0.10	U	0.10		0.10	=	0.10	=
Endrin aldehyde	0.10	Ω	0.10	Ω	0.10	Ω	0.10	'n	0.10	o D	0.10	) =	0.10	) <u> </u>
Endrin ketone	0.10	Ω	0.10	Ω	0.10	Ω	0.10	Þ	0.10	î	0.10	) <u>=</u>	0.10	) <u> </u>
Heptachlor	0.05	n	0.05	n	0.05	Ω	0.05	D	0.05	) =	0.05	) <b>=</b>	0.05	> =
Heptachlor epoxide	0.05	Ω	0.05	n	0.05	Ω	0.05	D	0.05	<u> </u>	0.05	) <u>=</u>	0.05	) <b>=</b>
Methoxychlor	0.50	Ω	0.50	Ω	5.00	D	0.50	n	0.50	n	0.50	) <u> </u>	0.50	) <u> </u>
Toxaphene	5.00	n	5.00	Ω	5.00	Ω	5.00	Ω	5.00	n	5.00	n	5.00	o D

<sup>a</sup> Sampled by ANL and USGS, analyzed by Enseco, CLP/HSL organochloride pesticides and PCBs. Provisional data, validation in progress.

Data qualifier:

U = analyte was analyzed for but not detected.

TABLE C.5b Pesticide and PCB Analyses for Surface Water, September 1993<sup>a</sup>

	Surface Water Concentrations (µg/L)
pound	SW-7

Compound	SW	-7
Aldrin	0.05	U
Aroclor 1016	1.00	U
Aroclor 1221	2.00	U
Aroclor 1232	1.00	U
Aroclor 1242	1.00	U
Aroclor 1248	1.00	U
Aroclor 1254	1.00	U
Aroclor 1260	1.00	U
α-ВНС	0.05	U
β-ВНС	0.05	U
δ-ВНС	0.05	U
γ-BHC (Lindane)	0.05	U
α-Chlordane	0.05	U
γ-Chlordane	0.05	U
4,4'-DDD	0.10	U
4,4'-DDE	0.10	U
4,4'-DDT	0.10	U
Dieldrin	0.10	U
Endosulfan I	0.05	U
Endosulfan sulfate	0.10	U
Endosulfan II	0.10	U
Endrin	0.10	U
Endrin aldehyde	0.10	U
Endrin ketone	0.10	U

## TABLE C.5b (Cont.)

	Surface W Concentrat (µg/L)	
Compound	SW	-7
Heptachlor	0.05	U
Heptachlor epoxide	0.05	U
Methoxychlor	0.50	U
Toxaphene	5.00	U

<sup>&</sup>lt;sup>a</sup> Sampled by USGS, analyzed by Enseco, CLP/HSL organochloride pesticides and PCBs. Provisional data, validation in progress.

Data qualifier:

U = analyte was analyzed for but not detected; detection limit given.

TABLE C.5c Pesticide and PCB Analyses for Surface Water, May 1994<sup>a</sup>

			Su	rface W	later Cor	ncentra	tions (µg	/L)		
Compound	S	W-7	SW	-10	SW	-11	SW-1	l Dup	SV	V-12
Aldrin	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
Aroclor 1016	1	U+	1	U+	1	U+	1	U+	1	U+
Aroclor 1221	2	U+	2	U+	2	U+	2	U+	2	U+
Aroclor 1232	1	U+	1	U+	1	U+	1	U+	1	U+
Aroclor 1242	1	U+	1	U+	1	U+	1	U+	1	U+
Aroclor 1248	1	U+	1	U+	1	U+	1	U+	1	U+
Aroclor 1254	1	U+	1	U+	1	U+	1	U+	I	U+
Aroclor 1260	1	U+	1	U+	1	U+	1	U+	I	U+
α-BHC	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
β-ВНС	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
δ-BHC	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
ү-ВНС	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
α-Chlordane	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
γ-Chlordane	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
4,4'-DDD	0.10	U+	0.10	U+	0.10	U+	0.10	U+	0.10	U+
4,4'-DDE	0.10	U+	0.10	U+	0.10	U+	0.10	U+	0.10	U+
4,4'-DDT	0.10	U+	0.10	U+	0.10	U+	0.10	U+	0.10	U+
Dieldrin	0.10	U+	0.10	U+	0.10	U+	0.10	U+	0.10	U+
Endosulfan I	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
Endosulfan sulfate	0.10	U+	0.10	U+	0.10	U+	0.10	U+	0.10	U+
Endosulfan II	0.10	U+	0.10	U+	0.10	U+	0.10	U+	0.10	U+
Endrin	0.10	U+	0.10	U+	0.10	U+	0.10	U+	0.10	U+
Endrin aldehyde	0.10	U+	0.10	U+	0.10	U+	0.10	U+	0.10	U+
Endrin ketone	0.10	U+	0.10	U+	0.10	U+	0.10	U+	0.10	U+
Heptachlor	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
Heptachlor epoxide	0.05	U+	0.05	U+	0.05	U+	0.05	U+	0.05	U+
Methoxychlor	0.50	U+	0.50	U+	0.50	U+	0.50	U+	0.50	U+
Toxaphene	5	U+	5	U+	5	U+	5	U+	5	U+

<sup>&</sup>lt;sup>a</sup> Sampled by ANL, analyzed by EA Laboratories, CLP/HSL organochloride pesticides and PCBs. Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

<sup>+ =</sup> surrogate recoveries were outside QC limit(s).

TABLE C.6a General Chemistry Analyses for Surface Water, May 1994<sup>a</sup>

			Sui	rface V	Vater Conc	entrati	ons (mg/L	) <sup>b</sup>		
Analyte	SW	-7	SW-	10	SW-	11	SW-1	1 Dup	SW-	-12
Alkalinity	60.8		222		207		210		248	
Bicarbonate	60.8		222		206		210		247	
Bromide	3.0		1.3		1.3		1.3		1.0	
Carbonate	1.0	U	1.0	U	1.0	U	1.0	U	1.0	U
Chloride	810		270		720		630	Ü	330	J
Cyanide	0.01	U	0.01	U	0.01	U	0.01	U *	0.01	U
Fluoride	0.1	U	0.1	U	0.1	U	0.1	U	0.1	U
Nitrate	0.1	U	0.1	U	0.1	U	3.6	Ü	2.2	U
Nitrite	5.0	U	2.5	U	2.5	U	2.5	U	0.5	U
Orthophosphate	0.1	U	0.1	U	0.1	U	0.1	U	0.1	U
pH (SI)	6.2		7.0		6.6	+	6.8	+	7.1	+
Phosphate, total	1.9		0.13		2.7		0.57	,	0.75	-
Sulfate	64		100		190		190		37	
TDS	1,400		775		1,510		1,460		790	
Turbidity (NTU)	7.5		2.6		69		28		13.8	
Total organic halogens (µg/L)	120		2,600		NT		NT		NT	

<sup>&</sup>lt;sup>a</sup> Sampled by ANL, analyzed by EA Laboratories, CLP/TAL metals.

U = analyte was analyzed for but not detected; detection limit given.

- + = holding time was not met.
- e cyanide sample was not preserved in field with NaOH.

NT = not tested.

<sup>&</sup>lt;sup>b</sup> Units are mg/L, except where indicated.

TABLE C.6b Explosive Compounds Analyses for Surface Water, May 1994<sup>a</sup>

			Su	rface	Water C	oncenti	rations (με	g/L)		
Compound	SW	'-7	SW	-10	SW-	11	SW-1	1 Dup	SW-	12
1,3-DNB	0.62	U	0.62	U	0.62	U	0.62	U	0.62	U
2,4-DNT	0.43	U	0.43	U	0.43	U	0.43	U	0.43	U
2,6-DNT	1.2	U	1.2	U	1.2	U	1.2	U	1.2	U
HMX	1.1	U	1.1	U	1.1	U	1.1	U	1.1	U
NB	0.45	U	0.45	U	0.45	U	0.45	U	0.45	U
RDX	1.7	U	1.7	U	0.48	JI	0.98	JI	1.0	JΙ
TETRYL	1.9	U	1.9	U	1.9	U	1.9	U	1.9	U
1,3,5-TNB	0.86	U	0.86	U	0.86	U	0.86	U	0.86	U
2,4,6-TNT	1.3	U	1.3	U	1.3	U	1.3	U	1.3	U

<sup>&</sup>lt;sup>a</sup> Sampled by ANL, analyzed by EA Laboratories.

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

I = interferences in sample cause the quantitation and/or identification to be suspect.

TABLE C.6c Chemical Warfare Agent (CWA) Degradation Products and Organosulfur Compounds Analyses for Surface Water, April 1993<sup>a</sup>

		Sı	ırface Wa	ater Co	ncentrat	ions (μ	g/L)	
Compound	SW	<u>-7</u>	SW-	-10	SW	/-12	SW	-13
CWA degradation products								
Diisopropyl methylphosphonate (DIMP)	0.43	U	0.45		0.43	U	0.43	IJ
Dimethyl methylphosphate (DMMP)	0.2	U	0.2	U	0.2	U	0.2	Ū
Thiodiglycol	10	U	10	U	10	U	10	U
Organosulfur compounds								
Benzothiozole (BTZ)	5.22	U	5.22	U	5.22	U	5.22	U
p-Chlorophenylmethyl sulfide (CPMS)	6.04	U	6.04	U	6.04	U	6.04	U
p-Chlorophenylmethyl sulfone (CPMSO2)	6.43	U	6.43	U	6.43	U	6.43	U
p-Chlorophenylmethyl sulfoxide (CPMSO)	11.9	U	11.9	U	11.9	U	11.9	U
Dimethyl disulfide (DMDS)	0.55	U	0.55	Ū	0.55	U	0.55	IJ
Dithiane (DITH)	1.57	U	2.49	-	1.57	IJ	1.57	U
1,1 Oxathiane (OXAT)	2.55	U	12.2		2.55	U	2.55	U

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and USGS, analyzed by EA Laboratories.

U = analyte was analyzed for but not detected; detection limit given.

TABLE C.7a Volatile Organics Analyses for Sediment Borings, April 1994<sup>a</sup>

DOR1-1 BOR1-2 BOR1-2 BOR1-1 (10-12)  15 U 43 B 14  15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 12  10 U 13 U 12  10 U 13 U 12  11 U 13 U 12  12 U 13 U 12  13 U 12  14 U 13 U 12  15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 13 U 12  10 U 13 U 12  11 U 13 U 13  11 U 13 U 13  11 U 13 U 13															
thane 15 U 43 B 14  15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 12  10 U 13 U 12  10 U 13 U 12  11 U 13 U 12  12 U 13 U 12  13 U 13 U 12  14 U 13 U 12  15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 13 U 12  10 U 13 U 12  11 U 13 U 12  12 U 13 U 12  13 U 13 U 12  14 U 13 U 12  15 U 13 U 13  17 U 13  18 U 13 U 12  19 U 13  10 U 13  11 U 13	Compound	BO (6.	1R1-1 -8 ft)	BOR (8-10	1-2 ) ft)	BO (10-	R1-3 12 ft)	BC (3.	BOR2-1 (3-5 ft)	BOJ (5-	BOR2-2 (5-8 ft)	BC (8-	BOR2-3 (8-10 ft)	BO -9)	BOR3-1 (6-8 ft)
thane 15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 13 U 12  11 U 13 U 12  11 U 13 U 12  12 U 13 U 12  14 U 13 U 12  15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 13 U 12  10 U 13  10 U 13  11 U 13  12 U 13  13 U 12  14 U 13  15 U 13 U 12  16 U 13  17 U 13  18 U 12  19 U 13  10 U 12  11 U 13  11 U 13  12 U 13  13 U 12  14 U 13  15 U 13  17 U 13  18 U 12  19 U 13  10 U 13  11 U 13  1	Acetone	15	ח	43	В	14	Ω	1,100	*	40	В	12	n	17	В
e (total)  15	Benzene	15	n	13	Ω	12	n	26	Ď	14	Ω	12	D	13	ņ
15   U   13   U   12     16   U   13   U   12     17   U   13   U   12     18   U   13   U   12     19   U   13   U   12     10   U   13   U   12     10   U   13   U   12     11   U   13   U   12     12   U   13   U   12     13   U   13   U   12     14   U   13   U   12     15   U   13   U   12     16   U   13   U   12     17   U   13   U   12     18   U   13   U   12     19   U   13   U   12     10   U   U   U   U     11   U   U   U   U     12   U   U   U   U     13   U   U   U   U     14   U   U   U     15   U   U   U     16   U   U   U   U     17   U   U   U     18   U   U   U     19   U   U   U     10   U   U   U     11   U   U   U     12   U   U     13   U   U   U     14   U   U   U     15   U   U   U     16   U   U   U     17   U   U     18   U   U   U     19   U   U     10   U   U     11   U   U     12   U   U     13   U   U     14   U   U     15   U   U     16   U   U     17   U     18   U   U     19   U     10   U     11   U   U     12   U     13   U   U     14   U     15   U     15   U     16   U     17   U     18   U   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U	Bromodichloromethane	15	n	13	D	12	Ω	26	* <u>n</u>	14	Ω	12	n	13	n
15   U   13   U   12     16   U   13   U   12     17   U   13   U   12     18   U   13   U   12     19   U   13   U   12     10   U   13   U   12     10   U   13   U   12     10   U   13   U   12     11   U   13   U   12     12   U   13   U   12     13   U   13   U   12     14   U   13   U   12     15   U   13   U   12     16   U   13   U   12     17   U   13   U   12     18   U   13   U   12     19   U   13   U   12     10   U   U   U     11   U   U   U     12   U   U   U   U     13   U   U   U     14   U   U   U     15   U   U   U     16   U   U   U     17   U   U   U     18   U   U   U     19   U   U   U     10   U   U   U     11   U   U   U     12   U   U   U     13   U   U   U     14   U   U   U     15   U   U   U     16   U   U   U     17   U   U     18   U   U   U     19   U   U     10   U   U     11   U   U     12   U   U     13   U   U     14   U   U     15   U   U     16   U   U     17   U   U     18   U   U     19   U     10   U     11   U   U     12   U     13   U   U     14   U     15   U     15   U     16   U     17   U     18   U   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     15   U     16   U     17   U     18   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U	Bromoform	15	Ω	13	n	12	Ω	26	*n	14	Ω	12	Ω	13	U
ide 15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 13 U 12  19 U 13 U 12  10 Opene 15 U 13 U 12  10 U 13 U 12  10 U 13 U 12  11 U 13 U 12  12 U 13 U 12  13 U 13 U 12  14 U 13 U 12  15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 13 U 12  10 U 13 U 12  11 U 13 U 12  11 U 13 U 13  11 U 13 U 13  11 U 13 U 13	Bromomethane	15	Ω	13	n	12	Ω	26	* <u></u>	14	Ω	12	n	13	ר
ide 15 U 13 U 12 12 14 15 U 1	2-Butanone	15	Ω	13	Ω	12	ב	150	*	14	n	12	n	13	Þ
ide 15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 13 U 12  10 U 13 U 12  10 U 13 U 12  10 U 13 U 12  11 U 13 U 12  12 U 13 U 12  15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 13 U 12  10 U 13 U 12  11 U 13 U 12  12 U 13 U 12  13 U 13 U 12  14 U 13 U 12  15 U 13 U 12  16 U 13 U 12  17 U 13 U 12  18 U 13 U 12  19 U 13 U 12  10 U 13 U 12  11 U 13	Carbon disulfide	15	Ω	13	D	12	Ω	26	*N	14	Þ	12	n	7	_
15   U   13   U   12     16   U   13   U   12     17   U   13   U   12     18   U   13   U   12     19   U   13   U   12     10   U   U   U     10   U   U   U     11   U   U   U     12   U   U   U     13   U   U   U     14   U   U   U     15   U   U   U     16   U   U   U     17   U   U     18   U   U   U     19   U   U     10   U     11   U   U     12   U   U     13   U   U     14   U   U     15   U   U     16   U   U     17   U     18   U   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     18   U     19   U     10   U     11     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U	Carbon tetrachloride	15	Ω	13	Ω	12	n	26	n*	4	n	12	Ω	13	D
15   U   13   U   12     16   U   13   U   12     17   U   13   U   12     18   U   13   U   12     19   U   13   U   12     10   U   13   U   12     11   U   13   U   12     12   U   13   U   12     13   U   13   U     14   U   13   U     15   U   13   U     15   U   13   U     15   U   13   U     16   U   13   U     17   U     18   U   13   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     18   U     18   U     19   U     10   U     11     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     18   U     18   U     18   U     18   U     19   U     10   U     11     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U	Chlorobenzene	15	D	13	Ω	12	Ω	26	*n	14	n	12	n	13	D
15   U   13   U   12     e                                 e	Chloroethane	15	n	13	Ω	12	Ω	26	*1	14	D	12	Ω	13	Ω
HS U H3 U  e H5 U H3 U  e (total) H5 U H3 U  e (total) H5 U H3 U  opene H5 U H3 U  opene H5 U H3 U  opene H5 U H3 U  one H5 U H3 U  one H5 U H3 U  othane H5 U H3 U  f	Chloroform	15	Ω	13	n	12	Ω	26	*N	14	Þ	12	D	13	Ω
e (total) 15 U 13 U 19 U e 15 U 13 U 19	Chloromethane	15	ח	13	Ω	12	n	26	*n	4	ລ	12	n	13	Þ
e (total)   15   U   13   U   U   U   U   U   U   U   U   U	Dibromochloromethane	15	n	13	Ω	12	D	26	*n	4	n	12	ח	13	n
e (total)   15   U   13   U   U   U   U   U   U   U   U   U	1,1-Dichloroethane	15	D	13	Ω	12	D	26	*1	14	D	12	Ω	13	Ω
e (total) 15 U 13 U 13 U 15 O 15 U 13 U 15 U 15 U 13 U 15 U 15 U 15 U	1,2-Dichloroethane	15	Ω	13	Ω	12	n	26	*0	14	n	12	Ω	18	
e (total) 15 U 13 U 13 U 15 U 15 U 13 U 15 U 13 U 15 U 13 U 15 U 15	1,1-Dichloroethene	15	n	13	D	12	Ω	26	* <u></u>	14	Ω	12	Ω	13	Ω
opene 15 U 13 U 19	1,2-Dichloroethene (total)	15	Ω	13	Ω	12	Ω	26	<u>*</u> 5	14	n	12	n	89	
opene 15 U 13 U 1  Interprete 15 U 1	1,2-Dichloropropane	15	Ω	13	Ω	12	Ω	56	*5	14	D	12	n	13	Ω
Propene	cis-1,3-Dichloropropene	15	Ω	13	n	12	n	26	*1	14	n	12	n	13	n
one 15 U 13 U 1 1	trans-1,3-Dichloropropene	1.5	n	13	n	12	Ω	26	*1	14	Ω	12	Ω	13	ח
one 15 U 13 U 1  e 15 U 13 U 1  octhane 15 U 13 U 1  octhane 15 U 13 U 1  IS U 13 U 1	Ethylbenzene	15	Ω	13	Ω	12	Ω	26	*0	14	Ω	12	Ω	13	D
e 15 U 13 U 1 1 Octaviane 15 U 13 U 1 1 Octaviane 15 U 13 U 1 I Octaviane 15 U 13 U 1 I Octaviane 15 U 13 U 1 U 1 U 1 Octaviane 15 U 13 U 1 U 1 U 1 U 1 U 1 U I Octaviane 15 U 13 U 1 U 1 U 1 U I Octaviane 15 U 13 U 1 U 1 U 1 U 1 U I Octaviane 15 U 13 U 1 U 1 U I U I U I U I U I U I U I U I	2-Hexanone	15	ם	13	n	12	D	26	*11	14	Ω	12	n	13	⊃
e 15 U 13 U 1 octhane 15 U 13 U 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	4-Methyl-2-pentanone	15	D	13	Ω	12	n	26	*0	14	Þ	12	Ω	13	D
15 U 13 U 1  16 U 13 U 1	Methylene chloride	15	n	13	Ω	12	D	26	*n	14	n	12	Ω	13	Ω
between 15 U 13 U 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	Styrene	1.5	ກ	13	n	12	D	26	*n	14	n	12	n	13	Ω
15 U 13 U 1 15 U 15 U 15 U 15 U 15 U 15	1,1,2,2-Tetrachloroethane	15	D	13	ت ت	7	D	52	* <u>∩</u>	14	Ω	12	Ü	13	$\supset$
15 U 13 U 1 Chloroethane 15 U 13 U 1	Tetrachloroethene	15	n	13	Ω	2	n	26	<u>*</u> 5	4	Þ	12	n	13	$\supset$
15 U 13 U	Toluene	15	Ω	13	n	12	Ω	150	٠	20		12	n	13	D
	1,1,1-Trichloroethane	1.5	n n	13	Ω	12	Þ	26	* <u></u>	14	Ω	12	ü	13	Ö

TABLE C.7a (Cont.)

					Sedime	nt Concen	trations (	Sediment Concentrations (µg/kg) at Various Depth Intervals	arious D	epth Inter	vals			
Compound	BC (6	BOR1-1 (6-8 ft)	BOI (8-1	BOR 1-2 (8-10 ft)	BO (10	BOR1-3 (10-12 ft)	BC (3	BOR2-1 (3-5 ft)	B(	BOR2-2 (5-8 ft)	) B(	BOR2-3 (8-10 ft)	BOJ	BOR3-1 (6-8 ft)
1,1,2-Trichloroethane	15	Ω	13	Ω	12	Ω	26	* *	4	=	15	=	12	=
Trichloroethene	15	Ω	13	Ω	12	Ξ	36	· *	: :	) [	3 5	<b>)</b> ;	CT (E)	)
Vinyl chloride	15	n	13	1	2	• =	25	) <u>*</u>	‡ :	<b>&gt;</b> ;	71 :	<b>)</b> ;	0.1	
Xylenes (total)	15	- 12	13	) =	2 2	) <u>=</u>	07 0	; • :	4 :	⊃ ;	12	<b>)</b>	13	D
		, İ			17		07	  -  -	4	0	12	n	13	٦į
				Sedimen	t Concen	trations (μ	ıg/kg) at	Sediment Concentrations (µg/kg) at Various Depth Intervals	pth Inter	vals				
	BO	BOR3-2	BOR4-1	4-1	BO	BOR4-2	BO	BOR5-1	2	BOR5-2	BODS 3	2 2	ı	
Compound	(8-1	(8-10 ft)	(5-8 ft)	3	(8-1	(8-10 ft)	(2-	(2-4 ft)	₹ 4	(4-6 ft)	(6-8 ft)	£ (£		
Acetone	12	U	20	Ω	19	B*	12	Ω	9	<u>~</u>	2	=		
Benzene	12	Ω	20	n	13	*n	12	n	12	i 🗅	7 2	> =		
Bromodichloromethane	12	Ω	20	ח	13	*n	12	n	12	· 1	2 2	· =		
Bromoform	12	Ω	20	ņ	13	*0	12	Ω	12	, ,	2 2	) <u>=</u>		
Bromomethane	12	Ω	20	Þ	13	*	12	D	12	'n	12	) <b>=</b>		
2-Butanone	12	Ω	20	Ω	13	* <u>n</u>	12	Ω	12	Ω	12	· =		
Carbon disulfide	12	n	20	Ω	13	*	12	n	12	· D	12	) =		
Carbon tetrachloride	12	Ω	20	n	13	*h	12	D	12	n	: 2	· =		
Chlorobenzene	12	Ω	20	*	13	*n	12	Ω	12	ָ ב	: 2	> =		
Chloroethane	12	Ω	20	Ω	13	*n	12	ם	12	n	12	) <u>=</u>		
Chloroform	12	Ω	20	Ω	13	*0	12	Ω	12	ם	2	· =		
Chloromethane	12	Ω	20	n	13	*n	12	Ω	12	ם י	2	· =		
Dibromochloromethane	12	Ω	20	Ω	13	* <u></u>	12	Ω	12	D	12	· =		
I, I-Dichloroethane	12	n	20	Ω	13	<u>*</u>	12	n	12	· D	: 2	) =		
1,2-Dichloroethane	12	Ω	20	Ω	13	* <u>n</u>	12	n	12	'n	2	) <u>=</u>		
1,1-Dichloroethene	12	Ω	20	Ω	13	*n	12	Ω	12	ב	12	) <u> </u>		
1,2-Dichloroethene (total)	12	Ω	2,200		360	*	12	n	12	n D	12	ם מ		
1,2-Dichloropropane	12	⊃	20	n	13	*n	12	Ω	12	n	12	Ω		
cis-1,3-Dichloropropene	12	Ω	20	Ω	13	*5	12	n	12	Ω	12	ņ		
trans-1,3-Dichloropropene	12	Ω	20	Ω	13	<u>*</u>	12	Ω	12	n N	12	n		

TABLE C.7a (Cont.)

BOR3-2   BOR4-1   BOR					Sedimen	t Conce.	Sediment Concentrations (µg/kg) at Various Depth Intervals	µg/kg) at	Various	Depth Inte	ervals			1	
12 U 20 U*  13 U 20 U*  oride 12 U 20 U*  lorocthane 31 09 *  ene 12 U 20 U*  lorocthane 12 U 20 U*  lorocthane 10 J 76  e 91 12 U 20 U*  lorocthane 10 J 6  e 12 U 20 U*  lorocthane 10 J 6  e 12 U 20 U*  lorocthane 12 U 20 U+  lorocthane 12 U 12 U+  lo	Compound	B(	DR3-2 10 ft)	BOI 5-5)	3.ft.)	BC (8-	0R4-2 10 ft)	B(	BOR5-1 (2-4 ft)	B V	BOR5-2 (4-6 ft)	BG (c	BOR5-3 (6-8 ft)	1	
ntanone 12 U 20 U* oride 12 U 20 U* oride 12 U 20 U 4 12 U 20 U 4 13 U 20 U 4 14 U 20 U 4 15 U 20 U 4 16 U 20 U 4 17 U 20 U 4 18 U 20 U 4 19 U 20 U 4 10 U 20 U 4 11 U 20 U 4 11 U 20 U 4 12 U 20 U 4 14 U 12 U 10 15 U 12 U 10 16 U 12 U 10 17 U 12 U 10 18 U 12 U 10 18 U 12 U 10 19 U 12 U 10 10 U 12 U 10 11 U 12 U 10 11 U 11 U 11 11 U 11 U 11 11 U 11 U	zene	12	Ω	20	*	13	*n	12	Ω	12	2 U	12	n		
ntanone 12 U 20 U*  loride 12 U 20 U 1*  lorocthane 31 0 20 U*  lorocthane 12 U 20 U*  lorocthane 12 U 20 U*  lorocthane 10 J 76  e 91 12 U 20 U*  lorocthane 10 J 0 1/  lorocthane 12 U 12 U+  lorocthane 12 U 1	one	12	D	20	*	13	*n	12	n	12	2 U	12	n		
oride 12 U 20 U*  lorocthane 31 99 **  ene 12 U 20 U*  12 U 20 U*  12 U 20 U*  12 U 20 U*  12 U 54 U  12 U 12 U  13 U 12 U  14 U 12 U  15 U 15 U  16 U 12 U  17 U 12 U  18 U 12 U  19 U 12 U  10 U 12 U  11 U 12 U  11 U 12 U  12 U 12 U  13 U 12 U  14 U  15 U 12 U  16 U  17 U 12 U  18 U  18 U  19 U  10 U  10 U  11 U  11 U  11 U  12 U  13 U  14 U  15 U  16 U  17 U  18 U  18 U  19 U  19 U  10 U  10 U  11 U  11 U  11 U  12 U  13 U  14 U  15 U  16 U	1-2-pentanone	12	Ω	20	*A	13	*n	12	Ü	12	2 U	12	Ŋ		
12   U   20   U*     13   U   20   U*     12   U   20   U*     12   U   20   U*     13   U   20   U*     14   U   20   U*     15   U   20   U*     16   U   20   U*     17   U   20   U*     18   U   20   U*     19   U   20   U*     10   U   12   U*     11   U   12   U*     12   U   12   U*     13   U   12   U*     14   U   U*     15   U   U*     16   U   U*     17   U   U*     18   U   U*     19   U   U*     10   U   U*     11   U   U*     12   U   U*     13   U   U*     14   U   U*     15   U   U*     16   U   U*     17   U   U*     18   U   U*     19   U   U*     10   U   U*     11   U   U*     12   U   U*     13   U   U*     14   U   U*     15   U   U*     15   U   U*     16   U   U*     17   U   U*     18   U   U*     19   U   U*     10   U   U*     11   U   U*     12   U   U*     13   U   U*     14   U   U*     15   U   U*     15   U   U*     16   U   U*     17   U   U*     18   U   U*     19   U*     10   U*     11   U   U*     12   U   U*     13   U   U*     14   U*     15   U   U*     16   U   U*     17   U   U*     18   U   U*     19   U*     10   U*     11   U   U*     12   U   U*     13   U   U*     14   U*     15   U   U*     16   U   U*     17   U   U*     18   U   U*     19   U*     10   U*     11   U*     12   U   U*     13   U   U*     14   U*     15   U   U*     16   U*     17   U*     18   U*     19   U*     10   U*     11   U*     12   U*     13   U*     14   U*     15   U*     16   U*     17   U*     18   U*     19   U*     10   U*     10   U*     11   U*     12   U*     13   U*     14   U*     15   U*     15   U*     16   U*     17   U*     18   U*     19   U*     10   U*     11   U*     12   U*     13   U*     14   U*     15   U*     15   U*     16   U*     17   U*     18   U*     18   U*     19   U*     10   U*     11   U*     12   U*     13   U*     14   U*     15   U*     15   U*     16   U*     17   U*     18   U*     18   U*     19   U*     10   U*     11   U*     12   U*     13   U*     14   U*     15   U*     15   U*     16   U*     17   U*     18   U*     18   U*     19   U*     1	ne chloride	12	Ω	20	ח	13	*n	12	Ω	12	2 U	12	Ω		
12   U   20   U*     12   U   20   U*     13   U   20   U*     14   U   20   U*     15   U   20   U*     16   J   76     17   U   54     18   U   54     19   U   54     10   U   12   U     11   U   12   U     12   U   12   U     13   U   12   U     14   U   12   U     15   U   12   U     16   U   12   U     17   U   U     18   U   U     19   U   U     10   U     11   U   U     12   U   U     13   U   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     15   U     16   U     17   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     15   U     16   U     17   U     18   U     19   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     19   U     19   U     10   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     19   U     10   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     16   U     17   U     18   U     18   U     18   U     19   U     10   U     11   U     12   U     13   U     14   U     15   U     15   U     15   U     16   U     17   U     18   U		12	D	20	*	13	*n	12	n	12	D 2	12	n		
ethane 12 U 20 U*  lu J 0 J 76  ethane 10 J 76  ethane 10 J 76  lu J 0 J 10  lu J 0	etrachloroethane	31		66	*	13	* <u>n</u>	12	D	12	n a	12	ם		
tethane 12 U 20 U*  ethane 10 J 76  e 91 12 J  12 U 54  12 U 54  12 U 54  12 U 64  12 U 12 U+  13 U 12 U+  14 U 12 U+  15 U 12 U+  16 U 12 U+  17 U 12 U+  18 U 12 U+  19 U 12 U+  10 U+  11 U 12 U+  11 U 12 U+  12 U 12 U+  13 U 12 U+  14 U 12 U+  15 U 12 U+  16 U+  17 U 12 U+  18 U+  18 U 12 U+  19 U+  19 U+  10 U+  10 U+  11 U+  12 U+  13 U+  14 U+  15 U+  16 U+  17 U+  18 U+  18 U+  19 U+	proethene	12	n	20	* <u></u>	13	*n	12	Ω	12	n a	12	n		
ethane 12 U 20 U 6 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		12	D	20	*n	13	*	12	Ω	12	n î	12	n		
be ethane 10 J 76  12 U 54  12 U 54  12 U 64  12 U 104  BOR5-4 BOR5-4Dup  (8-10 ft) (8-10 ft)  12 U 12 U+  13 U 12 U+  14 U 12 U+  15 U 12 U+  16 U 12 U+  17 U 12 U+  18 U 12 U+  19 U+  10 U+  11 U 12 U+  11 U 12 U+  12 U 12 U+  13 U 12 U+  14 U+  15 U 12 U+  16 U+  17 U 12 U+  18 U 12 U+  18 U 12 U+  19 U+	chloroethane	12	Ω	20	n	13	<b>*</b>	12	n	12	ם	12	n		
BORS-4 BORS-4Dup (8-10 ft) (8-10 ft)  BORS-4 BORS-4Dup (8-10 ft) (8-10 ft)  12 U 12 U+ 13 U 12 U+ 14 U+ 15 U 12 U+ 16 U+ 17 U+ 18 U+ 18 U+ 19 U+ 19 U+ 19 U+ 19 U+	chloroethane	10	_	76		13	<u>*</u>	12	n	12	n	12	Ω		
12 U 54  12 U 20 U*  BOR5-4 BOR5-4Dup  (8-10 ft) (8-10 ft)  12 U 12 U+  13 U 12 U+  14 U 12 U+  15 U 15 U+  16 U+  17 U 17 U+  18 U 18 U+  19 U+  10 U+  11 U+  12 U+  13 U+  14 U+  15 U+	ethene	16		12	-	13	*n	12	D	12	n	12	b		
Dound (8-10 ft) 20 U*  BOR5-4 BOR5-4Dup (8-10 ft) (8-10 ft)  12 U 12 U+ 13 U 12 U+ 14 U+ 15 U 12 U+ 16 U+ 17 U+ 18 U+ 18 U+ 18 U+ 19 U+ 19 U+ 19 U+	oride	12	n	54		13	*î	12	Ω	12	n	12	n		
BOR5-4 BOR5-4Dup (8-10 ft) (8-10 ft)  12 U 12 U+ 13 U 12 U+ 14 U+ 15 U 12 U+	(total)	12	ם ا	20	*	13	*n	12	ח	12	n i	12	n		
BORS-4 BORS-4Dup  (8-10 ft) (8-10 ft)  12 U 12 U+ 1  13 U 12 U+ 1  14 U 12 U+ 1  15 U 12 U+ 1  16 U 12 U+ 1  17 U 12 U+ 1  18 U 12 U+ 1  18 U 12 U+ 1						Sedimer	nt Concent	rations (µ	ıg/kg) at	Various E	epth Inte	rvals		:	
methane (8-10 ft) (8-10 ft)  12 U 12 U+ 1  13 U 12 U+ 1  14 U+ 1  15 U 12 U+ 1		BO	R5-4	BORS	-4Dup	BO	R6-1	BO	BOR6-2	ă	BOR6-3	B	BOR6-4	BO	BOR7-1
12   U   12   U+   12   U+   12   U+   12   U+   13   U+   13   U+   12   U+   U+   U+   U+   U+   U+   U+   U	Compound	(8-	10 ft)	(8-1	0 ft)	(2-	4 ft)	(4)	(4-6 ft)		(tj 8-9)	8)	(8-10 ft)	(2)	(2-4 ft)
12 U 12 U+ 3  methane 12 U 12 U+ 12  14 U 12 U+ 12  15 U 12 U+ 12  16 U 12 U+ 12  17 U 12 U+ 12  18 U 12 U+ 12  19 U 12 U+ 12		12	Ω	12	Ť,	12	Ω	12	ņ	12	D	12	n	14	Ω
methane 12 U 12 U+ 12 12 13 14 15 15 16 17 18 17 18 17 18 17 18 17 18 17 18 17 18 18 18 18 18 18 18 18 18 18 18 18 18		12	מ	12	τ <b>+</b> Ω	8	<u>_</u>	12	Ω	4	J	12	D	14	Ω
12 U 12 U+ 12 12 U 12 U+ 12 13 U 12 U+ 12 14 U 12 U+ 12 oride 12 U 12 U+ 12 15 U 12 U+ 12 17 U 12 U+ 12 18 U 12 U+ 12 19 U 12 U+ 12	thloromethane	12	⊃	12	1 1	12	Ω	12	Ω	12	Þ	12	D	14	D
12 U 12 U+ 12  12 U 12 U+ 12  13 U 12 U+ 12  oride 12 U 12 U+ 12  13 U 12 U+ 12  14 U 12 U+ 12  15 U 12 U+ 12	E	12	n	12	t n	12	Ω	12	n	12	n	12	ם	14	D
12 U 12 U+ 12  oride 12 U 12 U+ 12  oride 12 U 12 U+ 12  12 U 12 U+ 12  13 U 12 U+ 12	thane	12	n	12	r C	12	Ω	12	ח	12	n	12	Ω	14	⊃
oride 12 U 12 U+ 12  oride 12 U 12 U+ 12  12 U 12 U+ 12  13 U 12 U+ 12	າເ	12	Ω	12	Ú+	2	n	12	ם	12	n	12	ב	4	D
oride 12 U 12 U+ 12 12 12 13 14 15 15 15 15 15 15 15 15 15 15 15 15 15	sulfide	12	n	12	Ċ+	12	D	12	D	12	Ω	12	n	4	Э
12 U 12 U+ 12 12 U 12 U+ 12 13 U 15 U+ 15	trachloride	12	n	12	+ 1	2	D	12	ב	12	Ω	12	n	14	כ
c 12 U 12 U+ 12	ızene	<u></u>	n	12	+ 1	12	Ω	12	Ω	12	ב	12	n	14	ב
	anc	12	n	<u>.</u>	+11	<u>C1</u>	n	12	D	12	כ	12	n	14	ລ
U 12 U+ 12	ε	12	Ω	12	+11	2	Ω	13	□	12	n	12	ם	17	Ω

TABLE C.7a (Cont.)

Compound	BC (%	BOR5-4 (8-10 ft)	BOR5 (8-1	BOR5-4Dup (8-10 ft)	BOI (2-	BOR6-1 (2-4 ft)	BC (4-	BOR6-2 (4-6 ft)	BC (6.	BOR6-3 (6-8 ft)	BC -8)	BOR6-4 (8-10 ft)	BO (2-	BOR7-1 (2-4 ft)
Chloromethane	12	n	12	#	2	=	2	=	2	-	2	;		
2. h	•	. ;	1		1	)	71	5	71	)	71	<b>-</b>	14	$\supset$
Dibromochloromethane	12	n	12	÷	12	ב	12	Ω	12	D	12	Ω	14	n
I, I-Dichloroethane	12	Ω	12	n+	12	n	12	Ω	12	D	12	D	14	
1,2-Dichloroethane	12	n	12	tn	12	n	12	Ω	12	D	12	ı D	. 4	) <u>=</u>
i, I-Dichloroethene	12	Ω	12	n+	12	U	12	Ω	12	ם	12	ם מ	. 4	) =
,2-Dichloroethene (total)	12	n	12	Λ+	12	U	12	n	12	n	12	D	4	) =
1,2-Dichloropropane	12	D	12	C+	12	n	12	ם	12	n	12	D	. 4	· =
cis-1,3-Dichloropropene	12	D	12	Û+	12	Ω	12	n	12	n	12	n	. 4	=
trans-1,3-Dichloropropene	12	Ω	12	r O+	12	Ω	12	Ŋ	12	n	12	n	14	· =
Ethylbenzene	12	n	12	Û+	12	Ω	12	Ω	12	ח	12	D	4	) =
2-Hexanone	12	Ω	12	n+	12	Ω	12	Ω	12	n	12	۵	4	
4-Methyl-2-pentanone	12	Ω	12	^+	12	Ω	12	Ω	12	n	12	n	. 41	ì
Methylene chloride	12	Ω	12	†n	12	Ω	12	n	12	D	12	n	4	n
Styrene	12	Ω	12	n+	12	n	12	Ŋ	12	Ω	12	U	14	
1,1,2,2-Tetrachloroethane	12	Ω	12	r U+	12	Ω	12	מ	12	n	12	Ω	14	) =
Tetrachloroethene	12	Ω	12	U+	12	Ω	12	n	12	n	12	n	. 4	· =
Toluene	12	ח	12	n+	12	Ω	12	Ω	12	Ω	12	Ω	. 4	) =
1,1,1-Trichloroethane	12	ב	12	Ü+	12	Ω	12	D	12	n	12	n	4	
1,1,2-Trichloroethane	12	Ω	12	n+	12	n	12	U	12	Ω	12	D	4	=
Trichloroethene	12	Ω	12	n+	12	Ω	12	Ω	12	Ω	12	D	4	
Vinyl chloride	12	n	12	n+	12	Ω	12	Ω	12	Ω	12	Ω	14	n
Xylenes (total)	2	1	2	;										

TABLE C.7a (Cont.)

Compound	BOR7-2 (4-6 ft)	BOR7-3 (6-8 ft)	8	BOR7-4 (8-10 ft)	4 £	BOR8-1 (0-2 ft)	8-1 ft)	BOR8-2 (2-4 ft)	8-2 ft)	BOR8-3 (4-6 ft)	° 2	BOR8-3Dup (4-6 ft)	3Dup ft)	BO (6-	BOR8-4 (6-8 ft)	BOR8-5 (8-10 ft)	:8-5 ) ft)
Acetone	12 U	12	n	1,100	В	20	В	12	Ω	17	В	23	æ	20	8	88	æ
Benzene	12 U	12	Ω	12	Ω	13	Ω	12	Ω	12	n	12	Ω	12	: =	2	=
Bromodichloromethane	12 U	12	Ω	12	Ω	13	Ω	12	Ω	12	n	12	) D	12	) <b>=</b>	: 2	) =
Bromoform	12 U	12	Ω	12	n	13	Ω	12	n	12	n	12	D	1 2	) D	7	) =
Bromomethane	12 U	12	D	12	n	13	Ω	12	Ω	12	U	12	n	12	ם י	12	: =
2-Butanone	12 U	12	D	180		13	U	12	n	12	U	12	ם	12	ם ה	12	) 🗅
Carbon disulfide	12 U	12	n	12	n	13	n	12	Ω	12	n	12	Ω	12	ם י	12	n
Carbon tetrachloride	12 U	12	D	12	Ω	13	Ω	12	Ω	12	Ω	12	ם	12	n	12	D
Chlorobenzene	12 U	12	Ω	12	n	13	Ω	12	Ω	12	n	12	n	12	D	12	n
Chloroethane	12 U	12 1	ם	12	Ω	13	n	12	n	12	n	12	n	12	Ω	12	n
Chloroform	12 U	12	ם	12	Ω	13	D	12	n	12	Ω	12	Þ	12	D	12	D
Chloromethane	12 U	12	n	12	Ω	13	n	12	n	12	Ω	12	Ω	12	D	12	Ω
Dibromochloromethane	12 U	12	ב	12	n	13	n	12	U	12	n	12	n	12	Ω	12	Ω
1,1-Dichloroethane	12 U	12		12	Ω	13	Ω	12	n	12	Ω	12	D	12	Ω	12	D
1,2-Dichloroethane	12 U	12	.D	12	n	13	Ω	12	n	12	n	12	D	12	Ω	12	þ
1,1-Dichloroethene	12 U	12 [		12	Ω	13	Ω	12	n	12	n	12	n	12	D	12	D
1,2-Dichloroethene (total)	12 U	12 (	,	12	Ω	13	Ω	12	Ω	12	Ω	12	n	12	n	12	ם
1,2-Dichloropropane	12 U	12 1	J	12	Ω	13	Ω	12	Ω	12	n	12	n	12	n	12	n
cis-1,3-Dichloropropene	12 U	12 [	J	12	Ω	12	Ω	12	Ω	12	n	12	Ω	12	ם	12	n
trans-1,3-Dichloropropene	12 U	12 [	_	12	Ω	13	Ω	12	Ω	12	Ω	12	Ω	12	Ω	12	Ω
Ethylbenzene	12 U	12 (	)	12	Ω	13	n	12	n	12	Ω	12	Ω	12	Ω	12	Ω
2-Hexanone	12 U	12 (	_	12	Ω	13	D	12	U	12	n	12	n	12	ם	12	D
4-Methyl-2-pentanone	12 U	12 L		12	Ω	13	D	12	n	12	n	12	n	12	n	12	Ω
Methylene chloride	12 U	12 L	Ţ	12	Ω	13	Ω	12	Ω	12	ב	12	Ω	12	n	12	Ω
Styrene	12 U	12 C	_	12	Ω	13	D	12	n	12	n	12	Ω	12	Ω	12	Ω
1.1,2,2-Tetrachloroethane	12 U	12 C	-	12	ņ	13	ח	7	n	12	n	<u>~</u>	n	12	n	7	ם
Tetrachloroethene	12 U	12 C	_	12	Ω	13	Ω	<u>~~~</u>	n	12	ם	12	D	12	n	2	n
Тоlиепе	12 U	12 U	-	12	n	13	Ω	2	n	7	n	13	Ω	12	Ω	12	Ω

TABLE C.7a (Cont.)

					Sediment	Concent	rations (p	Sediment Concentrations (µg/kg) at Various Depth Intervals	ious Depth	Interval						
Compound	BOR7-2 (4-6 ft)		BOR7-3 (6-8 ft)	BOR7-4 (8-10 ft)	t) 4	BOR8-1 (0-2 ft)	Q -1	BOR8-2 (2-4 ft)	BOR8-3 (4-6 ft)	(8-3	BOR8-3Dup (4-6 ft)	3Dup ft)		BOR8-4 (6-8 ft)	BOR8-5 (8-10 ft)	8-5 (ft)
													1			
1,1,1-Trichloroethane	12 (	J E	2 U	12	Ω	13	n	12 U	12	Ω	2	=	12	=	12	=
1.1.2-Trichloroethane	1 21	ì		:	;	,	;			ı	ļ	)	ì	)	7	)
	71	-	0 7	71	_	13	D	12 U	12	ם	17	D	12	Ω	12	Ω
Trichloroethene	12 L		2 N	12	Ω	13	Ω	12 U	12	Ω	12	Ξ	12	11		. =
Vinyl chloride	12 [	12	2 U	12	U	13	Þ	12 11	1	· =	2	· =	: :	) <b>:</b>	1 5	<b>)</b>
Xylenes (total)	12 (	=	1	1	=	72	- =			;	1 :	<b>)</b>	71	>	71	<b>D</b>
			,	71		CI	2	12 O	17	<b>&gt;</b>	17	<b>-</b>	12	)	12	n

<sup>a</sup> Sampled by ANL, analyzed by Weston Gulf Coast, CLP/HSL volatile organics.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

B = analyte found in associated blank.

internal standard area counts are outside QC limits.

J = estimated value.

+ = surrogate recoveries were outside QC limit(s).

TABLE C.7b Volatile Organics Analyses for Sediment Borings, June 1995<sup>a</sup>

		- Programma .	Sedim	ent Concentrat	Sediment Concentrations (µg/kg) at 0-6 in.	0-6 in.		
Compound	TPSED 1	TPSED 2	TPSED 3	TPSED 4	TPSED 5	TPSED 6	TPSED 7	TPSED 8
Acetone	61	170	200	120	20 U	22 U	2	14 U
Benzene	32 U	U 71	24 U	24 U	20 U	·	15 U	·
Bromodichloromethane	32 U	17 U	24 U	24 U	20 U			-
Bromoform	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
Bromomethane	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
2-Butanone	32 U	17 U	24 U	200	20 U	22 U	15 U	14 U
Carbon disulfide	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
Carbon tetrachloride	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
Chlorobenzene	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
Chloroethane	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
Chloroform	32 U	17 U	24 U	24 U	20 U	22 U	15 U	
Chloromethane	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
Dibromochloromethane	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
1,1-Dichloroethane	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
1,2-Dichloroethane	32 U	17 U	24 U	24 U	20 U	22 U	к	14 U
1,1-Dichloroethene	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
1,2-Dichloroethene (total)	32 U	72	45	24 U	20 U	22 U	4	14 U
1,2-Dichloropropane	32 U	17 U	24 U	24 U	20 U	22 U	15 U	14 U
cis-1,3-Dichloropropene	32 U	U 71	24 U	24 U	20 U	22 U	15 U	
trans-1,3-Dichloropropene	32 U	I7 U	24 U	24 U	20 U	22 U	15 U	14 U
Ethyl benzene	32 U	U 71	24 U	24 U	20 U	22 U	15 U	14 U
2-Hexanone	32 U	17 U	24 U	24 U	20 U	22 U	ю	14 U
4-Methyl-2-pentanone	32 U	J 71	24 U	24 U	20 U	22 U	2	14 U
Methylene chloride	32 U	5 BJ	24 U	13	6 BJ	13	6	7

TABLE C.7b (Cont.)

					-	Sedime	ent Con	Sediment Concentrations (µg/kg) at 0-6 in.	/gn) suc	kg) at 0	-6 in.					
Compound	TPSED 1		TPSE	TPSED 2	TPSED 3	D 3	TPSED 4	JD 4	TPSED 5	ID 5	TPSED 6	90	TPSED 7	D 7	TPSED 8	8 O
Styrene	32	Ω	17	n	24	n	24	Ω	20	n	22	1	15	Ξ	7	<u>}-</u>
l, 1, 2, 2-Tetrachloroethane	32	Ω	17	U	23	_	24	Ω	20	11	2 2	· =	3 4	)	t =	) <b>:</b>
Tetrachloroethene	32	Ω	17	n	24	n	24	Ω	20	· =	22	) <u> </u>	<u>, , , , , , , , , , , , , , , , , , , </u>	Ξ	<u> </u>	)
Toluene	32	n	2	-	24	D	7	ŧ	) C	) <u> </u>	7 6	) <b>=</b>	. Y	)	<u> </u>	<b>&gt;</b> ;
1,1,1-Trichloroethane	32	Ω	17	Ξ	24	· =	. 7	11	07 6	) <u> </u>	7 6	) ;	2 ;	<b>)</b> ;	<b>1</b>	<b>)</b>
1,1,2-Trichloroethane	32	· =	17	) <u>=</u>	7	) <b>-</b>	† ?	) ;	07	<b>)</b> ;	77	<b>⊃</b> ;	5	<b></b>	4	n
Trichloroethene		) <u> </u>	7 :	) <b>,</b>	† (	٠	47	<b>)</b>	70	<b>D</b>	77	<b>-</b>	6		14	n
	75	<b>-</b>	Ξ		29		24	D	20	Ω	22	Ω	15		14	Ω
Vinyl chloride	32	n	54		18	ĭ	24	n	20	U	22	Ω	15	Ω	14	<u> </u>
Xylenes (total)	32	n	17	Ω	24	Ω	24	Ω	20	Ω	22	n	15	n	. 4	) <u> </u>
													,	)	-	

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

 $<sup>\</sup>mathbf{U}_{i}$  = analyte was analyzed for but not detected; detection limit given.

B = analyte was found in the associated blank.

J = estimated value.

TABLE C.8a Semivolatile Organics Analyses for Sediment Borings, April 1994<sup>a</sup>

			Similarity index coords in (82,484)									
Compound	BOR1-1 (6-8 ft)	3OR1-1 (6-8 ft)	BOR1-2 (8-10 ft)	1-2 ft)	BOR1-3 (10-12 ft)	1-3 2 ft)	BOR2-1 (3-5 ft)	2-1 (c)	BOR2-2 (5-8 ft)	2-2 ft)	BOR2-3 (8-10 ft)	2-3 ff)
-												
Acenaphthene	480	Þ	440	ב	410	Þ	820	D	460	D	400	n
Acenaphthylene	480	n	440	Þ	410	Ω	850	n	460	Ŋ	400	ר
Anthracene	480	n	440	D	410	Ω	850	D	460	Ω	400	$\Box$
Benz[a]anthracene	480	Ω	440	n	410	Ω	850	D	460	Þ	400	ח
Benzo[a]pyrene	480	D	440	D	410	Ω	850	Ω	460	n	400	U
Benzo[b]fluoranthene	480	Ω	440	n	410	D	850	Ω	460	n	400	Ω
Benzo[g,h,i]perylene	480	Ω	440	Ω	410	Ω	850	Ω	460	Ω	400	Ω
Benzo[k]fluoranthene	480	ם	440	n	410	n	850	Ω	460	n	400	
Bis(2-chloroethoxy)methane	480	Ŋ	440	Ω	410	n	850	Ω	460	n	400	ם
Bis(2-chloroethyl)ether	480	n	440	D	410	n	850	D	460	n	400	Ω
Bis(2-chloroisopropyl)ether	480	D	440	Ω	410	D	850	Ω	460	D	400	D
Bis(2-ethylhexyl)phthalate	180	ſ	78	ŗ	140	<b>-</b>	46	_	140	,-,	43	_
4-Bromophenyl phenyl ether	480	Ω	440	D	410	Ω	850	D	460	Þ	400	D
Butylbenzylphthalate	480	n	440	D	410	Ω	850	D	460	Þ	400	Ω
Carbazole	480	D	440	D	410	n	850	ר	460	Ω	400	D
4-Chloro-3-methylphenol	480	n	440	n	410	Ω	850	Þ	460	n	400	Ω
4-Chloroaniline	480	Ω	440	n	410	Ω	850	Ω	460	Ω	400	Ω
2-Chloronaphthalene	480	U	440	כ	410	Ω	850	Ω	460	Ω	400	n
2-Chlorophenol	480	Ŋ	440	D	410	Ω	850	Ω	460	n	400	D
4-Chlorophenyl phenyl ether	480	מ	440	n	410	Ω	850	Ω	460	D	400	D
Chrysene	480	D	440	Ω	410	Þ	850	n	460	Ω	400	ח
Di-n-butylphthalate	150	ſ	130	-	120	Ţ	260	Ţ	140	<b></b> -,	130	_
Di-n-octyl phthalate	480	n	440	D	410	n	850	Ω	460	D	400	n
Dibenz[a,h]anthracene	480	ח	440	Þ	410	Ω	850	Þ	460	ב	400	n
Dibenzofuran	480	٦	440	Ω	410	n	850	ב	460	D	400	n
1,2-Dichlorobenzene	480	n	440	ח	410	n	850	n	460	n	400	Ω
1,3-Dichlorobenzene	480	ח	440	n	410	ב	850	D	460	ח	400	⊃
1,4-Dichlorobenzene	480	Ξ	440	$\Box$	410	ח	850	n	460	n	400	ב

TABLE C.8a (Cont.)

Compaind	BO	BOR1-1	BOR1-2	7.3	BOR1-3	1-3	BOR2-1	12-1	BOR2-2	2-2	BOR2-3	2-3
nunduna	0)	0.10	(8-10 II)		(10-12 ft)	(E)	(3-5 ft)	3	(5-8 ft)	<b>a</b>	(8-10 ft)	<b>E</b>
3,3'-Dichlorobenzidine	480	Ω	440	Þ	410	Ω	. 850	1	460	Ξ	700	1
2,4-Dichlorophenol	480	Ω	440	D	410	Ξ	850	) <u>=</u>	960	) <u>=</u>	9	) :
Diethylphthalate	480	D	440	· =	410	) =	050	) <u>:</u>	904	) :	400	) 
Dimethylphthalate	480	Ξ	440	) =	21.7	) :	0.00	);	400	⊃ ;	400	>
2.4-Dimethylphenol	180	ב כ		<b>)</b> ;	014	) :	820	$\supset$	460	D	400	$\supset$
4 6. Dinitro 2 mothylphonol	1004	) :	440	<b>)</b>	410	D	820	כ	460	D	400	
.,o-Dinido-2-riichiyiphenol	1,200	⊃	1,100	n	1,000	Ŋ	2,100	Ω	1,200	ם	1,000	$\Box$
2,4-Uinitrophenol	1,200	n	1,100	Ω	1,000	n	2,100	D	1,200	Þ	1,000	D
2,4-Dinitrotoluene	480	n	440	D	410	Ω	850	n	460	n	400	
2,6-Dinitrotoluene	480	Ω	440	n	410	n	850		460	Ω	400	
Fluoranthene	480	n	440	n	410	n	850	D	460	D	400	· =
Fluorene	480	D	440	ח	410	Ω	850	Ω	460	Þ	400	
Hexachlorobenzene	480	Ω	440	Ω	410	n	850	Ω	460	ח	400	) =
Hexachlorobutadiene	480	Ω	440	Ω	410	n	850	D	460	=	400	=
Hexachlorocyclopentadiene	480	Ω	440	Ω	410	n	850	Þ	460	n	400	) <u>=</u>
Hexachloroethane	480	Ω	440	n	410	n	850	ב	460	'n	400	) <u>=</u>
Indeno[1,2,3-c,d]pyrene	480	Ω	440	Ω	410	U	850	Ω	460	=	400	) =
sophorone	480	n	440	D	410	n	850	n	460	· =	400	) <u>=</u>
2-Methylnaphthalene	480	Ω	440	Ω	410	Þ	850	D	460	· 🗀	400	) <u> </u>
2-Methylphenol	480	Ω	440	Ω	410	n	850	n	460	n	400	=
4-Methylphenol	480	Ω	440	n	410	n	850	ב	460	'n	400	· =
Naphthalene	480	Ω	440	D	410	Ω	850	n	460	) =	400	) <u>_</u>
2-Nitroaniline	1,200	Ω	1,100	n	1,000	Ω	2,100	ם	1,200	) D	1.000	<b>=</b>
3-Nitroaniline	1,200	Ω	1,100	ם	1,000	Ω	2,100	n	1,200	n	1.000	
4-Nitroaniline	1,200	n	1,100	D	1,000	Ω	2,100	D	1,200	D	1.000	=
Nitrobenzene	480	Ω	440	b	410	n	850	n	460	Þ	400	Ξ
2-Nitrophenol	480	n	440 L	ר	410	U	850	Ω	460	Þ	400	) =
4-Nitrophenol	1,200	n	1,100 L	Ŋ	1,000	Ω	2,100	Ω	1,200	ב	1.000	
N-Nitroso-di-n-propylamine	480	11	,	,								,

TABLE C.8a (Cont.)

			Sculli	ICIII COI	icenii ano	ns (µg/k	g) at vari	ons De	Securical Concentrations (pg/kg) at various Depth Intervals	SIIIS		
Compound	BOH (6-8	BOR 1-1 (6-8 ft)	BOR1-2 (8-10 ft)	1-2 ) ft)	BOR1-3 (10-12 ft)	1-3 2 ft)	BOR2-1 (3-5 ft)	2-1 ft)	BOR2-2 (5-8 ft)	2-2 ft)	BOR2-3 (8-10 ft)	£-3
N-Nitrosodiphenylamine	480	n	440	Ω	410	n	850	n	460	n	400	Ω
Pentachlorophenol	1,200	Ω	1,100	D	1,000	Ω	2,100	Ω	1,200	n	1,000	n
Phenanthrene	480	D	440	Ω	410	D	850	Ω	460	n	400	Ω
Phenol	480	D	440	n	410	Ω	850	Ω	460	Ω	53	_
Pyrene	480	n	440	Ω	410	Ω	850	n	460	n	400	D
1,2,4-Trichlorobenzene	480	D	440	Ω	410	U	850	n	460	b	400	Ω
2,4,5-Trichlorophenol	1,200	ר	1,100	Ω	1,000	D	2,100	n	1,200	n	1,000	Ω
2,4,6-Trichlorophenol	480	ח	440	D	410	D	850	n	460	n	400	Ω
	BOR3-I	3-1	BOR3-2	3-2	BOR4-1	+1	BOR4-2	1-2	BOR5-1	5-1	BOR5-2	2-2
Compound	(t) 8-9)	£)	(8-10 ft)	£)	(5-8 ft)	(£)	(8-10 ft)	£ (£	(2-4 ft)	£ (	(4-6 ft)	اج :
Acenaphthene	420	n	390	n	099	n	440	n	400	Ω	400	Ω
Acenaphthylene	420	n	390	n	099	Ω	440	Þ	400	D	400	Ω
Anthracene	420	D	390	n	099	Ω	440	n	400	D	400	n
Benz[a]anthracene	420	n	390	D	099	D	440	D	400	n	400	n
Benzo[a]pyrene	420	Ω	390	n	099	n	440	n	400	Ω	400	D
Benzo[b]fluoranthene	420	Ω	390	D	099	n	440	n	400	Ω	400	n
Benzo[g,h,i]perylene	420	D	390	n	099	Ω	440	Ω	400	Ω	400	D
Benzo[k]fluoranthene	420	D	390	D	099	n	440	Ŋ	400	ם	400	n
Bis(2-chloroethoxy)methane	420	n	390	ח	099	n	440	ב	400	n	400	ח
Bis(2-chloroethy1)ether	420	Ω	390	D	099	n	440	n	400	ח	400	D
Bis(2-chloroisopropyl)ether	420	⊃	390	n	099	Þ	440	ח	400	n	400	Ω
Bis(2-ethylhexyl)phthalate	420	D	44	n	099	n	72	•••	400	כ	400	$\supset$
4-Bromophenyl-phenylether	420	D	390	Ω	099	D	440	ח	400	n	400	$\supset$
Butylbenzylphthalate	420	D	390	ב	099	Ω	440	D	400	n	400	n

TABLE C.8a (Cont.)

			Sedin	nent Co	ncentratio	/gн) su	kg) at Vari	ous De	Sediment Concentrations (µg/kg) at Various Depth Intervals	slı	į	
Compound	BO (6-9)	BOR3-1 (6-8 ft)	BOR3-2 (8-10 ft)	:3-2 ) ft)	BOR4-1 (5-8 ft)	4-1 fi	BOR4-2 (8-10 ft)	.4-2	BOR5-1	5-1	BOR5-2	5-2
									7		0-+)	
Carbazole	420	Ω	390	'n	099	Ω	440	Ω	400	D	400	D
4-Chloro-3-methylphenol	420	Ω	390	Ω	099	D	440	n	400	Ω	400	n
4-Chloroaniline	420	n	390	n	099	n	440	Ω	400	n	400	D
2-Chloronaphthalene	420	n	390	n	099	D	440	n	400	Ω	400	
2-Chlorophenol	420	D	390	n	099	n	440	Ω	400	כ	400	1
4-Chlorophenyl phenyl ether	420	Ω	390	n	099	Ω	440	n	400	D	400	· D
Chrysene	420	D	390	'n	099	D	440	D	400	n	400	Ω
Di-n-butylphthalate	140	_	130	_	230	_	440	'n	170	~	190	·,
Di-n-octylphthalate	420	Ω	390	Ω	099	Ω	440	n	400	Ω	400	Ω
Dibenz[a,h]anthracene	420	Ω	390	Ω	099	Ω	440	Ω	400	n	400	n
Dibenzofuran	420	n	390	Ω	099	D	440	n	400	D	400	n
1,2-Dichlorobenzene	420	Ω	390	Ω	099	Ω	440	n	400	Ω	400	D
1,3-Dichlorobenzene	420	Ω	390	ם	099	n	440	n	400	Ω	400	n
1,4-Dichlorobenzene	420	n	390	Ω	099	D	440	Ω	400	n	400	D
3,3'-Dichlorobenzidine	420	Ω	390	D	099	D	440	n	400	Ω	400	D
2,4-Dichlorophenol	420	U	390	Ü	099	n	440	Ω	400	Ω	400	n
Diethylphthalate	420	Ω	390	Ŋ	099	Ω	440	Ω	400	Ω	400	n
Dimethyl phthalate	420	Ω	390	Ω	099	Ω	440	Ω	400	מ	400	D
2,4-Dimethylphenol	420	Ω	390	Ω	099	Ω	440	n	400	n	400	Ω
4,6-Dinitro-2-methylphenol	1,000	Ω	086	Ω	1,700	n	1,100	Ω	1,000	n	1,000	Þ
2,4-Dinitrophenol	1,000	Ω	086	n	1,700	n	1,100	Ω	1,000	D	1,000	'n
2,4-Dinitrotoluene	420	Ω	390	n	099	D	440	Ω	400	Ω	400	Ω
2,6-Dinitrotoluene	420	Ω	390	Ω	099	Þ	440	Þ	400	n	400	ם
Fluoranthene	420	Ω	390	Ω	099	Þ	440	n	400	Ω	400	Ω
Fluorene	420	Ω	390	n	099	Ŋ	440	Ω	400	n	400	n
Hexachlorobenzene	420	Ω	390	Ω	099	D	440	Ω	400	n	400	U
Hexachlorobutadiene	420	Ω	390	Ω	099	Ω	440	Ω	400	Ω	400	n
Hexachlorocyclopentadiene	420	Ω	390	n	099	Ω	440	Ω	400	Ω	400	D

TABLE C.8a (Cont.)

Compound	BOR3-1 (6-8 ft)	3-1 £	BOR3-2 (8-10 ft)	3-2	BOR4-1 (5-8 ft)	4-1	BOR4-2	t-2	BOR5-1	5-1	BOR5-2	2-5
						<u>}</u>	(0.10)		1.7	3	0-+)	2
Hexachloroethane	420	Ω	390	n	099	ם	440	D	400	Ω	400	D
Indeno[1,2,3-c,d]pyrene	420	n	390	n	099	n	440	n	400	n	400	D
Isophorone	420	n	390	n	099	Ω	440	n	400	Þ	400	n
2-Methylnaphthalene	420	D	390	D	099	n	440	ם	400	Ω	400	$\supset$
2-Methylphenol	420	Ω	390	n	099	n	440	n	400	Ω	400	$\Gamma$
4-Methylphenol	420	Þ	390	n	099	D	440	Ω	400	Þ	400	U
N-Nitrosodiphenylamine	420	n	390	ח	099	ר	440	D	400	ם	400	ח
N-Nitroso-di-n-propylamine	420	Ω	390	D	099	D	440	n	400	ח	400	D
Naphthalene	420	Ω	390	Ω	099	ח	440	Ω	400	D	400	ב
2-Nitroaniline	1,000	Ω	086	n	1,700	ב	1,100	Þ	1,000	n	1,000	
3-Nitroaniline	1,000	n	086	Ω	1,700	n	1,100	n	1,000	Ω	1,000	n
4-Nitroaniline	1,000	ח	086	Ω	1,700	n	1,100	n	1,000	n	1,000	
Nitrobenzene	420	Ω	390	D	099	D	440	ם	400	Ω	400	D
2-Nitrophenol	420	D	390	n	099	n	440	Þ	400	Ω	400	ב
4-Nitrophenol	1,000	Ω	086	n	1,700	D	1,100	ב	1,000	n	1,000	n
Pentachlorophenoì	1,000	n	086	Þ	1,700	Þ	1,100	n	1,000	ח	1,000	$\Box$
Phenanthrene	420	Ω	390	Ω	099	n	440	Ω	400	b	400	
Phenol	420	Ω	390	Ω	099	n	440	D	400	n	400	n
Pyrene	420	n	390	Þ	099	Ω	440	D	400	Ω	400	
1,2,4-Trichlorobenzenc	420	n	390	n	099	D	440	D	400	Ω	400	n
2,4,5-Trichlorophenol	1,000	Ω	086	Ω	1,700	Þ	1,100	n	1,000	ם	1,000	D
2.4.6-Trichlorophenol	750	-	000	;		:	•	;				

TABLE C.8a (Cont.)

			Sedime	ent Co	centration	пз (µg/kg	) at Vario	ous De	Sediment Concentrations (µg/kg) at Various Depth Intervals	ls		
Compound	BO (6-	BOR5-3 (6-8 ft)	BOR5-4 (8-10 ft)	£.5	BOR5-4Dup (8-10 ft)	4Dup ft)	BOR6-1 (2-4 ft)	5-1 ft)	BOR6-2 (4-6 ft)	6-2 ft)	BOR6-3 (6-8 ft)	6-3 ft)
Acenaphthene	400	n	390	Þ	400	Ω	410	Ü	400	۵	004	=
Acenaphthylene	400	Ω	390	Ω	400	Ω	410	Ω	400	'n	400	ם כ
Anthracene	400	U	390	n	400	n	410	n	400	Ω	400	n
Benz[a]anthracene	400	Ω	390	Ω	400	Ω	410	D	400	D	400	ם
Benzo[a]pyrene	400	Ŋ	390	D	400	n	410	D	400	Ω	400	· D
Benzo[b]fluoranthene	400	U	390	n	400	Ω	410	Ω	400	n	400	Þ
Benzo[g,h,i]perylene	400	n	390	Ω	400	Ω	410	D	400	n	400	Þ
Benzo[k]fluoranthene	400	D	390	Ω	400	Ω	410	Ŋ	400	n	400	Ω
Bis(2-chloroethoxy)methane	400	n	390	Ω	400	Ω	410	Ω	400	Ω	400	Ω
Bis(2-chloroethyl)ether	400	Ω	390	Ω	400	D	410	D	400	Ω	400	Ω
Bis(2-chloroisopropyl)ether	400	ב	390	Ω	400	D	410	D	400	Ω	400	D
Bis(2-ethylhexyl)phthalate	400	Ω	99	ſ	41	'n	410	Ω	400	Ω	400	n
4-Bromophenyl phenyl ether	400	Ω	390	n	400	Ω	410	Ω	400	Ω	400	n
Butylbenzylphthalate	400	Ω	390	Ω	400	Ω	410	D	400	n	400	D
Carbazole	400	Ω	390	n	400	Ω	410	D	400	n	400	D
4-Chloro-3-methylphenol	400	D	390	n	400	Ω	410	Ω	400	Ω	400	U
4-Chloroaniline	400	Ω	390	Ω	400	n	410	Ω	400	D	400	Ω
2-Chloronaphthalene	400	n	390	Ω	400	Ŋ	410	n	400	n	400	Ω
2-Chlorophenol	400	Ω	390	Ω	400	n	410	ם	400	U	400	n
4-Chlorophenyl phenyl ether	400	n	390	Ω	400	Ω	410	Ω	400	U	400	Ω
Chrysene	400	Ω	390	Ω	400	n	410	n	400	n	400	n
Di-n-butylphthalate	170	_	160	'n	140	ſ	170	ŗ	140	-	86	_
Di-n-octylphthalate	400	n	390	Ω	400	Ω	410	Ω	400	n	400	Ω
Dibenz[a,h]anthracene	400	Ω	390	Ω	400	n	410	n	400	Ω	400	Ω
Dibenzofuran	400	D	390	n	400	Ω	410	n	400	D	400	Ω

TABLE C.8a (Cont.)

Compound	BOR5-3 (6-8 ft)	.ft)	BOR5-4 (8-10 ft)	5.4 ff)	BOR5-4Dup (8-10 ft)	4Dup ft)	BOR6-1 (2-4 ft)	6-1 ft)	BOR6-2 (4-6 ft)	5-2 ft)	BOR6-3 (6-8 ft)	6-3
1,2-Dichlorobenzene	400	n	390	n	400	Ω	410	D	400	Ω	400	Ω
1,3-Dichlorobenzene	400	Þ	390	Ω	400	n	410	ם	400	D	400	
1,4-Dichlorobenzene	400	n	390	D	400	Ω	410	n	400	Ω	400	D
3,3'-Dichlorobenzidine	400	n	390	Ω	400	Þ	410	n	400	ם	400	Ω
2,4-Dichlorophenol	400	D	390	D	400	Ŋ	410	n	400	Ω	400	D
Diethylphthalate	400	ם	390	n	400	Ω	410	D	400	D	400	Ω
Dimethylphthalate	400	n	390	Ω	400	Ω	410	Ω	400	ח	400	n
2,4-Dimethylphenol	400	Ω	390	Ω	400	ב	410	D	400	n	400	Ω
4,6-Dinitro-2-methylphenol	1,000	Ω	086	Ω	066	Ω	1,000	D	1,000	Ω	1,000	$\supset$
2,4-Dinitrophenol	1,000	Ω	086	D	066	Ω	1,000	n	1,000	D	1,000	
2,4-Dinitrotoluene	400	Ω	390	כ	400	Ω	410	n	400	ח	400	D
2,6-Dinitrotoluene	400	Ω	390	Ω	400	Ω	410	ם	400	n	400	D
Fluoranthene	400	Ω	390	Ω	400	n	410	D	400	D	400	n
Fluorene	400	D	390	Ω	400	D	410	D	400	D	400	כ
Hexachlorobenzene	400	n	390	n	400	n	410	n	400	Ω	400	n
Hexachlorobutadiene	400	D	390	n	400	Ω	410	Ω	400	D	400	Ω
Hexachlorocyclopentadiene	400	D	390	n	400	n	410	Ω	400	n	400	Ω
Hexachloroethane	400	Ω	390	D	400	Ω	410	Ω	400	D	400	D
Indeno[1,2,3-c,d]pyrene	400	n	390	Ω	400	Ŋ	410	n	400	D	400	$\supset$
Isophorone	400	D	390	Ω	400	D	410	D	400	Ω	400	D
2-Methylnaphthalene	400	n	390	Ω	400	D	410	D	400	n	400	D
2-Methylphenol	400	Ω	390	Ω	400	n	410	n	400	Ω	400	ר
4-Methylphenol	400	D	390	n	400	ח	410	ח	400	n	400	Ω
Naphthalenc	400	n	390	n	400	D	410	Ω	400	Þ	400	ר
2-Nitroaniline	1,000	n	086	Ω	066	n	1,000	n	1,000	n	1,000	$\Box$
3-Nitroaniline	1,000	Ω	086	n	066	D	1,000	Ω	1,000	Ω	1,000	ח
4-Nitroaniline	1,000	<u> </u>	086	Ω	066	⊃	1,000	⊃	1,000	ت ت	1,000	ב
Nitrobenzene	400	Ξ	390	Ξ	400	Ξ	410	Ξ	400	-	400	=

TABLE C.8a (Cont.)

		Sediment (	Sediment Concentrations (µg/kg) at Various Depth Intervals	(µg/kg) :	ut Variou	s Depth	Interval	s		
Compound	BOR5-3 (6-8 ft)	BOR5-4 (8-10 ft)	BOR5-4Dup (8-10 ft)	dnC ()	BOR6-1 (2-4 ft)	- 1	BOR6-2 (4-6 ft)	2-7	BOR6-3 (6-8 ft)	6-3 ft)
2-Nitrophenol	400 U	n 068	400	n	410	Ω	400	۵	400	=
4-Nitrophenol	1,000 U	n 086	066	מ	000,1	n	1.000	) D	1.000	=
N-Nitroso-di-n-propylamine	400 U	390 U	400	Ω	410	Ŋ	400	ì	400	=
N-Nitrosodiphenylamine	400 U	390 U	400	n	410	Ω	400	ם	400	) <u></u>
Pentachlorophenol	1,000 U	O86	066	D D	000,	Ω	1,000	ם	1,000	n
Phenanthrene	400 U	390 U	400	Ω	410	Ω	400	D	400	n
Phenol	400 U	390 U	400	Ŋ	70	ſ	400	D	400	ר
Pyrene	400 U	390 U	400	Ŋ	410	D	400	D	400	Þ
1,2,4-Trichlorobenzene	400 U	390 U	400	U	410	Ω	400	Ω	400	Ω
2,4,5-Trichlorophenol	1,000 U	O86	066	U I	000,1	Ω	000		1.000	· =
2,4,6-Trichlorophenol	400 U	390 U	400	Ω	410	n	400	י ב	400	
	BOR6-4	BOR7-1	BOR7-2	_,	BOR7-3		BOR7-4	4	BOR8-1	
Compound	(8-10 ft)	(2-4 ft)	(4-6 ft)		(6-8 ft)		(8-10 ft)	2	(0-2 ft)	اء
Acenaphthene	390 U	380 J	390	Ω	400	D	400	n	420	Ξ
Acenaphthylene	390 U	130 J	390	Ω	400	D	400	· D	420	) D
Anthracene	390 U	280 J	390	Ω	400	ם	400	D	420	
Benz[a]anthracene	390 U	1,200	390	Ω	110	_	400	Þ	420	· =
Benzo[a]pyrene	390 U	006	390	D	. 9/	_	400	n	420	) <u>=</u>
Benzo[b]fluoranthene	390 U	1,700	390	ם	150	_	400	Ω	420	
Benzo[g,h,i]perylene	390 U	490	390	D	46		400	Ω	420	n
Benzo[k]fluoranthene	390 U	700	390	ם	56		400	Ω	420	ם
Bis(2-chloroethoxy)methane	390 U	460 U	390	D.	400	_	400	Ω	420	n
Bis(2-chloroethyl)ether	390 U	460 U	390	ם	400	D	400	Ω	420	n
Bis(2-chloroisopropyl)ether	390 U	460 U	390	ט.	400	_	400	n	420	n
Bis(2-ethylhexyl)phthalate	390 U	1,100	230 J	_	400	_	400	Ω	420	
								,	ļ	

TABLE C.8a (Cont.)

Compound	BO (8-]	BOR6-4 (8-10 ft)	BOR7-1 (2-4 ft)	(7-1 (ft)	BOR7-2 (4-6 ft)	7-2 ft)	BOR7-3 (6-8 ft)	7-3 ft)	BOR7-4 (8-10 ft)	7-4 ) ft)	BOR8-1 (0-2 ft)	8-1 ft)
4-Bromophenyl phenyl ether	390	Ω	460	ב	390	n	400	ח	400	n	420	D
Butylbenzylphthalate	390	Ω	460	n	390	Ω	400	n	400	D	420	n
Carbazole	390	D	120	,,	390	Ω	400	n	400	n	420	n
4-Chloro-3-methylphenol	390	D	460	n	390	Ω	400	Ω	400	Ω	420	ח
4-Chloroaniline	390	D	460	Ω	390	n	400	Π	400	n	420	n
2-Chloronaphthalene	390	Ω	460	Ω	390	Ω	400	Ŋ	400	ב	420	n
2-Chlorophenol	390	n	460	n	390	n	400	n	400	ח	420	Ω
4-Chlorophenyl phenyl ether	390	Ω	460	Ω	390	n	400	n	400	D	420	Þ
Chrysene	390	Ω	1,200		390	Þ	120	_	400	Ω	420	Þ
Di-n-butylphthalate	100	-	120	Г	110	-	110	ĭ	86	<b>-</b> 5	110	_
Di-n-octylphthalate	390	Ω	460	Ω	390	n	400	n	400	D	420	Ω
Dibenz[a,h]anthracene	390	n	150	_	390	Ŋ	400	D	400	U	420	ב
Dibenzofuran	390	n	310	-	390	Þ	400	n	400	Ω	420	n
1,2-Dichlorobenzene	390	n	460	Ω	390	D	400	D	400	Ŋ	420	Þ
1,3-Dichlorobenzene	390	n	460	Ω	390	U	400	n	400	U	420	D
1,4-Dichlorobenzene	390	Ω	460	Ω	390	n	400	Þ	400	D	420	ר
3,3'-Dichlorobenzidine	390	D	460	Ŋ	390	n	400	D	400	n	420	Þ
2,4-Dichlorophenol	390	Ω	460	n	390	D	400	ņ	400	D	420	ם
Diethylphthalate	390	D	460	'n	390	n	400	כ	400	n	420	D
Dimethylphthalate	390	D	460	Ω	390	Ω	400	n	400	D	420	n
2,4-Dimethylphenol	390	n	460	Ω	390	Ω	400	n	400	Þ	420	כ
4,6-Dinitro-2-methylphenol	096	D	1,200	Ω	096	n	1,000	n	1,000	ם	1,000	ב
2,4-Dinitrophenol	096	n	1,200	n	096	D	1,000	ב	1,000	n	1,000	b
2,4-Dinitrotoluene	390	ם	460	Ω	390	Ω	400	n	400	D	420	ם
2,6-Dinitrotoluene	390	ם	460	D	390	n	400	n	400	n	420	ם
Fluoranthene	390	コ	2,300		390	ח	180	_	400	D	420	ם
Fluorenc	390	ב	410	ī	390	Ω	400	⊃	400	n	420	Þ
Hexachlorobenzene	390	ב	460	ח	390	ם	400	n	400	⊃	420	D

TABLE C.8a (Cont.)

	ВО	BOR6-4	BOR7-1	7-1	BOR7-2	7-2	BOR7-3	7-3	ROR7-4	7-4	BOBS	2
Compound	(-8)	(8-10 ft)	(2-4 ft)	<b>a</b>	(4-6 ft)	£	(6-8 ft)	3	(8-10 ft)	[≘]	(0-2 ft)	ft)
Hexachlorobutadiene	390	Ω	460	n	390	D	400	n	400	Ξ	420	
Hexachlorocyclopentadiene	390	Ω	460	Ω	390	D	400	n	400	· =	420	) =
Hexachloroethane	390	Ω	460	n	390	n	400	) =	400	) <u>=</u>	024	) <u>:</u>
Indeno[1,2,3-c,d]pyrene	390	D	610		390	Ω	58	, –	400	) <u>=</u>	420	) <u> </u>
Isophorone	390	Ω	460	מ	390	n	400	n	400	) =	420	) <u>-</u>
2-Methylnaphthalene	390	Ω	210	Ţ	390	D	400	Ω	400	=	420	) =
2-Methylphenol	390	n	460	n	390	n	400	n	400	> =	420	> =
4-Methylphenol	390	n	460	n	390	Ω	400	'n	400	> =	420	=
Naphthalene	390	ם	460	Ω	390	Ω	400	n	400	) =	420	) <u>=</u>
2-Nitroaniline	096	n	1,200	n	096	n	1,000	D	1.000	· =	1 000	) =
3-Nitroaniline	096	n	1,200	Ω	096	Ω	1,000	D	1.000	) =	1 000	-
4-Nitroaniline	096	Ω	1,200	ח	096	Ω	1,000		1,000	- =	0001	-
Nitrobenzene	390	Ω	460	Ω	390	Ω	400	n	400	) D	420	) <u>=</u>
2-Nitrophenol	390	n	460	Ω	390	ח	400	n	400	n	420	- =
4-Nitrophenol	096	Ω	1,200	Þ	096	n	1,000	n	1,000	n	1.000	=
N-Nitroso-di-n-propylamine	390	Ω	460	n	390	D	400	Þ	400	· =	420	· =
N-Nitrosodiphenylamine	390	D	460	n	390	Þ	400	ח	400	· =	420	=
Pentachlorophenol	096	Ω	1,200	Ω	096	n	1.000		000 1	) <u>=</u>	1 000	2 =
Phenanthrene	390	n	1,400		390	Ŋ	400	) 🗅	400	> =	420	) <u>=</u>
Phenol	390	Ω	460	Þ	390	Ω	400	n	400	> =	420	) <u>=</u>
Pyrene	390	D.	1,800		390	D	160	ń	400	n	420	=
i,2,4-Trichlorobenzene	390	Ω	460	n	390	U	400	n	400	Ω	420	> =
2,4,5-Trichlorophenol	096	Ω	1,200	Ω	096	Ω	1,000	n	1,000	n	1.000	) <u></u>
2,4,6-Trichlorophenol	390	D	460	D	390	Ω	400	1		* }		:

TABLE C.8a (Cont.)

Compound	BOJ (2-4	BOR8-2 (2-4 ft)	BOR8-3 (4-6 ft)	3-3 ft)	BOR8-3Dup (4-6 ft)	3Dup ft)	BOR8-4 (6-8 ft)	£ 5	BOR8-5 (8-10 ft)	3-5 ft)
Acanonhihana	01.8	<u> </u>	3	:		,		;		
	410	)	400	)	400	$\supset$	390	)	410	Þ
Acenaphthylene	410	n	400	n	400	Ω	390	D	410	ח
Anthracene	410	n	400	n	400	n	390	Ω	410	Ω
Benz[a]anthracene	410	D	400	D	400	Ω	390	D	410	Ω
Benzo[a]pyrene	410	Ω	400	D	400	n	390	Ω	410	D
Benzo[b]fluoranthene	410	n	400	Ω	400	Ω	390	D	410	n
Benzo[g,h,i]perylene	410	D	400	n	400	Ω	390	Ω	410	ח
Benzo[k]fluoranthene	410	n	400	Ω	400	Ω	390	Ω	410	n
Bis(2-chloroethoxy)methane	410	Ω	400	Ω	400	n	390	Þ	410	n
Bis(2-chloroethyl)ether	410	n	400	Ω	400	D	390	D	410	n
Bis(2-chloroisopropyl)ether	410	D	400	D	400	Ŋ	390	n	410	n
Bis(2-ethylhexyl)phthalate	410	n	400	'n	400	Ω	50	_	410	מ
4-Bromophenyl phenylether	410	Ω	400	Ω	400	Ω	390	Ω	410	n
Butylbenzylphthalate	410	n	400	n	400	n	390	D	410	D
Carbazole	410	Ω	400	n	400	n	390	n	410	n
4-Chloro-3-methylphenol	410	Ŋ	400	D	400	Ω	390	n	410	Ω
4-Chloroaniline	410	Ω	400	D	400	n	390	D	410	Ω
2-Chloronaphthalene	410	Ω	400	$\supset$	400	Ω	390	D	410	Ω
2-Chlorophenol	410	n	400	n	400	D	390	ם	410	Ω
4-Chlorophenyl phenyl ether	410	Ω	400	n	400	Þ	390	n	410	Ω
Chrysene	410	Ŋ	400	n	400	n	390	n	410	Ω
Di-n-butylphthalate	110	<b>-</b>	150	<u>-</u>	200		390	n	19	ſ
Di-n-octylphthalate	410	ח	400	D	400	)	390	D	410	ח
Dibenz[a,h]anthracene	410	n	400	b	400	n	390	Ω	410	Ω
Dibenzofuran	410	n	400	n	400	Ω	390	n	410	n
1,2-Dichlorobenzene	410	n	400	n	400	n	390	n	410	Ω
1,3-Dichlorobenzene	410	n	400	n	400	Ω	390	n	410	D
1,4-Dichlorobenzene	410	D	400	כ	400	n	390	Ω	410	n

TABLE C.8a (Cont.)

Compound	BOJ	BOR8-2 (2-4 ft)	BOR8-3 (4-6 ft)	8-3 ft)	BOR8-3Dup (4-6 ft)	3Dup ft)	BOR8-4 (6-8 ft)	£ 4	BOR8-5 (8-10 ft)	8-5 ift)
3,3'-Dichlorobenzidine	410	n	400	n	400	n	390	D	410	Þ
2,4-Dichlorophenol	410	n	400	n	400	n	390	Ω	410	n
Diethylphthalate	410	n	400	n	400	Ω	390	Ω	410	n
Dimethylphthalate	410	Ω	400	n	400	Ω	390	Ω	410	n
2,4-Dimethylphenol	410	Ω	400	n	400	Þ	390	Ω	410	
4,6-Dinitro-2-methylphenol	1,000	ח	1,000	n	1,000	Ω	086	Ω	1,000	n
2,4-Dinitrophenol	1,000	Ω	1,000	Ω	1,000	Ω	086	Ω	1,000	Ω
2,4-Dinitrotoluene	410	Ω	400	D	400	Ω	390	n	410	D
2,6-Dinitrotoluene	410	n	400	n	400	Ω	390	Þ	410	ם
Fluoranthene	410	Ω	400	Ω	400	D	390	Þ	410	n
Fluorene	410	Ω	400	Ω	400	n	390	Þ	410	D
Hexachlorobenzene	410	Ω	400	Ω	400	n	390	D	410	D
Hexachlorobutadiene	410	Ω	400	n	400	n	390	D	410	n
Hexachlorocyclopentadiene	410	Ω	400	ח	400	Ω	390	n	410	n
Hexachloroethane	410	Ω	400	ח	400	n	390	Ω	410	D
Indcno[1,2,3-c,d]pyrene	410	Ω	400	Ω	400	Ω	390	n	410	n
Isophorone	410	D	400	Ω	400	n	390	n	410	Ω
2-Methylnaphthalene	410	b	400	n	400	Ω	390	Ω	410	Ω
2-Methylphenol	410	Ω	400	D	400	Ω	390	Ω	410	n
4-Methylphenol	410	Ω	400	Ω	400	b	390	Ŋ	410	n
Naphthalene	410	Ω	400	Ω	400	Ω	390	n	410	n
2-Nitroaniline	1,000	D	1,000	Ω	1,000	n	086	n	1,000	Þ
3-Nitroaniline	1,000	n	1,000	n	1,000	Ω	086	n	1,000	D
4-Nitroaniline	1,000	Ω	1,000	Ω	1,000	Ω	086	Ω	1,000	Þ
Nitrobenzene	410	D	400	Ω	400	Ω	390	D	410	n
2-Nitrophenol	410	D	400	D	400	Ω	390	Ω	410	Ω
4-Nitrophenol	1,000	n	1,000	n	1,000	D	086	ם	1,000	n

TABLE C.8a (Cont.)

	ROP8_2	6-8	BOD9 3	,	DOD6 35		2000	•	Ç	ţ
Compound	(2-4 ft)	£ (2)	(4-6 ft)	(a)	(4-6 ft)	(t)	(6-8 ft)	1 e	(8-10 ft)	Ç €
N-Nitroso-di-n-propylamine	410	n	400	ם	400	n	390	n	410	Ω
N-Nitrosodiphenylamine	410	n	400	D	400	Ω	390	D	410	Ω
Pentachlorophenol	1,000	n	1,000	n	1,000	n	086	n	1,000	Ω
Phenanthrene	410	D	400	ņ	400	D	390	n	410	Ω
Phenol	410	n	400	D	400	Ω	390	Ω	410	Ω
Pyrene	410	D	400	n	400	Ω	390	ם	410	Ω
1,2,4-Trichlorobenzene	410	Þ	400	D	400	n	390	ם	410	Ω
2,4,5-Trichlorophenol	1,000	n	1,000	D	1,000	Ω	086	Ŋ	1,000	Ω
2,4,6-Trichlorophenol	410		400		400	=	300	11	917	=

<sup>&</sup>lt;sup>a</sup> Samples by ANL, analyzed by Weston Gulf Coast, CLP/HSL semivolatile organics.

J = indicates an estimated value.

U = analyte was analyzed for but not detected; detection limit given.

TABLE C.8b Semivolatile Organics Analyses for Sediment Borings, June 1995<sup>a</sup>

			Sedin	ment Co	oncentratio	ns (µ	g/kg) at (	)-6 in.		
Compound	TPS	ED 2	TPS	ED 3	TPSE	ED 4	TPSE	ED 5	TPSI	ED 7
Acenaphthene	558	U	805	U	805	U	674	U	485	U
Acenaphthylene	558	U	805	U	805	U	674	U	485	U
Anthracene	558	U	805	U	805	Ū	674	U	485	U
Benz(a)anthracene	558	U	805	U	805	U	674	U	485	U
Benzo(a)pyrene	558	U	805	U	805	U	674	U	485	U
Benzo(b)fluoranthene	558	U	805	U	805	U	674	U	485	U
Benzo(g,h,i)perylene	558	U	805	U	805	U	674	U	485	U
Benzo(k)fluoranthene	558	U	805	U	805	U	674	U	485	Ü
Bis(2-chloroethoxy)methane	558	U	805	U	805	U	674	U	485	U
Bis(2-chloroethyl)ether	558	U	805	Ū	805	U	674	U	485	บ
Bis(2-ethylhexyl)phthalate	558	U	805	U	805	U	674	U	485	U
4-Bromophenyl phenyl ether	558	U	805	U	805	U	674	U	485	U
Butylbenzylphthalate	558	U	805	U	805	U	674	U	485	U
Carbazole	558	U	805	U	805	U	674	U	485	U
4-Chloro-3-methylphenol	558	U	805	Ū	805	U	674	U	485	U
4-Chloroaniline	558	U	805	U	805	U	674	U	485	U
2-Chloronaphthalene	558	U	805	U	805	U	674	U	485	U
2-Chlorophenol	558	U	805	Ū	805	U	674	U	485	U
4-Chlorophenyl phenyl ether	558	U	805	U	805	U	674	U	485	U
Chrysene	558	U	805	U	805	U	674	U	485	U
Di-n-butylphthalate	558	U	805	U	805	U	674	U	485	U
Di-n-octylphthalate	558	U	805	U	805	U	674	U	485	U
Dibenz(a,h)anthracene	558	U	805	U	805	U	674	U	485	U
Dibenzofuran	558	U	805	U	805	U	674	U	485	U
1,2-Dichlorobenzene	558	U	805	U	805	U	674	U	485	U
1,3-Dichlorobenzene	558	U	805	U	805	U	674	U	485	U
1,4-Dichlorobenzene	558	U	89	J	805	U	674	U	485	U
3,3'-Dichlorobenzidine	558	U	805	U	805	U	674	U	485	
2,4-Dichlorophenol	558	U	805	U	805	U	674	U	485	U U
Diethyl phthalate	558	U	805	U	805	U	674	U	485	U
Dimethyl phthalate	558	U	805	U	805	U	674	U	485	U
2,4-Dimethylphenol	558	U	805	U	805	U	674	U	485	U
4,6-Dinitro-2-methylphenol	1,400	U	2,020	U	2,020	U	1,690	U	1,220	
2,4-Dinitrophenol	1,400	U	2,020	U	2,020	U	1,690	U	1,220	U U
2,4-Dinitrotoluene	558	U	805	U	805	U	674	U	485	
2,6-Dinitrotoluene	558	U	805	U	805	U	674	U	485	U U

TABLE C.8b (Cont.)

			Sedin	nent C	oncentration	ıs (µg	g/kg) at 0	-6 in.		
Compound	TPSE	D 2	TPSI	ED 3	TPSE	D 4	TPSE	D 5	TPSE	D 7
Fluoranthene	61	J	805	U	805	U	674	U	485	U
Fluorene	558	U	805	U	805	U	674	U	485	U
Hexachlorobenzene	84	J	805	U	805	U	674	U	485	U
Hexachlorobutadiene	558	U	805	U	805	U	674	U	485	U
Hexachlorocyclopentadiene	558	U	805	U	805	U	674	U	485	U
Hexachloroethane	558	U	805	U	805	U	674	U	485	U
Indeno(1,2,3-cd)pyrene	558	U	805	U	805	U	674	U	485	U
Isophorone	558	U	805	U	805	U	674	Ū	485	U
2-Methylnaphthalene	558	U	805	U	805	U	674	Ū	485	U
2-Methylphenol	558	U	805	U	805	U	674	U	485	U
4-Methylphenol	558	U	805	U	805	U	674	U	485	U
Naphthalene	558	U	805	U	805	U	674	U	485	U
2-Nitroaniline	1,400	U	2,020	U	2,020	U	1,690	U	1,220	U
3-Nitroaniline	1,400	U	2,020	U	2,020	U	1,690	U	1,220	U
4-Nitroaniline	1,400	U	2,020	U	2,020	U	1,690	U	1,220	U
Nitrobenzene	558	U	805	U	805	U	674	U	485	U
2-Nitrophenol	558	U	805	U	805	U	674	U	485	U
4-Nitrophenol	1,400	U	2,020	U	2,020	U	1,690	U	1,220	U
N-Nitroso-di-n-propylamine	558	U	805	U	805	U	674	U	485	Ŭ
N-Nitrosodiphenylamine	558	U	805	U	805	U	674	U	485	U
2,2'-Oxybis-(1-chloropropane)	558	U	805	U	805	U	674	U	485	U
Pentachlorophenol	1,400	U	2,020	U	2,020	U	1,690	U	1.220	U
Phenanthrene	558	U	805	U	805	U	674	U	485	U
Phenol	558	U	805	U	805	U	674	U	485	U
Pyrene	64	J	805	U	805	U	674	U	485	U
1,2,4-Trichlorobenzene	558	U	805	U	805	U	674	U	485	U
2,4,5-Trichlorophenol	1,400	U	2,020	U	2,020	U	1,690	U	1,220	U
2,4,6-Trichlorophenol	558	U	805	U	805	U	674	U	485	U

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

U = analyte was analyzed for but not detected; detection limit given.

J = indicates an estimated value.

TABLE C.9a Inorganic Analyses for Sediment Borings, April 1994<sup>a</sup>

Dominion	BC	BORI-1	BOF	BOR1-2	BO	BOR1-3	BO	BOR2-1	BOI	BOR2-2	BOF	BOR2-3	BOI	BOR3-1	BOR 3-2	23.2
raraneter	9)	(6-8 ft)	(8-1	(8-10 ft)	(10-	(10-12 ft)	(3-	(3-5 ft)	(5-ì	(5-8 ft)	(8-1	(8-10 ft)	3 <u>-</u> 9	(6-8 ft)	(8-10 ft)	(F)
Metals																
Aluminum	11,800	z	10,100		8,110	z	9.820	z	10,000	Z	6 000	2	000	;		;
Antimony	9	CN	6.2	ND	5.4	NO.	10.5	: 2	, c, coo		066,0	2	0,550	z	4,400	Z
Arsenic	4.4		10.5					5	7.0	<u>.</u>	0.0	NO.	8.	Z O	5.0	S
Barium	17.6		64.6		2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2		† 6 †		0.0		1.4		96.0		0.31	
Rerullium	2.7		P. F. G		55.5		707		78.2		51		26.7		30.6	
D-1-:	70.0	;	0.97		0.36		0.35		0.36		89.0		0.28		0.28	
Cadimum	0.57	n	0.59	D	0.51	D	2.6		0.58	n	0.48	Ω	0.46	n	0.48	
Calcium	296		421		351		1,300		169		471		413		275	,
Chromium	18.4		20		13.5		36.3		14.1		11.9		7.0		7	
Cobalt	5.8		5.1		3.5		4.1		3.2		4 9					
Copper	17.8		18.2		10.8		319		11.4		11 3				7:7	
Iron	14,100		16,500		9,350		14,200		9.750		14 000		017.9		5.5	
Lead	13.1	@ @	20.6		12.3	@ N	318	8	13	@	7.00	N®	01/10		4,330	(
Magnesium	2,200		2,370		1.360		2.030		1 630	;	(27)		0.0	Z (3)	8.3	<b>S</b>
Manganese	55.5		62.1	@	35.8		215		000		000,1		766		973	
Mercury	: -	=		9 1	9.00	;	0.1.0		43.3		53		13.9		34.3	
Nickel	11:0	5	11.0	) )	60:0	<b>D</b>	0.38		0.14	Ω	0.09	Ω	0.09	Ω	0.11	ח
INICACI	17.7		11.3		9.5		15.2		7.2		10.6		3.3		8.1	
Potassium	696		826		669		962		732		602		447		473	
Selenium	0.36		0.28		0.16	Ω	1.0		0.36		0.15	n	0.18	1	0.17	Ξ
Silver	0.58		0.59	Ω	0.51		2.8		0.58	Ω	0.48	Ω	0.46	. =	37.0	) <u>=</u>
Sodium	1,020		801	ш	989		2,280		1,270		738		190	<b>,</b>	2,5	>
Thallium	0.32	Ω	0.33	Ω	0.26	Ω	0.59	Ω	0.31	Ω	0.25	Ξ	0.00		2,0	Ĩ
Vanadium	19.4		22.3		11.4		21		21.4		14.3	)	(m)	)	77.0	>
Zinc	56.9	ш	45.7		60.1	ш	445	ы	31.7	ш	64.4	田	11.9	ш	38.6	田
Other Parameters																
Cyanide (mg/kg)	69:0	n	0.57	U	0.55	Ω	1.2	n	0.69	1	950	Ξ	7.5 0	=	,	:
Colide (m)																

TABLE C.9a (Cont.)

					Commen	Comecun	Comment Commentations (highest at various Deput fillervals	हें) वा ४ वास	us Depui III	ICI V allS				
Parameter	BOR4-1 (5-8 ft)	1-1 (i)	BOR4-2 (8-10 ft)	24-2 5 ft)	BOR5-1 (2-4 ft)	.5-1 ft)	BOF (4-6	BOR5-2 (4-6 ft)	BOR5-3 (6-8 ft)	BOR5-3 (6-8 ft)	BOR5-4 (8-10 ft)	5-4 ) ft)	BOR5 (8-1	BOR5-4Dup (8-10 ft)
Metals														
Aluminum	3,480	z	1,850	z	9,830		12,400		6,780		7,520		080'9	
Antimony	9.0	N D	5.7	N	5.5	N S	4.9	N	5.5	N	5.6	N	5.2	N
Arsenic	6.7		1.9		3.4		3.4		3.2		3.9		2.1	
Barium	34.1		Ξ		50.3		51.8		15.8		36.9		38.5	
Beryllium	0.54		0.42		0.51		0.46		0.17		0.62		0.54	
Cadmium	0.85	Ω	0.54	n	0.52	Ω	0.46	n	0.52	ם	0.53	Ω	0.49	ם
Calcium	1,150		346		503		206		284		258		225	
Chromium	10.3		5.4		12.2		17.6		10.9		12		9.6	
Cobalt	14.4		2.7		7.4		6.6		6.2		5.9		5.4	
Copper	16.2		4.5		5.4		8.9		5.0		7.4		8.9	
Iron	10,600		4,800		10,900		17,400		12,000		12,000		9,480	
Lead	9.19	@ @	7.4	@ Z	11.2		8.7		5.7		10.7		6.2	
Magnesium	604		350		1,620		1,980		1,180		1,360		1,230	
Manganese	40.4		16.8		431	@	152	<b>@</b>	61.8	<b>©</b>	49.6	<b>@</b>	47.4	(9)
Mercury	0.15	n	0.12	Ω	0.08	@ D	0.08	@ N	0.16	@	0.1	00	0.1	U®
Nickel	8.5		3.1		9.4		10.8		10.4		11.7		10.7	
Potassium	402		218		624		925		604		533		556	
Selenium	0.73		0.25		0.28		0.16		0.22		0.19		0.18	Ω
Silver	0.85	n	0.54	Þ	0.52	Ü	0.46	Ω	0.52	Ω	0.49	Ω	1.0	Ω
Sodium	1130		226		592	田	989	ш	588	Э	648	Э	549	ы
Thallium	0.45	Ω	0.3	Ω	0.31	n	0.26	n	0.29	ņ	0.3	Ω	0.3	n
Vanadium	6.3		3.9		17.9		24.4		14.9		15		8.1	
Zinc	193	E	42.9	ш	26.5		32.7		24.8		46		36.6	
Other Parameters														
Cyanide (mg/kg)	0.92	ລ	0.62	Ω	0.58	ם	0.58	ח	9.0	ם	0.59	n	0.59	n
Solide (%)	•													

TABLE C.9a (Cont.)

Parameter	BOR6-1 (2-4 ft)	.(6-1 ft)	BOR6-2 (4-6 ft)	(6-2 ft)	BO (6-1	BOR6-3 (6-8 ft)	BO  (8-1	BOR6-4 (8-10 ft)	BO (2-	BOR7-1 (2-4 ft)	BOI (4-6	BOR7-2 (4-6 ft)	BOI (6-{	BOR7-3 (6-8 ft)
Metals														
Aluminum	9,180		7,680		5,990		10,100		6.430		7.530		0969	
Antimony	5.9	ND	5.6	N	5.3	N5	5.1	N C	6.3	Z	5.0	2	0,700	NI
Arsenic	3.1		3.8		2.1		2.6		1.9		2.2		3.1	Š
Barium	34.1		46.2		22.7		34.9		36.6		35		2.7	
Beryllium	0.37		0.51		0.32		0.62		0.49		0.44		0.73	
Cadmium	0.56	Ω	0.52	Ω	0.5	n	0.48	n	9.0	1	0.47	Ξ	0.55	Ξ
Calcium	427		493		369		614		754	)	\$04	)	478	)
Chromium	11.7		14.6		8.8		17.1		9.5		12.4		14.2	
Cobalt	4.6		5.6		3.5		5.8		3.5		4.9		2:17	
Copper	4.3		8.3		5.0		9.0		5.8		8.5		7. 8	
Iron	12,300		14,600		8,430		12,000		8,570		8.070		11 600	
Lead	6.7		10.3		4.4		7.4		7.7		2.6		10.4	
Magnesium	1,390		1,560		1,200		2,050		1,260		1,160		1.400	
Manganese	113	<b>©</b>	88.4	<b>@</b>	71.9	@	80.3	(9)	37.6	©	20.2	@	33.7	@
Mercury	0.11	N@	0.09	@ N	0.1	@N	0.1	00	0.11	0 0	0.1	11@	0.17	) [6
Nickel	8.8		8.6		9.9		11.9		7.6	1	7.4	)	12.6	) )
Potassium	563		682		457		989		467		462		537	
Selenium	0.2	Ω	0.18	n	0.19	Ω	0.19		0.34		0.18	D	0.26	
Silver	0.56	n	0.52	Ω	0.5	Ω	0.52		9.0	n	0.47	- 12	0.57	
Sodium	825	Э	762	Ξ	571	Э	803	ш	1,290	ъ	066	, п	878	Ţ
Thallium	0.32	Ω	0.29	Ω	0.3	n	0.27	Ω	0.33	n	0.3	ı D	0.31	1 =
Vanadium	16.4		16.4		12.1		17.7		12.6		9.3		19.9	)
Zinc	21.2		23.9		24		43.1		32.5		26.4		46.6	
Other Parameters														
Cyanide (mg/kg)	0.58	ū	0.56	מ	0.54	Ω	0.58		0.68	1	0.57	11	0.61	Ξ
Solide (92)										)		)	70.0	>

TABLE C.9a (Cont.)

	6		1											
Parameter	(8-1	80K/-4 (8-10 ft)	BO (0-	BOR8-1 (0-2 ft)	BOR8-2 (2-4 ft)	t8-2 : ft)	BO (4)	BOR8-3 (4-6 ft)	BOR (4-	BOR8-3Dup (4-6 ft)	BOI (6-1	BOR8-4 (6-8 ft)	B (8)	BOR8-5 (8-10 ft)
Metal														
Aluminum	8,670		9,090		3,690		2,230		2,640		2.530		1.900	
Antimony	5.5	N S	6.1	N	5.2	ND	5.2	ND	5.6	N S	5.8	ND	6.0	N
Arsenic	3.1		3.4		3.5		1.7		1.7		3.5		2.7	
Barium	35.5		88.3		15.5		9.4		11.5		11.9		9.6	
Beryllium	0.46		0.67		0.49		0.24		0.25		0.31		0.31	
Cadmium	0.52	Ω	0.58	n	0.49	Ω	0.49	n	0.53	Ω	0.54	n	0.56	n
Calcium	393		909		253		115		141		131		94.4	)
Chromium	16.3		13.3		7.1		4.9		5.2		4.9		4.0	
Cobalt	5.9		3.0		1.9		1.3		1.2		1.6		0.99	
Copper	7.7		21.9		3.9		2.3		3.2		2.6		2.2	
Iron	11,500		6,600		6,020		2,930		3,740		4,110		3,250	
Lead	9.6		19.9		3.5		2.9		4.2		3.6		2.4	
Magnesium	1,570		1,050		642		472		515		519		436	
Manganese	40.1	<b>©</b>	38.8	@	23	@	18.3	@	20.5	@	23.3	(9)	19.1	(9)
Mercury	0.08	en O	0.11	O.O.	0.11	O@	0.62	@	0.1	@N	0.12	e O	0.00	0
Nickel	12.6		6.2		3.5		2.6		2.7		4.1		2.8	n
Potassium	626		477		335		293		288		322		168	
Selenium	0.17	Ω	0.32		0.19	Ω	0.19	Ω	0.19	n	0.18	Ω	0.16	D
Silver	0.52	Ω	0.58	Ω	0.49	Ω	0.49	ח	0.53	n	0.54	מ	0.56	Ω
Sodium	861	ш	441	田	124	ш	49	ш	55.3	ਜ਼	38.4	Э	30.9	Э
Thallium	0.27	D	0.32	n	0.3	Ω	0.31	n	0.31	Þ	0.3	ם	0.27	Ω
Vanadium	16.9		14.5		8.7		4.8		5.6		6.4		4.9	
Zinc	32.5		63.7		27.2		15.4		16.9		13.7		16.5	

TABLE C.9a (Cont.)

			Sediment Concentrations (mg/kg) at Various Depth Intervals	rations (mg/kg) at Va	rious Depth Intervals		
Parameter	BOR7-4 (8-10 ft)	BOR8-1 (0-2 ft)	BOR8-2 (2-4 ft)	BOR8-3 (4-6 ft)	BOR8-3Dup (4-6 ft)	BOR8-4 (6-8 ft)	BOR8-5 (8-10 ft)
Other Parameters							
Cyanide (mg/kg)	0.54 U	0.6 U	U 75.0	0.57 U	0.58	11 35 0	11 23 0
Solids (%)	82.6	79.5	81.6	81.8	82.3	83.8	0.77

a Sampled by ANL, analyzed by Weston Gulf Coast, CLP/TAL metals and cyanide.

Data qualifiers:

N = spike and spike duplicate recovery of percent RPD were outside the control limits.

U = analyte was analyzed for but not detected; detection limit given.

@ = percent RPD of sample duplicate was outside of control limits.

E = serial dilution percent difference was outside QC limit.

TABLE C.9b Inorganic Analyses for Sediment Borings, June 1995<sup>a</sup>

Parameter	Sediment Concentrations (mg/kg) at 0-6 in.								
	TPSED 1		TPSED 2		TPSED 3		TPSED 4		
Metals									
Aluminum	26,900	*	20,800	*	5,260	*			
Antimony	15.9	BJN	10.9	JN	0.412	UJN	6,120		
Arsenic	12.3		14.1		6.98		0.599		
Barium	927	N	267	N	104	N	1.48		
Beryllium	1.05	ВЈ	0.89	J	0.379	BJ	50.4		
Cadmium	7.04	J	1.32	J	1.55	J	0.291		
Calcium	2,470		1,530		3,300		1.06		
Chromium	80.2	N	45.9	N	8.74	N	5,720		
Cobalt	9	BJ	8.08	BJ	3.15	UJ	7.19		
Copper	515		209		54.7		3.17	U	
Iron	33,400		25,700		9,870		15.3		
Lead	1,780	*	1,260	*	34.9	*	5,740		
Magnesium	5,110		3,080		994		24.2		
Manganese	160		91.2		84.7	В	1,760		
Mercury	1.71		0.748		0.159	В	66.4		
Nickel	35.2		16.2		5.08	U	0.247		
Potassium	1,640		1,080		246	В	9.04		
Selenium	1.76	JN	0.614	BJN	0.972	BJN	429		
Silver	3.74	JN	1.3	BJN	0.169	UJN	0.482		
Sodium	430	В	162	В	716	В	0.171	U	
Solids (%)	30.6		58.5		41.3		1,020	~	
Thallium	0.915	U	0.479	U	0.678	U	0.683	U	
Vanadium	50.2	J	31.5	J	15.4	J	14.1		
Zinc	3,410		1,410		240	-	107		
Other Parameters									
Cyanide (total)	3.27		NT		2.42		NT		
pH (SI)	7.22		6.47		6.32		6.23		
Solids (%)	NT		NT		NT		41		

Parameter	Sediment Concentrations (mg/kg) at 0-6 in.								
	TPSED 5		TPSED 6		TPSED 7		TPSED 8		
Metals									
Aluminum	5600	@J	6,380		9,750		5,450		
Antimony	0.344	UNJ	0.372	U	0.248	U	0.232	U	
Arsenic	1.65	В	2.91		3.02		1.11		

TABLE C.9b (Cont.)

			Sediment (	Concentra	ations (mg/k	g) at 0-6 i	n	
Parameter	TPS	ED 5	TPS	ED 6	TPS	ED 7	TPS	ED 8
Metals (Cont.)								
Barium	35.4	В	49.1		42.6		146	
Beryllium	0.323	В	0.414		0.491		0.464	
Cadmium	0.886	В	0.656	U	0.565		0.654	
Calcium	2,720		3,870		3,520		1,020	
Chromium	8.67		10.3		46.2		9.02	
Cobalt	2.63	U	4.88		3.76		2.23	
Copper	12.6		16		13.4		34	
Iron	4,920		9,620		35,600		5,780	
Lead	18		21		8.26		20.9	
Magnesium	1,700	Е	1,840		1,080		793	
Manganese	41.9	Е	80.9		40.9		31.4	
Mercury	0.101	U	0.112		0.072	U	0.056	U
Nickel	10.9		10.2		9.02	Ü	6.54	U
Potassium	410	В	473		394		277	
Selenium	0.46	В	1.08		0.289		0.271	
Silver	0.142	U	0.153	U	0.102	U	0.271	U
Sodium	1,130		1,780	Ü	328	C	180	U
Thallium	0.567	U	0.613	U	0.409	U	0.383	U
Vanadium	17.1		13.7	Ü	13.3	U	11.8	U
Zinc	69		66.7		51.8		28.7	
					51.0		20.1	
Other Parameters								
Cyanide (total)	2.02		NT		1.46		NT	
pH (SI)	6.15		NT		4.79		NT	
Solids (%)	49.4		45.7		68.4		73.1	

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

U = analyte was analyzed for but not detected; detection limit given.

\* = internal standard area counts were outside QC limits.

B = analyte was found in the associated blank.

J = estimated value.

N = spike and spike duplicate recovery of percent RPD were outside the control limits.

NT = not tested.

@ = percent RPD of sample duplicate was outside of the control limits.

E = serial dilution percent difference was outside QC limit.

TABLE C.10 Pesticide and PCB Analyses for Sediment, June 1995<sup>a</sup>

			Sedime	nt Co	ncentrati	ons (µ	g/kg) at	0-6 in		
Compound	TPSE	ED 2	TPSE	ED 3	TPSI	ED 4	TPSE	D 5	TPSE	D 7
Aldrin	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
alpha-BHC	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
alpha-Chlordane	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
Aroclor 1016	56.4	U	79.2	U	78.5	U	66.7	U	48.2	U
Aroclor 1221	114	U	161	U	159	U	135	U	97.8	U
Aroclor 1232	56.4	U	79.2	U	78.5	U	66.7	U	48.2	U
Aroclor 1242	56.4	U	79.2	U	78.5	U	66.7	U	48.2	U
Aroclor 1248	56.4	U	79.2	U	78.5	U	66.7	U	48.2	U
Aroclor 1254	56.4	U	79.2	U	78.5	U	66.7	U	48.2	U
Aroclor 1260	56.4	U	79.2	U	78.5	U	66.7	U	48.2	U
beta-BHC	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
4,4'-DDD	11		22		3.1		6.5	Ü	3	O
4,4'-DDE	7.3		16		8.2		9.3		4.2	
4,4'-DDT	1.6	J	7.92	U	7.85	U	6.67	U	4.82	U
delta-BHC	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
Dieldrin	5.64	U	7.92	U	7.85	U	6.67	U	4.82	U
Endosulfan I	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
Endosulfan II	5.64	U	7.92	U	7.85	U	6.67	U	4.82	U
Endosulfan sulfate	5.64	U	7.92	U	7.85	U	6.67	U	4.82	U
Endrin	5.64	U	7.92	U	7.85	U	6.67	U	4.82	U
Endrin aldehyde	5.64	U	7.92	U	7.85	U	6.67	U	4.82	U
Endrin ketone	5.64	U	7.92	U	7.85	U	6.67	U	4.82	U
gamma-Chlordane	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
Heptachlor	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
Heptachlor epoxide	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
Lindane (gamma-BHC)	2.91	U	4.08	U	4.05	U	3.43	U	2.48	U
Methoxychlor	29.1	U	40.8	U	40.5	U	34.3	U	24.8	U
Toxaphene	291	U	408	U	405	U	343	U	248	U

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

TABLE C.11 Explosive Compounds Analyses for Sediment, June 1995<sup>a</sup>

		- 1				Sedi	Sediment Concentrations (µg/kg) at 0-6 in.	entratio	ns (µg/kg)	at 0-6 in						
TPSED 1 TPSED 2	TPS	TPSED 2	2		TPSED 3	D 3	TPSED 4	74	TPSED 5	5	TPSED 6	9(	TPSED 7	D 7	TPSED 8	80
40.8 U 40.8 U	U 40.8 U	40.8 U	$\Box$		40.8	n	40.8	U	40.8	Þ	40.8	=	40 8	=	807	=
46.7 U 46.7 U	U 46.7 U	46.7 U	n		46.7	D	46.7	Ω	46.7	Ω	46.7	o D	46.7	) =	46.7	o =
37.2 U 37.2 U	U 37.2 U	37.2 U	$\Box$		37.2	Ω	37.2	Ω	37.2	D	37.2	Ω	37.2	) D	37.2	> =
2110 51.6 U	51.6 U	51.6 U	$\Box$		51.6	Ω	51.6	Ω	51.6	Ω	51.6	Ω	51.6	Ω	51.6	n
Ω	U 47.6 U	47.6 U	$\Box$		47.6	Ω	47.6	Ω	47.6	Ω	47.6	Ω	47.6	Ω	47.6	o D
Ω	U 70.5 U	70.5 U	D		70.5	Ω	70.5	Ω	70.5	Ω	70.5	Ω	70.5	Ω	70.5	'n
35.2 U 35.2 U	U 35.2 U	35.2 U	Ω		35.2	n	35.2	Ω	35.2	Ω	35.2	n	35.2	Ω	35.2	
10,000 U 10,000 U	U 10,000 U	10,000 U	n		10,000	n	10,000	Ω	10,000	n	10,000	Ω	10,000	n	10,000	
D	U 81.4 U	81.4 U	n		81.4	Ω	81.4	Ω	81.4	Ω	81.4	Ω	81.4	n	81.4	
D	U 81.7 U	81.7 U	Ω		81.7	Ω	81.7	Ω	81.7	Ω	81.7	n	81.7	n	81.7	î D
87.2 U 87.2 U	U 87.2 U	87.2 U	D		87.2	Ω	87.2	n	87.2	Ω	87.2	n	87.2	n	87.2	1
250 U 250 U	U 250 U	250 U	n		250	n	250	n	250	n	200	Ω	250	n	250	· =
50.9 U 50.9 U	U 50.9 U	50.9 U	n		50.9	U	50.9	n	50.9	D	50.9	Ω	50.9	- 1	50.0	=
163 U 163 U	U 163 U	163 U	Ω		163	Ω	163	n	163	Ω	163	· 🗀	163	· =	163	) <u> </u>
TN	L	L			N		9.63	n	7.98	Ω	8.62	n	5.76	) <u>=</u>	5 30	) <u> </u>
40.2 U 40.2 U	U 40.2 U	40.2 U	n		40.2	n	40.2	Ω	40.2	Ω	40.2	'n	40.2	î D	40.2	· =
35.6 U 35.6 U	U 35.6 U	35.6 U	ח	1	35.6	n	35.6	n	35.6	Ω	35.6	n	35.6	D	35.6	'n
																)

a Sampled by ANL and analyzed by GP Environmental Services.

NT = not tested.

U = analyte was analyzed for but not detected; detection limit given.

TABLE C.12 CWA Degradation Products and Organosulfur Compounds Analyses for Sediment, June 1995<sup>a</sup>

Compound         TPSED 1         TPSED 2         TPSED 3           on products         nnethylphosphonate (DIMP)         0.373         U         1.85         U         2.61           sthylphosphonic acid (IMPA)         8.15         U         6.73         U         9.54           phonic acid (IMPA)         8.15         U         6.73         U         9.54           phonic acid (IMPA)         8.15         U         6.73         U         9.54           phonic acid (IMPA)         8.15         U         0.195         U         0.276           mpounds         npmounds         3.53         U         2.51         U         3.56           nylmethyl sulfide (CPMS)         3.53         U         1.46         U         2.07           nylmethyl sulfoxide (CPMSO)         7.75         U         4.05         U         5.74           uylmethyl sulfoxide (CPMSO)         7.36         U         4.05         U         5.15           uylmethyl sulfoxide (CPMSO)         7.36         U         4.05         U         5.15					Scurincin Concentrations (µg/ng) at 0-0 III.	at 0-0 I						
sophonate (DIMP) 0.373 U 1.85 U 2.61 hate (DMMP) 0.435 U 3.51 U 5.15 hhonic acid (IMPA) 8.15 U 6.73 U 5.15 hhonic acid (IMPA) 8.15 U 6.73 U 9.54 id 8.15 U 1.18 U 1.67 hloride (CPMS) 3.53 U 2.51 U 3.56 sulfide (CPMS) 3.53 U 1.46 U 2.07 sulfoxide (CPMSO2) 7.75 U 3.85 U 5.15 hlorid (CPMSO) 7.36 U 4.05 U 5.15 hlorid (CPMSO) 7.36 U 3.51 U 5.15 hlorid (CPMSO) 7.36 U 3.51 U 5.15	I TPSED 2	SED 3	TPSED 4	D 4	TPSED 5	5	TPSED 6	9 Q	TPSED 7	3D 7	TPSED 8	D 8
osphonate (DIMP) 0.373 U 1.85 U 2.61 bhate (DMMP) 0.435 U 3.51 U 5.15 bhonic acid (IMPA) 8.15 U 6.73 U 9.54 id 8.15 U 0.195 U 0.195 U 0.276 lid 12.9 U 0.195 U 0.276 lid 3.53 U 2.51 U 3.56 sulfide (CPMS) 3.53 U 1.46 U 2.07 sulfone (CPMSO2) 7.75 U 3.85 U 5.74 sulfoxide (CPMSO) 7.36 U 4.05 U 5.15 U 5.15 U 5.15 lid												
bhate (DMMP) 0.435 U 3.51 U 5.15 phonic acid (IMPA) 8.15 U 6.73 U 9.54 id 8.15 U 1.18 U 1.67 l) 12.9 U 0.195 U 0.276 l) 3.53 U 2.51 U 3.56 sulfide (CPMS) 3.53 U 1.46 U 2.07 sulfoxide (CPMSO2) 7.75 U 3.85 U 5.15 MDS) 2.26 U 3.51 U 5.15	U 1.85 U	.61 U	2.64	n	2.18	Ω	2.36	Ω	1.58	n	1.48	Ω
id 8.15 U 6.73 U 9.54 id 8.15 U 0.195 U 1.67 )  12.9 U 0.195 U 0.276 )  sulfide (CPMS) 3.53 U 2.51 U 3.56 sulfoxide (CPMSO2) 7.75 U 3.85 U 5.74 sulfoxide (CPMSO) 7.36 U 4.05 U 5.15 MDS) 2.26 U 3.51 U 5.15	U 3.51 U	.15 U	2.79	n	2.64	Ω	2.29	Ω	1.54	n	1.66	Ŋ
id 8.15 U 1.18 U 1.67  12.9 U 0.195 U 0.276  3.53 U 2.51 U 3.56  sulfide (CPMS) 3.53 U 1.46 U 2.07  sulfone (CPMSO2) 7.75 U 3.85 U 5.44  sulfoxide (CPMSO) 7.36 U 4.05 U 5.74  MDS) 2.26 U 3.51 U 5.15	8.15 U 6.73 U	.54 U	TN		LN		L		LN		L	
sulfide (CPMS) 3.53 U 2.51 U 3.56 sulfide (CPMSO2) 7.75 U 3.85 U 5.74 sulfoxide (CPMSO) 7.36 U 4.05 U 5.14 MDS) 2.26 U 3.51 U 5.15	U 1.18 U	U 19.	1.69	Ω	4.1	Ω	1.52	Ω	1.01	n	0.948	ņ
3.53 U 2.51 U sulfide (CPMS) 3.53 U 1.46 U sulfone (CPMSO2) 7.75 U 3.85 U sulfoxide (CPMSO) 7.36 U 4.05 U MDS) 2.26 U 3.51 U	U 0.195 U	N 92	0.278	Ω	0.23	n	0.25	Ω	0.166	Ω	0.156	n
3.53 U 2.51 U 1 sulfide (CPMS) 3.53 U 1.46 U 1 sulfone (CPMSO2) 7.75 U 3.85 U 1 sulfoxide (CPMSO) 7.36 U 4.05 U MDS) 2.26 U 3.51 U												
3.53 U 1.46 U 7.75 U 3.85 U 7.36 U 4.05 U 2.26 U 3.51 U	U 2.51 U	.56 U	3.59	Ω	2.97	Ω	3.22	Ω	2.15	ב	2.01	Ω
7.75 U 3.85 U 7.36 U 4.05 U 2.26 U 3.51 U	U 1.46 U	U 70.	2.09	Ω	1.73	Ω	1.87	Ω	1.25	Ω	1.17	Ω
7.36 U 4.05 U 2.26 U 3.51 U	7.75 U 3.85 U	.44 U	5.49	Ω	4.54	n	4.93	Ω	3.28	Ω	3.08	n
2.26 U 3.51 U	7.36 U 4.05 U	.74 U	5.78	Ω	4.79	Ω	5.19	D	3.46	Ω	3.25	n
	U 3.51 U	15 U	2.79	D	2.64	Ω	2.29	n	1.54	Ω	1.66	Ω
0	4.81 U 0.227 U 0.3	22 U	0.324	n	0.269	Ω	0.291	D	0.194	n	0.182	כ
1,4-Oxathiane 2.8 U 1.85 U 2.61	U 1.85 U	0 I9	2.64	Ω	2.18	Ω	2.36	n	1.58	ח	1.48	ח

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

NT = not tested.

U = analyte was analyzed for but not detected; detection limit given.

TABLE C.13a Volatile Organics Analyses for Soil Borings, January 1993<sup>a</sup>

Compound	JBP1-E-2' (2 ft)	JBP1-E-2'Dup (2 ft)	2'Dup	JBP1-E-4' (4 ft)	JBP1-W-2' (2 ft)	2.	JBP1-W-4' (4 ft)	.W4' ft)	JBP2-E-2' (2 ft)	-E-2' ft)	JBP2-E-4' (4 ft)	<b>.</b> 4	JBP2-E-4'Dup (4 ft)	4'Dup
Acetone	16	14	m	33 B	77	æ	Ξ	ā	,					
Benzene	Ω 9	9	-		]	a =	: `	G ;	<del>,</del>	<b>n</b> :	116	20	8	Ф
Bromodichloromathona			) ;		0	<b>)</b>	0	<b>-</b>	9	Þ	9	Ω	9	n
		9	<b>-</b>	Ω 9	9	Ω	9	Ω	9	Ω	9	Ω	9	ח
Bromotorm	n 9	9	n	Ω 9	9	U	9	Ω	9	n	9	Ω	9	=
Bromomethane	12 U	12	U	12 U	12	Ω	12	Ω	12	ב		· =	, :	=
2-Butanone	12 U	12	Ω	12 U	12	D	12	Ŋ	12	1		) <u>=</u>	2 2	) =
Carbon disulfide	0 9	9	U	n 9	9	Ω	9		9	· =		) <u>=</u>	7 7	) ;
Carbon tetrachloride	0 9	9	Ω	0 9	9	7	٠	. =	· v	> =		) <u>:</u>	<b>o</b> (	> ;
Chlorobenzene	n 9	9	n	n 9	9	. 12	9 9	· =	o v	o =		o :	o <b>\</b>	⊃ :
Chloroethane	12 U	12	Ω	12 U	12	, =	, 5	) <u>=</u>	2	o <u>=</u>	0 5	o :	٠ :	o ;
2-Chloroethylvinyl ether	12 U	12	U	12 U		n	? 2	ם מ	2 2	o =		o :	7 5	<b>&gt;</b> :
Chloroform	Ω 9	9	Ω	n 9	9	Ω	ب د	· =	7 4	o =		) <u>:</u>	7 '	⊃ :
Chloromethane	12 U	12	n	12 U		, D	2	· =	2	o =	2 2	o :	۽ ه	⊃ :
Dibromochloromethane	n 9	1 9	U	n 9		=	ب ر	) <u>=</u>	7 7	o		o :	71	<b>&gt;</b> :
1,1-Dichloroethane	n 9	1 9	IJ			) <u>=</u>	) <b>v</b>	<b>)</b>	<b>)</b> (	<b>&gt;</b> ;		o ;	9	n
1,2-Dichloroethane	11 9		, <b>–</b>			o :	o (	<b>)</b> :	۰ م	o ;		Þ	9	)
1,1-Dichloroethene	n 9					¬ -	۰ ۵	); -	9 ,	D ;		D.	9	n
trans-1.2-Dichloroethene			`			¬ •	۰ ۵	<b>)</b>	9	ם	9	Þ	9	n
1.2-Dichloropronape	s =		` -			<b>&gt;</b> :	9	n	9	n	9	Ţ	9	Ω
cis-1 3-Dichloropropene			~ <b>-</b>			o ;	9	n	9	D	9	D	9	n
trans 1.3 Disklaman						n	9	Ω	9	n	Ω 9	_	9	Ω
tians-1,3-Dicnioropropene		n 9	_	Ω 9	9	n	9	Ω	9	Ω	0 9	1	9	Ω
Ethylbenzene		Ω 9	_	0 9	n 9	1	9	Ω	9	Ω	n 9	J	9	D
2-Hexanone	12 U	12 U	_	12 U	12 U	J	12	n	12	Ω	12 U	-	12	
4-Methyl-2-pentanone	12 U	12 U	_	12 U	12 U	1	12	Ω	12	ח		_	12	· =
Methylene chloride	4 J	6 BJ		6 B	7 B		8	BJ	9	BJ	9 9		, w	B
Styrene	N 9	0 9		Ω 9	0 9	_	9	U	9	Ω	n 9	_	9	

TABLE C.13a (Cont.)

			8	Sorehole Concentrat	Borehole Concentrations (µg/kg) at Various Depths	is Depths		
Compound	JBP1-E-2' (2 ft)	JBP1-E-2'Dup (2 ft)	JBP1-E-4' (4 ft)	JBP1-W-2' (2 ft)	JBP1-W-4' (4 ft)	JBP2-E-2' (2 ft)	JBP2-E-4' (4 ft)	JBP2-E-4'Dup (4 ft)
1,1,2,2-Tetrachloroethane	7	3	4 I	3 BJ	n 9	3 ј	6 J	n 9
Tetrachloroethene	N 9	n 9	0 9	n 9	N 9	n 9	Ω 9	
Toluene	2 J	3 J	N 9	n 9	N 9	0 9		
1,1,1-Trichloroethane	0 9	n 9	n 9	n 9	0 9	Ω 9		
1,1,2-Trichloroethane	N 9	0 9	0 9	n 9	n 9	Ω 9	n 9	Ω 9
Trichloroethene	28	5 J	0 9	n 9	0 9	n 9	N 9	
Vinyl acetate	12 U	12 U	12 U	12 U	12 U	12 U	12 U	
Vinyl chloride	12 U	12 U	12 U	12 U	12 U	12 U		
Xylenes (total)	6 U	12	6 U	6 U	6 U	6 U	0 9 n	Ω 9
			Borehole C	oncentrations (µg/kį	Borehole Concentrations (µg/kg) at Various Depths			1
Compound	JBP2-C-2' (2 ft)	JBP2-C-4' (4 ft)	JBP2-W-2' (2 ft)	JBP2-W-4' (4 ft)	JHDP-C-4' (4 ft)	JHDP-C-6' (6 ft)	JHDP-C-6'Dup (6 ft)	1
Acetone	19 B	30 B	18 B	17 B	771	92	113	
Benzene	7 U	n 9	n 9	n 9	61	83	58	
Bromodichloromethane	7 U	n 9	n 9	n 9	2 J	6 J	S J	
Bromoform	7 U	n 9	Ω .9	n 9	0 9	n 9	Ω 9	
Bromomethane	14 U	12 U	12 U	12 U	12 U	12 U	U II	
2-Butanone	14 U	12 U	12 U	12 U	12 U	12 U	11 U	
Carbon disulfide	7 U	N 9	0 9	0 9	Ω 9	l J	-	
Carbon tetrachloride	7 U	Ω 9	n 9	n 9	17	172	107	
Chlorobenzene	7 U	Ω 9	N 9	n 9	n 9	n 9	n 9	
Chloroethane	14 U	12 U	12 U	12 U	12 U	12 U	O 11	
2-Chloroethylvinyl ether	14 U	12 U	12 U	12 U	12 U	12 U	n n	
Chloroform	17	n 9	Ω 9	l J	142	445	444	
Chloromethane	14 U	12 U	12 U	12 U	12 U	12 U	n II	
Dibromochloromethane	) U	Ω 9	n 9	n 9	n 9	0 9	Ω 9	

TABLE C.13a (Cont.)

1   1   1   1   1   1   1   1   1   1	Compound	JBP	JBP2-C-2' (2 ft)	JBP2	JBP2-C-4' (4 ft)	JBP2-W-2' (2 ft)	W-2'	JBP2-W (4 ft)	JBP2-W-4' (4 ft)	JHDP-C-4'	C-4.	JHDP-C-6' (6 ft)	C-6'	JHDP-C-6'Dup (6 ft)	6'Dup )	
thinnee 2 1 1 6 U 6 U 6 U 31 78 77 77 77 77 77 77 77 77 77 77 77 77	1,1-Dichloroethane	7	n	9	Ω	9	ם	9	11	9	=					,
there	1,2-Dichloroethane	2	-	9	11	4	Ξ	, 4	· =	•	)	- :		ب		
Notice the part of the part	I.1-Dichloroethene	14		٠ ٧	) <b>:</b>		) ;	> <b>·</b>	<b>)</b> ;	04		53		62		
Marchenical Control	rans_1 2_Dichloroethene	t e		<b>5</b> (	ο,	ِ م	<b>-</b>	9	<b>D</b>	31		78		77		
rotyone 7 U 6 U 6 U 6 U 6 U 6 U 6 U 6 U 6 U 6 U	ans-1,2-Diction octions	76		n	_	91		62		3,860	Ω	1,710	Ω	3,220		
1   1   1   1   1   1   1   1   1   1	,2-Dichloropropane	7	ם	9	n	9	D	9	Ω	9	Ω	9	Ω	9		
1   1   1   1   1   1   1   1   1   1	is-1,3-Dichloropropene	7	n	9	Ω	9	n	9	D	9	n	9	D	. 9	· =	
14   U   12   U   13   U   14   U   12   U   12   U   13   U   14   U   12   U   U   12   U   U   12   U   U   U   U   U   U   U   U   U	rans-1,3-Dichloropropene	7	Ω	9	Ω	9	n	9	Ω	9	D	9	n	· •	) <u>=</u>	
14   U   12   U   14   U   12   U   12   U   14   U   12   U   U   12   U   U   12   U   U   12   U   U   U   U   U   U   U   U   U	thylbenzene	7	Ω	9	Ω	9	Ω	9	Ω	9	ם	9	n	· •	) <u>=</u>	
nitatione 14 U 12 U 12 U 12 U 12 U 12 U 11 U 11 U	-Hexanone	14	D	12	Ω	12	n	12	n	12	ח	12	Þ	· =	· =	
oride         7         B         5         BJ         6         B         45         B         46         B         41         B           nhorocethane         7         U         6	-Methyl-2-pentanone	14	Ď	12	n	12	Ω	12	U	12	n	12	· =	: :	) <u>=</u>	
1   U   G   U   G   U   G   U   G   U   G   U   G   U   G   G	<b>fethylene</b> chloride	7	В	5	<b>BJ</b>	9	В	9	BJ	45	В	40		. 14	Ω	
bloroethane 7 U 6 U 26 U 16 289,000 D 1,890,000 D 3,270,000 D 2,270,00 D 2,27	tyrene	7	n	9	Ω	9	Ω	9	Ω	9	Þ	y vo	1 =	. 9	a =	
ene         582         D         40         26         9         682         5,730         D         25,000         D           7         U         6         U	,1,2,2-Tetrachloroethane	7	Ω	9	Ω	26	Ω	16		289,000	Q	1.890.000	) C	3 270 000	٥	
1   U   6   U   C   U   C   U   C   U   C   U   C   U   C   U   C   U   C   U   C   U   C   U   C   U   C   U   C   U   C   U   C   U   U	etrachloroethene	582	D	40		26		6		682		5.730	n C	25,700	a C	
vethane         7         U         6         U </td <td>oluene</td> <td>7</td> <td>Ω</td> <td>9</td> <td>Ω</td> <td>9</td> <td>Ω</td> <td>9</td> <td>Ω</td> <td>2</td> <td>'n</td> <td>9</td> <td>=</td> <td>2011</td> <td>) <u>=</u></td> <td></td>	oluene	7	Ω	9	Ω	9	Ω	9	Ω	2	'n	9	=	2011	) <u>=</u>	
behane 7 U 6 U 6 U 6 U 6 U 2,100 D 7,860 D 8,540 D 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	1,1-Trichloroethane	7	n	9	Ω	9	n	9	Ω	9	n	y ve	) <u> </u>	· •	> =	
e 499 D 39 176 D 168 D 6,780 D 33,100 D 263,000 D 114 U 12	1,2-Trichloroethane	7	Ω	9	Ω	9	D	9	Ω	2,100	D	7 860	۵ ۵	8 \$40	ם כ	
14   U   12   U   12   U   12   U   12   U   12   U   12   U   11   U   U   11   U   U   U   U	ichloroethene	466	D	39		176	D	168	Ω	6.780		33.100	۵ ۵	043,000	ם ב	
14 U 12 U 12 U 15 U 315 170 302  7 U 6 U 6 U 6 U 6 U 6 U 6 U 6 U 6 U 6 U	inyl acetate	14	Ω	12	Ω	12	Ŋ	12	Ω	12	D	21	ı =	11	۵ =	
7   U   6   U   G   U   U	inyl chloride	14	Ω	12	D	12	Ω	12	n	315		170	)	302	Þ	
JVXP-C-4"       JSDP-C-4"       JSDP-C-6"       JBPM-A-3"       JBPM-A-1"         JVXP-C-4"       JSDP-C-4"       JSDP-C-6"       JBPM-A-3"       JBPM-A-1"         (4 ft)       (6 ft)       (6 ft)       (3 in.)       (3 in.)       (1 ft)         13       U       344       13       U       67       275       1,460       BJ         7       U       7       U       7       U       7       U       743       U	ylenes (total)	7	n	9	n	9 1 0	D .	9 1	n	9	ָם נו	9	Ŋ	9	Ω	
JVXP-C-4'         JVXP-C-4'         JSDP-C-6'         JBPM-A-3"         JBPM-A-3"Dup         JBPM-A-1'           (4 ft)         (6 ft)         (3 in.)         (3 in.)         (1 ft)           13         U         14         U         344         13         U         67         275         1,460         BJ           7         U         7         U         7         U         7         U         743         U							Boi	rehole Co	ncentrati	ns (µg/kg) a	at Variou	ss Depths			I I I	
13 U 14 U 344 13 U 67 275 1,460 7 U 7 U 7 U 27 U 30 U 743	Compound	JVXP (4 f	r.C-4'	JVXP- (6 fi	C-6'	JSDP-C (4 ft)	.4	JSDP-C ff (6 ff	.9-\ -0-	JBPM-A (3 in.)	-3"	JBPM-A-3' (3 in.)	'Dup	JBPM-A.	-1-	JBPM-B-3"
7 U 7 U 33 U 7 U 27 U 30 U 743	cetone	13	Ω	14	U	344		7	=	129		326			1	
2 0 30 0 743	nzene	7	n	7	Ω	33	11	<u>, , , , , , , , , , , , , , , , , , , </u>	) <u>=</u>	; ;	:	C17	;	1,460	F :	324
	Bromodichloromothone				)	,	)	-	<b>-</b>	17	<b>-</b>	30	_	743	=	20

TABLE C.13a (Cont.)

•	JVXI	JVXP-C-4'	JVX	JVXP-C-6'	JSDP-C-4"	C-4.	JSDP	JSDP-C-6'	JBPM-A-3"	A-3"	JBPM-A-3"Dup	3"Dup	JBPM-A-1'	<u>-1-</u>	JBPM-B-3"	-B-3"
Compound	4)	(4 ft)		(6 ft)	(4 ft)		9)	(e ft)	(3 in.)	_	(3 in.)		(1 ft)		(3 in.)	()
Вготобогт	7	Ω	7	n	33	Ŋ	7	n	27	n	30	n	743	n	29	ב
Bromomethane	13	U	14	U	29	Ω	13	Ω	54	Ω	59	n	1,490	D	57	⊃
2-Butanone	13	n	14	U	19	Ω	13	D	54	Ω	59	n	1,490	n	57	$\supset$
Carbon disulfide	7	n	7	Ω	33	Ω	7	ם	27	n	30	n	743	n	29	$\supset$
Carbon tetrachloride	7	Ω	7	Ω	33	Ω	7	Ω	27	ח	30	Ω	743	ב	29	n
Chlorobenzene	7	Ω	7	n	33	n	7	D	27	Ü	30	Ŋ	743	ח	29	
Chloroethane	13	Ω	14	Ω	29	Ω	13	n	54	Ω	59	ם	1,490	n	57	
2-Chloroethylvinyl ether	13	Ω	14	Ω	<i>L</i> 9	n	13	Ω	54	ב	59	D	1,490	n	57	⊃
Chloroform	7	Ω	9	ī	33	D	7	Ω	27	Ω	30	D	743	n	29	Þ
Chloromethane	13	Ω	14	Ω	29	Ω	13	Ω	54	Ω	59	Ω	1,490	ב	57	D
Dibromochloromethane	7	Ω	7	U	33	Ω	7	Ω	27	Ω	30	n	743	n	29	כ
1,1-Dichloroethane	7	Ω	7	Ω	33	Ω	7	n	27	Ω	30	Ω	743	Ω	29	Ω
1,2-Dichloroethane	7	Ω	7	n	33	Ω	7	U	27	n	30	Ω	743	Þ	29	D
1,1-Dichloroethene	7	D	7	Ω	33	n	7	Ω	27	n	30	Ω	743	D	29	Ω
trans-1,2-Dichloroethene	7	ם	7	n	78		7	Ω	27	n	30	Ω	743	Ω	50	
1,2-Dichloropropane	7	Ω	7	Ω	33	n	7	Ω	27	D	30	Ω	743	n	29	Ω
cis-1,3-Dichloropropene	7	Ω	7	Ω	33	n	7	n	27	n	30	n	743	ņ	29	Ω
trans-1,3-Dichloropropene	7	n	7	n	33	Ω	7	D	27	D	30	n	743	Ω	29	Ω
Ethylbenzene	7	Ω	7	Ω	33	Ω	7	n	27	Ω	30	U	743	Þ	29	Ω
2-Hexanone	13	Ω	14	n	29	n	13	n	54	ב	59	Ω	1,490	Þ	57	Ω
4-Methyl-2-pentanone	13	Ω	14	Ω	<i>L</i> 9	Ω	13	Ω	54	D	59	Ω	1,490	n	57	D
Methylene chloride	7	В	8	В	44	В	12	В	21	ВJ	36	В	467	ВЈ	36	83
Styrene	7	n	7	n	33	n	7	n	27	D	30	Ω	743	מ	29	D
1,1,2,2-Tetrachloroethane	16		39		15,200		26		123		118		1,460	D	13,200	Ω
Tetrachloroethene	7	Þ	7	n	27	'n	7	Ω	27	ח	13	ī	530	ſ	17	_
Toluene	بر.	-	4	J	33	⊃	4	ī		ſ	30	ח	743	n	29	n
1,1,1-Trichloroethane	7	n	7	Ω	33	n	7	ח	27	ם	30	n	743	n	29	Ω
1,1,2-Trichloroethane	7	n	7	D	81		7	ລ	33		19		367	٠,	80	
Frichloroethene	~,	_	16		345	<u> </u>	~	_	2.15		843		00096	_	712	

TABLE C.13a (Cont.)

					Borehole Concentrations (µg/kg) at Various Depths	ations (μg/kg)	at Various De	oths		
Compound	JVXP-C-4' (4 ft)	.4.	JVXP-C-6' (6 ft)	JSDP-C-4' (4 ft)	JSDP-C-6' (6 ft)	JBPM-A-3" (3 in.)		JBPM-A-3"Dup (3 in.)	JBPM-A-1' (1 ft)	JBPM-B-3" (3 in.)
Vinyl acetate	13	Ω	14 U	N 19	13 U	54	Þ	11 65	1 490	
Vinyl chloride	13	D	14 U	n <i>L</i> 9	13 U	54	n			
Xylenes (total)	13	U	7 U	33 U	7 U	27	n			
					Borehole Concentrations (µg/kg) at Various Depths	tions (µg/kg)	at Various Dep	ths	İ	ĺ
Compound	JBPM-B-1'	8-1.	JBPM-C-3" (3 in.)	JBPM-C-1' (1 ft)	JBPP-A-3" (3 in.)	JBPP-A-I'	JBPP-B-3" (3 in.)	3" JBPP-B-1' (1 ft)	JBPP-C-3" (3 in.)	JBPP-C-1' (1 ft)
Acetone	1,060	BJ	13 U	57	8 B1	32 B	4	, oc		
Benzene	731	Ω	n 9	N 9		n 9			17 0	
Bromodichloromethane	731	Ω	n 9	0 9	Ω 9	n 9		o vo		0 4
Вготоfогт	731	D	n 9	n 9	n 9	n 9	7 U	9		
Bromomethane	1,460	Þ	13 U	11 U	12 U	12 U	14 U	12		12 11
2-Butanone	1,460	n	13 U	11 U	12 U	12 U	14 U	12 U	12 U	
Carbon disulfide	731	ח	-	n 9	0 9	0 9	7 U	n 9	N 9	
Carbon tetrachloride	731	n	n 9	n 9	n 9	n 9	7 U	n 9	Ω 9	
Chlorobenzene	731	Ω	Ω 9	n 9	n 9	n 9	7 U	N 9		
Chloroethane	1,460	Ω	13 U	11 U	12 U	12 U	14 U	12 U		
2-Chloroethylvinyl ether	1,460	n	13 U	11 U	12 U	12 U	14 U	12 U	12 U	
Chloroform	731	Ω	Ω 9	0 9	N 9	n 9	7 U	10		
Chloromethane	1,460	n	13 U	11 U	12 U	12 U	14 U	12 U	12 U	
Dibromochloromethane	731	Ω	n 9	Ω 9	Ω 9	n 9	7 U	n 9		
1,1-Dichloroethane	731	Ω	n 9	n 9	n 9	0 9	7 U			
1,2-Dichloroethane	731	D	Ω 9	0 9	N 9	Ω 9	7 U	Ω 9		
I, I-Dichloroethene	731	Ω	n 9	Ω 9	n 9	n 9	7 U	- 11		
trans-1,2-Dichloroethene	731	n	n 9	n 9	n 9	Ω 9				
1,2-Dichloropropane	731	Ω	Ω 9	n 9	N 9	0 9	7 U			
cis-1,3-Dichloropropene	731	מ	n 9	0 9	N 9	0 9	7 U			
trans-1,3-Dichloropropene	731	ם	n 9	N 9	Ω 9	Ω 9	7 U	0 9	N 9	n 9

TABLE C.13a (Cont.)

Compound	JBPM-B-1' (1 ft)	-B-1'	JBPM-C-3" (3 in.)	.3"	JBPM-C-1' (1 ft)	JBPI (3	JBPP-A-3" (3 in.)	JBPI (1	JBPP-A-1' (1 ft)	JBPF (3	JBPP-B-3" (3 in.)	JBPP-B. (1 ft)	JBPP-B-1' (1 ft)	ЈВРР (3	JBPP-C-3" (3 in.)	JBP	JBPP-C-1' (1 ft)
Ethylbenzene	731	n	7	-	n 9	9	D	9	'n	7	Ω	9	ם	9	Ω	9	Ω
2-Hexanone	1,460	Ω	13 1	Ω	11 U	12	Ω	12	Þ	14	Ω	12	ם	12	ם י	12	) D
4-Methyl-2-pentanone	1,460	Ω	13 1	Ω	11 U	12	D	12	D	14	n	12	D	12	n	12	D
Methylene chloride	451	ВЈ	6	В	6 B	7	В	∞	В	∞	В	∞		\$	ſ	9	
Styrene	731	n	1 9	n	n 9	9	ם	9	Ω	7	n	9	n	9	n	9	n
1,1,2,2-Tetrachloroethane	15,200		7.1		41	7		5	ĭ	4	ſ	2	ľ	9	ב	6	
Tetrachloroethene	731	n	9	n	N 9	9	D	9	Ω	24		66		9	D	9	n
Toluene	731	Ω	7		N 9	9	Ω	2	•	7	ב	3	_	9	Ω	9	Ω
I,1,1-Trichloroethane	731	Ω	9	Ω	n 9	9	Ω	9	ם	7	Ω	9	n	9	D	9	n
1,1,2-Trichloroethane	731	Ω	9	Ω	0 9	9	ח	9	D	7	n	9	n	9	Ω	9	n
Trichloroethene	251	ſ	10		3 J	9	ſ	9	ם	6		31		9	D	9	n
Vinyl acetate	1,460	ב	13 (	n	11 U	12	Ω	12	ב	14	n	12	Ω	12	Ω	12	n
Vinyl chloride	1,460	ם	13 (	n	11 U	12	n	12	Þ	14	n	12	Ω	12	Ω	12	n
Xylenes (total)	731	n	6		11 9	ν.	1	9	-	7	=	4	1.1	4		•	-

<sup>a</sup> Sampled by Weston, analyzed by GP Environmental Services, TCL volatile organics-SW846 8240s; provisional data.

Data qualifiers:

U = analyte was analyzed for but not detected.

J = estimated value.

B = analyte was found in the associated blank.

D = sample was diluted for analysis.

TABLE C.13b Volatile Organics Analyses for Soil Borings, January 1994<sup>a</sup>

					Sc	il Con	Soil Concentrations (µg/kg) at Various Depth Intervals	(µg/kg)	at Var	ious Dep	th Interv	als				
Compound	0 6	CLP1 (0-6 in.)		CLP1 (6-24 in.)	P1 ( in.)	CLF (6-2	CLP1-RE (6-24 in.)	CI (24-7	CLP1 (24-40 in.)	9)	CLP2 (0-6 in.)	9	CLP2 (6-24 in.)	G. C.	CLP2-RE (6-24 in.)	
Acetone	21	щ	B	6	JB	24	æ	7	Ξ	13	=	ç	٩			
Benzene	15		ם	13	n	13	D	13	) D	13		2 2	a E	4 <u></u>	9 1	
Bromodichloromethane	15	٦	J	13	Ω	13	ם	13	n	13		3 5	) =	13	) [	
Bromoform	15	_	<u></u>	13	Ω	13	Ω	13	) D	; ;:		3 5	) <u> </u>	<u> </u>	) <u>:</u>	
Bromomethane	15	_	l l	13	Ω	13	U	13	n	13	) <u> </u>	3 2	) <u> </u>	C 1	) =	
2-Butanone	15	ر	_	13	Ŋ	13	n	13	D	13	n	13	) <u> </u>	3 2	> =	
Carbon disulfide	15	)	_	13	Ω	13	n	13	n	13	n	3 2	) <u> </u>	2 5	) =	
Carbon tetrachloride	15	)	_	13	Ω	13	D	13	n	13	n	13	2 1	2 :	=	
Chlorobenzene	15	)	_	∞	-	Ŋ	'n	7	-	13	ב	13	) [	2	) <u>=</u>	
Chloroethane	15	ם		13	Ω	13	Ω	13	n	13	D	13	n D	13	) =	
Chloroform	15	D		13	n	13	Ω	13	n	13	D	13	- 1	2	) =	
Chloromethane	15	Ω		13	Ω	13	Ω	13	n	13	ר	13	ם מ	<u> </u>	=	
Dibromochloromethane	15	Þ		13	Ω	13	U	13	Ω	13	n	13	n	13	<u> </u>	
1,1-Dichloroethane	15	Ω		13	Ω	13	n	13	ם	13	Ω	13	Ω	13	Þ	
1,2-Dichloroethane	15	Ω		13	Ω	13	U	13	Ω	13	n	13	Ω	13	ם	
1,1-Dichloroethene	15			13	D	13	U	13	Ω	13	D	13	Þ	13	) <u> </u>	
1,2-Dichloroethene (total)	15	n		13	Ω	13	U	13	n	13	n	. 13	D	13	) D	
cis-1,3-Dichloropropane	15			13	D	13	U	13	n	13	n	13	n	13	n	
1,2-Dichloropropane	15	D		13	Ω	13	D	13	n	13	n	13	U	13	<u> </u>	
trans-1,3-Dichloropropene	15			13	Ω	13	U	13	Ω	13	Ω	13	ב	13		
Ethylbenzene	15	Ω		13	Ω	13	Ω	13	Ω	13	n	13	Ω	13		
2-Hexanone	15			13	Ŋ	13	Ω	13	Ω	13	U	13	Ω	13	- =	
4-Methyl-2-pentanone	15	Ω		13	n	13	n	13	Ω	13	ב	13	Ω		) =	
Methylene chloride	15			13	U	13	Ω	13	Ω	13	D	13	n	2 2	=	
Styrene	15	Ω		13	Ω	13	ū	13	Ω	13	Þ	13	) [	2 :	=	
1,1,2,2- Tetrachloroethane	15	$\Box$		[3	n	13	U	13	Ω	13	Þ	13	) []	: :	=	
Tetrachloroethene	15	n		3	J	8		13	Ω	13	Ω	13	n D	13	מ	

TABLE C.13b (Cont.)

			Soil Concentration	Soil Concentrations (µg/kg) at Various Depth Intervals	us Depth Interva	SI		
Compound	CLP1 (0-6 in.)	CLP1 (6-24 in.)	CLP1-RE (6-24 in.)	CLP1 (24-40 in.)	CLP2 (0-6 in.)	CL.P2 (6-24 in.)	CLP2-RE (6-24 in.)	
Toluene	15 U	4 J	4 J	5 J	13 U	13 U	. O EI	
1,1,1-Trichloroethane	15 U	13 U	13 U	13 U	13 U	13 U	13 U	
1,1,2-Trichloroethane	15 U	13 U	13 U	13 U	13 U	13 U	13 U	
Trichloroethene	5 J	5 J	4 Ĭ	3	13 U	13 U		
Vinyl chloride	15 U	13 U	13 U	13 U	13 U	13 U	13 U	
Xylenes (total)	15 U	13 U	13 U	13 U	13 U	13 U	13 U	
		S	oil Concentration	Soil Concentrations (µg/kg) at Various Depth Intervals	ıs Depth Interval	S		 
	CLP2	CLP2-RE	CLP3	CLP3	CLP3-RE	CLP3-Dup	CLP3	
Compound	(24-48 in.)	(24-48 in.)	(0-6 in.)	(6-24 in.)	(6-24 in.)	(6-24 in.)	(24-48 in.)	
Acetone	150 U	220 B	55 B	23 B	27 B	15 U	29 U	
Benzene	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
Bromodichloromethane	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
Bromoform	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
Bromomethane	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
2-Butanone	26	36	18 U	15 U	15 U	15 U	29 U	
Carbon disulfide	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
Carbon tetrachloride	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
Chlorobenzene	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
Chloroethane	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
Chloroform	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
Chloromethane	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
Dibromochloromethane	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
1,1-Dichloroethane	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
1,2-Dichloroethane	15 U	15 U	18 U	15 U	15 U	15 U	29 U	
1,1-Dichloroethene	15 U	15 U	18 U	15 U	15 U	15 U	29 U	

TABLE C.13b (Cont.)

CLP3 (6-24 in.) (6-24 in.) (24-48 in.)  5						,			-
there (total) 15 U 15 U 18 U 15 U 15 U 15 U 15 U 15 U	Compound	CLP2 (24-48 in.)	CLP2-RE (24-48 in.)	CLP3 (0-6 in.)	CLP3 (6-24 in.)	CLP3-RE (6-24 in.)	CLP3-Dup (6-24 in.)	CLP3 (24-48 in)	
ropane 15 U 15 U 18 U 18 U 15 U 15 U 15 U 15 U	1,2-Dichloroethene (total)								
Propose 15 U 15 U 18 U 15 U 15 U 15 U 15 U 15 U	1.3 15:11:								
Noticy proper 15 U 15 U 18 U 18 U 15 U 15 U 15 U 15 U	1,2-Dichloropropane			-				29	
loropropene 15 U 15 U 18 U 15 U 15 U 15 U 15 U 15 U	cis-1,3-Dichloropropane			18 U					
15   U   15   U   18   U   15   U   U   15   U   U   15   U   U   15   U   15   U   U   15   U   U   15   U   U   15   U   U   U   U   U   U   U   U   U	trans-1,3-Dichloropropene			18 U					
15 U 15 U 16 U 17 U 17 U 17 U 17 U 17 U 17 U 17	Ethylbenzene			18 11					
nitanone 15 U 15 U 18 U 15 U 15 U 15 U 15 U 15 U	2-Hexanone			18 U					
oride   15   U   15   U   18   U   15   U   U   15   U   U   15   U   U   15   U   U   U   U   U   U   U   U   U	4-Methyl-2-pentanone			18 U					
15   U   15   U   18   U   15   U   U   15   U   U   15   U   U   15   U   U   U   U   U   U   U   U   U	Methylene chloride								
15   U   15   U   18   U   15   U   U   15   U   U   15   U   U   U   U   U   U   U   U   U	Styrene			18 U					
certe 15 U 15 U 18 U 15 U 15 U 15 U 15 U 15 U	1,1,2,2-Tetrachloroethane			18 U					
Sethane 15 U 15 U 18 U 15 U 15 U 15 U 15 U 15 U	Tetrachloroethene								
octhane 15 U 15 U 18 U 15 U 15 U 15 U 15 U 15 U	Toluene	8 J	6 J						
pethane 15 U 15 U 18 U 15 U 15 U 15 U 15 U 15 U	1,1,1-Trichloroethane		_		15 U				
c	1,1,2-Trichloroethane								
15 U 15 U 18 U 15 U 15 U 15 U 15 U 15 U	Trichloroethene				15 U				
15   U   15   U   15   U   15   U   15   U   15   U   15   U   15   U   29   U	Vinyl chloride								
CLP3-RE CLP4 CLP4 CLP4 CLP4 CLP5 CLP5 CLP5 CLP5 CLP5 CLP5 CLP5 CLP5	Xylenes (total)	į	į	ĺ					
CLP3-RE CLP4 CLP4 CLP4 CLP5 CLP5 CLP5 CLP5 CLP5 CLP5 CLP5 CLP5				Soil Conce	ntrations (µg/kg) a	t Various Depth	Intervals	! ! ! ! !	  -  -  -
29     U     14     U     13     U     13     U     12     U     14     U       29     U     14     U     13     U     26     U     13     U     12     U     14     U       29     U     14     U     13     U     26     U     13     U     12     U     14     U       29     U     14     U     13     U     26     U     13     U     14     U       29     U     14     U     13     U     26     U     13     U     14     U       29     U     14     U     13     U     26     U     13     U     14     U		CLP3-RE	CLP4	CLP4	CI P4	CIDS	ž	Š	
29 U 14 U 16 B 62 B 13 U 12 U 110 B 13 nothane 29 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 14 U 13 U 15 U 14 U 15 U 15 U 15 U 15 U 15 U 15	Compound	(24-48 in.)	(0-6 in.)	(6-24 in.)	(24-48 in.)	(0-6 in.)	(6-24 in.)	(24-48 in.)	CLP5-Dup (24-48 in.)
29 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 15 U 14 U 13 U 26 U 15 U 17 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 14 U 13 U 15 U 15 U 15 U 15 U 15 U 15 U 15	Acetone								
nethane 29 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 15 U 15 U 15 U 15 U 15 U 15 U 15	Benzene		_		_	-			
29 U 14 U 13 U 26 U 13 U 12 U 14 U 13 U 26 U 13 U 17 U 14 U 13 U 26 U 13 U 17 U 17 U 17 U 18 U 18	3 romodichloromethane								
29 U 14 U 13 U 26 U 13 U 13 U 13 U 13 U 13 U 13 U 13 U 1	3romoform								
	3romomethane				26 11			_	C :

TABLE C.13b (Cont.)

Compound	CLP.	CLP3-RE (24-48 in.)	CLP4 (0-6 in.)	.P4 in.)	CI (6-2	CLP4 (6-24 in.)	CL (24-4	CLP4 (24-48 in.)	<u>5</u>	CLP5 (0-6 in.)	C-9)	CLP5 (6-24 in.)	CL (24-4	CLP5 (24-48 in.)	CLP:	CLP5-Dup (24-48 in.)
2-Butanone	29	ם	14	n	13	IJ	26	12		=	2	  -	2		12	=
Carbon disulfide	29	D	14	D	13	11	29		2	· =	: 2	· =	7	<u>-</u>	; ;	) <u> </u>
Carbon tetrachloride	29	þ	14	'n	13	) D	36	11	: ::	> =	1 5	) <u>=</u>	14	) <u> </u>	3 2	) <u>=</u>
Chlorobenzene	29	D	14	n	13	D	26	ם י	13	ם י	17	o D	. 4	) <u>=</u>	2 2	> =
Chloroethane	29	U	14	b	13	Ω	26	n	13		12	; <b>&gt;</b>	1 4	) D	13	) <u>=</u>
Chloroform	29	Ω	14	Ω	13	Ω	26	n	13	Ω	12	D	14	Þ	13	n
Chloromethane	29	n	14	n	13	D	26	Ω	13	n	12	ח	14	D	13	n
Dibromochloromethane	29	n	14	D	13	þ	26	Ω	13	D	12	ם	14	D	13	D
1,1-Dichloroethane	29	Ω	14	Ω	13	Ω	26	U	13	n	12	D	14	Ŋ	13	D
1,2-Dichloroethane	29	ם	14	n	13	D	26	D	13	D	12	D	14	Ω	13	n
1,1-Dichloroethene	29	ם	14	Ω	13	D	26	n	13	ח	12	n	14	Ω	13	D
1,2-Dichloroethene (total)	29	D	14	n	13	D	009		91		13		180		83	
1,2-Dichloropropane	29	Ω	14	Ω	13	ם	26	ם	13	n	12	n	14	Ω	13	D
cis-1,3-Dichloropropane	29	D	14	D	13	D	26	D	13	n	12	D	14	Ω	13	Ω
trans-1,3-Dichloropropene	29	Ω	14	Ω	13	n	26	Ω	13	n	12	D	14	Ω	13	Ω
Ethylbenzene	29	Ω	14	Ω	13	U	26	n	13	D	12	D	4	ב	13	Ω
2-Hexanone	29	n	14	Ŋ	13	Ω	26	Ŋ	13	Ŋ	12	n	14	D	13	n
4-Methyl-2-pentanone	29	מ	14	n	13	n	26	n	13	ח	12	Ω	14	ב	13	n
Methylene chloride	29	D	14	D	13	Ω	26	ח	13	D	12	D	14	D	13	Ω
Styrene	29	n	14	Ω	13	D	26	Ω	13	D	12	Ω	14	n	13	Ω
1,1,2,2-Tetrachloroethane	29	n	14	D	13	D	5,600	ш	13	D	12	Ω	14	n	13	⊃
Tetrachloroethene	29	n	14	D	13	D	1,500	ш	13	D	12	n	14	D	13	ņ
Toluene	29	D	14	Ω	13	D	66		12	-	27		33		9	_
1,1,1-Trichloroethane	29	D	14	ລ	13	Ŋ	26	n	13	n	12	n	14	Ω	13	$\Box$
1,1,2-Trichloroethane	29	D	14	n	13	ם	500		13	ລ	12	נ	14	n	13	D
Trichloroethene	29	D	14	Ω	13	ם	3,100		22		11		6	_	.ب	_
Vinyl chloride	56	כ	14	ב	13	D	26	Ω	13	⊃	12	Ŋ	14	D	13	n
Values Getal	ç															

TABLE C.13b (Cont.)

	į					Š	Soil Concentrations (µg/kg) at Various Depth Intervals	ntration	; (µg/kg	) at Vario	ous Depth	1 Interva	ıls				
Compound	5 6	CLP6 0-6 in.)	ي ک	CLP6 (6-24 in.)	2	ت ا	CLP7 (0-6 in)	ט ק	CLP7	Š	CLP7	0 (	CLP8	CLP	CLP8-RE	٥	CLP9
								5	( III.)	7	-40 III.)	2	(0-6 in.)	9-0	(0-6 in.)	ė	(0-6 in.)
Acetone	13	U	20	_		16	n	13	Ω	13	Ω	:	Ξ	3.	Ξ	2	=
Benzene	13	C	13		n	16	Ω	13	D	13	Ω	; ;	) <u> </u>	3	) <b>=</b>	1 2	) =
Bromodichloromethane	13	n	13		D	16	n	13	Ξ	13	- =	3 2	) <u> </u>	7 - 7	ב כ	7 5	> <b>:</b>
Bromoform	13	כ	13		D	16	Ω	2	· =		) <u> </u>	7 7	) <u> </u>	7 6	) :	7 5	o :
Bromomethane	13	D	13		D	16	'n	13	) =	3 5	) <u>=</u>	3 5	) <u>=</u>	21.	o :	7 5	) :
2-Butanone	13	Þ	13		D	16	D	2	· =	13	) <u>=</u>	3 2	) <u> </u>	21	)	7 5	<b>&gt;</b> ;
Carbon disulfide	13	n	4			16	- 11	3 2	) <u>=</u>	3 5	) =	7 5	) <u>:</u>	10	) :	71	o ;
Carbon tetrachloride	13	D	13		D	91	) D	13	) D	3 5	) <u>=</u>	3 5	> =	3 5	) <u>;</u>	7 :	) :
Chlorobenzene	13	n	13		n	16	D	<u></u>	Ξ	1 2	· =	3 2	) <u>=</u>	21.	) <b>:</b>	7 [	> :
Chloroethane	13	n	13		D	16	n	13	) [	3 1	) =	7 7	) <u>:</u>	7 7	o	71 1	<b>&gt;</b> :
Chloroform	13	Ω	13		b	16	D	13	) D	3 3	) =	3 5	> <b>=</b>	3 17	> =	17	) <u>;</u>
Chloromethane	13	Ω	13	Ω	-	16	D	13	n	13	) D	3 5	) <b>=</b>	3 5	> =	71	> =
Dibromochloromethane	13	ר	13	D	_	16	n	13	ח	13	1	3 1.	) <u>=</u>	7 7	) <u>=</u>	2 2	) <u> </u>
1,1-Dichloroethane	13	n	13	n	_	16	U	13	ם	13	'n	31	> =	3 2	) <u>=</u>	7 2	<b>&gt;</b> =
1,2-Dichloroethane	13	Ω	13	ט	_	16	U	13	Ω	13	Ω	31	n	31	) D	27	) =
1,1-Dichloroethene	13	n	13	ב	_	16	U	13	Ŋ	13	n	31	n	3	) =	2	) <b>=</b>
1,2-Dichloroethene (total)	11	-	48			16	Ω	13	D	13	ב	28	, –	; oc	) <u>-</u>	2 6	·
1,2-Dichloropropane	13	D	13	כ	_	16	U	13	n	13	Ŋ	31	Ω	31	n	12	. =
cis-1,3-Dichloropropane	13	n	13	$\supset$		91	Ω	13	Ω	13	D	31	Ω	31	n	12	ם י
trans-1,3-Dichloropropene	13	Þ	13	Þ		16	Ω	13	n	13	U	31	Ω	31	n	12	Ω
Ethylbenzene	13	n	13			16	Ω	13	n	13	D	31	n	31	Ω	12	Ω
2-Hexanone	13	Ω	13	Ω		16	Ω	13	Ŋ	13	ב	31	n	31	Ω	12	'n
4-Methyl-2-pentanone	13	Ω	13	$\Box$		16	Ω	13	Ω	13	D	31	D	31	D	12	ם
Methylene chloride	13	D	13	D		16	Ω	13	Ω	13	n	31	D	31	Ω	12	n
Styrene	13	Ω	13	n		91	n	13	Ω	13	D	31	D	31	Ω	12	n
1,1,2,2-Tetrachloroethane	13	n	13	$\Box$		91	n	13	n	13	Ω	31	Ω	31	n	17	n
Tetrachloroethene	13	D	13	n		91	Ω	13	D	13	Ŋ	31	Ω	31	n	12	'n
Toluene	17		24			91	Ω	13	Ω	13	Ω	31	D	31	n	12	U

TABLE C.13b (Cont.)

								, Lo., 10)		concentration (he) we i mixed Copulation vals	THE PAIN					
Compound	9-0) TO	CLP6 0-6 in.)	CL (6-24	CLP6 (6-24 in.)	CI (0-6	CLP7 (0-6 in.)	CT (6-2	CLP7 (6-24 in.)	CI (24-4	CLP7 (24-48 in.)	CLP8 (0-6 in.)	P8 in.)	CLP8 (0-6	CLP8-RE (0-6 in.)	5 <u>9</u>	CLP9 (0-6 in.)
1,1,1-Trichloroethane	13	Ŋ	13	D	16	n	13	Ω	13	D	31	Ω	31	n	12	D
1,1,2-Trichloroethane	13	Ω	13	כ	16	U	13	Ω	13	Ŋ	31	n	31	Ω	12	n
Trichloroethene	\$	_	5	ſ	16	n	13	n	13	n	31	D	31	Þ	11	,
Vinyl chloride	13	Ω	13	D	16	Ω	13	ח	13	n	31	n	31	D	12	ח
Xylenes (total)	13	D	13	ח	16	Ω	13	ם	13	n	31	Ω	31	Ξ	2	1

<sup>a</sup> Sampled by ANL, analyzed by Westion Gulf Coast, TCL volatile organics-SW846 8240s.

Data qualifiers:

U = analyte was analyzed for but not detected.

B = analyte was found in the associated blank.

J = estimated value.

E = analyte concentration exceeded calibration range of instrument; concentration estimated.

TABLE C.13c Volatile Organics Analyses for Soil Borings, May 1994<sup>a</sup>

		Soil Cc	Soil Concentrations (µg/kg) at Various Depth Intervals	Various Depth Intervals		
Compound	TBSPBOR1 (0-2 ft)	TSBPBOR1-RE (0-2 ft)	TBSPBOR1-Dup (0-2 ft)	TSBPBORI-Dup-RE (0-2 ft)	TBSPBOR1 (2-4 ft)	TSBPBOR1- RE (2-4 ft)
Acetone	350 B*	120 B	17 11*+			11 61
Benzenc				10 B	/3 B	0 7
Bromodichloromethane	12 U*	12 U		12 0	2 J	12 U
Bromoform	12 U*	12 U		·	12 0	12 0
Bromomethane	12 U*	12 U	12 U*+		12 11	12 0
2-Butanone	37 *	12 U	12 U*+	12 U	12 11	0 71
Carbon disulfide	12 U*	12 U	12 U*+	_		12 11
Carbon tetrachloride	12 U*	12 U	12 U*+	12 U	12 11	12 61
Chlorobenzene	12 U*	12 U	12 U*+	12 U	12 11	17 11
Chloroethane	12 U*	12 U	12 U*+	12 U	12 11	0 71
Chloroform	12 U*	12 U	12 U*+	12 U	12 11	12 0
Chloromethane	12 U*	12 U	12 U*+	12 U	,-	12 0
Dibromochloromethane	12 U*	12 U	12 U*+	12 U		12 0
1,1-Dichloroethane	12 U*	12 U	12 U*+	12 U		12 0
1,2-Dichloroethane	12 U*	12 U	12 U*+	12 U	_	12 U
1,1-Dichloroethene	12 U*	12 U	12 U*+	12 U	12 U	12 U
1,2-Dichloroethene (total)		10 J	170 *+	14	7 J	
1,2-Dichloropropane	12 U*	12 U	12 U*+	12 U	12 11	17 11
cis-1,3-Dichloropropene	12 U*	12 U	12 U*+	12 U	12 U	
trans-1,3-Dichloropropene	12 U*	12 U	12 U*+	12 U	12 U	
Ethylbenzene	12 U*	12 U	12 U*+	12 U	12 U	
2-Hexanone	12 U*	12 U	12 U*+	12 U	12 11	
4-Methyl-2-pentanone	12 U*	12 U	12 U*+	12 U	12 11	
Methylenc chloride	4 J*	83	12 U*+	12 U	12 U	
						1

TABLE C.13c (Cont.)

TBSPBOR1 TSBPBOR (0-2 ft) (0-2			Soil Cc	Soil Concentrations (µg/kg) at Various Depth Intervals	Various Depth Intervals		
12   U*   12   U*   12   U*   12   U*   12   U*   12   U*   U*   12   U*   U*   U*   U*   U*   U*   U*   U	Compound	TBSPBOR1 (0-2 ft)	TSBPBOR1-RE (0-2 ft)	TBSPBOR1-Dup (0-2 ft)	TSBPBOR1-Dup-RE (0-2 ft)	TBSPBOR1 (2-4 ft)	TSBPBOR1- RE (2-4 ft)
ene	Styrene	_	12 U	12 1)*+	11 61	17 11	11 61
ethane	1,1,2,2-Tetrachloroethane	* 49	f 6			~ ~	
12   U*   12   U*   12   U*   12   U*   U*   U*   U*   U*   U*   U*   U	Tetrachloroethene	_		12 U*+	12 U	j j	12 11
ethane	Toluene	_	12 U	12 U*+	12 U	12 U	
e 50 * 4 12 12 12 12 12 12 12 12 12 12 12 12 12	1,1,1-Trichloroethane	٦	12 U	12 U*+	12 U	12 U	
ound  TBSPBOR1	1,1,2-Trichloroethane	_	12 U	12 U*+	12 U	12 U	12 U
12 U* 12 U  12 U* 12 U  12 U* 12 U  13 U	Trichloroethene	* 05	4 J	37 *+	10 J	12 U	12 U
TBSPBOR1 TBSPBOR  ound (4-6 ft) (6-8 ft)  280 B 42  12 U 5 J  12 U 12 U  12 U 12 U  12 U 12 U  12 U 12 U  13 U 12 U  14 J  oride 12 U 12 U	Vinyl chloride	_	12 U	12 U*+	12 U	12 U	12 U
TBSPBOR1 TBSPBOF (4-6 ft) (6-8 ft) (6-	Xylenes (total)	7	į	12 U*+	12 U		
TBSPBORI  (4-6 ft)  280 B  12 U  12 U  12 U  12 U  12 U	•		Soil Co	Soil Concentrations (µg/kg) at Various Depth Intervals	'arious Depth Intervals		
(4-6 ft)  280 B  12 U  12 U  12 U  12 U  12 U  12 U		TROBBORI	Tocasar	i a Cado at		1 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	TBSPBOR2-
280 B 42 12 U 5 12 U 12 12 U 12 12 U 12 12 U 12 12 U 12 12 U 12 12 U 12 12 U 12 12 U 12 13 U 12 14 4	Compound	(4-6 ft)	(6-8 ft)	(8-10 ft)	1 BSPBOR1 (10-12 ft)	TBSPBOR2 (0-2 ft)	Dup (0-2 ft)
12 U 5 12 U 12 12 U 12 12 U 12 12 U 12 6 J 4 1	Acetonc		42	42	30	12 U	12 13
12 U 12 12 U 12 12 U 12 12 U 12 6 J 4 1	Benzene	12 U	5 J	12 U	12 U	12 U	
12 U 12 12 U 12 12 U 12 6 J 4	Bromodichloromethane	12 U		12 U	12 U	12 U	
12 U 12 12 U 12 6 J 4	Bromoform	12 U		12 U	12 U	12 U	
12 U 12 6 J 4 12 U 12	Bromomethane	12 U		12 U	12 U	12 U	
6 J 4 12 U 12	2-Butanone	12 U		12 U	12 U	12 U	
12 U 12	Carbon disulfide	f 9	4 J	12 U	12 U	12 U	
	Carbon tetrachloride	12 U		12 U	12 U	12 U	
D 71	Chlorobenzene	12 U	12 U	12 U	12 U	12 U	12 U*

TABLE C.13c (Cont.)

Compound	TBSPBOR1 (4-6 ft)	TBSPBOR1	TBSPBOR1	TBSPBOR1	TBSPBOR2	TBSPBOR2- Dup
		(1100)	(0-10.11)	(10-12 It)	(0-2 ft)	(0-2 ft)
Chloroethane	12 U	12 U	12 U	17 11	17 11	5
Chloroform	12 U	12 U	12 11			
Chloromethane	12 U	12 U	17 11	7 7 7	) ; ) ;	
Dibromochloromethane	12 U		12 11	12 0	12 O	
1,1-Dichloroethane	12 U	12 U	12 U	27 21	0 71	*0 71
1,2-Dichloroethane	12 U	4 J	12 U	12 U	2 2 2	
1,1-Dichloroethene	12 U	15	12 U	12 U	12 21	12 0
1,2-Dichloroethene (total)	89	4,400	1,500	11 I	11 61	_
1,2-Dichloropropane	12 U	12 U	12 U	12 U	12 0	77 11*
cis-1,3-Dichloropropene	12 U	12 U	12 U	12 U	12 11	
trans-1,Dichloropropene	12 U	12 U	12 U	12 U		
Ethylbenzene	12 U	19	12 U	12 U	12 U	
2-Hexanone	12 U	12 U	12 U	12 U	12 11	
4-Methyl-2-pentanone	12 U	12 U	12 U	12 U		
Methylene chloride	12 U	12 U	12 U	12 U		
Styrene	12 U	12 U	12 U	12 11	2 2 2	
1,1,2,2-Tetrachloroethane	9 ј	7 3	009		2 2	_
Tetrachloroethene	12 U	12 U	41	12 11	12 11	*f 7:
Toluene	12 U	S 8	12 U		7 7 6	
1,1,1-Trichloroethane	12 U	12 U	12 U	2 21	0 21	12 04
1,1,2-Trichloroethane	12 U	3 J	140			
Trichloroethene	4 J	370	550	<u> </u>		
Vinyl chloride	19	180	95	11 (1		
Xylenes (total)	,		<b>&gt;</b>		0 71	12 0

TABLE C.13c (Cont.)

Compound	TSBPBOR2-REDup (0-2 ft)	TBSPBOR2 (2-4 ft)	TBSPBOR2 (4-6 ft)	TBSPBOR2 (6-8 ft)	TBSPBOR2 (8-10 ft)	22 -
Acetone	12 U	12 U	97 JB	120	130	~
Benzene	12 U	12 U	120 U	120 U		
Bromodichloromethane	12 U	12 U	120 U	120 U	120	, –
Bromoform	12 U	12 U	120 U	120 U	120	
Bromomethane	12 U	12 U	120 U	120 U	120	, -
2-Butanone	12 U	12 U	120 U	120 U	120 L	
Carbon disulfide	12 U	12 U	120 U	120 U	120 L	_
Carbon tetrachloride	12 U	12 U	120 U	120 U	120 L	_
Chlorobenzene	12 U	12 U	120 U	120 U	120 U	_
Chloroethane	12 U	12 U	120 U	120 U	120 U	<b>-</b>
Chloroform	12 U	12 U	120 U	120 U	120 ה	_
Chloromethane	12 U	12 U	120 U	120 U	120 U	-
Dibromochloromethane	12 U	12 U	120 U	120 U	120 U	-
1,1-Dichloroethane	12 U	12 U	120 U	120 U	120 U	_
1,2-Dichloroethane	12 U	12 U	120 U	120 U	120 U	_
1,1-Dichloroethene	12 U	12 U	120 U	120 U	120 U	_
1,2-Dichloroethene (total)	12 U	12 U	120 J	460	8,400	
1,2-Dichloropropane	12 U	12 U	120 U	120 U	120 U	_
cis-1,3-Dichloropropene	12 U	12 U	120 U	120 U	120 U	
trans-1,3-Dichloropropene	12 U	12 U	120 U	120 U	120 U	-
Ethylbenzene	12 U	12 U	120 U	120 U	120 U	_
2-Hexanone	12 U	12 U	120 U	120 U	120 U	
4-Methyl-2-pentanone	12 U	12 U	120 11	130 11	11 001	

TABLE C.13c (Cont.)

SPBOR2   TBSPBOR2   TBSPBOR2   TBSPB (4-6 ft)   (6-8 ft)   (8-10 (6-8 ft) (6-8 ft) (6-10 (6-10 (6-8 ft) (6-8			Soil C	Soil Concentrations (µg/kg) at Various Depth Intervals	arious Depth Intervals		
oride   12   U   12   U   120   U   U   U   U   U   U   U   U   U	Compound	TSBPBOR2-REDup (0-2 ft)	TBSPBOR2 (2-4 ft)	TBSPBOR2 (4-6 ft)	TBSPBOR2 (6-8 ft)	TBSPBOR2 (8-10 ft)	
12   U   12   U   12   U   120   U   U   U   U   U   U   U   U   U	Methylene chloride	12 U					
hidrocubane 14 130 4,600 24,000 220,000 20,000 cere 12 U 12 U 12 U 12 U 120 U 120 U 120 U 120 Certhane 12 U 12 U 12 U 120 U 120 U 120 U 120 Certhane 12 U 12 U 12 U 120	Styrcne	12 U					
12   U   12   U   12   U   120   U   U   U   U   U   U   U   U   U	1,1,2,2-Tetrachloroethane	14	130				
ochane	Tetrachloroethene	12 U			000,47	720,000	
ethane 12 U 12 U 12 U 120 U 12	Toluene	12 U					
ochane 12 U 12 U 140 150 150 150 150 150 150 150 150 150 15	1,1,1-Trichloroethane						
e         12         U         12         U         350         1,200         21,000           12         U         12         U         12         U         120         U         120           12         U         12         U         120         U         120         U         120           Soil Concentrations (µg/kg) at Various Depth Intervals           Soil Concentrations (µg/kg) at Various Depth Intervals           13         U*         (2-4 ft)         (4-6 ft)         (6-8 ft)         (8-10 ft)           13         U*         (2-4 ft)         (4-6 ft)         (6-8 ft)         (8-10 ft)           13         U*         12         U         58         U         14           13         U*         12         U*         58         U         12	1,1,2-Trichloroethane	12 U					
12   U   12   U   12   U   120   U   U   120   U   U   U   U   U   U   U   U   U	Trichloroethene			350	1 200	1,600	
12   U   12   U   12   U   12   U   120   U   U   120   U   U   U   U   U   U   U   U   U	Vinyl chloride						
HBOR1   HBOR	Kylenes (total)						
13   U*   15   B*   120   U   58   U   14     13   U*   12   U*   120   U   58   U   12     13   U*   12   U*   120   U   58   U   12     13   U*   12   U*   120   U   58   U   12     13   U*   12   U*   120   U   58   U   12     14   U*   U*   U*   U*   U*   U*   U*   U	7	HBORI	HBOR1	HBOR1	HBOR1	HBOR1	HBOR2
13         U*         15         B*         120         U         58         U         14           13         U*         12         U*         120         U         58         U         12           13         U*         12         U*         120         U         58         U         12           ane         13         U*         12         U*         120         U         58         U         12           Iffide         13         U*         12         U*         120         U         58         U         12	Compound	(0-2 ft)	(2-4 ft)	(4-6 ft)	(6-8 ft)	(8-10 ft)	(0-2 ft)
13 U*   12 U*   120 U   58 U   12   12   13   14   12   14   120 U   58 U   12   12   13   14   12   14   120 U   58 U   12   13   14   12   14   120 U   58 U   12   13   14   12   14   120 U   58 U   12   13   14   12   14   120 U   58 U   12   13   14   12   14   120 U   58 U   12   13   14   12   14   120 U   58 U   12   13   14   12   14   13   14   13   14   13   14   13   14   13   14   13   14   14	cetone	13 U*					
Oromethane         13         U*         12         U*         120         U         58         U         12           I3         U*         12         U*         120         U         58         U         12           Iffide         13         U*         12         U*         120         U         58         U         12           Iffide         13         U*         12         U*         120         U         58         U         12           see         13         U*         5         J*         120         U         58         U         12         I           ine         13         U*         12         U*         58         U         12         I           ine         13         U*         12         U*         58         U         12         I           ine         13         U*         12         U*         58         U         12         I           ine         13         U*         12         U*         58         U         12         I           ine         13         U*         12         U         58         U	enzene						
ane 13 U* 12 U* 120 U 58 U 12 13 U* 12 U* 120 U 58 U 12 13 U* 12 U* 120 U 58 U 12 13 U* 12 U* 120 U 58 U 12 13 U* 12 U* 120 U 58 U 12 13 U* 12 U* 120 U 58 U 12 13 U* 12 U* 120 U 58 U 12 14 15 U* 16 U 58 U 17 18 19 19 10 10 10 11 11 12 U* 12 U* 13 U* 14 U* 15 U* 15 U* 16 U 58 U 17 18 19 10 10 11 11 12 U* 12 U* 13 U* 14 U* 15 U* 15 U* 16 U 16 U 17 U 18 U 18 U 19 U 19 U 19 U 19 U 19 U 19 U 19 U 19	romodichloromethane			•			
ane 13 U* 12 U* 120 U 58 U 12 12 U* 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* U* U 58 U 12 U 12 U* U 120 U 58 U 12 U 12 U 12 U 13 U* U* U* U* U* U* U* U* U* U* U* U* U*	romoform						
Hide 13 U* 12 U* 120 U 58 U 12 L 12 U 14 I 120 U 58 U 12 I 12 U 14 I 120 U 58 U 12 I 12 U 12 I 12 U 13 U* I 12 U 14 I 120 U 58 U I 12 U 13 U* I 12 U 14 I 120 U 58 U I 12 U 14 I 12 U 14 I 120 U 58 U I 12 U 14 I 120 U 58 U I 12 U 14 I 120 U 58 U I 12 U 15 U 15 U 15 U 15 U 15 U 15 U 15 U	romomethane			120 U			
ilfide 13 U* 12 U* 120 U 58 U 12 C 12 C 12 C 12 C 12 C 12 C 12 C 12	-Butanone			120 11			
techloride 13 U* 12 U* 120 U 58 U 12 U 120 U 58 U 12 U 12 U 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* 120 U 58 U 12 U 12 U* 12 U* 120 U 58 U 12 U 12 U* U 12 U* U 12 U 12 U 13 U 14 U 15 U 15 U 15 U 15 U 15 U 15 U 15	arbon disulfide			-			
The contraction of the contract	arbon tetrachloride		•	,			
ie 13 $U*$ 12 $U*$ 120 $U$ 58 $U$ 12 13 $U*$ 12 $U*$ 120 $U$ 58 $U$ 12	hlorobenzene		·				
13 U* 12 U* 120 II 58 II	hloroethane						0 21
	hloroform			120 11			O ;

TABLE C.13c (Cont.)

Compound	HBOR1 (0-2 ft)	HBOR1 (2-4 ft)	HBOR1 (4-6 ft)	HBOR1 (6-8 ft)	HBOR1 (8-10 ft)	HBOR2 (0-2 ft)
Chloromethane	13 U*	12 U*	120 U	11 85	17 11	1 61
Dibromochloromethane	13 U*	12 U*	120 U	28 11	12 21	21 2
1,1-Dichloroethane	13 U*	12 U*	120 U	n 85	12 11	12 1
1,2-Dichloroethane	13 U*	12 U*	120 U	58 U	12 U	12 U
1,1-Dichloroethene	13 U*	12 U*	120 U	58 U	12 U	12 U
1,2-Dichloroethene (total)	120 U*	* 11	120 U	58 U	12 U	12 U
1,2-Dichloropropane	13 U*	12 U*	120 U	58 U	12 U	12 U
cis-1,3-Dichloropropene	13 U*	12 U*	120 U	58 U	12 U	12 U
trans-1,3-Dichloropropene	13 U*	12 U*	120 U	58 U	12 U	12 U
Ethylbenzene	13 U*	12 U*	120 U	58 U	12 U	12 U
2-Hexanone	13 U*	12 U*	120 U	58 U	12 U	12 U
4-Methyl-2-pentanone	13 U*	12 U*	120 U	58 U	12 U	12 U
Methylene chloride	13 U*	12 U*	120 U	58 U	12 U	12 U
Styrene	13 U*	12 U*	120 U	58 U	12 U	12 U
1,1,2,2-Tetrachloroethane	18 U*	* 44	120 U	17 J	96	12 U
Tetrachloroethene	13 U*	12 U*	120 U	58 U	12 U	12 U
Toluene	13 U*	12 U*	120 U	58 U	12 U	12 U
I, I, 2-Trichloroethane	13 U*	12 U*	120 U	58 U	12 U	12 U
1,1,1-Trichloroethane	13 U*	12 U*	120 U	09	22	12 U
Trichloroethene	130 U*	*I 6	120 U	58 U	30	, E
Vinyl chloride	13 U*	12 U*	120 U	58 U	12 U	12 U
Xylenes (total)	13 11*	*:1 (1	170 11	11 05	:	

TABLE C.13c (Cont.)

I		Soil Co	Soil Concentrations (µg/kg) at Various Depth Intervals	arious Depth Intervals		
Compound	HBOR2 (2-4 ft)	HBOR2 (4-6 ft)	HBOR2 (6-8 ft)	HBOR2	TBNPBOR1	TBNPBOR1
				(01.01-0)	(0-7 11)	(2-4 It)
Acetone	12 U	12 U	16 B	26 B	12 B	13 B
Benzene	12 U	12 U	12 U	12 11		
Bromodichloromethane	, 12 U	12 U	12 U	2 2		12 0
Bromoform	12 U	12 U	17 11	1 2		0 7 7
Bromomethane	12 U	12 U	12 U	12 0	12 04	
2-Butanone	12 U	12 U	12 U	_		
Carbon disulfide	12 U	12 U	12 U	12 11		0 71
Carbon tetrachloride	12 U	12 11	11 21	2 2		
Chlorobenzene	12 U	12 11	12 21	12 0		
Chloroethane	12 U	12 11	2 2 2	0 :		12 U
Chloroform	12 U		0 21	0 71		12 U
Chloromethane		2 2		17 0	12 U	10 J
Dibromochloromethons	7 7 7		12 U	12 U	12 U	12 U
1 Distriction	17 O	12 U	12 U	12 U	12 U*	12 U
i, I-Dichloroethane	12 U	12 U	12 U	12 U	12 U	12 U
1,2-Dichloroethane	12 U	12 U	12 U	12 U	12 U	12 U
l, I-Dichloroethene	12 U	12 U	12 U	12 U	12 U	11 21
1,2-Dichloroethene (total)	12 U	12 U	12 U	12 U	) 6	
1,2-Dichloropropane	12 U	12 U	12 U	12 U	12 11*	12 11
cis-1,3-Dichloropropene	12 U	12 U	12 U	12 U		_
trans-1,3-Dichloropropene	12 U	12 U	12 U	12 U	•	17 1
Ethylbenzene	12 U	12 U	12 U	17 11		
2-Hexanone	12 U	12 U	12 U	12 11		7 .
4-Methyl-2-pentanone	12 U	12 U	12 U	12 U	,	7 7 6
Methylene chloride	12 U	12 U	12 U	12 U		
Styrene	12 U	12 U	12 U	12 U		
1,1,2,2-Tetrachloroethane	12 U	12 U	12 U	5 J		
					;	`

TABLE C.13c (Cont.)

		Soil C	Soil Concentrations (µg/kg) at Various Depth Intervals	rious Depth Intervals		
7	HBOR2	HBOR2	HBOR2	HBOR2	TBNPBOR1	TBNPBOR1
Compound	(2-4 II)	(4-6 ft)	(6-8 ft)	(8-10 ft)	(0-2 ft)	(2-4 ft)
Tetrachloroethene	12 U	12 U	12 U	12 U	*f 6	19
Toluene	12 U	12 U	12 U	12 U	12 U*	12 U
1,1,1-Trichloroethane	12 U	12 U	12 U	12 U	12 U*	12 U
1,1,2-Trichloroethane	12 U	12 U	12 U	12 U	12 U*	12 U
Trichloroethene	12 U	12 U	12 U	5 J	83 *	230
Vinyl chloride	12 U	12 U	12 U	12 U	12 U	12 U
Xylenes (total)	12 U	12 U	12 U	12 U	12 U*	
•	Soil Concentration	Soil Concentrations (µg/kg) at Various Depth Intervals	Depth Intervals			
	TBNPBOR	TBNPBOR	TRNPROR 1			
Compound	(4-6 ft)	(6-8 ft)	(8-10 ft)			
Acetone	22 B	12 U	12 U			
Benzene	12 U	12 U	12 U			
Bromodichloromethane	12 U	12 U	12 U			
Bromoform	12 U	12 U	12 U			
Bromomethane	12 U	12 U	12 U			
2-Butanone	12 U	12 U	12 U			
Carbon disulfide	12 U	12 U	12 U			
Carbon tetrachloride	12 U	12 U	12 U			
Chlorobenzene	12 U	12 U	12 U			
Chloroethane	12 U	12 U	12 U			
Chloroform	f 6	12 U	12 U			
Chloromethane	12 U	12 U	12 U			
Dibromochlorómethane	12 U	12 U	12 U			

TABLE C.13c (Cont.)

	Soil Concentration	Soil Concentrations (µg/kg) at Various Depth Intervals	Depth Intervals		
Compound	TBNPBOR1 (4-6 ft)	TBNPBOR1 (6-8 ft)	TBNPBOR1 (8-10 ft)		
1,1-Dichloroethane	12 U	12 U	12 U		
1,2-Dichloroethane	12 U	12 U			
1,1-Dichloroethene	12 U	12 U	12 U		
1,2-Dichloroethene (total)	490	190	23		
1,2-Dichloropropane	12 U	12 U	12 U		
cis-1,3-Dichloropropene	12 U	12 U	12 U		
trans-1,3-Dichloropropene	12 U	12 U	12 U		
Ethylbenzene	12 U	12 U	12 U		
2-Hexanone	12 U	12 U	12 U		
4-Methyl-2-pentanone	12 U	12 U	12 U		
Methylene chloride	12 U	12 U	12 U		
Styrene	12 U	12 U			
1,1,2,2-Tetrachloroethane	160	17	, 4		
Tetrachloroethene	2 J	12 U	12 U		
Toluene	12 U	12 U	12 U		
1,1,1-Trichloroethane	12 U	12 U	12 U		
1,1,2-Trichloroethane	12 U	12 U	_		
Trichloroethene	140	58			
Vinyl chloride	12 U	12 U	12 U		
Xylenes (total)	12 U	12 U	n		
		Soil Con	Soil Concentrations (μg/kg) at Various Depth Intervals	Various Depth Intervals	
Compound	VXBOR1 (0-2 ft)	VXBOR1 (4-6 ft)	VXBOR1 (6-8 ft)	VXBOR1 VXBOR1 (8-10 ft) (9.5-10 ft)	VXBOR2 (0-2 ft)
Acetone	12 U	26	14 U	12 U 12 U	11 21
Benzene	12 U	12 U	14 U		

TABLE C.13c (Cont.)

Č	VXBOR1	VXBOR1	VXBOR1	VXBORI	VXBOR1	VXBOR2
Compound	(0-2 tt)	(4-6 ft)	(6-8 ft)	(8-10 ft)	(9.5-10 ft)	(0-2 ft)
Bromodichloromethane	12 U	12 U	14 U	12 U	12 11	17 11
Вготобогт	12 U	12 U	14 U	12 U	12 11	11 61
Bromomethane	12 U	12 U	14 U	12 11	11 61	12 01
2-Butanone	12 U	12 U	14 U		12 C1	12 0
Carbon disulfide	12 U	12 U	14 U	12 17	12 51	21 21 21
Carbon tetrachloride	12 U	12 U	14 U	12 U	12 11	27 21
Chlorobenzene	12 U	12 U	14 U	12 U	12 U	12 11
Chloroethane	12 U	12 U	14 U	12 U	12 U	12 11
Chloroform	12 U	12 U	14 U	12 U	12 U	12 11
Chloromethane	12 U	12 U	14 U	12 U	12 U	12 11
Dibromochloromethane	12 U	12 U	14 U	12 U	12 U	
1,2-Dichloroethane	12 U	12 U	14 U	12 U	12 U	12 U
1,1-Dichloroethane	12 U	12 U	14 U	12 U	12 U	12 U
1,1-Dichloroethene	12 U	12 U	14 U	12 U	12 U	12 [1]
trans-1,2-Dichloroethene	12 U	12 U	14 U	12 U	12 U	12 11
cis-1,2-Dichloroethene	12 U	12 U	14 U	12 U	12 U	12 U
1,2-Dichloropropane	12 U	12 U	14 U	12 U	12 U	12 U
trans-1,3-Dichloropropene	12 U	12 U	14 U	12 U	12 U	12 U
cis-1,3-Dichloropropene	12 U	12 U	14 U	12 U	12 U	12 11
Ethyl benzene	12 U	12 U	14 U	12 U	12 U	12 1
2-Hexanone	12 U	12 U	14 U	12 U	12 U	12 U
4-Methyl-2-pentanone	12 U	12 U	14 U	12 U	12 U	12 U
Methylene chloride	1.7	12 U	10 J	12 U	12 U	12 U
Styrene	12 U	12 U	14 U	12 U	12 U	12 U
1.1,2,2-Tetrachloroethane	11 61	11 61	17 11			

TABLE C.13c (Cont.)

7	VXBOR1	VXBOR1	VXBOR1	VXBOR1	VXBOR1	VXBOR2
Compound	(0-2 ft)	(4-6 ft)	(6-8 ft)	(8-10 ft)	(9.5-10 ft)	(0-2 ft)
Tetrachloroethene	12 U	12 U	14 U	12 11	17 11	5
Toluene	12 U	12 U	14 U			
1,1,2-Trichloroethane	12 U	12 U				0 71
1,1,1-Trichloroethane	12 U	12 11				
Trichloroethene	12 U	î 21		0 71		12 U
Vinyl acetate	12 U			12 0	12 0	
Vinyl chloride	12 U	12 U			7 7 61	
m- and p-Xylene	12 U	12 U				12 0
o-Xylene	12 U	12 U	14 U			
	VXBOR2	VXBOR2	ETROR! ETROPI P	ETBOD 1 D	, d Odaki	
Compound	(2-4 ft)	(4-6 ft)	(0-2 ft)	(0-2 ft)	(2-4 ft)	F1BOK1-Dup (2-4 ft)
Acetone	12 U	12 11	11 11			
Benzene	12 U					0 11
Bromodichloromethane	12 U			17 CI		11 U
Bromoform	12 U		-		12 0	
Bromomethane	12 U	12 U	12 U			
2-Butanone	12 U	12 U	12 U			
Carbon disulfide	12 U	12 U	12 U			
Carbon tetrachloride	12 U	12 U	12 U			11 0
Chlorobenzene	12 U	12 U	12 U	12 U		11 12
Chloroethane	12 U	12 U	, 12 U	12 U		
Chloroform	12 U	12 U	12 U	12 U		
Chloromethane	12 U	12 U	12 U	11 11		: :

TABLE C.13c (Cont.)

Fanouaco, J	VXBOR2	VXBOR2	FTBORI	FTBOR1-Dup	FTBOR1	FTBOR1-Dup
Compound	(11 4-7)	(4-0 It)	(0-2 it)	(0-2 ft)	(2-4 ft)	(2-4 ft)
Dibromochloromethane	12 U	12 U	12 U	12 U	12 U	11 U
1,1-Dichloroethane	12 U	12 U	12 U	12 U	12 U	11 U
1,2-Dichloroethane	12 U	12 U	12 U	12 U	12 U	11 U
1,1-Dichloroethene	12 U	12 U	12 U	12 U	12 U	11 U
cis-1,2-Dichloroethene	12 U	12 U	12 U	12 U	12 U	11 U
trans-1,2-Dichloroethene	12 U	12 U	12 U	12 U	12 U	11 U
1,2-Dichloropropanc	12 U	12 U	12 U	12 U	12 U	11 U
cis-1,3-Dichloropropene	12 U	12 U	12 U	12 U	12 U	11 U
trans-1,3-Dichloropropene	12 U	12 U	12 U	12 U	12 U	11 U
Ethyl benzene	12 U	12 U	12 U	12 U	12 U	11 U
2-Hexanone	12 U	12 U	12 U	12 U	12 U	11 U
4-Methyl-2-pentanone	12 U	12 U	12 U	12 U	12 U	11 U
Methylene chloride	12 U	12 U	12 U	12 U	10 J	11 U
Styrene	12 U	12 U	12 U	12 U	12 U	11 U
1,1,2,2-Tetrachloroethane	12 U	12 U	12 U	12 U	12 U	11 U
Tetrachloroethene	12 U	12 U	12 U	12 U	12 U	11 U
Toluene	12 U	12 U	12 U	12 U	12 U	11 U
1,1,2-Trichloroethane	12 U	12 U	12 U	12 U	12 U	11 U
1,1,1-Trichloroethane	12 U	12 U	12 U	12 U	12 U	11 U
Trichloroethene	12 U	12 U	12 U	12 U	12 U	111 U
Vinyl acetate	12 U	12 U	12 U	12 U	12 U	111 U
Vinyl chloride	12 U	12 U	12 U	12 U	12 U	11 U
m- and p-Xylene	12 U	12 U	12 U	12 U	12 U	11 U
o-Xylene						

TABLE C.13c (Cont.)

1	Soil	Concentrations (µg/kg)	Soil Concentrations (µg/kg) at Various Depth Intervals	S
	FTBOR1	FTBOR1	FTBOR1	FTBOR1
Compound	(4-6 ft)	(6-8 ft)	(8-10 ft)	(10-12 ft)
Acetone	12 U	12 U	17 B	11 61
Benzene	12 U	12 U	14 U	12 U
Bromodichloromethane	12 U	12 U	14 U	12 U
Bromoform	12 U	12 U	14 U	12 U
Bromomethane	12 U	12 U	14 U	
2-Butanone	12 U	12 U	14 U	
Carbon disulfide	12 U	12 U	14 U	12 U
Carbon tetrachloride	12 U	12 U	14 U	12 U
Chlorobenzene	12 U	12 U	14 U	12 11
Chloroethane	12 U	12 U	14 U	12 11
Chloroform	12 U	12 U	14 U	ŕ
Chloromethane	12 U	12 U	14 U	
Dibromochloromethane	12 U	12 U	14 U	
1,1-Dichloroethane	12 U	12 U	14 U	
1,2-Dichloroethane	12 U	12 U	14 U	_
1,1-Dichloroethene	12 U	12 U	14 U	
cis-1,2-Dichloroethene	12 U	12 U	14 U	12 U
trans-1,2-Dichloroethene	12 U	12 U	14 U	12 U
1,2-Dichloropropane	12 U	12 U	14 U	
cis-1,3-Dichloropropene	12 U	12 U	14 U	12 U
trans-1,3-Dichloropropene	12 U	12 U	14 U	12 U
Ethyl benzene	12 U	12 U	14 U	12 U
2-Hexanone	12 U	12 U	14 U	12 U
4-Methyl-2-pentanone	12 U	12 U	14 U	12 U
Methylene chlaride	12 U	12 U	14 U	12 U
Styrene	12 U	12 U	14 U	12 U

TABLE C.13c (Cont.)

	FTBOR1	FTBOR1	FTBOR1	FTBOR1
Compound	(4-6 ft)	(6-8 ft)	(8-10 ft)	(10-12 ft)
1,1,2,2-Tetrachloroethane	12 U	12 U	5 J	12 U
Tetrachloroethene	12 U	12 U	14 U	12 U
Toluene	12 U	12 U	14 U	12 U
1,1,1-Trichloroethane	12 U	12 U	14 U	12 U
I, 1, 2-Trichloroethane	12 U	12 U	14 U	12 U
Trichloroethene	12 U	12 U	14 U	12 U
Vinyl acetate	12 U	12 U	NT	12 U
Vinyl chloride	12 U	12 U	14 U	12 U
m- and p-Xylene	12 U	12 U	14 U	12 U
o-Xylene	12 U	12 U	14 U	12 U

<sup>&</sup>lt;sup>a</sup> Sampled by ANL, analyzed by Weston Gulf Coast or ANL/ACL, CLP/HSL volatile organics.

U = analyte was analyzed for but not detected; detection limits given.

= internal standard area counts were outside QC limits.

= estimated value.

= analyte was found in the associated blank.

= surrogate recoveries were outside QC limits.

NT = not tested.

TABLE C.13d Volatile Organics Analyses for Soil Borings, May 1995<sup>a</sup>

	Soi	Soil Concentrations (µg/kg) at Various Depth Intervals	at Various Depth Inter	vals
Compound	HBOR4 (4-6 ft)	HBOR4 (6-8 ft)	HBOR4 (8-10 ft)	HBOR4 (14-16 ft)
Acetone	29	27	2.100	780 *F
Benzene	12.1 U	11.8 U	56.8 U	
Bromodichloromethane	12.1 U	11.8 U	56.8 U	·
Bromoform	12.1 U	11.8 U	56.8 U	_
Bromomethane	12.1 U	11.8 U	56.8 U	12.3 UJ
2-Butanone	12.1 U	11.8 U	S6.8 U	·
Carbon disulfide	12.1 U	11.8 U	56.8 U	•
Carbon tetrachloride	12.1 U	11.8 U	56.8 U	,
Chlorobenzene	12.1 U	11.8 U	56.8 U	
Chloroethane	12.1 U	11.8 U	56.8 U	
Chloroform	12.1 U	11.8 U	56.8 U	_
Chloromethane	12.1 U	11.8 U	56.8 U	
Dibromochloromethane	12.1 U	11.8 U	56.8 U	
1,1-Dichloroethane	12.1 U	11.8 U	56.8 U	_
1,2-Dichloroethane	12.1 U	11.8 U	56.8 U	
1,1-Dichloroethene	12.1 U	11.8 U	56.8 U	12.3 UJ
1,2-Dichloroethene (total)	12.1 U	11.8 U	56.8 U	450 *E
1,2-Dichloropropane	12.1 U	11.8 U	56.8 U	12.3 UJ
cis-1,3-Dichloropropene	12.1 U	11.8 U	56.8 U	12.3 UJ
trans-1,3-Dichloropropene	12.1 U	11.8 U	56.8 U	
Ethyl benzene	12.1 U	11.8 U	56.8 U	12.3 UJ
2-Hexanone	12.1 U	11.8 U	56.8 U	12.3 UJ

TABLE C.13d (Cont.)

	HBOR4	HBOR4	HBOR4	HBOR4	
Compound	(4-6 ft)	(6-8 ft)	(8-10 ft)	(14-16 ft)	
4-Methyl-2-pentanone	12.1 U	11.8 U	56.8 U	12.3 UJ	
Methylene chloride	6 BJ	4 BJ	38 J	£ 8	
Styrene	12.1 U	11.8 U	56.8 U	12.3 UJ	
1,1,2,2-Tetrachloroethane	12.1 U	11.8 U	56.8 U	480 @JE	
Tetrachloroethene	12.1 U	11.8 U	56.8 U	32 J	
Toluene	12.1 U	11.8 U	56.8 U	12.3 UJ	
1,1,1-Trichloroethane	12.1 U	11.8 U	56.8 U	12.3 UJ	
1,1,2-Trichloroethane	12.1 U	11.8 U	56.8 U	57 J	
Trichloroethene	12.1 U	11.8 U	56.8 U	450 *E	
Vinyl chloride	12.1 U	11.8 U	56.8 U	12.3 UJ	
Xylenes (total)	12.1 U	11.8 U	56.8 U	12.3 UJ	
		Soil Concentration	Soil Concentrations (µg/kg) at Various Depth Intervals	s Depth Intervals	
	TBNPBOR2	TBNPBOR2	TBNPBOR2	TBNPBOR3	TBNPBOR3
Compound	(0-2 ft)	(2-6 ft)	(10-12 ft)	(0-2 ft)	(2-4 ft)
Acetone	10 J	6 J	1020 B	67 B	210 J*
Benzene	12 U	11.6 U	61.7 U	12 U	67.6 UJ*
Bromodichloromethane	12 U	11.6 U	61.7 U	12 U	67.6 UJ*
Bromoform	12 U	11.6 U	61.7 U	12 U	67.6 UJ*
Bromomethane	12 U	11.6 U	61.7 U	12 U	67.6 UJ*

TABLE C.13d (Cont.)

Compound	TBNPBOR2 (0-2 ft)	TBNPBOR2 (2-6 ft)	TBNPBOR2 (10-12 ft)	TBNPBOR3 (0-2 ft)	TBNPBOR3 (2-4 ft)
Carbon disulfide	2 J	11.6 11	61.7 11	1 7	
Carbon tetrachloride	12 U		61.7 U	12 11	*III 9 L9
Chlorobenzene	12 U	11.6 U	61.7 U	12 U	
Chloroethane	12 U	11.6 U	61.7 U	12 U	_
Chloroform	12 U	11.6 U	61.7 U	12 U	
Chloromethane	12 U	11.6 U	61.7 U	12 U	•
Dibromochloromethane	12 U	11.6 U	61.7 U	12 U	-
1,1-Dichloroethane	12 U	11.6 U	61.7 U	12 U	•
1,2-Dichloroethane	12 U	11.6 U	61.7 U	12 U	
1,1-Dichloroethene	12 U	11.6 U	61.7 U	12 U	67.6 UJ*
1,2-Dichloroethene (total)	4 J	11.6 U	61.7 U	22	
1,2-Dichloropropane	12 U	11.6 U	61.7 U	12 U	•
cis-1,3-Dichloropropene	12 U	11.6 U	61.7 U	12 U	
trans-1,3-Dichloropropene	12 U	11.6 U	61.7 U	12 U	
Ethyl benzene	12 U	11.6 U	61.7 U	12 U	
2-Hexanone	12 U	11.6 U	61.7 U	12 U	
4-Methyl-2-pentanone	12 U	11.6 U	61.7 U	12 U	
Methylene chloride	10 BJ	8 BJ	31 BJ	11.9 BJ	
Styrene	12 U	11.6 U	61.7 U	12 U	
1,1,2,2-Tetrachloroethane	12 U	11.6 U	61.7 U	12 U	
Tetrachloroethene	6 J	11.6 U	61.7 U	2 J	
Toluene	17 11	11 / 11			

TABLE C.13d (Cont.)

		Soil Concentrations (µg/kg) at Various Depth Intervals	s (µg/kg) a	-					
Compound	TBNPBOR2 (0-2 ft)	TBNPBOR2 (2-6 ft)	TBNPBOR2 (10-12 ft)	BNPBOR2 (10-12 ft)	TBNPBOR3 (0-2 ft)	NPBOR3 (0-2 ft)		TBNPBOR3 (2-4 ft)	30R3 ft)
1,1,1-Trichloroethane	12 U	11.6 U	61.7	D	12	Ω		9.79	nı*
1,1,2-Trichloroethane	12 U	11.6 U	61.7	U	12	Ω		9.79	m*
Trichloroethene	f 8	11.6 U	61.7	Ω	13			105	*,
Vinyl chloride	12 U	11.6 U	61.7	n	12	Ω		37	*
Xylenes (total)	12 U	11.6 U	61.7	n	12	U	1	67.6	*101
	Soi	Soil Concentrations (µg/kg) at Various Depth Intervals	/kg) at Var	ious Dep	th Interva	ls			
	TBNPBOR3	TBNI	TBNPBOR3	TBNF	TBNPBOR3	Г	TBNPBOR3	OR3	
Compound	(4-6 ft)	9)	(6-8 ft)	(8-1	(8-10 ft)		(10-12 ft)	ft)	
Acetone	70.5 UJ	1,500		6,000	<b>©</b>		140	В	
Benzene	51 J	19		63.3	Ω		61.5	UJ	
Bromodichloromethane	70.5 UJ	59.5	Ω	63.3	Ω		61.5	UJ	
Bromoform	70.5 UJ	59.5	Ω	63.3	D		61.5	UJ	
Bromomethane	70.5 UJ	59.5	Ω	63.3	Ω		61.5	UJ	
2-Butanone	70.5 UJ	59.5	Ŋ	63.3	Ω		61.5	UJ	
Carbon disulfide	70.5 UJ	59.5	n	63.3	Ω		61.5	UJ	
Carbon tetrachloride	70.5 UJ	59.5	n	63.3	Ω		61.5	UJ	
Chlorobenzene	23,000 ~J,E	009,6	<u>©</u>	140		_	1,000	<u>, , , , , , , , , , , , , , , , , , , </u>	
Chloroethane	70.5 UJ	59.5	D	63.3	Ω		61.5	UJ	
Chloroform	l 61	59.5	D	63.3	Ω		61.5	Ω	
Chloromethane	70.5 UJ	59.5	n	63.3	n		61.5	UJ	

TABLE C.13d (Cont.)

Compound	TBNPBOR3 (4-6 ft)	TBNPBOR3	TBNPBOR3	TBNPBOR3
		(2100)	(0-10.11)	(10-171)
Dibromochloromethane	70.5 UJ	59.5 U	63.3 U	615 111
1,1-Dichloroethane	70.5 UJ	59.5 U	63.3 U	61.5 111
1,2-Dichloroethane	70.5 UJ	59.5 U	63.3 U	
1,1-Dichloroethene	70.5 UJ	59.5 U	63.3 U	,
1,2-Dichloroethene (total)	f 086	2,000	230	
1,2-Dichloropropane	70.5 UJ	59.5 U	63.3 U	61.5 UJ
trans-1,3-Dichloropropene	70.5 UJ	59.5 U	63.3 U	
cis-1,3-Dichloropropene	70.5 UJ	59.5 U	63.3 U	_
Ethyl benzene	250 J	6,600 @	06	
2-Hexanone	70.5 UJ	59.5 U	63.3 U	61.5 UJ
4-Methyl-2-pentanone	70.5 UJ	59.5 U	63.3 U	_
Methylene chloride	23 BJ	59.5 U	27 BJ	
Styrene	70.5 UJ	59.5 U	63.3 U	
1,1,2,2-Tetrachloroethane	70.5 UJ	59.5 U	63.3 U	,
Tetrachloroethene	70.5 UJ	750	10 J	
Toluene	320 J	4,200 @	55 J	099
1,1,1-Trichloroethane	70.5 UJ	59.5 U	63.3 U	61.5 111
.,1,2-Trichloroethane	70.5 UJ	59.5 U	63.3 U	_
Trichloroethene	58 J	59.5 U	63.3 U	_
Vinyl chloride	750 J	59.5 U	63.3 U	-
Xylenes (total)	1 089	0000	(	

TABLE C.13d (Cont.)

7	VXBOR3	VXBOR4	VXBOR4	VXBOR5	VXBOR5
Compound	(2-4 tt)	(2-4 ft)	(4-6 ft)	(0-2 ft)	(4-6 ft)
Acetone	18	130 J	260	12 U	160
Benzene	42	63.6 UJ	61.2 U	12 U	65.1 U
Bromodichloromethane	. 12 U	63.6 UJ	61.2 U	12 U	65.1 U
Bromoform	12 U	63.6 UJ	61.2 U	12 U	65.1 U
Bromomethane	12 U	63.6 UJ	61.2 U	12 U	65.1 U
2-Butanone	12 U	63.6 UJ	61.2 U	12 U	65.1 U
Carbon disulfide	11 J	63.6 UJ	61.2 U	2 J	65.1 U
Carbon tetrachloride	12 U	63.6 UJ	61.2 U	12 U	65.1 U
Chlorobenzene	12 U	63.6 UJ	61.2 U	12 U	65.1 U
Chloroethane	12 U	63.6 UJ	61.2 U	12 U	65.1 U
Chloroform	12 U	63.6 UJ	61.2 U	12 U	65.1 U
Chloromethane	12 U	63.6 UJ	61.2 U	12 U	65.1 U
Dibromochloromethane	12 U	63.6 UJ	61.2 U	12 U	65.1 U
1,1-Dichloroethane	12 U	63.6 UJ	61.2 U	12 U	65.1 U
1,2-Dichloroethane	12 U	63.6 UJ	61.2 U	12 U	65.1 U
1,1-Dichloroethene	12 U	63.6 UJ	61.2 U	12 U	65.1 U
,2-Dichloroethene (total)	5 J	63.6 UJ	61.2 U	12 U	65.1 U
1,2-Dichloropropane	12 U	63.6 UJ	61.2 U	12 U	65.1 U
cis-1,3-Dichloropropene	12 U	63.6 UJ	61.2 U	12 U	65.1 U
trans-1,3-Dichloropropene	12 U	63.6 UJ	61.2 U	12 U	65.1 U
Ethyl benzene	58	1,300 J@E	2,900 @	12 U	65.1 U
2-Hexanone	12 U	111 9 89	11 (17		,

TABLE C.13d (Cont.)

- Function C	VXBOR3	VXBOR4	ΛX	VXBOR4	VXBOR5		VXBOR5
Compound	(2-4 ft)	(2-4 ft)	4)	(4-6 ft)	(0-2 ft)	(4	(4-6 ft)
4-Methyl-2-pentanone	12 U	63.6 UJ	61.2	11	17 11	189	1
Methylene chloride	12 U	53 J	52			30	) <u>-</u>
Styrene	12 U	63.6 UJ	61.2	. []	17 11	65.1	, <u> </u>
1,1,2,2-Tetrachloroethane	12 U	63.6 UJ	61.2	n		65.1	) <u> </u>
Tetrachloroethene	12 U	63.6 UJ	61.2	Ω	12 U	17	) —
Toluene	12 U	63.6 UJ	61.2	Ω		65.1	. [
1,1,1-Trichloroethane	12 U	63.6 UJ	61.2	Ω	12 U	65.1	) =
1,1,2-Trichloroethane	12 U	63.6 UJ	61.2	Ω		65.1	) =
Trichloroethene	15	63.6 UJ	61.2	Ω		× ×	<b>-</b>
Vinyl chloride	12 U	63.6 UJ	61.2	11	·	65.1	, <u> </u>
Xylenes (total)	27	1.200 J	2 300	ı		1.00	> ;
	So	Soil Concentrations (µg/kg) at Various Depth Intervals		th Intervals	İ		
			*				
Compound	VXBOR5 (6-8 ft)	VXBOR5 (8-10 ft)	VXBOR5 (10-12 ft)	VXBOR5		VXBOR5	
			(11 71-01)	(17-14-11)		(14-10 II)	
Acetone	20	210	14	\$ 550 *	1,100	0	
Benzene	12.4 U	12.4 U	12.5 U	12.5 U	61.8	11 8	
Bromodichloromethane	12.4 U	12.4 U	12.5 U	12.5 U	61.8	) I	
Bromoform	12.4 U	12.4 U	12.5 U	,	61.8		
Bromomethane	12.4 U	12.4 U	, -		61.8		
2-Butanone	11 7 C1				.10		
2-Dulallone							

TABLE C.13d (Cont.)

	VXBOR5	VXBOR5	VXBOR5	VXBOR5	VXBOR5
Compound	(6-8 ft)	(8-10 ft)	(10-12 ft)	(12-14 ft)	(14-16 ft)
Carbon disulfide	12.4 U	12.4 U	12.5 U	11 J	61.8 U
Carbon tetrachloride	12.4 U	12.4 U	12.5 U	12.5 U	
Chlorobenzene	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
Chloroethane	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
Chloroform	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
Chloromethane	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
Dibromochloromethane	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
1,1-Dichloroethane	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
1,2-Dichloroethane	12.4 U	12.4 U	12.5 U	2 J	61.8 U
1,1-Dichloroethene	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
1,2-Dichloroethene (total)	2 J	12.4 U	3 J	3 3	61.8 U
1,2-Dichloropropane	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
cis-1,3-Dichloropropene	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
trans-1,3-Dichloropropene	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
Ethyl benzene	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
2-Hexanone	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
4-Methyl-2-pentanone	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
Methylene chloride	6 BJ	7 BJ	11 BJ	8 BJ	28 J
Styrene	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
1,1,2,2-Tetrachloroethane	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
Fetrachloroethene	5 J	3 J	4 J	12.5 U	61.8 U
Toluene	12.4 U	12.4 U	12.5 U	12.5 U	61.8 11

TABLE C.13d (Cont.)

	So	Soil Concentrations (µg/kg) at Various Depth Intervals	/kg) at Various Dep	oth Intervals	
Compound	VXBOR5 (6-8 ft)	VXBOR5 (8-10 ft)	VXBOR5 (10-12 ft)	VXBOR5 (12-14 ft)	VXBOR5 (14-16 ft)
1,1,1-Trichloroethane	12.4 U	12.4 U	12.5 U	12.5 U	618 11
1,1,2-Trichloroethane	12.4 U	12.4 U	12.5 U	12.5 U	61.8
Trichloroethene	12.4 U	12.4 U	12.5 U		61.8
Vinyl chloride	12.4 U	12.4 U	12.5 U	3 J	61.8 U
Xylenes (total)	12.4 U	12.4 U	12.5 U	12.5 U	61.8 U
	Soil Concentrations (µg/kg) at 0-6 in.	ug/kg) at 0-6 in.	Ī		
Compound	SQPS 5	SQPS-8	ı		
Acetone	es.	13 BJ*			
Benzene	12 U	24 U			
Bromodichloromethane	12 U	24 U			
Bromoform	12 U	24 U			
Bromomethane	12 U	24 U			
2-Butanone	12 U	24 U			
Carbon disulfide	12 U	38 *			
Carbon tetrachloride	12 U	24 U			
Chlorobenzene	12 U	24 U			
Chloroethane	12 U	24 U			
Chloroform	12 U	24 U			
Chloromethane	12 U	24 U			

TABLE C.13d (Cont.)

Compound	IC S	4 00	200	0
Compound	DC	ours 5	SQFS-8	δ-ς
Dibromochloromethane	12	U	24	n
1,1-Dichloroethane	12	Ω	24	Ω
1,2-Dichloroethane	12	n	24	n
1,1-Dichloroethene	12	Ω	24	D
1,2-Dichloroethene (total)	12	Ω	24	Ω
1,2-Dichloropropane	12	Ω	24	Ω
cis-1,3-Dichloropropene	12	Ω	24	Ω
trans-1,3-Dichloropropene	12	Ω	24	Ω
Ethyl benzene	12	Ω	24	n
2-Hexanone	12	Ω	24	Ω
4-Methyl-2-pentanone	12	Ω	24	D
Methylene chloride	9		11	BJ*
Styrene	12	Ω	24	D
1,1,2,2-Tetrachloroethane	12	U	24	n
Tetrachloroethene	12	Ω	24	Ω
Toluene	12	Ω	24	n
1,1,1-Trichloroethane	12	Ω	24	Ω
1,1,2-Trichloroethane	12	Ω	24	Ω
Trichloroethene	12	Ω	24	Ω
Vinyl chloride	12	Ω	24	Ω
Xylenes (total)	12		č	,

See next page for footnotes.

## TABLE C.13d (Cont.)

<sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limits given.

B = analyte was found in the associated blank.

= estimated value.

@ = value obtained from a 1:125 dilution.

E = concentration exceeded the calibration range of the instrument.

= QC result outside the laboratory control limits.

~ = value obtained from a 1:100 dilution.

TABLE C.14a Semivolatile Organics Analyses for Soil Borings, January 1993<sup>a</sup>

Compound	JBP1-E-2' (2 ft)	-2,	JBP1-E-2'Dup (2 ft)	John John John John John John John John	JBP1-E-4' (4 ft)	E-4'	JBP1-W-2' (2 ft)	W-2'	JBP1-W-4 (4 ft)	.W-4'	JBP2-E-2' (2 ft)	E-2'	JBP2-E-4 (4 ft)	4.
Acenaphthene	402	Ω	392	Ω	402	n	402	n	409	Ω	396	ñ	396	=
Acenaphthylene	402	Ω	392	Ω	402	Ω	402	n	409	n	396	'n	396	o D
Anthracene	402	n	392	n	402	n	402	Ω	409	n	396	n	396	n
Benz[a]anthracene	234	ī	392	Ω	402	Ω	402	n	409	Ω	396	n	396	n
Benzoic acid	2,010	Ω	1,960	Ω	2,010	Ω	2,010	Ω	2,050	n	1,980	Ω	1,980	) D
Benzo[a]pyrene	137	_	392	Ω	402	Ω	402	Ω	409	Þ	396	Ω	396	Þ
Benzo[b]fluoranthene	221	Ţ	392	ב	402	Ω	402	n	409	n	396	Ω	396	D
Benzo[g,h,i]perylene	402	n	392	n	402	n	402	Ω	409	Ω	396	Ω	396	Ω
Benzo[k]fluoranthene	92	_	392	Ω	402	n	402	D	409	Ω	396	Ω	396	n
Benzyl alcohol	805	n	785	Ω	805	D	805	D	818	Ω	792	n	792	n
Bis(2-chloroethoxy)methane	402	D	392	ח	402	Ω	402	Ω	409	Ω	396	Ω	396	n
Bis(2-chloroethyl)ether	402	D	392	Ω	402	ì	402	Þ	409	n	396	Ŋ	396	n
Bis(2-chloroisopropyl)ether	402	n	392	ב	402	Ω	402	$\supset$	409	n	396	Ω	396	Ω
Bis(2-ethylhexyl)phthalate	402	n	392	Ω	402	Ω	402	Ω	409	D	396	Ω	396	D
4-Bromophenyl phenyl ether	402	n	392	n	402	Ω	402	n	409	n	396	n	396	Þ
Butyl benzyl phthalate	402	Ω	392	Ŋ	402	D	402	Ω	409	D	396	Ω	396	D
4-Chloro-3-methylphenol	805	n	785	n	805	Ω	805	D	818	Ω	792	n	792	D
4-Chloroaniline	805	n	785	ב	805	Ω	805	n	818	Ω	792	C	792	n
2-Chloronaphthalene	402	n	392	Ω	402	Ω	402	Ω	409	D	396	Ω	396	ב
2-Chlorophenol	405	n	392	n	402	n	402	n	409	Ω	396	Ω	396	n
4-Chlorophenyl phenyl ether	402	Ω	392	n	402	Ω	402	n	409	D	396	n	396	$\supset$
Chrysene	335	ſ	392	Ω	402	Ω	402	Ω	409	n	396	Ω	396	n
Di-n-butylphthalate	402	Ω	392	Ω	402	n	402	Ω	409	n	77	<u>-</u>	396	n
Di-n-octylphthalate	402	n	392	Ω	402	Ω	402	D	409	D	396	Ω	396	n
Dibenz[a,h]anthracene	402	n	392	Ω	402	Ω	402	Ω	409	n	396	n	396	n
Dibenzofuran	402	D	392	Ω	402	Ω	402	n	409	<b>5</b>	396	n	396	b

TABLE C.14a (Cont.)

				ļ	Boreho	le Conce	Borehole Concentrations (µg/kg) at Various Depths	ug/kg)	t Various	Depth					
Compound	JBP1-E-2'	3-2,	JBP1-E-2'Dup (2 ft)	2'Dup	JBP (4	JBP1-E-4' (4 ft)	JBP1	JBP1-W-2' (2 ft)	JBP1-W-4' (4 ft)	P1-W-4' (4 ft)	JBP2-E-2 (2 ft)	.E-2'	JBP2-E-4 (4 ft)	.4.	1
1,2-Dichlorobenzene	402	Ω	392	n	402	Ω	402	Ξ	400	=	306	1		:	1
1,3-Dichlorobenzene	402	Ω	392	n	402	n	402	n	409	) <u>=</u>	396	) <u>=</u>	306	) <u>:</u>	
1,4-Dichlorobenzene	402	Ω	392	n	402	n	402	Ω	409	) <b>=</b>	30%	) <u>=</u>	305	) <b>:</b>	
3,3'-Dichlorobenzidine	805	n	785	Ω	805	n	805	Ω	818	) <b>=</b>	797	) <u>=</u>	060	> =	
2,4-Dichlorophenol	402	Ω	392	Ω	402	D	402	'n	409	) D	366	> =	90E	> =	
Diethylphthalate	402	Ω	392	n	402	D	402	Ω	409	D	396	) <u>=</u>	30%	) =	
2,4-Dimethylphenol	402	n	392	n	402	Ω	402	Ω	409	D	396	n	396	) <u>=</u>	
Dimethylphthalate	402	n	392	n	402	Ω	402	Ω	409	Ω	396	Ω	396	ì	
4,6-Dinitro-2-methylphenol	2,010	Ω	1,960	n	2,010	Ω	2,010	Ω	2,050	Ω	1,980	'n	1.980	) <u>=</u>	
2,4-Dinitrophenol	2,010	Ω	1,960	Ω	2,010	Ω	2,010	Ω	2,050	Ω	1,980	n	1.980	) <u>=</u>	
2,4-Dinitrotoluene	402	Ω	392	Ω	402	D	402	Ω	409	Ω	396	n	396	) =	
2,6-Dinitrotoluene	402	Ω	392	Ω	402	ಗ	402	Ω	409	D	396	n	396	) <u>=</u>	
Fluoranthene	306	ĭ	392	Ω	402	n	402	Ω	409	Ω	396	n	396	> =	
Fluorene	402	Ω	392	Ω	402	U	402	Ω	409	Ω	396	n	396	) <u>=</u>	
Hexachlorobenzene	402	n	392	Ω	402	Ω	402	Ω	409	Ω	396	'n	396	> =	
Hexachlorobutadiene	402	Ω	392	n	402	Û	402	n	409	D	396		306	) <u>=</u>	
Hexachlorocyclopentadiene	405	Ω	392	n	402	Ω	402	n	409	D	368	> =	306	) <u> </u>	
Hexachloroethane	402	Ω	392	Ω	402	n	402	Ω	490	ı D	396	) <u>=</u>	306	) <u> </u>	
Indeno[1,2,3-c,d]pyrene	402	Ω	392	Ω	402	Ω	402	Ω	409	ח	396	) <u>=</u>	306	) <u>=</u>	
Isophorone	402	Ω	392	n	402	U	402	Ω	409	n	396	) <u>=</u>	30%	) <u>=</u>	
2-Methylnaphthalene	471		392	n	402	n	402	Ω	409		396	; =	30%	) <u>=</u>	
2-Methylphenol	402	U	392	Ω	402	n	402	D	409	· =	366	) <b>=</b>	306	> =	
4-Methylphenol	405	D	392	Ω	402	Ω	402	ņ	409	, 1	366	> =	306	> =	
Naphthalene	107	ſ	392	Ω	402	Ω	402	Ω	409	<u> </u>	366	) <u>=</u>	306	) <u>=</u>	
2-Nitroaniline	2,010	Ω	1,960	n	2,010	Ω	2,010	Ω	2.050	· =	1.980	) <u>=</u>	1 080	) <u>=</u>	
3-Nitroaniline	2,010	Ω	1,960	Ω	2,010	Ω	2,010	Ω	2,050	n	1,980	n D	1,980	) D	

TABLE C.14a (Cont.)

Compound														
	JBP1-E-2' (2 ft)	-2,	JBP1-E-2'Dup (2 ft)	2'Dup	JBP (4	JBP1-E-4' (4 ft)	JBP1-W-2' (2 ft)	W-2' ft)	JBP1-W-4'	W-4'	JBP2-E-2' (2 ft)	.E-2'	JBP2-E-4' (4 ft)	4
4-Nitroaniline	2,010	n	1,960	n	2.010	Þ	2.010	1	2.050	ī	1 980	=	1 080	=
Nitrobenzene	402	Ω	392	Ω	402	- =	407	> =	409	) <u>=</u>	306	)	1,700	) :
2-Nitrophenol	402	Ω	392	Ω	402		402	> =	400	) <u>=</u>	306	) <u>=</u>	305	) <u>;</u>
4-Nitrophenol	2,010	Ω	1,960	ר ה	2.010	o D	2.010	) <u>=</u>	2.050	> =	1 980	o <u>t</u>	1 080	) =
N-Nitroso-di-n-propylamine	402	Ω	392	Ω	402	n	402	'n	409	o D	396	> =	396	) <u>=</u>
N-Nitrosodiphenylamine	402	n	392	n	402	n	402	n	409	n	396	o D	396	) =
Pentachlorophenol	2,010	n	1,960	n	2,010	Ω	2,010	n	2,050	Ω	1,980	ר	1.980	
Phenanthrene	376	_	392	n	402	Ω	402	Ω	409	Ω	396	Ω	396	n
Phenol	402	n	392	⊃	402	Ω	402	D	409	'n	396	n	396	n
Pyrene	280		392	Ω	405	n	402	n	409	Ω	396	D	396	þ
1,2,4-Trichlorobenzene	402	n	392	n	402	ח	402	Ω	409	Ω	396	ח	396	D
2,4,5-Trichlorophenol	402	n	392	n	402	Ω	402	n	409	Þ	396	D	396	כ
2,4,6-Trichlorophenol	402	U	392		402	ח	402	n	409	n	396	Ω	396	Ω
					Borehole	Concen	trations (µ	g/kg) at	Borehole Concentrations (µg/kg) at Various Depths	epths				
	JBP2-E-4'Dup	Jup	JBP2-C-2'	-2	JBP2-C-4'	-C-4	JBP2-W-2'	N-2'	JBP2-W-4'	V4'	JHDP-C-4	C-4.	JHDP-C-6'	-6,
Compound	(4 ft)		(2 ft)		4)	(4 ft)	(2 ft)		(4 ft)		(4 ft)		(6 ft)	
Acenaphthene	392	Ŋ	455	Þ	392	ח	396	n	392	Þ	396	ח	402	
Acenaphthylene	392	Ω	455	D	392	D	396	Ω	392	D	396	n	402	D
Anthracene	392	Ω	455	Ω	392	n	396	n	392	n	396	Ω	402	ח
Benz[a]anthracene	392	Ω	455	n	392	Ω	396	n	392	n	396	Ω	402	ב
Benzo[a]pyrene	392	Ω	455	n	392	ח	396	n	392	n	396	Ð	402	כ
Benzo[b]fluoranthene	392	Ω	455	Ω	392	D	396	Ω	392	n	396	Ω	402	n
Benzo[g.h.i]perylene	392	D	455	n	392	נ	396	Ω	392	Ω	396	ח	402	Ω
Benzoic acid	1,960	n	2,280	n	1,960	n	118	_	1,960	D	175	<b>-</b>	74	_

TABLE C.14a (Cont.)

				Borehol	Borehole Concentrations (µg/kg) at Various Depths	rations (	ıg/kg) at	Various	Depths					l
Compound	JBP2-E-4'Dup (4 ft)		JBP2-C-2' (2 ft)	JBPZ	JBP2-C-4' (4 ft)	JBP2-W-2'	.W-2'	JBP2-W-4' (4 ft)	W-4' t)	JHDP-C-4' (4 ft)	C-4'	JHDP-C-6'	9.0	1
Benzo[k]fluoranthene	392 U	455	D	392	1	306	=	303	=	ò	:		,	1
Benzyl alcohol	785 U	910		785	) <u>-</u>	797	) <u>=</u>	785	ם כ	090	ב כ	402	<b>)</b> ;	
Bis(2-chloroethoxy)methane	392 U	455	n	392	) <u> </u>	306	) <u> </u>	7 65	ב כ	761	o ;	SDS :	)	
Bis(2-chloroethyl)ether	392 U	455		392	) <u>=</u>	306	) <u>;</u>	302	ב כ	396	) ;	402	D ;	
Bis(2-chloroisopropyl)ether	392 U	455		392	) <u>-</u>	396	o =	392	o =	396	) I	402	D :	
Bis(2-ethylhexyl)phthalate	392 U	455	Ω	392	n	396	o D	392	ם כ	396	) <u> </u>	402	)	
4-Bromophenyl phenyl ether	392 U	455	n	392	Ω	396	Ω	392	n	396	) <u>=</u>	402	> <b>=</b>	
Butyl benzyl phthalate	392 U	455	מ	392	U	396	Ω	392	Ω	396	· =	407	) <u> </u>	
4-Chloro-3-methylphenol	785 U	910	Ω	785	U	792	D	785	n	792	) <u> </u>	805	=	
4-Chloroaniline	785 U	910	n	785	Ω	792	D	785	n	792	n n	805	) <u>=</u>	
2-Chloronaphthalene	392 U	455	Ω	392	n	396	Ω	392	Ω	396	, D	402	=	
2-Chlorophenol	392 U	455	Ω	392	Ω	396	U	392	Ú	396	· =	407	) <u> </u>	
4-Chlorophenyl phenyl ether	392 U	455	Ω	392	Ω	396	Ω	392	'n	396	2 1	407	) <u>=</u>	
Chrysene	392 U	455	Ω	392	U	396	Ω	392	n	396	) <u>=</u>	407	) <u>:</u>	
Di-n-butylphthalate	392 U	455	Ω	392	Ω	396	D	392	Ω	396	) <u> </u>	402	) <b>=</b>	
Di-n-octylphthalate	392 U	455	Ω	392	n	396	Ω	392	D	396	) <u> </u>	707	) <u>=</u>	
Dibenz[a,h]anthracene	392 U	455	Ω	392	Ω	396	Ω	392	ם	396	) <u> </u>	407	<b>&gt;</b> =	
Dibenzofuran	392 U	455	Ω	392	Ω	396	Ω	392	U	396	· 11	402	) <u>=</u>	
1,2-Dichlorobenzene	392 U	455	Ω	392	Ω	396	D	392	Ω	396	n	402	) =	
I,3-Dichlorobenzene	392 U	455	D	392	Ω	396	Ω	392	Ω	396	· =	402	) <u>=</u>	
1,4-Dichlorobenzene	392 U	455	Ω	392	Ω	396	n	392	, D	396	) <u>=</u>	402	) <u>=</u>	
3,3'-Dichlorobenzidine	785 U	910	D	785	Ω	792	n	785	D	792	) <u> </u>	805	· =	
2,4-Dichlorophenol	392 U	455	Ω	392	Ω	396	Ω	392	Ω	396	) I	402	> =	
Diethylphthalate	392 U	455	Ω	392	Ω	396	Ω	392	n	396	) <u>=</u>	402	) <b>=</b>	
2,4-Dimethylphenol	392 U	455	Ω	392	U	396	n	392	Ω	396	) D	402	) <b>=</b>	
Dimethylphthalate	392 U	455	n	392	Ω	396	Ŋ	392	Ω	396	n	402	n	

TABLE C.14a (Cont.)

Compound														
	JBP2-E-4'Dup (4 ft)	r'Dup	JBP2-C-2' (2 ft)	2-2.	JBP2 (4	JBP2-C-4' (4 ft)	JBP2-W-2 (2 ft)	W-2' ft)	JBP2-W-4 (4 ft)	.W-4' ft)	JHDP-C-4 (4 ft)	-C-4'	JHDP-C-6 (6 ft)	.9-
4,6-Dinitro-2-methylphenol	1,960	U	2,280	D	1,960	n	1,980	n	1,960	Ω	1.980	ĥ	2.010	=
2,4-Dinitrophenol	1,960	D	2,280	Ω	1,960	Þ	1,980	Ω	1,960	n	1,980	ם מ	2.010	) =
2,4-Dinitrotoluene	392	n	455	n	392	Ω	396	Ω	392	n	396	Þ	402	) =
2,6-Dinitrotoluene	392	Ω	455	Þ	392	Ω	396	Ω	392	n	396	- 1	402	· =
Fluoranthene	392	Ω	455	n	392	D	396	Ω	392	n	396	מי	402	> =
Fluorene	392	Ω	531		392	Ω	396	n	392	Ω	396	Ω	402	Þ
Hexachlorobenzene	392	Ω	455	n	380	_	845		392	n	396	Ŋ	402	n
Hexachlorobutadiene	392	n	455	n	392	n	396	n	392	Ω	396	Ω	402	Þ
Hexachlorocyclopentadiene	392	n	455	D	392	n	396	n	392	n	396	n	402	n
Hexachloroethane	392	Ω	455	ח	392	Ω	396	D	392	n	396	D	628	
Indeno[1,2,3-c,d]pyrene	392	D	455	Ω	392	Ω	396	Ω	392	ח	396	n	402	D
Isophorone	392	n	455	D	392	n	396	Ω	392	n	396	Ω	402	Ω
2-Methylnaphthalene	392	n	455	Ω	392	Ω	396	n	392	n	396	Ω	402	n
2-Methylphenol	392	Ω	455	n	392	Ω	396	Ω	392	Ω	396	Ω	402	Ω
4-Methylphenol	392	Ω	455	Ω	392	Ω	396	Ω	392	n	396	Ω	402	Ω
Naphthalene	392	n	50	ī	392	Ω	396	n	392	Ω	396	n	402	$\Box$
2-Nitroaniline	1,960	n	2,280	Ω	1,960	Ω	1,980	n	1,960	Ω	1,980	Ω	2,010	n
3-Nitroaniline	1,960	U	2,280	n	1,960	Ω	1,980	n	1,960	D	1,980	n	2,010	Þ
4-Nitroaniline	1,960	D	2,280	Ω	1,960	n	1,980	n	1,960	n	1,980	ב	2,010	כ
Nitrobenzene	57	J.	455	n	57	_	396	Ω	392	n	396	Ω	402	D
2-Nitrophenol	392	Ω	455	n	392	Ω	396	Ω	392	n	396	n	402	D
4-Nitrophenol	1,960	D	2,280	ח	1,960	D	1,980	Ω	1,960	Ω	1,980	Ω	2,010	Ω
N-Nitroso-di-n-propylamine	392	Ω	455	D	392	n	396	D	392	n	396	ם	402	D
N-Nitrosodiphenylamine	392	n	455	כ	392	Ω	396	n	392	Ω	396	n	402	Ω
Pentachlorophenol	1,960	Ω	2,280	n	1,960	Ω	1,980	Ω	1,960	Ω	1,980	כ	2,010	$\supset$
Phenanthrene	392	n	52	_	43	_	396	Ω	392	⊃	396	<u>ن</u>	402	D

TABLE C.14a (Cont.)

			Borehole Conce	ntrations (µg/kg)	Borehole Concentrations (μg/kg) at Various Depths			
Compound	JBP2-E-4'Dup (4 ft)	JBP2-C-2' (2 ft)	JBP2-C-4' (4 ft)	JBP2-W-2' (2 ft)	JBP2-W-4' (4 ft)	JHDP-C-4'	JHDP-C-6'	1.9
Phenol	392 U	455 U	392 11	30,5	307			
Pyrenc	392 U	455 11	362 11	306		390 U	402	<b>-</b>
1,2,4-Trichlorobenzene	392 U	_	392 1		392 0		402	<b>&gt;</b> ;
2,4,5-Trichlorophenol	392 U					396 U	402	D ;
2,4,6-Trichlorophenol	392 U	455 U	392 U	į	į		402	ם כ
			Borehole Concer	ntrations (µg/kg) a	Borehole Concentrations (µg/kg) at Various Depths		 	1
Compound	JHDP-C-6'Dup (6 ft)	JVXP-C-4' (4 ft)	JVXP-C-6' (6 ft)	JSDP-C-4' (4 ft)	JSDP-C-6'	JBPM-A-3"	JBPM-A-3"Dup	dng
				(ar. r.)	(0.11)	(3 III.)	(3 m.)	
Acenaphthene	369 U	435 U	465 U	442 U	432 U	392 U	392	=
Acenaphthylene	369 U	435 U	465 U	442 U	432 U		392	) <u>=</u>
Anthracene	369 U	435 U	465 U	442 U	432 U		392	) =
Benz[a]anthracene	N 69E	435 U	465 U	442 U	432 U	392 U	392	> =
Benzo[a]pyrene		435 U	465 U	442 U	432 U	392 U	392	
Benzo[b]fluoranthene	369 U	435 U	465 U	442 U	432 U	392 U	392	· =
Benzolg,n,ijperylene	369 U	435 U	465 U	442 U	432 U	392 U	392	1
Benzoic acid	78 J	91 J	2,330 U	2,210 U	2,160 U	1,960 U	1.960	n n
Benzo[k]Huoranthene	369 U	435 U	465 U	442 U	432 U	392 U	392	, 12
Benzyl alcohol		871 U	930 U	884 U	864 U	785 U	785	)   
Bis(2-chloroethoxy)methane		435 U	465 U	442 U	432 U	392 U	392	) <u>=</u>
Bis(2-chloroethyl)ether	369 U	435 U	465 U	442 U	432 U		392	) <u> </u>
Bis(2-chloroisopropyl)ether	_	435 U	465 U	442 U	432 U	392 U	392	) <b>=</b>
B1s(2-ethylhexyl)phthalate		435 U	465 U	442 U	432 U		392	) <u>=</u>
4-Bromophenyl phenyl ether		435 U	465 U	442 U	432 U	392 U	392	) D
Butyi benzyi phthalate	369 U	435 U	465 U	442 U	432 U	392 U	392	. b

TABLE C.14a (Cont.)

i														
Compound	JHDP-C-6'Dup (6 ft)	3'Dup	JVXP-C-4' (4 ft)	C-4.	XVL	JVXP-C-6' (6 ft)	JSDP-C (4 ft)	JSDP-C-4' (4 ft)	JSDI (6	JSDP-C-6' (6 ft)	JBPM-A-3" (3 in.)	-A-3" n.)	JBPM-A-3"Dup (3 in.)	3"Dur .)
4-Chloro-3-methylphenol	739	Ω	871	n	930	D	884	n	864	Ω	785	Ξ	785	=
4-Chloroaniline	739	n	871	Ω	930	Ω	884	n	864	• =	785	· =	787	) =
2-Chloronaphthalene	369	n	435	Ω	465	D	442	) D	432	) <u>=</u>	397	) <u>=</u>	207	) =
2-Chlorophenol	369	Ω	435	n	465	Ω	442	Þ	432	'n	392	) <u>=</u>	300	) <u> </u>
4-Chlorophenyl phenyl ether	369	Þ	435	Ω	465	n	442	Ω	432	Ω	392	ם ח	392	
Chrysene	369	Ω	435	n	465	Ω	442	n	432	Ω	392	'n	392	î
Di-n-butylphthalate	369	n	435	Ω	70	ъ	442	Ω	432	U	392	Ω	392	n
Di-n-octylphthalate	369	Ω	435	Ω	465	n	442	Ω	432	n	392	Ω	392	D
Dibenz[a,h]anthracene	369	n	435	Ω	465	Ω	442	Ω	432	n	392	Ω	392	D
Dibenzofuran	369	D	435	Ω	465	D	442	n	432	D	392	n	392	Þ
1,2-Dichlorobenzene	369	Ω	435	n	465	n	442	Ω	432	Ω	392	Ω	392	Ω
1,3-Dichlorobenzene	369	D	435	Ω	465	Ω	442	Ω	432	Ω	392	n	392	D
1,4-Dichlorobenzene	369	n	435	⊃	465	n	442	Ω	432	D	392	D	392	Ω
3,3'-Dichlorobenzidine	739	n	871	n	930	n	884	Þ	864	Ω	785	D	785	Ω
2,4-Dichlorophenol	369	Ω	435	ר	465	n	442	n	432	n	392	D	392	Ω
Diethylphthalate	369	n	435	Ω	465	Ω	442	Ω	432	n	392	Ω	392	
2,4-Dimethylphenol	369	n	435	n	465	Ω	442	Ω	432	ח	392	Ω	392	n
Dimethylphthalate	369	Ω	435	D	465	Ω	442	n	432	n	392	D	392	D
4,6-Dinitro-2-methylphenol	1,850	Ω	2,180	n	2,330	Ω	2,210	Ω	2,160	n	1,960	⊃	1,960	ב
2,4-Dinitrophenol	1,850	Ω	2,180	Ω	2,330	n	2,210	n	2,160	D	1,960	Ω	1,960	n
2,4-Dinitrotoluene	369	n	435	כ	465	n	442	D	432	n	392	Þ	392	
2,6-Dinitrotoluene	369	D	435	n	465	D	442	D	432	ח	392	n	392	ח
Fluoranthene	369	Ω	435	Ω	465	D	442	Ω	432	Þ	77	_	392	Þ
Fluorene	369	Ω	435	Ω	465	D	442	ם	432	ב	392	Ω	392	⊃
Hexachlorobenzene	369	Ω	435	n	465	Э	442	ח	432	n	392	Ω	392	ם
Hexachlorobutadiene	369	⊃	435	n	465	D	442	Ξ	432	Ξ	303	-	6	

TABLE C.14a (Cont.)

	Cadill	2		:										
Compound	JHDF-C-6 Dup (6 ft)	o.Dup	JVXP-C-4 (4 ft)	C 4.	) XX	JVXP-C-6' (6 ft)	JSDI (4	JSDP-C-4' (4 ft)	JSDE 9)	JSDP-C-6' (6 ft)	JBPM-A-3" (3 in.)	A-3" 1.)	JBPM-A-3"Dup (3 in.)	.3"Dup
Hexachlorocyclopentadiene	369	Ω	435	n	465	Þ	442	Ξ	432	=	302	=		.  ;
Hexachloroethane	282	ſ	435	Ξ	465	- 1		) [	70.	) ;	766	<b>&gt;</b> ;	266	<b>-</b>
Indeno[1,2,3-c,d]pyrene	369	1	435	) =	244	) <u> </u>	7	) ;	437	⊃ ;	392	<b>-</b>	392	D
Isophorone	340	) <u>=</u>	73.6	) <u>:</u>	407	); 	447	⊃ ;	432	<b>D</b>	392	ח	392	D
2-Methylnaphthalene	369	) <u>-</u>	435	o <u>=</u>	465	) <u>:</u>	442	) :	432	n :	392	n .	392	n
2-Methylphenol	369	n	435	o D	465	> =	442	) <u> </u>	437	)	139	-, ;	392	D ;
4-Methylphenol	369	n	435	D	465	) D	442	) =	432	> =	303	) <u>:</u>	392	);
Naphthalene	369	Ω	435	n	465	Ω	442	) II	432	> =	7 87	> <u>-</u>	760	) :
2-Nitroaniline	1,850	n	2,180	Ω	2,330	n	2,210	) D	2.160	) <u>=</u>	1 960	, <u>=</u>	292 1 060	) =
3-Nitroaniline	1,850	D	2,180	Ω	2,330	Ω	2,210	Ω	2.160	) <u> </u>	1 960	> <b>=</b>	1,500	) <u>:</u>
4-Nitroaniline	1,850	Ω	2,180	Ω	2,330	Ω	2,210	Ω	2,160	'n	1.960	> <b>=</b>	1 960	) =
Nitrobenzene	369	Ω	435	Ω	465	Ω	442	Ω	432	Ω	392	· =	307,	) <u> </u>
2-Nitrophenol	369	Ω	435	Ω	465	U	442	Ω	432		392	· =	300	) =
4-Nitrophenol	1,850	Ω	2,180	Ω	2,330	n	2,210	Þ	2.160	) <u>=</u>	1 960	> =	1 060	) <u>;</u>
N-Nitroso-di-n-propylamine	369	Ω	435	Ω	465	n	442	Ω	432	) D	392	) <u>=</u>	307	)
N-Nitrosodiphenylamine	369	Ω	435	n	465	U	442	Ω	432	ם כ	392	) <u>=</u>	307	) <u>=</u>
Pentachlorophenol	1,850	n	2,180	Þ	2,230	n	2,210	Ω	2,160	D	1.960	) <u>=</u>	1960	) <u>=</u>
Phenanthrene	369	Ω	435	Ω	465	U	442	Ω	432	n	311	),	397	) <u>=</u>
Phenol	369	n	435	Ω	465	Ω	442	Ω	432	Ω	392	. 🗅	392	) <u>=</u>
Pyrene	369	Ω	435	Ω	465	Ω	442	n	432	Ω	215	·	392	> =
1,2,4-Trichlorobenzene	369	Ω	435	n	465	n	442	Ω	432	Ω	392	Ω	392	) =
2,4,5-Trichlorophenol	369	Ω	435	Ω	465	Ω	442	Ω	432	n	392	Ω	392	n
2.4.6-Trichlorophenol	0,0													

TABLE C.14a (Cont.)

Compound														
	JBPM-A-1' (1 ft)	۸-1.	JBPM-B-3" (3 in.)	B-3"	JBPN (1	JBPM-B-1' (1 ft)	JBPM-C-3' (3 in.)	-C-3"	JBPM-C-1 (1 ft)	F-C-1'	JBPP-A-3" (3 in.)	A-3" 1.)	JBPP-A-1 (1 ft)	
Acenaphthene	392	Ω	376	Ω	386	Ω	419	n	379	n	386	Þ	392	Ω
Acenaphthylene	392	D	376	Ω	386	Ω	419	D	379	n	386	Ω	392	D
Anthracene	392	Ω	376	Ω	386	Ω	419	Ω	379	D	386	n	392	ר
Benz[a]anthracene	392	n	376	Ω	386	Ω	419	Ω	379	n	386	Ω	392	n
Benzo[a]pyrene	392	Ω	376	Ω	386	D	419	n	379	D	386	ח	392	Ω
Benzo[b]fluoranthene	392	Ω	376	D	386	n	419	Ω	379	n	386	n	392	C
Benzo[g,h,i]perylene	392	Ω	376	D	386	n	419	Ω	379	D	386	Ω	392	Ω
Benzoic acid	1,960	Ω	1,880	Ω	1,930	Ω	2,100	ם	1,900	Ω	1,930	D	1,960	Ω
Benzo[k]fluoranthene	392	Ω	376	n	386	Ω	419	n	379	n	386	Ω	392	Ω
Benzyl alcohol	785	n	752	Ω	772	Ω	838	n	759	Ω	772	Ω	785	D
Bis(2-chloroethoxy)methane	392	Ω	376	Ω	386	Ω	419	D	379	n	386	n	392	$\supset$
Bis(2-chloroethyl)ether	392	Þ	376	Ω	386	Ω	419	n	379	n	386	Ω	392	ר
Bis(2-chloroisopropyl)ether	392	Ω	376	Ω	386	c	419	Ω	379	D	386	Ω	392	D
Bis(2-ethylhexyl)phthalate	392	n	190	-	386	Ω	419	Ω	379	n	386	n	392	D
4-Bromophenyl phenyl ether	392	Ω	376	Ω	386	Ω	419	n	379	D	386	Ω	392	C
Butyl benzyl phthalate	392	Ω	376	Ω	386	n	419	Ω	379	Ω	386	n	392	$\Box$
4-Chloro-3-methylphenol	785	Ω	752	D	772	D	838	n	759	D	772	Ω	785	כ
4-Chloroaniline	785	Ω	752	D	772	D	838	Ω	759	n	772	D	785	n
2-Chloronaphthalene	392	n	376	Ω	386	D	419	Ω	379	D	386	U	392	Ω
2-Chlorophenol	392	ר	376	Ω	386	n	419	n	379	n	386	D	392	D
4-Chlorophenyl phenyl ether	392	Ω	376	n	386	n	419	D	379	n	386	D	392	Þ
Chrysene	392	n	376	Ω	386	n	419	D	379	n	386	Ω	392	n
Di-n-butylphthalate	77	<u></u>	376	Ω	386	Ω	419	n	379	Л	386	Ω	392	D
Di-n-octylphthalate	392	D	376	D	386	Ω	419	Ω	379	n	386	Ω	392	D
Dibenz[a,h]anthracene	392	n	376	כ	386	Ω	419	Ω	379	D	386	ם	392	ם
Dibenzofuran	392	Ξ	376	D	386	Ω	419	D	379	<u> </u>	386	Ü	392	$\Box$

TABLE C.14a (Cont.)

Compound	JBPM-A-1	A-1.	JBPM-B-3" (3 in.)	.B-3"	JBP (	JBPM-B-1' (1 ft)	JBPN (3	JBPM-C-3" (3 in.)	JBPN (1	JBPM-C-1' (1 ft)	JBPP (3)	JBPP-A-3" (3 in.)	JBPP-A-1 (1 ft)	A-1'
1,2-Dichlorobenzene	392	Ω	376	n	386	Þ	419	11	370	=	306	=	6	:
1,3-Dichlorobenzene	392	n	376	n	386	Ω	419	· =	370	) <u>;</u>	200	ב כ	760	);
1,4-Dichlorobenzene	392	n	376	Þ	386	) D	419	> =	370	) <u>:</u>	300	) ;	392	⊃ ;
3,3'-Dichlorobenzidine	785	Ω	752		777	) <u>=</u>	838	> =	750	) <u>:</u>	330	);	392	⊃ ;
2,4-Dichlorophenol	392	n	376	n	386	) D	419	) =	379	) <u> </u>	788	> <b>=</b>	(8) (2)	) :
Diethylphthalate	49	_	376	n	386	ח	419	'n	379	) <u>=</u>	386	<b>&gt;</b> =	365	) <u>:</u>
2,4-Dimethylphenol	392	n	376	n	386	Ω	419	n	379	'n	386	) =	392	o <u>=</u>
Dimethylphthalate	392	n	376	n	386	Ω	419	Ω	379	Ω	386	) [	300	=
4,6-Dinitro-2-methylphenol	1,960	Ω	1,880	n	1,930	Ω	2,100	n	1,900	Ω	1.930	ם כ	096 1	> =
2,4-Dinitrophenol	1,960	Ω	1,880	Ω	1,930	Ω	2,100	D	1,900	Ω	1.930	ם כ	1.960	) <u>=</u>
2,4-Dinitrotoluene	392	Ω	376	n	386	Ω	419	n	379	n	386	) <u> </u>	397	) <b>=</b>
2,6-Dinitrotoluene	392	Ω	376	Ω	386	Ω	419	n	379	Ω	386	· 1	300	) <b>=</b>
Fluoranthene	392	ח	376	D	386	Ω	419	n	379	Ω	386	) <u> </u>	300	) <u> </u>
Fluorene	392	D	376	Ω	386	D	419	n	379	Ω	386	· 1	300	) <u>=</u>
Hexachlorobenzene	392	Ω	376	Ω	386	n	419	ם	379	'n	386		365	) <u>=</u>
Hexachlorobutadiene	392	n	376	n	386	Ω	419	Ω	379	n	386	) =	300	) <u> </u>
Hexachlorocyclopentadiene	392	Ω	376	Ω	386	n	419	n	379	Ω	386	: D	392	) <u>=</u>
Hexachloroethane	498		376	D	386	Ω	419	Ω	379	Ω	386	Þ	392	) <b>=</b>
Indeno[1,2,3-c,d]pyrene	392	Ω	376	n	386	Ω	419	Ω	379	U	386	n	392	) <u>=</u>
Isophorone	392	Ω	376	n	386	Ω	419	Ω	379	Ω	386	n	392	) <u>=</u>
2-Methylnaphthalene	392	ב	376	Ω	386	Ŋ	419	Ω	379	n	386	· =	300	) <u> </u>
2-Methylphenol	392	Ú	376	Ω	386	U	419	D	379	Ω	386	· 11	392	) <b>=</b>
4-Methylphenol	392	Ω	376	n	386	Ω	419	D	379	n	386	n	392	) =
Naphthalene	392	n	376	Ω	386	Ω	419	Ω	379	Ω	386	U	392	) I
2-Nitroaniline	1,960	n	1,880	n	1,930	n	2,100	Ω	1,900	n	1,930	Ω	1,960	n
3-Nitroaniline	1 060		1 000											

TABLE C.14a (Cont.)

					Poleno	ie Conce	borenote Concentrations (µg/kg) at Various Deptins	ıg/kg) a	it Various	Depths				
Compound	JBPM-A-1'	A-1.	JBPM-B-3" (3 in.)	B-3"	JBP	JBPM-B-1' (1 ft)	JBPM-C-3" (3 in.)	-C-3" n.)	JBPM-C (1 ft)	JBPM-C-1' (1 ft)	JBPP-A-3" (3 in.)	A-3" n.)	JBPP-A-1'	A-1.
4-Nitroaniline	1,960	n	1,880	n	1,930	ņ	2,100	Ω	1,900	Ω	1,930	Þ	1.960	n
Nitrobenzene	392	n	376	n	386	n	419	D	379	n	386	D	392	n
2-Nitrophenol	392	Ω	376	D	386	n	419	Ω	379	Ω	386	n	392	n
4-Nitrophenol	1,960	Ω	1,880	n	1,930	Þ	2,100	n	1,900	n	1,930	D	1,960	) D
N-Nitroso-di-n-propylamine	392	D	376	Ω	386	Ω	419	n	379	n	386	Ω	392	Ω
N-Nitrosodiphenylamine	392	Ω	376	D	386	Ω	419	Ω	379	Ω	386	n	392	Ω
Pentachlorophenol	1,960	Ω	1,880	D	1,930	Ω	2,100	n	1,900	n	1,930	n	1,960	Ω
Phenanthrene	392	D	376	n	386	Ω	419	Ω	379	Ω	386	n	392	Ω
Phenol	392	D	376	Ω	386	n	419	n	379	n	386	ח	392	Ω
Pyrene	392	Ω	376	n	386	ח	419	Ω	379	n	386	n	392	D
1,2,4-Trichlorobenzene	392	C	376	D	386	n	419	Ω	379	n	386	D	392	n
2,4,5-Trichlorophenol	392	n	376	n	386	$\supset$	419	n	379	n	386	Ω	392	ח
2,4,6-Trichlorophenol	392	חן	376	ח	386	n	419	n	379	٦	386	n	392	D
		Boreho	Borchole Concentrations (µg/kg) at Various Depths	tions (	ug/kg) at	Various ]	Depths							
	JBPP-B-3"	<u>.</u>	JBPP-B-1	<u> </u>	JBPI	JBPP-C-3"	JBPP-C-1	C-1.						
Compound	(3 in.)		(1 ft)		(3	(3 in.)	(1 ft)	<u>.</u>						
Acenaphthene	455	ח	389	n	392	ב	392	n						
Acenaphthylene	455	D	389	n	392	Ω	392	ח						
Anthracene	455	ח	389	Ω	392	Ω	392	n						
Benz[a]anthracene	455	כ	389	n	392	ח	392	D						
Benzo[a]pyrene	455	ב	389	Þ	392	U	392	n						
Benzo[b]fluoranthene	455	n	389	Ω	392	Ω	392	Ω						
Benzo[g,h,i]perylene	455	n	389	n	392	n	392	Ω						
Benzoic acid	2,280	n	1,950	n	1,960	Ω	1,960	ח						

TABLE C.14a (Cont.)

i	JBPP-B-3"	3-3"	JBPP-B-1	B-1,	JBPI	JBPP-C-3"	JBPP	JBPP-C-1'
Compound	(3 in.)		(1 ft)	£	(3	(3 in.)	(1 ft)	£)
Benzo[k]fluoranthene	455	Ω	389	Ω	392	ם	392	Ω
Benzyl alcohol	910	D	778	Ω	785	Ŋ	785	n
Bis(2-chloroethoxy)methane	455	n	389	D	392	U	392	n
Bis(2-chloroethyl)ether	455	Ω	389	Ω	392	Ω	392	Ď
Bis(2-chloroisopropyl)ether	455	Ω	389	n	392	n	392	ח
Bis(2-ethylhexyl)phthalate	455	Ω	389	U	392	Ω	392	n
4-Bromophenyl phenyl ether	455	Ω	389	n	392	n	392	Ω
Butyl benzyl phthalate	455	n	389	n	392	Ω	392	Ω
4-Chloro-3-methylphenol	910	Ω	778	n	785	Ŋ	785	Ω
4-Chloroaniline	910	Ω	778	Ω	785	n	785	ח
2-Chloronaphthalene	455	Ω	389	n	392	ם	392	Ω
2-Chlorophenol	455	Ω	389	n	392	Ω	392	Ω
4-Chlorophenyl phenyl ether	455	Ω	389	Ω	392	Ω	392	Ω
Chrysene	455	Ω	389	Ω	392	D	392	Ω
Di-n-butylphthalate	56	J	389	n	392	ם	392	n
Di-n-octylphthalate	455	Ω	389	Ω	392	Ω	392	Ω
Dibenz[a,h]anthracene	455	Ω	389	n	392	Ω	392	Ω
Dibenzofuran	455	D	389	Ω	392	Ω	392	Ω
1,2-Dichlorobenzene	455	Ω	389	Ω	392	Ω	392	n
1,3-Dichlorobenzene	455	Ω	389	Ω	392	Ω	392	Ω
1,4-Dichlorobenzene	455	Ω	389	Ω	392	Ω	392	n
3,3'-Dichlorobenzidine	910	Ω	778	Ω	785	n	785	n
2,4-Dichlorophenol	455	D	389	Ω	392	Ω	392	Ω
Dicthylphthalate	455	Ω	389	Ω	392	n	392	Ω
2,4-Dimethylphenol	455	Ω	389	n	392	n	392	n
Dimethylphthalate	455	n	389	Ω	392	Ŋ	392	Ω

TABLE C.14a (Cont.)

Compound	JBPP-B-3" (3 in.)	3-3"	JBPP-B-1'	8-1.	JBPP (3)	JBPP-C-3" (3 in.)	JBPP-C-1'	-C-1-
4,6-Dinitro-2-methylphenol	2,280	n	1,950	Ω	1,960	D	1,960	n
2,4-Dinitrophenol	2,280	Ω	1,950	n	1,960	n	1,960	Ω
2,4-Dinitrotoluene	455	Ω	389	Ω	392	n	392	n
2,6-Dinitrotoluene	455	n	389	Ω	392	Ω	392	Ω
Fluoranthene	455	Ω	389	Ω	392	Ω	392	Ω
Fluorene	455	Ω	389	Ω	392	n	392	Ω
Hexachlorobenzene	455	Ω	389	Ω	392	Ω	392	D
Hexachlorobutadiene	455	Ω	389	Ω	392	n	392	Ω
Hexachlorocyclopentadiene	455	D	389	Ω	392	Ω	392	n
Hexachlorocthane	455	Ω	389	Ω	392	Ω	392	
Indeno[1,2,3-c,d]pyrene	455	D	389	D	392	Ω	392	n
Isophorone	455	n	389	n	392	Ω	392	Ω
2-Methylnaphthalene	161	۳	389	n	392	Ω	392	$\supset$
2-Methylphenol	455	n	389	n	392	Ω	392	$\supset$
4-Methylphenol	455	Ω	389	Ω	392	Ω	392	D
Naphthalenc	309	_	389	Ω	392	D	392	D
2-Nitroaniline	2,280	Ω	1,950	Ω	1,960	D	1,960	$\supset$
3-Nitroaniline	2,280	n	1,950	Þ	1,960	Ŋ	1,960	Ω
4-Nitroaniline	2,280	n	1,950	Ω	1,960	n	1,960	Ω
Nitrobenzene	455	Ω	389	כ	392	Ω	392	$\cap$
2-Nitrophenol	455	Ω	389	n	392	Ω	392	ח
4-Nitrophenol	2,280	D	1,950	n	1,960	Ω	1,960	n
N-Nitroso-di-n-propylamine	455	n	389	Ω	392	Ω	392	$\supset$
N-Nitrosodiphenylamine	455	Ω	389	D	392	n	392	D
Pentachlorophenol	2,280	Ω	1,950	n	1,960	n	1,960	Ω
Phononthrone	ť	,	300	;				

TABLE C.14a (Cont.)

		portugic Concentrations (µg/kg) at various Depths			i			
Compound	JBPP-B-3" (3 in.)	3"	JBPP-B-1'	3-1-	JBPP-C.	JBPP-C-3" (3 in.)	JBPP-C-1' (1 ft)	C-1.
Phenol	455	Ω	389	Ω	392	n	392	ח
Pyrcne	455	D	389	n	392	U	392	n
1,2,4-Trichlorobenzene	455	Ω	389	Ω	392	n	392	n
2,4,5-Trichlorophenol	455	Ω	389	Ω	392	n	392	n
2,4,6-Trichlorophenol	455	Ω	389	Ω	392	Ω	392	n

<sup>a</sup> Sampled by Weston, analyzed by GP Environmental Services, CLP/HSL volatile organics; provisional data.

Data qualifiers:

U =analyte was analyzed for but not detected.

J = estimated value.

TABLE C.14b Semivolatile Organics Analyses for Soil Borings, January 1994<sup>a</sup>

Compound Acenaphthene Acenaphthylene Anthracene Benz[a]anthracene Benzo[b]fluoranthene Benzo[k]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-chloroisopropyl)ether Bis(2-chloroisopropyl)ether Bis(2-chloroisopropyl)ether	CLP1 (0-6 in.)	Ы								
Acenaphthene Acenaphthene Anthracene Benz[a]anthracene Benzo[a]pyrene Benzo[b]fluoranthene Benzo[k],i]perylene Benzo[k]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroethyl)pther Bis(2-chloroethyl)pther Busylbenzylphthalate	9-0)		CLPI	ΡΙ	CLP1	Į,	ב	CI.P2	G 15	3
Acenaphthene Acenaphthylene Anthracene Benz(a]anthracene Benzo[a]pyrene Benzo[b]fluoranthene Benzo[k]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-chylhexyl)phthalate 4-Bromophenyl phenyl ether	200	in.)	(6-24 in.)	in.)	(24-4	(24-40 in.)	(0-6 in.)	in.)	(6-24 in.)	in.)
Acenaphthylene Anthracene Benzo[a]anthracene Benzo[b]fluoranthene Benzo[k,i]perylene Benzo[k]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-chylorosty)phthalate 4-Bromophenyl phenyl ether	3	Ω	420	Ω	420	Ω	420	n	430	Ω
Anthracene Benz[a]anthracene Benzo[a]pyrene Benzo[b]fluoranthene Benzo[k]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-chylhexyl)phthalate 4-Bromophenyl phenyl ether	500	n	420	Ω	420	Ω	420	ח	430	Ω
Benzo[a]pyrene Benzo[b]fluoranthene Benzo[b]fluoranthene Benzo[k]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-chloroisopropyl)ether Bis(2-chylhexyl)phthalate 4-Bromophenyl phenyl ether	200	n	420	n	420	n	420	Ω	430	Ω
Benzo[a]pyrene Benzo[b]fluoranthene Benzo[g,h,i]perylene Benzo[k]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chlorospropyl)ether Bis(2-chloroisopropyl)ether Bis(2-chlylhexyl)phthalate 4-Bromophenyl phenyl ether	200	ם	420	Ω	420	Ω	420	Ω	430	D
Benzo[b]fluoranthene Benzo[g,h,i]perylene Benzo[K]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroisopropy])ether Bis(2-chloroisopropy])ether Bis(2-chloroisopropy] ether Bis(2-chlylhexyl)phthalate 4-Bromophenyl phenyl ether	200	Ω	420	Ω	420	Ω	420	Ω	430	n
Benzo[g,h,i]perylene Benzo[k]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-chlylhexyl)phthalate 4-Bromophenyl phenyl ether	200	U	420	n	420	Ω	420	n	430	Ω
Benzo(k]fluoranthene Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-ethylhexyl)phthalate 4-Bromophenyl phenyl ether Butylbenzylphthalate	200	Ω	420	D	420	n	420	D	430	ח
Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-ethylhexyl)phthalate 4-Bromophenyl phenyl ether Butylbenzylphthalate	200	Ω	420	Ω	420	מ	420	Ω	430	Ω
Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-ethylhexyl)phthalate 4-Bromophenyl phenyl ether Butylbenzylphthalate	200	Ω	420	U	420	Ω	420	D	430	Ω
Bis(2-chloroisopropyl)ether Bis(2-ethylhexyl)phthalate 4-Bromophenyl phenyl ether Butylbenzylphthalate	200	n	420	Ω	420	D	420	n	430	ם
Bis(2-ethylhexyl)phthalate 4-Bromophenyl phenyl ether Butylbenzylphthalate	200	n	420	Ω	420	Ω	420	n	430	ם
4-Bromophenyl phenyl ether Butylbenzylphthalate	200	JB	420	JB	420	JB	420	JB	430	JB
Butylbenzylphthalate	200	Ω	420	D	420	n	420	ב	430	Þ
	200	n	420	D	420	Þ	420	ב	430	n
Carbazole	200	n	420	ב	420	n	420	D	430	ר
4-Chloro-3-methylphenol	200	Ω	420	n	420	Ω	420	ב	430	Ω
4-Chloroaniline	200	Ω	420	Ω	420	n	420	Ω	430	n
2-Chloronaphthalene	200	Ω	420	Ω	420	n	420	n	430	D
2-Chlorophenol	200	U	420	D	420	Ω	420	D	430	D
4-Chlorophenyl phenyl ether	200	n	420	ņ	420	Ω	420	ם	430	Ω
Chrysene	500	n	420	Ω	420	D	420	Ω	430	n
Di-n-butyIphthalate	200	JB	420	JB	420	JB	420	JB	430	JB
Di-n-octyl phthalate	200	U	420	Ω	420	Ω	420	U	430	n
Dibenz[a,h]anthracene	200	n	420	n.	420	n	420	Ω	430	ח
Dibenzofuran	200	n	420	Ω	420	ם	420	n	430	$\Box$
1,2-Dichlorobenzene	200	n	420	ב	420	n	420	n	430	D
,3-Dichlorobenzene	200	Ω	420	Ω	420		420	n	430	n
,4- Dichlorobenzene	200	Ω	420	n	420	n	420	n	430	n

TABLE C.14b (Cont.)

			Soil Con	centrati	ons (µg/kg)	at Vario	Soil Concentrations (µg/kg) at Various Depth Intervals	itervals		
	Ū	CLP1	IJ	CLPI	Ö	CLPI	כ	<u>8</u>	5	8
Compound	(0-(	(0-6 in.)	(6-24 in.)	in.)	(24-7	(24-40 in.)	9-0)	(0-6 in.)	(6-24 in.)	r.z. .in.)
3,3'-Dichlorobenzidine	200	Ω	420	Ω	420	ם	420	=	430	=
2,4- Dichlorophenol	200	Ω	420	n	420		420	) =	054	> <b>:</b>
Diethylphthalate	200	Ω	420	Ω	420	_	420	· =	430	>
2,4-Dimethylphenol	200	Ω	420	Þ	420	. 🗅	420	· -	430	-
Dimethylphthalate	200	Ω	420	n	420	Ω	420	. =	430	, <u>=</u>
4,6-Dinitro-2-methylphenol	1,200	n	1,000	U	1,000	n	1.000	) =	100	> =
2,4-Dinitrophenol	1,200	D	1,000	n	1,000	Ω	1,000	o D	1.100	) <u>=</u>
2,4-Dinitrotoluene	200	Ω	420	Ω	420	D	420	Ω	430	) =
2,6-Dinitrotoluene	200	n	420	Ω	420	n	420	· =	430	> =
Fluoranthene	500	Ω	420	n	420	Ω	420	; D	430	- =
Fluorene	200	Ω	420	n	420	n	420	Ω	430	> =
Hexachlorobenzene	200	'n	420	Ω	420	ſ	420	·	430	· -
Hexachlorobutadiene	200	D	420	D	420	ם	420	Ω	430	, =
Hexachlorocyclopentadiene	200	Ω	420	n	420	D	420	n	430	) =
Hexachloroethane	200	Ω	420	D	420	n	420	ם	430	) =
Indeno[1,2,3-c,d]pyrene	200	n	420	Ω	420	n	420	· D	430	> =
Isophorone	200	Ω	420	n	420	Ŋ	420	n	430	> =
2-Methylnaphthalene	200	Ω	420	Ω	420	D	420	1	430	) <b>-</b>
2-Methylphenol	200	Ω	420	Ŋ	420	D	420	· –	430	•
4-Methylphenol	200	Ω	420	n	420	n	420		430	
Naphthalene	200	Ω	420	Ω	420	Ω	420	Þ	430	-
2-Nitroaniline	1,200	n	1,000	Ω	1,000	Ω	1,000	Ω	100	, =
3-Nitroaniline	1,200	Ω	1,000	Ω	1,000	n	1,000	, D	1.100	) <u>=</u>
4-Nitroaniline	1,200	n	1,000	n	1,000	D	1,000	o D	1.100	) =
Nitrobenzene	200	n	420	D	420	n	420	n	430	· =
2-Nitrophenol	200	n	420	Ω	420	n	420	D	430	) <u> </u>
4-Nitrophenol	1,200	Ω	1,000	Ω	1,000	Ω	1,000	Ω	1,100	· D
N-Nitroso-di-n-propylamine	200	D	420	Ω	420	Ω	420	Ω	430	Ω
N-Nitrosodiphenylamine	200	Ω	420	Ω	420	Ω	420	Ω	430	<u>-</u>

TABLE C.14b (Cont.)

			Soil Con	centratio	ns (µg/kg)	at Vario	Soil Concentrations (µg/kg) at Various Depth Intervals	tervals		
	ぴら	CLPI	CLPI	PI	CLPI	E :	CLP2	P2	CLP2	27
Compound	9-0)	(0-6 in.)	(6-24 in.)	(in.)	(24-4	(24-40 in.)	(0-6 in.)	in.)	(6-24 in.)	in.)
Pentachlorophenol	1,200	n	1,000	Ω	1,000	Ω	1,000	Ω	1,100	n
Phenanthrene	200	Ω	420	Ω	420	Þ	420	D	430	Ω
Phenol	200	n	420	Ω	420	n	420	ĭ	430	_
Pyrene	200	n	420	Ω	420	Ω	420	D	430	n
1,2,4-Trichlorobenzene	200	Ω	420	D	420	Ω	420	Ω	430	n
2,4,5-Trichlorophenol	1,200	Ω	1,000	Þ	1,000	Ω	1,000	D	1,100	n
2,4,6-Trichlorophenol	500	b	420	ח	420	o¦	420	ם חור	430	ם
·			Soil Conc	entration	із (µg/kg)	at Variou	Soil Concentrations (µg/kg) at Various Depth Intervals	ervals		
	CLP2	P2	CLP3	23	CLP3	33	CLP3-Dup	anq	CLP3	77
Compound	(24-48 in.)	3 in.)	(0-6 in.)	in.)	(6-24 in.)	in.)	(6-24 in.)	in.)	(24-48 in.)	in.)
Acenaphthene	200	ם	290	ם	490	D	2,500	n	970	Ω
Acenaphthylene	200	Ω	290	Ω	490	Ω	2,500	ם	926	Ω
Anthracene	500	n	290	n	490	Þ	2,500	Ω	026	Ω
Benz[a]anthracene	200	Ω	590	n	490	Ω	2,500	מ	970	n
Benzo[a]pyrene	500	D	290	Þ	490	D	2,500	ם	026	D
Benzo[b]fluoranthene	200	Ω	290	n	490	Ω	2,500	n	970	D
Benzo[g,h,i]perylene	200	n	290	Ω	490	Ω	2,500	Ω	970	n
Benzo[k]fluoranthene	200	D	280	ם	490	Ω	2,500	Ω	970	ח
Bis(2-chloroethoxy)methane	200	Ω	290	Ω	490	Ω	2,500	Ω	026	D
Bis(2-chloroethyl)ether	200	n	290	Ω	490	D	2,500	ח	970	Ω
Bis(2-chloroisopropy1)ether	200	n	290	Ω	490	n	2,500	Þ	970	Ω
Bis(2-ethylhexyl)phthalate	200	Ω	290	JB	260	JB	2,500	Ŋ	460	JВ
4-Bromophenyl phenyl ether	200	$\Box$	290	Ω	490	D	2,500	D	970	n
Butylbenzylphthalate	500	Ω	290	Ω	490	Ω	2,500	n	970	n
Carbazole	200	ם	290	Ω	490	ם	2,500	D	970	Ω
4-Chloro-3-methylphenol	200	Π	260	Ω	490	n	2,500	U	970	n
4-Chloroaniline	200	ח	290	Ω	740	ם	2,500	D	970	Ξ

TABLE C.14b (Cont.)

			Soil Con	centratic	ns (µg/kg)	at Vario	Soil Concentrations (µg/kg) at Various Depth Intervals	tervals		
	ט	CLP2	CLP3	P3	ō	CLP3	CLP3-Dup	-Dup	CLP3	23
Compound	(24-7	(24-48 in.)	(0- 6 in.)	in.)	(6-2	(6-24 in.)	(6-24 in.)	in.)	(24-48 in.)	3 in.)
2-Chloronaphthalene	200	n	290	n	490	n	2.500	D	026	=
2-Chlorophenol	200	Ω	590	Ω	490	Ω	2,500	n	970	> =
4-Chlorophenyl phenyl ether	500	Ω	290	Ω	490	Ω	2,500	Ω	970	î D
Chrysene	200	n	590	'n	490	Ω	2,500	n	970	· D
Di-n-butylphthalate	200	n	290	JB	220	JB	2,500	n	380	18
Di-n-octyl phthalate	500	מ	290	Ω	490	Ω	2,500	n	970	2 0
Dibenz[a,h]anthracene	200	Ω	290	n	490	Ω	2,500	D	026	, D
Dibenzofuran	500	Ω	290	Ω	490	Ω	2,500	D	970	ב
1,2-Dichlorobenzene	500	Ω	290	n	490	Ω	2,500	n	420	٠ ,
1,3-Dichlorobenzene	200	Ω	290	Ω	490	D	2,500	n	210	
1,4-Dichlorobenzene	200	n	590	n	490	n	2,500	n	970	. n
3,3'-Dichlorobenzidine	200	Ω	290	Ω	490	Ω	2,500	Ω	970	ם
2,4-Dichlorophenol	200	Ω	290	Ω	490	Ω	2,500	n	970	n
Diethylphthalate	200	n	290	Ω	2,000		2,500	Ω	970	n
2,4-Dimethylphenol	200	Ω	290	n	490	n	2,500	n	970	n
Dimethylphthalate	200	Ω	290	Ω	490	Ω	2,500	Ω	970	Ω
4,6-Dinitro-2-methylphenol	1,200	Ω	1,500	Ω	1,200	n	6,200	n	2,400	Ω
2,4-Dinitrophenol	1,200	Ω	1,500	Ω	1,200	Ω	6,200	Ω	2,400	Ω
2,4-Dinitrotoluene	200	Ω	590	Ω	490	Ω	2,500	Ω	970	D
2,6-Dinitrotoluene	200	Ω	290	Ω	490	Ŋ	2,500	n	970	n
Fluoranthene	200	Ω	290	-	490	n	2,500	Ω	970	Ω
Fluorene	200	Ω	290	Ω	490	n	2,500	Ω	970	n
Hexachlorobenzene	200	n	290		3,100		2,500	n	710	<b>-</b>
Hexachlorobutadiene	200	Ω	290	Ω	490	Ω	2,500	Ω	970	Ω
Hexachlorocyclopentadiene	200	D	290	Ω	490	Ŋ	2,500	Ω	970	U
Hexachloroethane	200	D	290	_	490	n	2,500	n	970	Ω
Indeno[1,2,3-c,d]pyrene	200	n	290	Ω	490	Ω	2,500	n	970	Ω
Isophorone	200	Ω	290	n	490	U	2,500	Ω	970	n
2-Methylnaphthalene	200	n	290	Ω	200	<b>-</b>	2,500	Ω	26	<b>,,</b>

TABLE C.14b (Cont.)

i	CLP2	P2	CLP3	33	CLP3	P3	CLP3-Dup	-Dup	CLP3	P3
Compound	(24-48 in.)	3 in.)	(0-6 in.)	in.)	(6-24 in.)	in.)	(6-24 in.)	in.)	(24-4	(24-48 in.)
2-Methylphenol	200	n	590	ח	130	ſ	2,500	Ω	140	_
4-Methylphenol	200	n	590	Ω	180	_	2,500	Ω	190	_
Naphthalene	200	Ω	590	Ω	06	-	2,500	n	970	n
2-Nitroaniline	1,200	Ω	1,500	Ω	1,200	Ω	6,200	Ω	2,400	n
3-Nitroaniline	1,200	Ω	1,500	ב	1,200	Ω	6,200	n	2,400	n
4-Nitroaniline	1,200	D	1,500	n	1,200	Ω	6,200	n	2,400	D
Nitrobenzene	200	n	290	n	490	Ω	2,500	Ω	970	n
2-Nitrophenol	200	Ω	290	n	490	n	2,500	Ω	1,400	
4-Nitrophenol	1,200	Ω	1,500	ם	1,200	D	6,200	Ω	2,400	n
N-Nitroso-di-n-propylamine	200	n	590	n	490	b	2,500	Ω	970	$\supset$
N-Nitrosodiphenylamine	200	Ω	290	D	240	_	2,500	Ω	970	ņ
Pentachlorophenol	1,200	Ω	1,500	Ω	1,200	Ω	6,200	n	2,400	Þ
Phenanthrene	200	n	290	Ω	490	n	2,500	n	940	Ω
Phenoi	200	n	290	Ω	490	Ω	2,500	n	970	Ω
Pyrene	200	U	290	-	53	ſ	2,500	n	970	ר
1,2,4-Trichlorobenzene	200	n	290	Ω	490	Ŋ	2,500	n	970	ם
2,4,5-Trichlorophenol	1,200	ם	1,500	D	1,200	D	6,200	Ω	2,400	Ω
2,4,6-Trichlorophenol	200	Ω	690	1	700	1	2 500	=	ţ	;

TABLE C.14b (Cont.)

Accomposing Composing Comp																			
40 1 440 1 430 1 4300 1 4300 1 420 1	Compound	υ <u>δ</u>	LP4 5 in.)	CI (6-2-	CP4 4 in.)	CLI	74 in )	CLP	4-RE	נד	.PS	CLP	5-RE		۲۶	CC	73	CLPS	-Dup
470 14 440 14 4300 1 4300 1 4300 1 420 1 4								17)	0 1111.)	3-0)	m.)	5	) In.)	(6-24	in.)	(24-4)	8 in.)	(24-4	8 in.)
470 U 440 U 4300 U 4300 U 4300 U 420	Acenaphthene	470		440		4,300	ņ	4,300	Ω	420	מ	420		400	Н	740	1	7	:
470   1   440   1   4,300	Acenaphthylene	470		440		4,300	Ω	4,300		420	Ξ	420		<u> </u>	) <b>:</b>		<b>:</b>	. 440	> :
checked with the control of the cont	Anthracene	470		440	n	4,300	D	4.300		420	> =	2 5		3 5	<b>&gt;</b> ;	440	<b>)</b>	440	n
From the control of t	Benz[a]anthracene	69	-	440	Π	4 300	- =	4 200		2 6	> :	77	<b>)</b>	3	<b>D</b>	440	Þ	440	D
rene 200 1 410 1 4300 1 4300 1 4300 1 420 1 420 1 420 1 420 1 440 1 420 1 440	Benzofalpyrene	470		2.7	) ;	000,4	o ;	0000,4	)	420	<b>n</b>	450	Þ	400	Ω	440	Ω	440	D
The contribution of the co	Benzolhiduomathono	1,0		440	o :	4,300	D	4,300	Ω	420	Ω	420	Ω	400	Ω	440	n	440	n
Fig. 470 1 440 1 4300 1 4300 1 4300 1 420	penzolojnuorantuene	200	<b>-</b> ,	440	ב	4,300	Ω	4,300	n	420	Ω	420	Þ	400	n	440		440	-
470         U         440         U         4300	Benzo[g,h,i]perylene	470	n	440	n	4,300	D	4,300	Ω	420	n	420	Ω	400	· =	440	) <u>=</u>	7	) :
Highlight Harmon (470) U 440 U 4300 U 4300 U 420 U 420 U 420 U 400 U 400 U 440 U 440 U 440 U 4300 U 4300 U 420 U 420 U 420 U 420 U 400 U 440 U 440 U 4300 U 4300 U 420 U	Benzo[k]fluoranthene	470	Ω	440	Ω	4,300	Ω	4,300	ם	420	D	420	1	400	) <u>=</u>	940	> =	7	⊃ :
tether 470 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 440 U 440 U 4,300 U 4,300 U 420	Bis(2-chloroethoxy)methane	470	Ω	440	Ω	4,300	D	4,300	Ω	420	Ω	420	· =	2 5	) <u>:</u>	1	<b>&gt;</b> :	044	<b>&gt;</b> ;
typy bether         470         U         440         U         4300         U         420         U         440         U         440         U         4300         U         420         U         420         U         420         U         420         U         420         U         440         U         440         U         4300         U         420         U         420         U         420         U	Bis(2-chloroethyl)ether	470	Ŋ	440	Ω	4,300	D	4,300	n	420	· =	420	) <u>=</u>	3 5	<b>&gt;</b>	0 4	o :	440	<b>)</b>
hithhaliate [170] JB 140 JB 4,300 U 4,300 U 4,200 U 4,20 U	Bis(2-chloroisopropyl)ether	470	Ω	440	Ω	4,300	Ω	4.300	Ω	420	· =	120	) <u> </u>	3 5	> :	044	<b>)</b>	440	<b>&gt;</b>
and         470         U         440         U         430         U         44	3is(2-ethylhexyl)phthalate	170	JB	140	JB	4,300	Ω	440	. Bī	870	) et	002	ם כ	3 5	> ;	440	э <b>;</b>	440	D :
ate 470 U 440 U 4300 U 4300 U 420 U 420 U 420 U 420 U 420 U 440 U 440 U 440 U 4300 U 4300 U 420 U 420 U 420 U 420 U 440 U 440 U 4300 U 4300 U 420 U 420 U 420 U 420 U 440 U 440 U 4300 U 4300 U 420 U 420 U 420 U 420 U 440 U 440 U 4300 U 4300 U 420 U 420 U 420 U 420 U 420 U 440 U 440 U 4300 U 4300 U 42	-Bromophenyl phenyl ether	470	Ω	440	Ω	4,300	U	4.300	Ω	420	ı =	000	a	3 5	o ;	061	97 ;	140	118
470 U 440 U 4300 U 4300 U 420 U 420 U 420 U 420 U 420 U 420 U 440 U 440 U 4300 U 4300 U 42	3uty1benzy1phthalate	470	n	440	Ω	4,300	Ω	4.300	· :=	420	) <b>=</b>	27 6	o :	<u> </u>	o ;	440	<b>D</b>	440	Þ
phenol         470         U         440         U         430         U         430         U         430         U         430         U         430         U         420         U <th< td=""><td>arbazole</td><td>470</td><td>n</td><td>440</td><td>Ω</td><td>4.300</td><td>Ω</td><td>4.300</td><td>· =</td><td>2 6</td><td>&gt; =</td><td>024</td><td>) ;</td><td>9 5</td><td>o ;</td><td>440</td><td>&gt;</td><td>440</td><td>Ω</td></th<>	arbazole	470	n	440	Ω	4.300	Ω	4.300	· =	2 6	> =	024	) ;	9 5	o ;	440	>	440	Ω
470 U 440 U 4,300 U 4,300 U 4,300 U 4,200 U 4,200 U 4,20 U	-Chloro-3-methylphenol	470	n	440	n	4,300	D	4 300	· =	720	) <b>=</b>	77	<b>&gt;</b> :	9 5	o ;	440	Þ	440	Þ
The first series of the fi	-Chloroaniline	470	Þ	440	=	4 300	· =	200.7	> :	770	<b>&gt;</b> ;	470	<b>-</b>	904	D	440	n	440	n
enylether 470 U 440 U 4300 U 4300 U 420 U 420 U 400 U 400 U 440 U 440 U 4300 U 4300 U 420 U 420 U 420 U 400 U 400 U 440 U 440 U 4300 U 4300 U 42	-Chloronaphthalene	470	: =	440	> =	7300	> =	000,4	> :	420	o :	420	D	400	Þ	440	n	440	כ
enylether 470 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 400 U 440 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 400 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 400 U 440 U 440 U 4,300 U 4,300 U 420	-Chloropheno!	470	) <u>=</u>	2 9	o :	000,4	<b>)</b> ;	4,300	<b>)</b>	420	n	420	Ω	400	Ω	440	U	440	D
110 1 440 U 4,300 U 4,300 U 420 U 420 U 420 U 400 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 4,300 U 4,400	-Chlorophenyl phenyl ether	077	) <u>=</u>	7 9	> <b>:</b>	4,300	o ;	4,300	D	420	Þ	420	Ω	400	n	440	Ω	440	Ω
110 J 440 U 4,300 U 4,300 U 42	become	0/+ :	ο,	044	<b>&gt;</b>	4,300	D	4,300	Ω	420	n	420	Ω	400	Ω	440	U	440	Ω
Each of the control o	in ysene	011	<b>-</b> ,	440	ם	4,300	Ω	4,300	Ω	420	U	420	n	400	Ω	440	Þ	440	· =
ene 470 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 420 U 400 U 440 U 440 U 440 U 440 U 440 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 440 U 440 U 4300 U 4300 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 440 U 440 U 42	ı-n-Dutyipntnalate	230	ar Br	120	JB	4,300	n	4,300	Ω	180	JB	200	JB	85	JB	001	JB	120	. 🛎
ene 470 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 400 U 440 U 440 U 440 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 420 U 400 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 400 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 420 U 420 U 440	r-n-octylpnthalate	470	Þ	440	Ω	4,300	Ω	4,300	Ω	420	Ω	420	Ω	400	=	440	=	7 7	; =
te 470 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 400 U 440 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 400 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 420 U 420 U 420 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 420 U 440 U 440 U 440 U 4300 U 420 U 420 U 420 U 420 U 440 U 440 U 440 U 4300 U 4300 U 4300 U 4300 U 4300 U 4300 U 440 U 440 U 440 U 440 U 4300 U 4300 U 4300 U 440 U 440 U 440 U 440 U 4300 U 4300 U 440 U 440 U 440 U 440 U 4300 U 4300 U 440 U 440 U 440 U 440 U 440 U 440 U 4300 U 440 U 440 U 440 U 440 U 4300 U 440 U 4300 U 440 U 440 U 440 U 440 U 4300 U 4300 U 440 U 440 U 440 U 4300 U 440 U 4300 U 44	ibenz[a,h]anthracene	470	Ω	440	n	4,300	Ω	4,300	U	420	Ω	420	'n	400	) <u>=</u>	440	) <b>=</b>	2 4	) =
le 470 U 440 U 4,300 U 4,300 U 420 U 420 U 400 U 440 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 400 U 440 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 400 U 440 U 440 U 440 U 4,300 U 420 U 420 U 420 U 420 U 440 U 440 U 440 U 440 U 4300 U 4300 U 4300 U 440	ibenzofuran	470	D	440	Ω	4,300	Ω	4,300	Ω	420	D	420	1	400	· =	270	> =	4	> :
le 470 U 440 U 4,300 U 4,300 U 420 U 420 U 400 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 400 U 440 U 440 U 4,300 U 4,300 U 420 U 420 U 420 U 420 U 440 U 440 U 440 U 4300 U 4300 U 4300 U 440 U 440 U 440 U 440 U 4300 U 4300 U 4300 U 440	2-Dichlorobenzene	470	Ω	440	n	4,300	Ω	4,300	Ω	420	n	420	=	400	) =	740	<b>)</b>	7	> :
ine 470 U 440 U 4,300 U 4,300 U 420 U 420 U 400 U 440 U 440 U 4,300 U 4,300 U 420 U 400 U 440 U 440 U 440 U 4300 U 4300 U 4300 U 4300 U 4300 U 4300 U 4300 U 4300 U 4300 U 4300 U 4300 U 440 U 4	3-Dichlorobenzene	470	Ω	440	U	4,300	Û	4,300	Ω	420	Þ	420	) <b>=</b>	40	> =	9 4	o :	0 4	<b>&gt;</b> :
ine 470 U 440 U 4,300 U 4,300 U 420 U 420 U 440 U 440 U 440 U 4300 U 4300 U 11 4,300 U 440	4-Dichlorobenzene	470	Ω	440	Ω	4,300	Ω	4,300	n	420	Ξ	420	· =	\$ \$	) <u>:</u>	2 4	o :	0 5	<b>&gt;</b> ;
470 U 440 U 4300 II 4300 II 440 U 440 U 440	3'-Dichlorobenzidine	470	n	440	Ω	4,300	D	4,300	Ω	420	) D	420	) <b>=</b>	3 5	o ::	0 4	) :	440	D ;
	4-Dichlorophenol	470	Ω	440	Ω	4.300	=	4 300	=	į	) =	2 5	<b>)</b> ;	} !	<b>5</b>	044	<b>D</b>	440	D

TABLE C.14b (Cont.)

Compound	CI (0-6	CLP4 (0-6 in.)	CI (6-2,	CLP4 (6-24 in.)	CLP4 (24-48 in.)	4 in.)	CLP4-RE (24-48 in.)	-RE in.)	9-0) CI	CLP5 (0-6 in.)	CLF (0-6	CLP5-RE (0-6 in.)	CL (6-2-	CLP5 (6-24 in.)	CLP5 (24-48 in.)	.P5 8 in.)	CLP5-Dup (24-48 in.)	-Dup 3 in.)
, Diethylphthalate	470	Ω	440	ב	4,300	n	4.300	Þ	420	1	420	11	004	=	9	-	440	=
2,4-Dimethylphenol	470	Ξ	440	Ξ	4 300	Ξ	7 300	- =	i ç		3 5		9	) ;	÷ :	<b>.</b> :	1	)
Dimethylphthalate	470	· =	440	> =	4 200	) <u>;</u>	300,4	) :	120	<b>)</b> ;	074		400	) :	440	Þ	440	Þ
4 6-Dinitro 2-mathylohanol	2 -	) :	1	) :	005,+	o ;	4,300	<b>)</b>	470	<b>-</b>	420		400	Þ	440	n	440	$\supset$
+,0-r,min.0-z-meuryipnenoi	1,200	⊋	91,1	<u> </u>	11,000	Þ	11,000	D	1,100	n	1,100	Ω	066	Ω	1,100	n	1,100	Þ
2,4-Dinitrophenol	1,200	Ω	1,100	U	11,000	n	11,000	Ω	1,100	ם	1,100	ב	066	'n	1.100	Ω	1.100	
2,4-Dinitrotoluene	470	Ω	440	Ŋ	4,300	U	4,300	ח	420	ם	420	n	400	· 🗅	440	> =	440	-
2,6-Dinitrotoluene	470	Ω	440	Ω	4,300	Ω	4,300	D	420	n	420		400	) I	440	) <u>=</u>	440	> =
Fluoranthene	150	ſ	62	-	4,300	n	4,300	ב	420	1	420		400	· <u>-</u>	2 6	) [	2 5	) =
Fluorene	470	מ	440	Ω	4,300	Ω	4.300	ם	420	- 11	420		400	) <u>=</u>	2 7	) [	7	> =
Hexachlorobenzene	430	J	440	U	4,300	D	4,300	۵	420	D	420		400	) =	440	) <u>-</u>	2 4	2
Hexachlorobutadiene	470	n	440	Þ	4,300	Þ	4,300	Þ	420	Þ	420	n	400	) =	440	) <u>=</u>	440	=
Hexachlorocyclopentadiene	470	D	440	n	4,300	Ω	4,300	n	420	ב	420	D	400	· ::	440	) <u>=</u>	440	) <u>=</u>
Hexachloroethane	470	Þ	440	U	4,300	Þ	4,300	Þ	120	D	8	ſ	400	o D	440	) D	440	) =
Indeno[1,2,3-c,d]pyrene	470	Ω	440	Ω	4,300	n	4,300	D	420	D	420	Þ	400	n	440	· ⊃	440	î
Isophorone	470	D	440	Ω	4,300	Ω	4,300	U	420	D	420	Þ	400	Ω	440	Þ	440	
2-Methylnaphthalene	470	Ŋ	200	Ω	4,300	Ω	4,300	n	420	מ	420	ם	400	n	440	n	440	
2-Methylphenol	470	ם	110	Ω	4,300	U	4,300	D	420	D	420	Ω	400	Ω	440	n	440	
4-Methylphenol	470	Ω	440	Ω	4,300	Ω	4,300	Ω	420	D	420	ב	400	ņ	440	n	440	'n
Naphthalene	470	Ü	120	Þ	4,300	n	4,300	ח	420	Ω	420	Ω	400	D	440	n	440	ב
2-Nitroaniline	1,200	n	1,100	Ω	11,000	D	11,000	n	1,100	n	1,100	n	066	ລ	1.100	Þ	1.100	₽
3-Nitroaniline	1,200	U	1,100	U	11,000	n	11,000	Þ	1,100	D	1,100	n	066	D	1.100	Ω	001	
4-Nitroaniline	1,200	n	1,100	Ω	11,000	Ω	11,000	Ω	1.100	n	1,100	n	066	n	1,100	D	1.100	
Nitrobenzene	470	Ω	440	n	4,300	n	4,300	U	420	ח	420	ם	400	ם	440	D	440	$\supset$
2-Nitrophenol	470	Ω	440	Ω	4.300	Ω	4,300	n	420	D	420	n	400	D	440	ם	440	
4-Nitrophenol	1,200	D	1,100	Ω	11.000	U	11,000	Ω	1.100	ם	1,100	D	066	n	1,100	ח	1.100	
N-Nitroso-di-n-propylamine	470	Þ	440	n	4,300	Ð	4,300	D	420	ח	420	Þ	400	n	440	D	440	D
N-Nitrosodiphenylamine	470	Ω	410	ב	4.300	Ω	4,300	Ω	420	מ	420	'n	007	Ω	440	Ω	440	$\supset$
Pentachlorophenol	1,200	Ω	1.100	D	11,000	Ω	11,000	ח	1,100	ח	1,100	þ	066	D	1,100	Ω	1.100	⊃
Phenanthrene	65	_	120	ī	4,300	n	4,300	Ŋ	420	n	420	n	907	ב	440	Ω	740	$\supset$
Phenol	420	n	440	n	4,300	D	4,300	D	420	n	420	n	007		4.10	Ξ	957	=

TABLE C.14b (Cont.)

Pyrone         CRPA         CLPA         <								Soil Conc	Soil Concentrations (µg/kg) at Various Depth Intervals	μg/kg) at	Various D	epth Inter	vals						
130   1   10   10   10   10   10   10	Compound	(0-6)	P4 in.)	CLP (6-24 i	4 n.)	CLP4 (24-48 i	п.)	CLP4-F (24-48 i	₹E n.)	CLP (0-6 i	.5 n.)	CLP5 (0-6)	-RE in.)	CLP (6-24	5 in.)	CLI (24-48	P5 3 in.)	CLPS-	Dup in.)
1,000   1,10	Pyrene	130	-	98	~	4,300	Ω	4.300	1	750	=	9	:	9	,				
1,200   Lange   Lang	1,2,4-Trichlorobenzene	470	Ξ	440	=	7 300	. :		) ;	27	5	074	>	94	<b>-</b>	440	D	440	Ω
Table   1.20   1.1   1.10   1.1   1.10   1.1	2.4 S. Treichlorophorop	04 .	<b>)</b> ;	) }	<b>&gt;</b>	4,300	<b>-</b>	4,300	Þ	420	⊃	420	Þ	400	ņ	440	Ω	440	Þ
CLP   CLP	2,4,3-1 itemorophenol	1,200	D	1,100	n	11,000	Ω	11,000	Ω	1,100	Ω	1,100	n	066	11	2	1	2	) =
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	2,4,6-Trichlorophenol	470	ם בור בור	ı	D .	4,300	n	4,300	U	420	n	420	n	400	ם ה	440	ם מ	440	) <b>=</b>
Cumpound   CLPs   CLP								Soil Conce	entrations (p	ug/kg) at	Various De	pth Interv	als	; ; 1 ]	]	 	[ [ [ ]	1	i !
CLPS         CLPS <th< td=""><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></th<>																			
hthere 880 U 420 U 420 U 510 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 820 U 820 U 820 U 420 U	Compound	CLP (0-6 ii	n.)	CLP6 (6-24 ir	); (:	CLP6-R (6-24 in	E	CLP7 (0-6 in.	_	CLP (6-24 i	7 n.)	CLP7- (6-24)	RE n.)	CLP (24-48	7 in.)	CLP (0-6 i	& <u>_</u>	CLP M-6 ii	٠ .
httylene 880 U 420 U 420 U 510 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 510 U 420 U 420 U 510 U 420 U 420 U 510 U 420 U 510 U 420 U 510 U 420 U 510 U 420 U 510 U 510 U 420 U 510	Acenaphthene	880	E	420	=	000			;										
The control of the co	Acenaphthylene	088	. =	3 5	) <u>:</u>	074	<b>&gt;</b> ;	010	<b>-</b>	420	n	420	Þ	420	Ω	2,000	þ	820	Ω
Secondary   Seco	Anthropona	000	> ;	074	<b>.</b>	470	<b>-</b>	210	n	420	n	420	Ω	420	Ω	2,000	þ	820	Ω
Sign   Sign	Anunacene	088	)	450	D	420	Ω	210	Ω	420	U	420	ם	420	U	2.000	Ω	820	- =
Plystrate   880   U   420   U   42	Benz[ajanthracene	880	D	98	-	28	ſ	510	Ω	420	n	420	n	420	n	2.000	) =	820	) <u>=</u>
Higherylating   State   Lange   Lang	Benzolalpyrene	880	D	420	Ω	420	Ω	510	U	420	n	420	n	420	· D	2000	) <b>=</b>	070	> =
hilperylene 880 U 420 U 420 U 510 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 510 U 420 U	Benzo[b]fluoranthene	880	n	420	Ω	420	n	510	Ω	420	U	420	Ω	420	· =		) =	070	> <b>:</b>
Houcranthene   SS   U   420   U   420   U   510   U   420   U	Benzo[g,h,i]perylene	880	Ω	420	n	420	Ω	510	Ω	420	D	420	· 1	420	) <u>=</u>	2000	) <u>:</u>	020	o :
orocethoxy)methane 880 U 420 U 420 U 510 U 420 U 420 U 510 U 420 U 420 U 420 U 510 U 420 U 420 U 510 U 420 U 510 U 420 U 420 U 510 U 420 U 510 U 420 U 510 U 420 U 510 U 420 U 510 U 420 U 510 U 420 U 510 U 420 U 510 U	Benzo[k]fluoranthene	880	Ω	420	n	420	Ω	510	D	420		75	) <b>:</b>	2	) :	2,000	<b>)</b>	820	<b>)</b>
orocetry) behits 880 U 420 U 420 U 510 U 420 U 420 U 620 U 420 U 420 U 420 U 420 U 620 Bis(2-chloroethoxy)methane	880	U	420	U	420	D	510	· =	420	) <u>=</u>	420	o :	470	<b>)</b> ;	2,000	<b>D</b>	820	Þ	
ylbersylphthalate         880         U         420         U	Bis(2-chloroethyl)ether	880	Ω	420	Ω	420	11	\$10	· =	2 5	) <u>:</u>	420	<b>&gt;</b> ;	420	<b>-</b>	2,000	Þ	820	Þ
ythexyliphthalate         180         1B         160         1B         130         1B         170         1B         140         1B         420         0         420         0         420         0         420         0         2000         0         820           phenyl phenyl ether         880         U         420         U <td>Bis(2-chloroisopropyl)ether</td> <td>880</td> <td>n</td> <td>420</td> <td>D</td> <td>420</td> <td>) II</td> <td>510</td> <td>) <u>=</u></td> <td>2 6</td> <td><b>)</b></td> <td>024</td> <td>o :</td> <td>420</td> <td>D ;</td> <td>2,000</td> <td>D</td> <td>820</td> <td>n</td>	Bis(2-chloroisopropyl)ether	880	n	420	D	420	) II	510	) <u>=</u>	2 6	<b>)</b>	024	o :	420	D ;	2,000	D	820	n
phenyl phenyl ether 880 U 420 U 420 U 510 U 420 U 420 U 510 U 420	Bis(2-ethylhexyl)phthalate	180	JB	160	JB	130	, at	02.1	, <u>e</u>	07.	o 8	6 6	<b>5</b>	420	o ;	2,000	Ω	820	Ω
2ylphthalate         880         U         420         U	4-Bromophenyl phenyl ether	880	Ω	420	n	420	U	510	! =	000	3 =	î ç	gr :	S 5	g ;	480	118	120	JB
Sko U 420 U 420 U 510 U 420 U 420 U 510 U 420 U 520 U	Butylbenzylphthalate	880	n	420	Ω	420		\$10	· =	22.	) <b>:</b>	420	<b>&gt;</b> ;	420	o ;	2,000	D	820	D
3-methylphenol 880 U 420 U 420 U 510 U 420 U 420 U 420 U 420 U 5200 U 820 U 300 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 820 U 420  Carbazole	880	n	420	11	420		015	· =	2 5	<b>)</b>	420	<b>)</b>	470	<b>-</b>	2,000	D	820	n	
uniline 880 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 820 U 820 U 420 U 820 U 820 U 420 U 820	4-Chloro-3-methylphenol	880	=	420	) <u>=</u>	3 5	> <b>:</b>	010	o ;	420	<b>-</b>	420	Ď	420	n	2,000	Ω	820	Ω
appropriate 880 U 420 U 420 U 420 U 420 U 420 U 420 U 420 U 820 U 820 U 420 U 420 U 820 U 820 U 420 U 420 U 420 U 420 U 820 U 820 U 420 U 420 U 420 U 420 U 820 U 820 U 420 U 420 U 420 U 420 U 820 U 820 U 820 U 420 U 420 U 420 U 420 U 420 U 820 U	4-Chloroaniline	088	) <u> </u>		o :	074	o ;		<b>5</b>	420	Þ	420	Ω	420	Ω	2,000	Ω	820	U
otheroid 880 U 420 U 420 U 510 U 420 U 420 U 420 U 2,000 U 820 U 420 U 2,000 U 820 U 420 U 420 U 420 U 2,000 U 820 U 420 U 420 U 420 U 420 U 420 U 420 U 820 U 820 U 820 U 97 J 64 J 510 U 420 U 420 U 420 U 2,000 U 820 U 820 U	2-Chloronophyholene	000	o ;		<b>)</b>	420	<b>-</b>		n	420	Ω	420	Ω	420	n	2,000	n	820	D
Menyl phenyl ether 880 U 420 U 420 U 510 U 420 U 420 U 420 U 2,000 U 820 Henyl phenyl ether 880 U 420 U 420 U 420 U 420 U 820 U 820 U 87 J 64 J 510 U 420 U 420 U 420 U 2,000 U 820	2 Chlosethered	989	o ;		ے ت	420	)		Ω	420	Ω	420	n	420	n	2,000	n	820	Ω
which properties the state of t		088	<b>5</b>		)	420	ב		Ω	420	n	420	Ω	420	n	2,000	D	820	. =
880 U 97 J 64 J 510 U 420 U 420 U 420 U 2,000 U 820	4-Cillotophenyi phenyi ether	088	o :		D	420	כ		n	420	Ω	420	D	420	כ	2,000	Ω	820	n
	Ciii y seine	880	<b>-</b>		_	64	_		ם	420	Ω	420	Ω	420	ם	2,000	n	820	

TABLE C.14b (Cont.)

ı	כד	CLP6	O	CLP6	CLP	CLP6-RE	CLP7	P7	Ū	CLP7	C	CLP7-RE		CLP7		CLP8		כ	01.P9
Compound	)-O)	(0-6 in.)	;-9)	(6-24 in.)	(6-24 in.)	l in.)	(0-6 in.)	in.)	(6-2	(6-24 in.)	-9)	(6-24 in.)		(24-48 in.)	n.)	(0-6 in.)	in.)	00	(0-6 in.)
Di-n-butylphthalate	880	Ω	160	) JB	170	JB	190	JB	94	JB	420	Ω (	_	92	JB	2.000	Ω	120	H
Di-n-octylphthalate	880	Ω	420	ם	420	Ω	510	ם	420	n	420	n	_	420	n	2.000	<u> </u>	820	: =
Dibenz[a,h]anthracene	880	D	420	Ω	420	n	510	D	420	Ω	420		-	420	n	2,000	ı D	820	) ⊃
Dibenzofuran	880	D	420	n	420	Ω	510	n	420	Þ	420	n (		420	Ω	2.000	. 11	820	· 🗀
1,2-Dichlorobenzene	880	D	420	ם	420	D	510	Ω	420	כ	420	Ω		420	- 1	2 000	, =	820	) =
1,3-Dichlorobenzene	880	U	420	Þ	420	Þ	510	n	420	Ω	420		_	420	) I	2000	> =	820	) <u>:</u>
I,4-Dichlorobenzene	880	U	420	n	420	n	510	Ω	420	Ω	420			420	; <u>D</u>	2.000	) <u>=</u>	820	) <u> </u>
3,3'-Dichlorobenzidine	880	n	420	D	420	ם	510	n	420	Ω	420	د		420	n	2.000	· 11	820	) =
2,4-Dichlorophenol	880	n	420	n	420	n	510	Ω	420	n	420	· D		420	n	2.000	n	820	) <u>=</u>
Diethylphthalate	880	n	420	Ω	420	Ω	510	U	420	n	420	D		420	n	2,000	n	820	n
2,4-Dimethylphenol	880	D	420	D	420	Ω	510	D	420	n	420	n		420	ם	2,000	Ω	820	D
Dimethylphthalate	880	Ω	420	Þ	420	n	510	Ω	420	ח	420	ח	•	420	Ω	2,000	D	820	
4,6-Dinitro-2-methylphenol	2,200	Ω	1,000	D	1,000	n	1,300	D	1,000	Ω	1,100	D _	1,0	000,	Ω	5,100	Ω	2,000	ב
2,4-Dinitrophenol	2,200	D	1,000	Ω	1,000	n	1,300	ם	1,000	Þ	1,100	n	);	000,1	D	5,100	n	2,000	n
2,4-Dinitrotoluene	880	Ω	420	Þ	420	Þ	510	ב	420	Þ	420	Þ	7	420	U	2,000	Ω	820	Þ
2,6-Dinitrotoluene	880	Ω	420	Ω	420	n	510	Ω	420	Ω	420	Þ	,	420	Ω	2,000	Þ	820	ח
Fluoranthene	880	D	140	ſ	140	ī	510	Ω	420	Ω	420	D	7	420	ם	2,000	Ω	820	Þ
Fluorene	880	ח	420	n	420	ם	510	Ω	420	Ω	420	n	7	420	Ω	2,000	D	820	Ω
Hexachlorobenzene	880	Ω	65	_	75	'n	510	Ω	420	n	420	Ω	7	420	Ω	490	'n	200	n
Hexachlorobutadiene	880	Þ	420	Ω	420	n	510	n	420	D	420	D	7	420	D	2,000	Ω	820	ח
Hexachlorocyclopentadiene	880	D	420	Ω	420	ם	510	D	420	n	420	D	ч	420	n	2,000	n	820	Þ
Hexachloroethane	880	n	420	Ω	420	n	510	Ω	420	n	420	D	4	420	Ω	2,000	ם	820	Ω
Indeno[1,2,3-c,d]pyrene	880	Þ	420	Ω	420	Ω	510	n	420	Þ	420	Ω	4	420	ם	2,000	D	820	Þ
Isophorone	880	⊃	420	Ω	420	n	510	n	420	ח	420	Ω	-13	420	Ω	2.000	ב	820	Ξ
2-Methylnaphthalene	880	D	120	-	420	D	510	Ω	420	n	420	n	4	420	n	2,000	n	820	ב
2-Methylphenol	880	⊃	80	-	7.5	ſ	510	D	420	D	420	D	4	420	ב	2,000	D	820	n
4-Methylphenol	880	n	92	_	16	n	510	ח	420	ח	420	Þ	4	420	⊃	2,000	Þ	820	Þ
Naphthalene	880	ם	19	►,	63	7	510	ם	420	D	420	⊃	7	420	D	2.000	Ω	820	Ω
2-Nitroaniline	2,200	ב	1,000	Ω	1.000	Ω	1,300	Ω	1,000	D	1,100	D	0.1	000.1	n	5,100	Ω	2,000	n
3-Nitroaniline	2,200	⊃	1,000	ם	1,000	D	1,300	ם	1.000	5	1,100	D	000.1	8	ם	5.100	ם	2,000	Ω

TABLE C.14b (Cont.)

						Soil	Concentratio	ns (µg/kg) at	Soil Concentrations (µg/kg) at Various Depth Intervals	h Intervals						
Compound	CLP6 (0-6 in.)	n.)	CLP6 (6-24 in.)		CLP6-RE (6-24 in.)	2)	CLP7 (0-6 in.)	CLP7 (6-24 in.)	77 in.)	CL <i>P7-</i> RE (6-24 in.)		CLP7 (24-48 in.)	7 9-0	CLP8	CLP9	8.5
4-Nitroaniline	2.200	Ω	000	-	:	•									0-0)	
Nitrobenzene	880	· =	2001	· :	0 000,	1,300	⊃ 8	000,1	n n	1,100	J 1,000	0 n	5,100	Ω	2.000	
2-Nitrophenol	088	) <b>=</b>	750	o :	420 0	δ.	510 U	420	D	420 L	J 420	0 n	2,000	n	820	=
4-Nitrophenol	30.	> <b>:</b>	1,000	· ·	420 U	S	510 U	420	Ω	420 t	J 420	n o	2,000	n	820	· =
N-Nitroso din nanalamina	007,2	<b>)</b> ;	000,1	<b>~</b>	) 000,	1,300	n 00	1,000	U I	1,100	J 1,000	n (	\$ 100	· =	250	) <u>:</u>
N Missondiak and a missondiak	980	<b>-</b>	420	n	420 U	51	510 U	420	Ω	420	J 420	- 1	000	) <b>:</b>	000,4	> ;
osocipieny idmine	880	Þ	420	ם	420 U	510	10 U	420	11	1 000			7,000	י כ	079	>
Pentachlorophenol	2,200	Ω	1,000	U I.	000	1 300	1	1 2	· • :	074	420	0	2,000	D	820	Þ
Phenanthrene	880	n	74		7 17	7.	> :	J.,000	0	1,100	1,000	n c	5,100	Ω	2,000	Ω
Phenol	880	Ω	420	_		010	o ;	420	n	420 L	1 420	n c	2,000	Ω	820	n
Pyrene	880	ב	2 081	<b>.</b>	0 -	010	n ;	420	n	420 U	420	n c	2,000	Ω	820	Ω
1,2,4-Trichlorobenzene	880	- 12			140	510	0	420	Ω	420 U	420	n c	2,000	D	4	-
2,4,5-Trichlorophenol	2.200	) <u>-</u>			4.5	210	D 0	420	n	420 U	420	n c	2,000	Ω	820	Ω
2,4,6-Trichlorophenol	880	) <u>:</u>	420	1,	000,	1,300	י מ ס	1,000	U 1,	U, 100	1,000	D ~	5,100	Ω	2,000	n
					0 07	NC		420	Ω	420 U	420	n	2,000	Ω	820	П
	000	0	470		420 U	510		D O	0 U 420	U 420 U	ū	U 420 U 420 U	U 420 U 420 U 420	U 420 U 420 U 420 U	U 420 U 420 U 420 U	U 420 U 420 U 420 U 2,000 U

<sup>a</sup> Sampled by ANL, analyzed by Weston Gulf Coast, CLP/HSL semivolatile organics.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

B = analyte was found in the associated blank.

TABLE C.14c Semivolatile Organics Analyses for Soil Borings, May 1994<sup>a</sup>

			Sc	oil Conce	Soil Concentrations (µg/kg) at Various Depth Intervals	µg/kg)	at Variou	s Depth	Intervals			
Compound	TBSPBOR1 (0-2 ft)	3OR1	TBSPBOR1 Dup (0-2 ft)	t1 Dup t)	TBSPBOR1 (2-4 ft)	OR1	TBSPBOR1 (4-6 ft)	OR1	TBSPBOR1 (6-8 ft)	OR1	TBS]	TBSPBOR1 (8-10 ft)
Acenaphthene	380	n	380	Ω	410	U	400	Ŋ	410	n	410	ņ
Acenaphthylene	380	Ω	380	U	410	n	400	n	410	Ω	410	Ω
Anthracene	380	Ω	380	Ω	410	Ω	61	ŗ	410	n	410	n
Benz[a]anthracene	42	ſ	380	Ω	410	n	400	Ω	410	Ω	410	Þ
Benzo[a]pyrene	380	Ω	380	n	410	Ω	400	Ω	410	Ω	410	Ω
Benzo[b]fluoranthene	380	Ω	380	n	410	Ω	400	Ω	410	Ω	410	U
Benzo[g,h,i]perylene	380	Ω	380	Ω	410	n	400	Ω	410	Ω	410	U
Benzo[k]fluoranthene	380	Ω	380	n	410	D	400	Ω	410	Ω	410	Ω
Bis(2-chloroethoxy)methane	380	Ω	380	n	410	D	400	n	410	Ω	410	Ω
Bis(2-chloroethyl)ether	380	Ω	380	n	410	D	400	n	410	Ω	410	n
Bis(2-chloroisopropyl)ether	380	Ω	380	Ω	410	Ω	400	Ω	410	Ω	410	n
Bis(2-ethylhexyl)phthalate	240	-	380	Ω	410	Ω	400	Ω	54	<b>—</b>	410	n
4-Bromophenyl phenyl ether	380	Ω	380	U	410	Ω	400	D	410	Ω	410	Ω
Butylbenzylphthalate	380	Ω	380	n	410	Ω	400	Ω	410	Ω	410	D
Carbazole	380	Ω	380	U	47	<u>-</u>	88	J	410	Ω	410	Ω
4-Chloro-3-methylphenol	380	Ω	380	U	410	n	400	Ω	410	Ω	410	n
4-Chloroaniline	380	Ω	380	n	410	Ω	400	Ω	410	n	410	n
2-Chloronaphthalene	380	Ω	380	Ω	410	Ω	400	n	410	n	410	n
2-Chlorophenol	380	Ω	380	Ω	410	n	400	D	410	D	410	Ω
4-Chlorophenyl phenyl ether	380	n	380	U	410	Ω	400	Ω	410	Þ	410	Ω
Chrysene	43	_	380	Ω	410	n	400	Ω	410	D	410	n
Di-n-butylphthalate	390	В	290	JB	310	JB	290	JB	320	JB	110	JB
Di-n-octyl phthalate	380	n	380	Ω	410	Ω	400	n	410	n	410	Ω
Dibenz[a,h]anthracene	380	Ŋ	380	Ω	410	n	400	Ω	410	Ω	410	n

TABLE C.14c (Cont.)

	TRSPROP	NOP 1	TREBODORI	2	100	,	i					
Compound	(0-2 ft)	(E)	(0-2 ft)	dna	(2-4 ft)	E)	TBSPBOR I (4-6 ft)	OR1	TBSPBORI (6-8 ft)	30R1 ft)	TBS (8-	TBSPBOR1 (8-10 ft)
Dibenzofuran	380	n	38	_	410	ם	400		710	=	;	
1,2-Dichlorobenzene	380	Ŋ	380		710	)	2 5	) ;	410		410	$\supset$
1,3-Dichlorobenzene	380	Ω	380	) <u> </u>		ב כ	400	); 	410	<b>D</b>	410	n
1,4-Dichlorobenzene	380	ם ה	380	) <del>[</del>	410	) =	400	); 	410	Þ	410	Ω
3,3'-Dichlorobenzidine	380	n	380	) E	410	> =	004	)	410	D ;	410	Ω
2,4-Dichlorophenol	380	Ω	1 086	. <u> </u>	410	) <u>=</u>	400	> <b>=</b>	410	) :	410	D;
Diethylphthalate	190	_	99	_	410	) D	79	> <b>-</b>	410	)	410	<b>)</b> ;
2,4-Dimethylphenol	380	Ω	380	Ω	410	11	) S	, :	410	<b>&gt;</b> ;	410	<b>-</b>
Dimethylphthalate	380	Ω	·	Þ	410	) =	400	> =	410	) :	410	) (
4,6-Dinitro-2-methylphenol	096	Ω	940	J	1,000	ם	1.000	) <u>=</u>	1 000	) <u>;</u>	410	) :
2,4-Dinitrophenol	096	Ω	940 L	J	1,000	n	1,000	n	1,000	) <b>=</b>	1,000	)
2,4-Dinitrotoluene	380	Ω	380	J	410	Ω	400	n	410	) <u>=</u>	710	> <b>=</b>
2,6-Dinitrotoluene	380	Ω	380	J	410	Ω	400	Û	410	) <u>=</u>	10	) <u>;</u>
Fluoranthene	73	ſ	63 J		410	Ω	400	· I	21.7	) <u>:</u>	017	<b>)</b> ;
Fluorene	380	Ω	380 U	<u></u>	410	· =	× ×	) <b>-</b>	410	)	410	<b>)</b> ;
Hexachlorobenzene	200	_	210 J		410	, D	400	, <u>=</u>	410	) <u> </u>	410	) ;
Hexachlorobutadiene	380	U	380 U	<b></b>	410	Ω	400	) <u> </u>	110	> <b>:</b>	410	o ;
Hexachlorocyclopentadiene	380	Ω	380 U	_	410	11	400	) <u> </u>	410	) ;	410	<b>)</b>
Hexachloroethane	580		350 J		410	) <u> </u>	2 5	) <u> </u>	410	o ;	410	<b>)</b>
Indeno[1,2,3-c,d]pyrene	380	Ω	380 11	_	710	> =	000	<b>)</b> ;	410	) 	410	Ď
Isophorone	380	. 1	780		01+	<b>)</b>	400	<b>-</b>	410	Þ	410	Ŋ
2-Methylnanbthalene	000	۰, د	300 0		410	)	400	D	410	D	410	Ω
2-Methylabonol	011	- ;	230 J		410	Ω	400	Ŋ	410	D	410	Ω
treety ipitetion	280	<b>-</b>	380 II		710	<u>, , , , , , , , , , , , , , , , , , , </u>	007	,				

TABLE C.14c (Cont.)

	المستوجة المستقدية											
Compound	TBSPBOR (0-2 ft)	SOR 1 ft)	TBSPBOR1 Dup (0-2 ft)	R I Dup ft)	TBSPBOR (2-4 ft)	OR1 ft)	TBSPBOR (4-6 ft)	SOR1	TBSPBOR (6-8 ft)	OR1	TBS)	TBSPBOR1 (8-10 ft)
4-Methylphenol	380	Ω	380	Ω	410	Ω	400	Ω	410	Ω	410	Ω
Naphthalene	380	D	61	r	410	Ω	400	Ω	410	D	410	Ω
2-Nitroaniline	096	Ω	940	Ω	1,000	Ω	1,000	Ω	1,000	Ω	1,000	Ω
3-Nitroaniline	096	Ω	940	Ω	1,000	D	1,000	Ω	1,000	Ω	1,000	n
4-Nitroaniline	096	n	940	Ω	1,000	Ω	1,000	n	1,000	n	1,000	D
Nitrobenzene	380	n	380	Ω	410	Ω	400	n	410	n	410	Þ
2-Nitrophenol	380	D	380	Ŋ	410	Ω	400	n	410	Ω	410	n
4-Nitrophenol	096	Ω	940	Ω	1,000	Ω	1,000	D	1,000	n	1,000	n
N-Nitroso-di-n-propylamine	380	D	380	D	410	U	400	n	410	ח	410	Ω
N-Nitrosodiphenylamine	650		950		64	۳,	120	Г	44	,	410	D
Pentachlorophenol	096	Ω	940	D	1,000	Þ	1,000	Ω	1,000	Ω	1,000	Ω
Phenanthrene	66	-	130	J	53	_	62	<del>ب</del>	410	n	410	n
Phenol	380	Ω	380	n	410	n	400	n	410	Ω	410	n
Pyrene	63	'n	74	J	42	ſ	400	n	410	D	410	Ω
1,2,4-Trichlorobenzene	130	ŗ.	340	ш,	410	Ω	400	Ω	410	Þ	410	Ω
2,4,5-Trichlorophenol	096	n	940	Ω	1,000	n	1,000	n	1,000	n	1,000	n
2,4,6-Trichlorophenol	380	ב	380	Ω	410	n	400	ח	410	11	410	1

TABLE C.14c (Cont.)

Compound	TBSP (10-	TBSPBOR1 (10-12 ft)	TBSI (0-	TBSPBOR2 (0-2 ft)	TBSPB(	TBSPBOR2 Dup (0-2 ft)	TBSP (2-	TBSPBOR2 (2-4 ft)	TBSPBO)	TBSPBOR2 (4-6 ft)	TBSPBOR2 (6-8 ft)	OR2 ft)	TBSPBOR2	BOR2
Acenanhthene	300	=	000	;										
, condiminate	086	)	390	)	390	Ŋ	390	n	390	Ω	400	n	400	
Acenaphthylene	390	N	390	Ω	390	n	390	n	390	Ω	400	11	007	) <u> </u>
Anthracene	390	Ω	390	U	390	Ω	390	n	390	) <u> </u>	400	) <u> </u>	904	)
Benz[a]anthracene	390	Ω	390	Ω	390	Ω	390	ū	390	) <b>=</b>	400	> =	9 6	)
Benzo[a]pyrene	390	Ω	390	Ω	390	Ω	390	o D	390	) <u> </u>	400	> =	400	
Benzo[b]fluoranthene	390	Ω	390	Ω	390	U	390	n	330	ם מ	400	) <u> </u>	400	) <u> </u>
Benzo[g,h,i]perylene	390	Ω	390	Ω	390	Ω	390	Ω	390	n	400	) <u>=</u>	400	) =
Benzo[k]fluoranthene	390	Ω	390	Ω	390	Ω	390	Ω	390	D	400	· =	907	) <u> </u>
Bis(2-chloroethoxy)methane	390	Ŋ	390	Ω	390	Ω	390	Ω	390	n	400	> =	90	) <u>;</u>
Bis(2-chloroethyl)ether	390	n	390	r	390	n	390	Ω	390	ם	400	) =	400	) <u> </u>
Bis(2-chloroisopropyl)ether	390	n	390	Ω	390	n	390	D	390	Ω	400	· 1	400	) <u> </u>
Bis(2-ethylhexyl)phthalate	390	D	390	Ω	390	U	390	Ω	390	U	400	'n	400	) <u> </u>
4-Bromophenyl phenyl ether	390	Þ	390	Ŋ	390	Ω	390	U	390	n	400		400	) <u> </u>
Butylbenzylphthalate	390	n	390	Ω	390	Ω	390	U	390	Ω	400	n	400	=
Carbazole	390	Ω	390	n	390	D	390	U	390	n	400	· 1	400	) =
4-Chloro-3-methylphenol	390	n	390	Ω	390	Ω	390	Ω	390	Ω	400	, D	400	) <u>=</u>
4-Chloroaniline	390	n	390	Ω	390	Ω	390	Ω	390	Ω	400	n	400	) =
Z-Chloronaphthalene	390	D	390	n	390	Ω	390	U	390	Ŋ	400	, D	400	) <u>=</u>
2-Chlorophenol	390	Ω	390	Ω	390	Ω	390	U	390	Ŋ	400		400	) <u> </u>
4-Chlorophenyl phenyl ether	390	Ω	390	Ω	390	Ω	390	Ω	390	U	400	1	8 5	) ;
Chrysene	390	Ω	390	Ω	390	Ω	390	Ω	390	n	400	) <u>=</u>	400	) <u>=</u>
Di-n-butylphthalate	150	JB	230	JB	190	JB	230	H	096	d.	27.0	) £	2 4	> \$
								)	22	q	740	25	2	Υ

TABLE C.14c (Cont.)

Compound	TBSP (10-	TBSPBOR1 (10-12 ft)	TBSPBO (0-2 ft)	TBSPBOR2 (0-2 ft)	TBSPBOR2 Dup (0-2 ft)	R2 Dup ft)	TBSPBOR2 (2-4 ft)	30R2 ft)	TBSPBOR2 (4-6 ft)	30R2 ft)	TBSPBOR2 (6-8 ft)	OR2 ft)	TBSPBOR2 (8-10 ft)	OR2 ft)
Dibenz[a,h]anthracene	390	n	390	U	390	Ω	390	D	390	n	400	Ω	400	Ω
Dibenzofuran	390	U	390	Ω	390	Ω	390	Ω	390	n	400	D	400	D
1,2-Dichlorobenzene	390	Ω	170	J	390	Ω	390	Ω	390	n	400	Ω	400	Û
1,3-Dichlorobenzene	390	Ω	390	Ω	390	Ω	390	U	390	Ω	400	Ω	400	Ω
1,4-Dichlorobenzene	390	ND	390	n	390	D	390	U	390	Ω	400	n	400	Þ
3,3'-Dichlorobenzidine	390	Ω	390	Ω	390	Ω	390	n	390	Ω	400	Ω	400	Ω
2,4- Dichlorophenol	390	n	390	Ŋ	390	Ω	390	n	390	Ω	400	n	400	n
Diethylphthalate	390	Ω	390	Ω	390	Ω	390	Ω	390	Ω	400	Þ	400	Ω
2,4-Dimethylphenol	390	n	390	n	390	Ω	390	n	390	Ω	400	D	400	Ω
Dimethylphthalate	390	Ŋ	390	Ω	390	Ω	390	Ω	390	Ω	400	D	400	D
4,6-Dinitro-2-methylphenol	086	D	970	Ω	096	n	086	Ω	086	Ω	066	n	1,000	D
2,4-Dinitrophenol	086	n	970	D	096	Ω	086	Ω	086	n	066	n	1,000	Ω
2,4-Dinitrotoluene	390	Ω	390	D	390	U	390	D	390	Ω	400	Ω	400	n
2,6-Dinitrotoluene	390	U	390	D	390	Ω	390	Ω	390	Ω	400	Ω	400	D
Fluoranthene	390	n	390	n	390	Ω	390	n	390	D	400	Ω	400	Þ
Fluorene	390	D	390	D	390	n	390	Ω	390	n	400	Ω	400	D
Hexachlorobenzene	390	Ω	390	Ω	390	Ω	390	Ŋ	390	Ω	400	Ω	400	Ω
Hexachlorobutadiene	390	n	390	Ω	390	Ω	390	n	390	Ŋ	400	Þ	400	n
Hexachlorocyclopentadiene	390	Ω	390	Ω	390	n	390	n	390	n	400	n	400	D
Hexachloroethane	390	D	390	Ω	390	Ω	390	Ω	390	Ω	400	Ω	58	<del>ب</del>
Indeno[1,2,3-c,d]pyrene	390	Ω	390	n	390	Ω	390	Ω	390	Ω	400	n	400	D
Isophorone	390	Ω	390	Ω	390	Ω	390	Ω	390	Ω	400	n	400	Ω
2-Methylnaphthalene	390	Ω	390	Ω	71	ŗ	390	Ω	390	Ω	400	Ω	400	n

TABLE C.14c (Cont.)

Compound 2-Methylphenol 4-Methylphenol Naphthalene	TBSPBOR1 (10-12 ft) 390 U													
2-Methylphenol 4-Methylphenol Naphthalene	390	BOR1 2 ft)	TBSF (0-	TBSPBOR2 (0-2 ft)	TBSPBOR2 Dup (0-2 ft)	Dup	TBSPBOR2 (2-4 ft)	30R2 ft)	TBSPBOR2 (4-6 ft)	BOR2 ft)	TBSPBOR2 (6-8 ft)	OR2 ft)	TBSPBOR2 (8-10 ft)	OR2 ft)
4-Methylphenol Naphthalene		n	390	D	3 300		390	1	300	1.1	904	=	100	;
Naphthalene	390	Ω	390	11	300	, -	300	) <u>=</u>	000	)	904	<b>)</b> ;	400	<b></b>
	390	n	390	n	3.000	o ⊢	390	o ::	390	) <u>;</u>	400	);	400	n :
2-Nitroaniline	086	Ω	970	ב	1 096	,	080	o =	080	ב כ	904	) :	400	) 
3-Nitroaniline	086	Ω	970	n	1 096	, -	080	) <u> </u>	080	o	066	) ;	1,000	<b>)</b> ;
4-Nitroaniline	086	Ω	970	n	1 096	, -	080	) <u> </u>	080	o =	066	) :	1,000	);
Nitrobenzene	390	Ω	390	Ŋ	3 068	· -	390	> <b>=</b>	300	ב כ	980	) <b>:</b>	1,000	o ;
	390	Ω	390	Ω	390 L	<u> </u>	390	) 	390	o =	100	) <u> </u>	004	)
	086	N S	970	n	1 096	J	086	n	980	) <u> </u>	060	) <u>=</u>	1	)
ine	390	U	390	Ω	390 r	l	390		390	) =		) <u> </u>	1,000	<b>&gt;</b> :
N-Nitrosodiphenylamine	390	Ω	390	Ω	390 C	1	390	î	390	) <u> </u>	000	o =	400	) <u>;</u>
Pentachlorophenol	086	N	970	U	n 096	1	086	. n	086	) )	066	o =	400	) <u> </u>
hrene	390	Ω	390	U	390 U	_	390	Ω	390	n	400	> =	400	) <b>=</b>
	390	Ω	390	U	390 U	1	390	Ω	390	U	400	) <u> </u>	400	) <u>=</u>
	390	Ω	390	Ω	390 U	1	390	U	390	U	400	· =	90	) <u> </u>
ဍ	390	N	390	Ω	390 U	<b>-</b>	390	Ŋ	390	, <u>(</u> 1	400	> =	8	o
	086	n	026	Ω	Ω 096	_	086	Ω	086	n	066	) =	000	o =
2,4,6-Trichlorophenol	390	U	390	n	390 U	<b>J</b> ear-	390	U	390	U	400	n	400	> <b>=</b>

TABLE C.14c (Cont.)

	1											
Compound	(0-2 ft)	£ £	(2-4 ft)	.f.)	HBORI (4-6 ft)	£.	HB (6-1)	HBOR1 (6-8 ft)	HBOR1 (8-10 ft)	)R1	HB (0-;	HBOR2 (0-2 ft)
Acenaphthene	430	Ω	400	Ω	400	Ŋ	390	Ω	380	n	390	n
Acenaphthylene	430	D	400	Ω	400	Ω	390	Ω	380	Ω	390	U
Anthracene	430	D	400	Ω	400	Ω	390	Ω	380	n	390	U
Benz[a]anthracene	310	Ŀ	400	Ω	400	Ω	390	Ω	380	D	390	U
Benzo[a]pyrene	400	<u>-</u>	400	n	400	Ω	390	Ω	380	U	390	U
Benzo[b]fluoranthene	099		400	Ω	400	D	390	Ω	380	Ω	20	<b>-</b>
Benzo[g,h,i]perylene	110	ь.	400	Ω	400	D	390	n	380	Ω	390	D
Benzo[k]fluoranthene	120	ŗ	400	ņ	400	n	390	Ω	380	Ω	390	n
Bis(2-chloroethoxy)methane	430	Ω	400	Ω	400	Þ	390	D	380	n	390	$\supset$
Bis(2-chloroethyl)ether	430	Ω	400	Ω	400	D	390	n	380	Ω	390	$\Box$
Bis(2-chloroisopropyl)ether	430	Ω	400	Ω	400	D	390	Ω	380	n	390	
Bis(2-ethylhexyl)phthalate	430	D	400	Ω	42	J	950		42	<u>-</u>	39	<b>—</b>
4-Bromophenyl phenyl ether	430	Ω	400	Ω	400	n	390	n	380	n	390	n
Butylbenzylphthalate	430	Ω	49	-	46	L	54	~	46	_	390	n
Carbazole	430	Ω	400	U	400	n	390	Ω	380	n	390	n
4-Chloro-3-methylphenol	430	Ω	400	n	400	Ω	390	Ω	380	Ω	390	Ω
4-Chloroaniline	430	Ω	400	Ω	400	Ω	390	n	380	n	390	n
2-Chloronaphthalene	430	Ω	400	Ω	400	Ŋ	390	n	380	Ω	390	$\Omega$
2-Chlorophenol	430	n	400	Ω	400	D	390	Ω	380	D	390	$\Box$
4-Chlorophenyl phenyl ether	430	Ω	400	Ω	400	D	390	D	380	D	390	$\Box$
Chrysene	390	_	400	n	400	D	390	Ω	380	n	390	$\supset$
Di-n-butylphthalate	290	JB	260	JB	270	JB	210	JB	210	JB	210	JB
Di-n-octylphthalate	430	Ω	400	Ω	400	⊃	190	_	380	n	390	$\supset$
Dibenz[a,h]anthracene	55	~	400	D	400	D	390	n	380	D	390	-

TABLE C.14c (Cont.)

	HBOR1	R1	HB	HBOR1	HBOR1	)R1	H	HBOR 1	H	HRORI		LaOah
Compound	(0-2 ft)	(t)	(2-7	(2-4 ft)	(4-6 ft)	ft)	9)	(6-8 ft)	8)	(8-10 ft)	9	(0-2 ft)
Dibenzofuran	430	Ω	400	n	400	Ω	390		380	=	300	1
,2-Dichlorobenzene	430	Ŋ	400	Ω	400		360	) <u> </u>	380	) =	066	) ;
,3-Dichlorobenzene	430	n	400	Ω	400	- 1	300	> =	200	) <u>;</u>	086	);
,4-Dichlorobenzene	430	n	400	D	400	) =	300	) <u> </u>	200	) <u>;</u>	390	) ;
3,3'-Dichlorobenzidine	430	Ω	400	D	400	'n	390	) <u> </u>	380	) <u> </u>	300	) <u>;</u>
2,4-Dichlorophenol	430	Ω	400	Ω	400	n	390	Þ	380	) <u>=</u>	360	) <u> </u>
Diethylphthalate	430	Ω	400	Ω	400	U	390	n	380	î Î	360	) =
2,4-Dimethylphenol	430	n	400	Ω	400	n	390	n	380		340	) =
Dimethylphthalate	430	Ω	400	Ω	400	Ω	390	Ω	380	n D	390	ם כ
4,6-Dinitro-2-methylphenol	1,100	Ω	066	D	1,000	U	096	n	950	Ω	086	) =
2,4-Dinitrophenol	1,100	Ω	066	Ω	1,000	n	096	Ω	950		980	) <u> </u>
2,4-Dinitrotoluene	430	Ω	400	Þ	400	Ŋ	390	D	380	· =	300	) =
2,6-Dinitrotoluene	430	Ω	400	Ω	400	Ω	390	n	380	> =	300	) =
Fluoranthene	260	,	400	D	400	n	390	n	380	î D	63	) <u>-</u>
Fluorene	430	Ω	400	Ω	400	Ω	390	Ω	380	'n	390	, =
Hexachlorobenzene	430	n	400	n	400	Ω	390	n	380	Ω	390	· =
Hexachlorobutadiene	430	Ω	58	ī	400	Ω	390	D	380	) <u> </u>	300	) <u> </u>
Hexachlorocyclopentadiene	430	Ω	400	n	400	Ω	390	'n	380	· =	300	) <u>=</u>
Hexachloroethane	430	n	400	Ω	400	Þ	390	D	380	) <u> </u>	300	) <u>=</u>
Indeno[1,2,3-c,d]pyrene	120	Ţ	400	Ω	400	Ω	390	Ω	380	) <u> </u>	390	) <u> </u>
Isophorone	430	n	400	Ω	400	Ω	390	n	380	1	360	) <u> </u>
2-Methylnaphthalene	430	n	400	Ω	400	Ω	390	Ω	380	· 🗅	360	) <u>=</u>
2-Methylphenol	430	Ω	400	Ω	400	n	390	n	380	Ω	390	) <u>=</u>
4-Methylphenol	000											

TABLE C.14c (Cont.)

			Soil	Concer	Soil Concentrations (µg/kg) at Various Depth Intervals	ıg/kg)	at Vario	us Deg	oth Inter	vals		
Compound	HBOR1	ft)	HBOR1 (2-4 ft)	OR1 (ft)	HBOR1	R1	HB(6-8	HBOR 1 (6-8 ft)	HBOR1 (8-10 ft)	OR1	HB (0-	HBOR2 (0-2 ft)
Naphthalene	430	n	400	n	400	Ŋ	390	n	380	n	390	n
2-Nitroaniline	1,100	Ω	066	Ω	1,000	U	096	n	950	n	086	Ω
3-Nitroaniline	1,100	Ω	066	Ω	1,000	U	096	D	950	Ω	086	n
4-Nitroaniline	1,100	n	066	Ω	1,000	n	096	D	950	Ω	086	Ω
Nitrobenzene	430	D	400	n	400	Ω	390	n	380	n	390	Ω
2-Nitrophenol	430	n	400	Ω	400	n	390	D	380	Ω	390	Ω
4-Nitrophenol	1,100	n	066	Ω	1,000	n	096	n	950	Ω	086	Ω
N-Nitroso-di-n-propylamine	430	D	400	Ω	400	Ω	390	n	380	n	390	n
N-Nitrosodiphenylamine	430	D	400	Ω	400	D	390	Þ	380	Ω	390	Ω
Pentachlorophenol	1,100	D	066	n	1,000	Ω	096	n	950	n	086	Ω
Phenanthrene	73	<b>-</b>	400	D	400	Ω	390	Ω	380	Ω	390	n
Phenol	430	n	400	n	400	Ω	390	Ω	380	Ω	390	U
Pyrene	200	_	400	Ω	400	Ω	390	n	380	Ω	55	Ţ
1,2,4-Trichlorobenzene	430	n	400	n	400	N	390	Ω	380	Ω	390	D
2,4,5-Trichlorophenol	1,100	U	066	n	1,000	Ω	096	D	950	Ω	086	Ω
2,4,6-Trichlorophenol	430	ח	400	n	400	D	390	D	380	D	390	n
								 	! ! !	     		 

TABLE C.14c (Cont.)

	S	oil Conc	entratio	ns (µg/l	(g) at Var	ious De	Soil Concentrations (µg/kg) at Various Depth Intervals	'als	
i	HE	HBOR2	HB	HBOR2	HBOR2	)R2	HBOR2	)R2	
Compound	(2)	(2-4 ft)	4)	(4-6 ft)	(6-8 ft)	ft)	(8-10 ft)	(tr)	
Acenaphthene	390	Ω	390	Ω	400	Ω	420	n	
Acenaphthylene	390	n	390	Ω	400	n	420	n	
Anthracene	390	Ω	390	Ω	400	n	420	'n	
Benz[a]anthracene	390	U	390	Ω	400	n	420	n	
Benzo[a]pyrene	390	*n	390	Ω	400	n	420	Ω	
Benzo[b]fluoranthene	390	*n	390	Ω	400	Ω	420	D	
Benzo[g,h,i]perylene	390	*	390	Ω	400	Ω	420	D	
Benzo[k]fluoranthene	390	*	390	Ω	400	n	420	Ω	
Bis(2-chloroethoxy)methane	390	Ω	390	Ω	400	D	420	U	
Bis(2-chloroethyl)ether	390	Ω	390	Ω	400	n	420	n	
Bis(2-chloroisopropyl)ether	390	ח	390	D	400	Ω	420	n	
Bis(2-ethylhexyl)phthalate	200	-	41	J	400	Ω	420	Ω	
4-Bromophenyl phenyl ether	390	n	390	D	400	Ω	420	Ω	
Butylbenzylphthalate	390	Ω	390	Ω	400	Ω	420	Ω	
Carbazole	390	n	390	n	400	Ω	420	n	
4-Chloro-3-methylphenol	390	Ω	390	Ω	400	D	420	Ω	
4-Chloroaniline	390	Ω	390	D	400	Ω	420	Ω	
2-Chloronaphthalene	390	Ω	390	n	400	Ω	420	Ω	
2-Chlorophenol	390	Ω	390	n	400	n	420	Ω	
4-Chlorophenyl phenyl ether	390	n	390	Ω	400	Ω	420	D	
Chrysene	390	Ω	390	Ω	400	U	420	Ω	
Di-n-butylphthalate	270	JB	190	JB	180	JB	190	Ţ	
Di-n-octylphthalate	390	*	390	n	400	Ω	420	n	
Dibenz[a,h]anthracene	390	*	390	Ω	400	Ω	420	Ω	

TABLE C.14c (Cont.)

	Sc	oil Conc	entration	ıs (µg/k	Soil Concentrations (µg/kg) at Various Depth Intervals	ous De	pth Interv	als
Compound	HB (2-	HBOR2 (2-4 ft)	HB (4-(	HBOR2 (4-6 ft)	HBOR2 (6-8 ft)	R2 ft)	HBOR2 (8-10 ft)	R2 ft)
Dibenzofiran	390	11	390	1	007	1	007	1
1.2-Dichlorobenzene	390	) =	360	) <u> </u>	400	) <u>=</u>	027	) <u> </u>
1,3-Dichlorobenzene	390	n	390	) D	400	> >	420	ם כ
1,4-Dichlorobenzene	390	Ω	390	Ω	400	Ω	420	n
3,3'-Dichlorobenzidine	390	n	390	Ω	400	n	420	n
2,4-Dichlorophenol	390	Ω	390	Ω	400	Ω	420	Ω
Diethylphthalate	390	n	390	Ω	400	Ω	420	Ω
2,4-Dimethylphenol	390	Ŋ	390	Ω	400	Ω	420	D
Dimethylphthalate	390	Ω	390	n	400	Ω	420	Ω
4,6-Dinitro-2-methylphenol	970	Ω	096	D	1,000	Ω	1,000	Ω
2,4-Dinitrophenol	970	Ω	096	n	1,000	Ω	1,000	Ω
2,4-Dinitrotoluene	390	Ω	390	D	400	D	420	Ω
2,6-Dinitrotoluene	390	Ω	390	n	400	n	420	Ω
Fluoranthene	390	Ω	390	Ω	400	Ω	420	U
Fluorene	390	Ω	390	Ω	400	D	420	n
Hexachlorobenzene	390	Ŋ	390	Ω	400	Ω	420	Ω
Hexachlorobutadiene	390	U	390	D	400	Ω	420	Ω
Hexachlorocyclopentadiene	390	n	390	Ω	400	D	420	Ω
Hexachloroethane	390	Ω	390	U	400	Ω	420	Ω
Indeno[1,2,3-c,d]pyrene	390	*1	390	Ŋ	400	D	420	Ω
Isophorone	390	Ω	390	n	400	n	420	D
2-Methylnaphthalene	390	Ω	390	n	400	Ω	420	D
2-Methylphenol	390	n	390	n	400	D	420	n
4-Methylphenol	390	Ω	390	D	400	D	420	n

TABLE C.14c (Cont.)

Compound	HB (2-	HBOR2 (2-4 ft)	HB (4-(	HBOR2 (4-6 ft)	HBOR2 (6-8 ft)	)R2 ft)	HBOR2 (8-10 ft)	.R2
Naphthalene	390	=	300	=	9	1		;
2-Nitroaniline	070	) <u> </u>	060	) <u> </u>	400	)	420	⊃ ;
3_Nitrospiline	010	) ;		<b>)</b> ;	1,000	<b>)</b>	1,000	)
	0/6	<b></b>	960	D	1,000	ב	1,000	Þ
4-Nitroaniline	970	n	096	n	1,000	Ω	1,000	D
Nitrobenzene	390	n	390	Ω	400	Ω	420	D
2-Nitrophenol	390	n	390	n	400	Ω	420	1
4-Nitrophenol	970	Ω	096	n	1,000	Þ	1.000	
N-Nitroso-di-n-propylamine	390	Ω	390	Ω	400	D	420	n
N-Nitrosodiphenylamine	390	n	390	Ω	400	Ŋ	420	n
Pentachlorophenol	970	Ω	096	Ω	1,000	Ω	1,000	Þ
Phenanthrene	390	D	390	Ω	400	Ω	420	D
Phenol	390	Ŋ	390	n	400	D	420	n
Pyrene	75	_	390	Ω	400	n	420	Ω
1,2,4-Trichlorobenzene	390	D	390	Ω	400	n	420	
2,4,5-Trichlorophenol	026	Ω	096	Ω	1,000	n	1,000	Ω
2,4,6-Trichlorophenol	390	n	390	Ω	400	Ω	420	D

TABLE C.14c (Cont.)

410 U 410 U 410 U 410  410 U 410 U 410 U 410  410 U* 410 U 410  410 U* 410 U 410  410 U* 410 U 410  5th 410 U* 410 U 410  alate 210 U 410 U 410  alate 210 J 100 J 93  dether 410 U 410 U 410  alate 210 J 100 J 93  dether 410 U 410 U 410  alot 410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410	Compound	TBNPBOR1 (0-2 ft)	OR1	TBNPBOR1 (2-4 ft)	3OR1	TBNPBOR1 (4-6 ft)	30R1 ft)	TBNPBOR1 (6-8 ft)	30R1 ft)	TBNPBOR1 (8-10 ft)	30R1
410       U       410       U       410         410       U       410       U       410         410       U*       410       U       410         410       U*       410       U       410         410       U*       410       U       410         410       U	Acenaphthene	410	Ω	410	Ω	410	n	410	Ŋ	400	Ω
410 U 410 U 410 U 410  410 U* 410 U 410  410 U* 410 U 410  410 U* 410 U 410  410 U* 410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410	Acenaphthylene	410	D	410	n	410	Ω	410	n	400	D
410 U* 410 U 410  410 U* 410 U 410  410 U* 410 U 410  410 U* 410 U 410  410 U* 410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410	Anthracene	410	Ω	410	Ω	410	U	410	n	400	D
410 U* 410 U 410 410 U* 410 U 410 410 U* 410 U 410 410 U* 410 U 410 410 U 410 U 410 210 J 100 J 93 7 410 U 410 U 410	Benz[a]anthracene	410	Ω	410	Ω	410	U	410	n	400	n
410 U* 410 U 410  410 U* 410 U 410  410 U* 410 U 410  410 U 410 U 410  210 J 100 J 93  7 410 U 410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410  410 U 410	Benzo[a]pyrene	410	*n	410	n	410	n	410	D	400	Ω
410 U* 410 U 410 U 410 U 410 410 U 410 U 410 U 410 410 U 410 U 410 U 410 210 J 100 J 93 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U 410 U 410 410 U* 410 U 410	Benzo[b]fluoranthene	410	*n	410	D	410	n	410	Ω	400	U
410       U*       410       U       410         410       U       410       U       410         410       U       410       U       410         210       J       100       J       93         410       U       410       U       410         410       U <td< td=""><td>Benzo[g,h,i]perylene</td><td>410</td><td>*n</td><td>410</td><td>n</td><td>410</td><td>Ω</td><td>410</td><td>D</td><td>400</td><td>Ω</td></td<>	Benzo[g,h,i]perylene	410	*n	410	n	410	Ω	410	D	400	Ω
410       U       410       U       410         410       U       410       U       410         210       J       100       J       93         410       U       410       U       410         410       U	Benzo[k]fluoranthene	410	*	410	n	410	Ω	410	D	400	n
410       U       410       U       410         410       U       410       U       410         210       J       100       J       93         410       U       410       U       410	Bis(2-chloroethoxy)methane	410	D	410	n	410	n	410	n	400	Þ
410       U       410       U       410         210       J       100       J       93         410       U       410       U       410	Bis(2-chloroethyl)ether	410	Ω	410	Ω	410	D	410	n	400	n
210 J 100 J 93  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410  410 U 410 U 410	Bis(2-chloroisopropy1)ether	410	Ω	410	Ω	410	D	410	D	400	n
410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U*       410       U       410         410       U*       410       U       410         410       U*       410       U       410	Bis(2-ethylhexyl)phthalate	210	Ĺ.	100	7	93	⊬	45	,	160	J
410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U*       410       U       410         410       U*       410       U       410         410       U*       410       U       410	4-Bromophenyl phenyl ether	410	Ω	410	n	410	Ω	410	n	400	D
410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U*       410       U       410	Butylbenzylphthalate	410	Ω	410	Ω	410	Ω	410	D	400	Ω
410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U*       410       U       410         410       U*       410       U       410	Carbazole	410	D	410	Ω	410	Ω	410	Ω	400	n
410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         410       U*       410       U       410         410       U*       410       U       410	4-Chloro-3-methylphenol	410	Ω	410	Ω	410	n	410	D	400	Ω
410       U       410       U       410         410       U       410       U       410         410       U       410       U       410         310       JB       290       JB       190         410       U*       410       U       410	4-Chloroaniline	410	Ω	410	Ω	410	Ω	410	Þ	400	D
410 U 410 U 410 410 U 410 U 410 310 JB 290 JB 190 410 U* 410 U 410	2-Chloronaphthalene	410	D	410	n	410	D	410	Ŋ	400	
410 U 410 U 410 410 U 410 U 410 310 JB 290 JB 190 410 U* 410 U 410	2-Chlorophenol	410	D	410	Ω	410	Ω	410	D	400	Þ
410 U 410 U 410 310 JB 290 JB 190 410 U* 410 U 410 410 U* 410	4-Chlorophenyl phenyl ether	410	D	410	Ω	410	Ω	410	Ω	400	D
310 JB 290 JB 190 410 U* 410 11* 410 11* 410 410	Chrysene	410	D	410	Ω	410	Ω	410	D	400	D
410 U* 410 U 410 410 U* 410 H	Di-n-butylphthalate	310	JB	290	JB	190	JB	200	JB	210	JB
410 11* 410 11 410	Di-n-octylphthalate	410	*	410	Ω	410	Ω	410	Ω	400	Ω
	Dibenz[a,h]anthracene	410	*	410	n	410	Ω	410	n	400	n

TABLE C.14c (Cont.)

		Š	oil Conce	ntration	s (µg/kg)	at Vari	Soil Concentrations (µg/kg) at Various Depth Intervals	Interva	ıls	
Compound	TBNPBOR1 (0-2 ft)	BOR1 ft)	TBNPBOR1 (2-4 ft)	SOR1 ft)	TBNPBOR1 (4-6 ft)	OR1 ft)	TBNPBOR1 (6-8 ft)	80R1 ft)	TBNPBOR1 (8-10 ft)	OR1 ft)
Dibenzofuran	410	n	410	=	410	=	710	=	1 8	
1,2-Dichlorobenzene	410	n	410	'n	410	) <u> </u>	410	> <u>=</u>	400	o
1,3-Dichlorobenzene	410	Ω	410	D	410	n	410	) <u> </u>	400	> =
1,4-Dichlorobenzene	410	n	410	Ω	410	n	410	n	400	a D
3,3'-Dichlorobenzidine	410	Ω	410	Ω	410	n	410	Ω	400	o D
2,4-Dichlorophenol	410	Ω	410	Ω	410	Ω	410	D	400	n
Diethylphthalate	410	Ω	410	Ω	410	n	410	Ω	400	Ω
2,4-Dimethylphenol	410	Ω	410	Ω	410	n	410	n	400	n
Dimethylphthalate	410	Ω	410	D	410	n	410	Ω	400	) D
4,6-Dinitro-2-methylphenol	1,000	Ω	1,000	D	1,000	n	1,000	Ω	1,000	n
2,4-Dinitrophenol	1,000	Ω	1,000	Ω	1,000	n	1,000	n	1,000	n
2,4-Dinitrotoluene	410	Ω	410	Ω	410	U	410	Ω	400	'n
2,6-Dinitrotoluene	410	Ω	410	n	410	n	410	Ω	400	ם ה
Fluoranthene	410	Ω	410	n	410	Ω	410	Ω	400	n
Fluorene	410	Ω	410	Ω	410	Ω	410	n	400	Ω
Hexachlorobenzene	410	Ω	920		180	ſ	97	F.	400	'n
Hexachlorobutadiene	410	n	410	Ω	410	Ω	410	Ω	400	n
Hexachlorocyclopentadiene	410	Ω	410	Ω	410	D	410	n	400	n
Hexachloroethane	410	Ω	410	Ω	410	Ω	410	U	400	n
Indeno[1,2,3-c,d]pyrene	410	*n	410	Ω	410	Ω	410	n	400	Ω
Isophorone	410	Ω	410	Ω	410	Ω	410	Ω	400	Ω
2-Methylnaphthalene	230	<u>-</u>	42	7	410	Ω	410	Ω	400	Ω

TABLE C.14c (Cont.)

Compound	TBNPBOR1 (0-2 ft)	OR1	TBNPBORI (2-4 ft)	SOR1	TBNPBORI (4-6 ft)	30R1 ft)	TBNPBORI (6-8 ft)	OR1	TBNPBOR1 (8-10 ft)	30R1
2-Methylphenol	410	Ω	410	n	410	U	410	Ω	400	Ω
4-Methylphenol	410	n	410	Ŋ	410	Ω	410	D	400	n
Naphthalene	68	ŗ	42	_	410	Ω	410	Ω	400	$\Gamma$
2-Nitroaniline	1,000	Ω	1,000	Ω	1,000	n	1,000	Ω	1,000	U
3-Nitroaniline	1,000	Ω	1,000	n	1,000	Ω	1,000	n	1,000	D
4-Nitroaniline	1,000	Ω	1,000	n	1,000	Ω	1,000	Ω	1,000	n
Nitrobenzene	55	_	410	Ω	410	Ω	410	n	400	n
2-Nitrophenol	410	Ω	410	D	410	Ω	410	D	400	n
4-Nitrophenol	1,000	Ω	1,000	Ω	1,000	Ω	1,000	Ω	1,000	n
N-Nitroso-di-n-propylamine	410	D	410	n	410	n	410	n	400	D
N-Nitrosodiphenylamine	410	Ω	160	Ţ	410	Ŋ	410	Ω	400	n
Pentachlorophenol	1,000	Ω	1,000	Ω	1,000	Ω	1,000	Ω	1,000	D
Phenanthrene	250	ĭ	410	Ω	410	n	410	Ω	400	n
Phenol	410	Ω	410	D	410	n	410	D	400	n
Pyrene	79	<b>-</b>	410	n	410	n	410	D	400	Ω
1,2,4-Trichlorobenzene	410	Ω	410	Ω	410	n	410	n	400	Ω
2,4,5-Trichlorophenol	1,000	Ω	1,000	n	1,000	Ω	1,000	n	1,000	Ω
2,4,6-Trichlorophenol	43		410	Ω	410		410	11	700	-

See next page for footnotes.

## TABLE C.14c (Cont.)

<sup>a</sup> Sampled by ANL, analyzed by Weston Gulf Coast or ANL/ACL, CLP/HSL semivolatile organics.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

B = analyte was found in the associated blank.

N = spike and spike duplicate recovery or percent RPD were outside the control limits.

\* = internal standard area counts were outside QC limits.

TABLE C.14d Semivolatile Organics Analyses for Soil Borings, May 1995<sup>a</sup>

			Soil	Concer	itrations (	µg/kg)	Soil Concentrations (µg/kg) at Various Depth Intervals	Depth	Intervals			
Compound	VXBORI	OR1	VXBORI	9R1	VXBOR1	OR1	VXBORI	)R1	VXBOR2	)R2	VXBOR2	OR2
	7-0)	OIT.	+-7)	61	0-0)	(31	(31.01-8)		(11 7-0)		(2-4 It)	<b>E</b>
Acenaphthene	380	Ŋ	410	Ω	460	Ŋ	390	n	410	n	380	n
Acenaphthylene	380	n	410	U	460	Ω	390	D	410	U	380	n
Anthracene	380	Ω	410	n	460	Ω	1,400		410	D	380	n
Benz[a]anthracene	380	Ω	410	Ω	460	Ω	390	D	410	n	380	Ω
Benzo[a]pyrene	380	Ω	410	n	460	Ω	390	n	410	Ω	380	Ω
Benzo[b]fluoranthene	380	D	410	n	460	U	390	Ω	410	n	380	n
Benzo[g,h,i]perylene	380	Ω	410	n	460	Ω	390	Ω	410	n	380	Ω
Benzo[k]fluoranthene	380	Ω	410	Ω	460	Ω	390	Ω	410	ח	380	Ω
Bis(2-chloroethoxy)methane	380	n	410	n	460	Ω	390	Ω	410	Ω	380	D
Bis(2-chloroethyl)ether	380	Ω	410	U	460	n	390	Ω	410	D	380	Ω
Bis(2-ethylhexyl)phthalate	1,600	В	1,400	В	2,500	В	1,100	В	1,300	В	096	В
4-Bromophenyl phenyl ether	380	Ω	410	n	460	n	390	Ω	410	Ω	380	n
Butylbenzylphthalate	750	В	260	JB	460	D	390	U	410	U	380	Ω
Carbazole	380	n	410	n	460	Ω	390	Ω	410	D	380	Ω
4-Chloro-3-methylphenol	380	n	410	Ω	460	Ω	390	Ω	410	U	380	Ω
4-Chloroaniline	380	D	410	n	460	n	390	U	410	Ω	380	Ω
2-Chloronaphthalene	380	n	410	Ω	460	n	390	Ω	410	n	380	Ω
2-Chlorophenol	380	D	410	n	460	Ω	390	n	410	n	380	n
4-Chlorophenyl phenyl ether	380	D	410	n	460	Ω	390	Ω	410	n	380	Ω
Chrysene	380	D	410	D	460	D	390	Ω	410	n	380	D
Di-n-butylphthalate	200	JB	1,200	В	350	JB	460	В	160	JB	380	ב
Di-n-octylphthalate	380	ח	410	Ω	460	Ω	390	Ŋ	410	U	380	Ω
Dibenz[a,h]anthracene	380	Ω	410	Ω	460	D	390	Ω	410	n	380	Ω
Dibenzofuran	380	Ω	410	Ω	460	n	390	n	410	n	380	Ω

TABLE C.14d (Cont.)

$ \begin{array}{c c c c c c c c c c c c c c c c c c c $				Soil	Conce	ntrations (	(µg/kg)	at Various	Depth	Soil Concentrations (µg/kg) at Various Depth Intervals			
the 380 U 410 U 460 U 390 U 410 U 380  ine 380 U 410 U 460 U 390 U 410 U 380  ine 380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  iphenol 950 U 1,000 U 1,200 U 990 U 1,000 U 380  380 U 410 U 460 U 390 U 1,000 U 380  isolatione 380 U 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U 380  isolatione 410 U 460 U 390 U 410 U	Compound	VXB(	OR1	VXB(	OR1 ft)	VXB (6-8	OR1 ft)	VXB(	OR1 (ft)	VXB(0-2	OR2 ft)	VXB (2-4	OR2 ft)
Fig. 380 U 410 U 460 U 390 U 410 U 380 U 380 U 410 U 380 U 390 U 410 U 380 U 380 U 410 U 460 U 390 U 410 U 380 U 380 U 410 U 460 U 390 U 410 U 380 U 380 U 410 U 460 U 390 U 410 U 380 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 380 U 480 U 390 U 410 U 380 U 480 U 390 U 410 U 380 U 480 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 390 U 410 U 380 U 410 U 440 U 440 U 380 U 410 U 440 U 380 U 410 U 440 U 380 U 410 U 440 U 440 U 440 U 440 U 4	1,2-Dichlorobenzene	380	D	410	Ŋ	460	n	390	U	410	=	380	=
ine 380 U 410 U 460 U 390 U 410 U 380 U 410 U 380 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 1,200 U 990 U 1,000 U 950 U 1,000 U 1,200 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 380 U 410 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U 440 U	1,3-Dichlorobenzene	380	n	410	Ω	460	n	390	· 🗅	410	) <u>=</u>	380	) =
ine 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 Jehenol 950 U 1,000 U 1,200 U 990 U 1,000 U 950 380 U 1,000 U 1,200 U 390 U 1,000 U 950 380 U 410 U 460 U 390 U 410 U 380	1,4-Dichlorobenzene	380	Ω	410	D	460	Ω	390	n	410	) <u>=</u>	380	) <u> </u>
380 U 410 U 460 U 390 U 410 U 380  180 U 410 U 460 U 390 U 410 U 380  180 U 410 U 460 U 390 U 410 U 380  180 U 410 U 460 U 390 U 410 U 380  180 U 1,000 U 1,200 U 990 U 1,000 U 950  380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  a 380 U 410 U 440 U 460 U 390 U 410 U 380  a 380 U 410 U 440 U 440 U 380  a 380 U 410 U 440 U 440 U 380  a 380 U 410 U 440 U 440 U 380  a 380 U 410 U 440 U 440 U 380  a 380 U 410 U 440 U 440 U 380  a 380 U 410 U 440 U 440 U 380  a 380 U 410 U 440 U 440 U 380  a 440 U 440 U 440 U 440 U 440 U 380  a 440 U 44	3,3'-Dichlorobenzidine	380	Ω	410	Ω	460	Ω	390	) D	410	) <u> </u>	380	) <u>=</u>
380 U 410 U 460 U 390 U 410 U 380  Jacob U 410 U 460 U 390 U 410 U 380  Jacob U 1,000 U 1,200 U 990 U 1,000 U 950  380 U 410 U 460 U 390 U 1,000 U 950  380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  atdiene 410 U 460 U 390 U 410 U 380  atdiene 410 U 460 U 390 U 410 U 380  atdiene 410 U 460 U 390 U 410 U 380  atdiene 410 U 460 U 390 U 410 U 380  atdiene 410 U 460 U 390 U 410 U 390  atdiene 410 U 460 U 390 U 410 U 390  atdiene 410 U 460 U 390 U 410 U 390  atdiene 410 U 460 U 390 U 390 U 300  atdiene 410 U 460 U 390 U 390 U 300  atdiene 410 U 460 U 390 U 390	2,4-Dichlorophenol	380	Ω	410	n	460	Ω	390	U	410	î	380	) <u>=</u>
380   U   410   U   460   U   390   U   410   U   380     380   U   410   U   460   U   390   U   410   U   380     410   U   1,200   U   1,200   U   1,000   U   1,000   U   1,000     380   U   1,000   U   1,200   U   390   U   1,000   U   380     380   U   410   U   460   U   390   U   410   U   380     380   U   410   U   460   U   390   U   410   U   380     410   U   460   U   390   U   410   U   380     410   U   460   U   390   U   410   U   380     410   U   460   U   390   U   410   U   380     410   U   460   U   390   U   410   U   380     410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   460   U   390   U   410   U   380     510   U   410   U   440	Diethylphthalate	380	n	410	ח	460	Ω	390	n	410	n	380	) =
380 U 410 U 460 U 390 U 1,000 U 950 U 900	2,4-Dimethylphenol	380	Ω	410	Ω	460	Ω	390	n	410	n	380	) D
lphenol 950 U 1,000 U 1,200 U 990 U 1,000 U 950 O 1,000 O 950 O 1,000 U 950 O 1,000 U 950 O 1,000 O 1,	Dimethylphthalate	380	Ω	410	Ω	460	Ω	390	D	410	Ω	380	) <u>=</u>
950 U 1,000 U 1,200 U 990 U 1,000 U 950 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 2 380 U 410 U 460 U 390 U 410 U 380 4410 U 460 U 390 U 410 U 380 4410 U 460 U 390 U 410 U 380 6ne 6ne 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380 780 U 410 U 460 U 390 U 410 U 380	4,6-Dinitro-2-methylphenol	950	Ω	1,000	D	1,200	Ω	066	Ω	1,000	n	950	) <u></u>
380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 tadiene 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 ene 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380 380 U 410 U 460 U 390 U 410 U 380	2,4-Dinitrophenol	950	Ω	1,000	n	1,200	Ω	066	Ω	1,000	D	950	) <u> </u>
sne         380         U         410         U         460         U         390         U         410         U         390         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         41	2,4-Dinitrotoluene	380	Ω	410	n	460	Ω	390	n	410	n	380	) <u> </u>
380 U 410 U 660 U 390 U 410 U 380  sene 380 U 410 U 460 U 390 U 410 U 380  diene 380 U 410 U 460 U 390 U 410 U 380  opentadiene 380 U 410 U 460 U 390 U 410 U 380  llpyrene 380 U 410 U 460 U 390 U 410 U 380  alene 380 U 410 U 460 U 390 U 410 U 380  alene 380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380  380 U 410 U 460 U 390 U 410 U 380	2,6-Dinitrotoluene	380	Ω	410	Ω	460	Ω	390	n	410	Ω	380	) D
sene         380         U         410         U         980         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         380         U         380         U         380         U         380         U         380         U         3	Fluoranthene	380	Ω	410	Ω	460	n	390	n	410	n	380	'n
zene         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         380         U           alene         380         U         410         U         460         U         390         U         410         U         380         U <td< td=""><td>Fluorene</td><td>380</td><td>Ω</td><td>410</td><td>Ω</td><td>086</td><td></td><td>390</td><td>Ω</td><td>410</td><td>Ω</td><td>380</td><td>'n</td></td<>	Fluorene	380	Ω	410	Ω	086		390	Ω	410	Ω	380	'n
diene         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U           alene         380         U         410         U         460         U         390         U         410         U         380         U           380         U         410         U         460         U         390         U         410         U         380         U           380         U         410         U         460         U         390         U         410         U         380         U	Hexachlorobenzene	380	Ω	410	D	460	n	390	Ω	410	n	380	n
opentadiene         380         U         410         U         460         U         390         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U         410         U         380         U         410         U         460         U         390         U         410         U         380         U           alene         380         U         410         U         460         U         390         U         410         U         380         U           380         U         410         U         460         U         390         U         410         U         380         U           380         U         410         U         460         U         390         U         410         U         380         U	Hexachlorobutadiene	380	Ω	410	n	460	Ω	390	U	410	n	380	'n
llpyrene 380 U 410 U 460 U 390 U 410 U 380 U	Hexachlorocyclopentadiene	380	D	410	n	460	Ω	390	D	410	D	380	Ω
Jlpyrene       380       U       410       U       460       U       390       U       410       U       380       J         alene       380       U       410       U       460       U       390       U       410       U       380       U         380       U       410       U       460       U       390       U       410       U       380       U         380       U       410       U       460       U       390       U       410       U       380       U	Hexachloroethane	380	Ω	410	Ω	460	Ω	390	Ω	410	n	380	Ω
380 U 410 U 460 U 390 U 410 U 380 u 380 u 380 u 380 u 380 u 380 u 380 u 410 U 460 U 390 U 410 U 380 u 380 u 380 u 410 U 460 U 390 U 410 U 380 u	Indeno[1,2,3-c,d]pyrene	380	Ω	410	Ω	460	Ω	390	Ω	410	D	380	Ω
alene 380 U 410 U 460 U 390 U 410 U 380 U 410 U 460 U 390 U 410 U 460 U 390 U 410 U	Isophorone	380	Ω	410	Ω	460	Ω	390	D	410	n	380	Ω
380 U 410 U 460 U 390 U 410 U 380 U 410 U	2-Methylnaphthalene	380	n	410	n	460	Ω	390	Ω	410	Ω	380	Ω
380 U 410 U 460 U 390 U 410 U	2-Methylphenol	380	Ω	410	n	460	Ω	390	Ω	410	Ω	380	n
	4-Methyiphenol	380	Ω	410	n	460	Ω	390	n	410	n	380	Ŋ

TABLE C.14d (Cont.)

Compound	VXBOR (0-2 ft)	R 1	VXBOR (2-4 ft)	R1	VXBOR (6-8 ft)	OR1	VXBOR (8-10 ft)	OR1	VXBOR2 (0-2 ft)	OR2 ft)	VXB (2-4	VXBOR2 (2-4 ft)
Naphthalene	380	n	410	Ω	460	Ω	390	D	410	n	380	Þ
2-Nitroaniline	950	Ω	1,000	Ω	1,200	Ω	066	Ω	1,000	Ω	950	ņ
3-Nitroaniline	950	Ω	1,000	U	1,200	Ω	066	n	1,000	Ω	950	n
4-Nitroaniline	950	U	1,000	n	1,200	Ω	066	Ω	1,000	Ω	950	n
Nitrobenzene	380	n	410	Ω	460	n	390	n	410	Ω	380	n
2-Nitrophenol	380	Ω	410	n	460	Ω	390	n	410	Ω	380	D
4-Nitrophenol	950	Ω	1,000	n	1,200	D	066	D	1,000	Ω	950	
N-Nitroso-di-n-propylamine	380	Ω	410	Ω	460	Ω	390	n	410	n	380	Ω
N-Nitrosodiphenylamine	380	U	410	Ω	460	n	390	Ω	410	Ω	380	Ω
2,2'-Oxybis-(1-chloropropane)	380	U	410	n	460	Ω	390	n	410	n	380	Ω
Pentachlorophenol	950	n	1,000	D	1,200	D	066	Ω	1,000	D	950	$\Box$
Phenanthrene	380	Ω	3,500		2,000		1,200		410	Ω	380	n
Phenol	380	n	410	Ω	460	ח	390	Ω	410	Ω	380	n
Pyrene	380	n	280	j	460	n	160	_	410	D	380	D
1,2,4-Trichlorobenzene	380	D	410	n	460	n	390	Ω	410	U	380	D
2,4,5-Trichlorophenol	950	Ŋ	1,000	n	1,200	Ω	066	D	1,000	D	950	N
2,4,6-Trichlorophenol	380	n	410	Ω	460	Ω	390	n	410	n	380	n

TABLE C.14d (Cont.)

			Soil	Conce	ntrations	(µg/kg	Soil Concentrations (µg/kg) at Various Depth Intervals	ous De	pth Inter	vals		
7	VXBOR2	OR2	VXBOR2	)R2	VXBOR2	)R2	VXBOR2	)R2	FTBOR1	)R1	FTBO	FTBOR1 Dup
Compound	(4-6 ft)	£()	(6-8 ft)	æ	(8-10 ft)	£	(10-12 ft)	£	(0-2 ft)	ft)	-0)	(0-2 ft)
Acenaphthene	390	Ω	410	n	420	Ω	390	Ω	390	11	300	11
Acenaphthylene	390	Ω	410	Ω	420	n	390	n	390	) <u> </u>	300	) <u> </u>
Anthracene	390	D	410	Ω	420	n	390	D	390	) <u> </u>	300	) <u>=</u>
Benz[a]anthracene	390	Ω	410	Ω	420	Ω	390	'n	390	) <u> </u>	390	o =
Benzo[a]pyrene	390	Ω	410	n	420	Ω	390	Ω	390	n	390	) D
Benzo[b]fluoranthene	390	D	410	Ω	420	Ω	390	n	390	D	390	ם מ
Benzo[g,h,i]perylene	390	Ω	410	Ω	420	Ω	390	n	390	Ω	390	n
Benzo[k]fluoranthene	390	n	410	Ω	420	n	390	Ω	390	Ω	390	n 11
Bis(2-chloroethoxy)methane	390	n	410	D	420	Ω	390	n	390	n	390	· []
Bis(2-chloroethyl)ether	390	n	410	Ω	420	Ω	390	Ω	390	n	390	o D
Bis(2-ethylhexyl)phthalate	1,500		2,000		4,300	В	4,100	В	1,700	В	4,600	В
4-Bromophenyl phenyl ether	390	Ω	410	D	420	Ω	390	Ω	390	Ω	390	D
Butylbenzylphthalate	390	Ω	410	Ω	420	Ω	390	Ω	390	n	390	'n
Carbazole	390	n	410	Ω	420	n	390	Ω	390	n	390	, <u>1</u> 1
4-Chloro-3-methylphenol	390	Ω	410	Ω	420	D	390	Ω	390	Ω	390	n D
4-Chloroaniline	390	Ω	410	n	420	Ω	390	D	390	Ω	390	n
2-Chloronaphthalene	390	Ω	410	n	420	Ω	390	n	390	n	390	U
2-Chlorophenol	390	Ω	410	Ω	420	Ω	390	Ω	390	n	390	Ω
4-Chlorophenyl phenyl ether	390	n	410	n	420	Ω	390	Ω	390	Ω	390	Ω
Chrysene	390	D	410	Ω	420	Þ	390	Ω	390	Ω	390	Ω
Di-n-butylphthalate	140	JB	410	Ω	100	ŗ	210	L.	110	_	170	<b>,</b>
Di-n-octylphthalate	390	Ω	410	Ω	420	Ω	390	n	390	n	390	Ω
Dibenz[a,h]anthracene	390	Ω	410	Ω	420	n	390	Ω	390	Ω	390	n
Dibenzofuran	390	Ω	410	Ω	420	Ω	390	Ω	390	Ω	390	Ω

TABLE C.14d (Cont.)

Compound	VXBOR2 (4-6 ft)	)R2 ft)	VXBOR2 (6-8 ft)	R2	VXBOR2 (8-10 ft)	R2 ft)	VXBOR2 (10-12 ft)	)R2	FTBOR1 (0-2 ft)	R1 ft)	FTBO (0-	FTBOR1 Dup (0-2 ft)
1.2-Dichlorobenzene	390	<u>-</u>	710	_	420	1.1	300	11	Coc	;	Š	:
		)	1	)	420	)	390	)	090	5	390	>
1,3-Dichlorobenzene	390	ח	410	Ω	420	Ω	390	n	390	Ω	390	n
1,4-Dichlorobenzene	390	Ω	410	n	420	n	390	n	390	Ω	390	n
3,3'-Dichlorobenzidine	390	Ω	410	Ω	420	n	390	n	390	U	390	D
2,4-Dichlorophenol	390	D	410	n	420	Ω	390	Ω	390	Ω	390	D
Diethylphthalate	390	Ω	410	n	420	Ω	390	n	390	n	390	Ω
2,4-Dimethylphenol	390	D	410	Ω	420	Ω	390	Ω	390	n	390	Ω
Dimethylphthalate	390	n	410	Ω	420	n	390	Ω	390	n	390	n
4,6-Dinitro-2-methylphenol	066	Ω	1,000	n	1,100	U	086	n	086	n	066	n
2,4-Dinitrophenol	066	Ω	1,000	Ω	1,100	n	086	Ω	086	n	066	n
2,4-Dinitrotoluene	390	U	410	n	420	D	390	n	390	n	390	n
2,6-Dinitrotoluene	390	Ω	410	Ω	420	D	390	U	390	Ŋ	390	n
Fluoranthene	390	Ω	410	Ω	420	n	390	Ω	390	Ω	390	Ω
Fluorene	390	Ω	410	Ω	420	D	390	n	390	n	390	U
Hexachlorobenzene	390	Ω	410	D	420	n	390	n	390	Ω	390	U
Hexachlorobutadiene	390	Ω	410	Ω	420	D	390	Ω	390	n	390	n
Hexachlorocyclopentadiene	390	Ω	410	n	420	n	390	n	390	Ω	390	Ω
Hexachloroethane	390	Ω	410	D	420	D	390	Ω	390	D	390	Ω
Indeno[1,2,3-c,d]pyrene	390	Ω	410	Ω	420	n	390	Ω	390	D	390	n
Isophorone	390	Ω	410	Ω	420	D	390	Ω	390	n	390	n
2-Methylnaphthalene	390	Ω	410	Ω	420	D	390	n	390	n	390	n
4-Methylphenol	390	Ω	410	Ŋ	420	n	390	Ω	390	Ω	390	Ω
2-Methylphenol	390	n	410	Ω	420	n	390	Ω	390	n	390	Ω
Naphthalene	390	n	410		420	n	390	Þ	390	Ω	390	Ω

TABLE C.14d (Cont.)

			Soil 6	Conce	ntrations	(µg/kg	) at Vari	ous De	Soil Concentrations (µg/kg) at Various Depth Intervals	/als		
Compound	VXBOR2 (4-6 ft)	)R2 ft)	VXBOR2 (6-8 ft)	22	VXBOR2 (8-10 ft)	)R2 (ft)	VXBOR2 (10-12 ft)	)R2 2 ft)	FTBOR1 (0-2 ft)	1R.1 (t)	FTBO (0-	FTBOR1 Dup (0-2 ft)
2-Nitroaniline	066	n	1,000	Ω	1,100	Ω	086	Þ	080	1	000	=
3-Nitroaniline	066	Ω	1,000	n	1,100	D	086	n	086	) D	066	) <u> </u>
4-Nitroaniline	066	Ω	1,000	Ω	1,100	Ω	086	n	086	n	066	o =
Nitrobenzene	390	Ω	410	n	420	Ω	390	Ω	390	þ	390	) <u> </u>
2-Nitrophenol	390	Ω	410	n	420	Ω	390	U	390	n	390	) <u>=</u>
4-Nitrophenol	066	D	1,000	Ω	1,100	Ω	086	n	980	'n	066	) <u> </u>
N-Nitroso-di-n-propylamine	390	Ω	410	Ω	420	D	390	n	390	ם כ	360	) <b>=</b>
N-Nitrosodiphenylamine	390	n	410	Ω	420	D	390	n	390		340	) <u>=</u>
2,2'-Oxybis-(1-chloropropane)	390	n	410	n	420	Ω	390	þ	390	) <u> </u>	340	)
Pentachlorophenol	066	Ω	1,000	n	1,100	Ω	086	n	086	) <u> </u>	066	) <u> </u>
Phenanthrene	390	Ω	410	Ω	420	Ω	390	D	390	) <u> </u>	300	> <b>=</b>
Phenol	390	Ω	410	Ω	420	n	390	'n	390	ם ב	390	o =
Pyrene	390	Ω	410	Ω	420	Þ	390	Ω	390	) [	390	) <u> </u>
1,2,4-Trichlorobenzene	390	Ω	410	D	420	n	390	n	390	n	360	) <u> </u>
2,4,5-Trichlorophenol	066	Ω	1,000	Ω	1,100	Ω	086	Û	086	n	066	) <b>=</b>
2,4,6-Trichlorophenol	390	n	410	Ω	420	n	390		390	· =	300	) <u>=</u>
	1 1 1 1 1	1 1 1 1 1 1	111111	1	11111	1		1	1	֓֞֝֞֝֞֜֝֞֜֝֓֓֓֞֝֓֓֓֓֞֜֝֓֓֓֓֞֝֓֓֓֞֝֜֜֝֝֓֡֜֝֜֝֡֜֝֡֡֜֝֡֜֜֝֡֡֜֜֝	020	>

TABLE C.14d (Cont.)

			Soil	Conce	ntrations	(µg/kg)	Soil Concentrations (µg/kg) at Various Depth Intervals	is Dept	h Interva	SI		
Parison Modern	FTBORI	<b>D</b> 4	FTBOR1 Dup	dno	FTBOR1	181	FTBOR1	3.	FTBORI	)R1	FTBORI	R1
Compound	(1I <del>5-</del> -7)		(2-4 II)		(4-0 11)	£	(6-8 tt)	£	(8-10 ft)	(£)	(10-12 ft)	£
Acenaphthene	380	Ω	380	U	390	Ω	400	Ω	460	Ų	410	Ω
Acenaphthylene	380	Ω	380	n	390	n	400	U	460	n+	410	n
Anthracene	380	Ω	380	n	390	n	400	Ω	460	Ú+	410	Þ
Benz[a]anthracene	380	Ω	380	n	390	n	400	n	460	Ω+	410	n
Benzo[a]pyrene	380	n	380	U	390	D	400	Ω	460	T+	410	Ω
Benzo[b]fluoranthene	380	Ω	380	D	390	n	400	n	460	Ů,	410	Ω
Benzo[g,h,i]perylene	380	Ω	380	Ω	390	Ω	400	n	460	n+	410	Ω
Benzo[k]fluoranthene	380	Ω	380	U	390	Ω	400	Ω	460	Ų	410	Þ
Bis(2-chloroethoxy)methane	380	Ω	380	D	390	D	400	Ω	460	Ú+	410	Ω
Bis(2-chloroethyl)ether	380	Ω	380	U	390	D	400	Ω	460	Û+	410	U
Bis(2-ethylhexyl)phthalate	4,300	В	4,100	В	4,500	В	3,200	В	460	Ü+	3,200	В
4-Bromophenyl phenyl ether	380	n	380	D	390	D	400	D	460	n+	410	Ω
Butylbenzylphthalate	380	n	380	n	390	D	400	Ω	460	t'U	410	D
Carbazole	380	D	380	n	390	Ω	400	n	460	T+	410	n
4-Chloro-3-methylphenol	380	D	380	Ŋ	390	n	400	Ŋ	460	n+	410	Ω
4-Chloroaniline	380	n	380	Ŋ	390	Ω	400	n	460	t n	410	Ω
2-Chloronaphthalene	380	D	380	D	390	n	400	n	460	T+	410	n
2-Chlorophenol	380	Ŋ	380	b	390	n	400	Ω	460	U+	410	Ω
4-Chlorophenyl phenyl ether	380	n	380	b	390	n	400	n	460	n+	410	Ω
Chrysene	380	ם	380	b	390	Ω	400	D	460	† 1	410	n
Di-n-butylphthalate	380	מ	140	<b></b>	130	_	120	_	200	JB+	410	U
Di-n-octylphthalate	380	ב	380	<b>.</b>	390	n	400	$ \cap $	460	n+	410	D
Dibenz[a,h]anthracene	380	D	380	D	390	Ω	400	Ω	460	n+	410	U
Dibenzofuran	380	ר	380	7	390	n	400	n	460	r T	410	n

TABLE C.14d (Cont.)

	FTBORI	12	FTBOR1 Dup	dng	FTBOR	OR1	FTBOR1	)R.I	FTBOR1	OR1	FTBOR1	OR1
Compound	(2-4 ft)		(2-4 ft)		(4-6 ft)	ft)	(6-8 ft)	Œ	(8-10 ft)	0 ft)	(10-12 ft)	2 ft)
1,2-Dichlorobenzene	380	Ω	380	Ω	390	Ω	400	Ω	460	n+	410	<b>—</b>
1,3-Dichlorobenzene	380	Ω	380	Ω	390	D	400	Ω	460	n+ D	410	) <u> </u>
1,4-Dichlorobenzene	380	Ω	380	U	390	Ω	400	n	460	t n	410	) =
3,3'-Dichlorobenzidine	380	Ω	380	n	390	n	400	Ω	460	n+	410	) =
2,4-Dichlorophenol	380	Ω	380	U	390	Ω	400	n	460	r D	410	) 🗀
Diethylphthalate	380	n	380	U	390	Ω	400	D	460	Ü+	410	
2,4-Dimethylphenol	380	U	380	U	390	Ω	400	n	460	Ú+	410	י ה
Dimethylphthalate	380	Ω	380	n	390	Ω	400	Ω	460	n+	410	$\mathbf{n}$
4,6-Dinitro-2-methylphenol	096	U	950	Ω	086	Ω	1,000	Ω	1,100	n+	1.000	
2,4-Dinitrophenol	096	Ω	950	Ω	086	Ω	1,000	Ω	1,100	†n	1,000	
2,4-Dinitrotoluene	380	Ω	380	Ω	390	Ω	400	U	460	n+	410	Ω
2,6-Dinitrotoluene	380	Ω	380	U	390	Ω	400	ם	460	1°	410	D
Fluoranthene	380	Ω	380	U	390	Ω	400	n	460	n+	410	D
Fluorene	380	n	380	U	390	Ω	400	D	460	n+	410	Ω
Hexachlorobenzene	380	Ω	380	n	390	Ω	400	n	460	ů+	410	n
Hexachlorobutadiene	380	n	380	Ω	390	Ω	400	n	460	r n	410	
Hexachlorocyclopentadiene	380	n	380	n	390	Ω	400	ח	460	Ú+	410	D
Hexachloroethane	380	D	380	n	390	n	400	n	460	U+	410	D
Indeno[1,2,3-c,d]pyrene	380	b	380	Ŋ	390	Ω	400	n	460	n+	410	n
Isophorone	380	D	380	n	390	Ω	400	Ω	460	r U	410	n
2-Methylnaphthalene	380	b	380	b	390	n	400	n	460	Û+	410	Ω
2-Methylphenol	380	IJ	380	D	390	n	400	Ŋ	460	r U	410	U
4-Methylphenol	380	Ü	380	ם	390	Ω	400	n	460	r n	410	Ŋ
Naphthalene	380	-	360	<u>+</u>		,						

TABLE C.14d (Cont.)

•		201	Concer	Ittations	(µg/kg)	Son Concentrations (µg/kg) at Various Depth Intervals	ns Depr	II MIRCI VA			
Compound	FTBOR1 (2-4 ft)	FTBOR1 Dup (2-4 ft)	(Dup	FTBOR (4-6 ft)	0R.1 ft)	FTBOR (6-8 ft)	f) Ri	FTBOR1 (8-10 ft)	OR1 ) ft)	FTBOR1 (10-12 ft)	2 ft)
2-Nitroaniline	N 096	950	n	086	n	1,000	Ω	1,100	Ú+	1.000	Ω
3-Nitroaniline	N 096	950	Ω	086	D	1,000	Ω	1,100	t'n	1.000	n
4-Nitroaniline	N 096	950	Ω	086	Ω	1,000	D	1,100	n+	1,000	n
Nitrobenzene	380 U	380	Ω	390	Ω	400	Ω	460	t'U	410	
2-Nitrophenol	380 U	380	Ω	390	Ω	400	Ω	460	n+	410	n
4-Nitrophenol	O 096	950	Ω	086	Ω	1,000	n	1,100	Ú+	1,000	D
N-Nitroso-di-n-propylamine	380 U	380	U	390	Þ	400	D	460	n+	410	
N-Nitrosodiphenylamine	380 U	380	n	390	n	400	Ω	460	Û+	410	Ω
2,2'-Oxybis-(1-chloropropane)	380 U	380	Ω	390	Ω	400	n	460	n+	410	
Pentachlorophenol	O 096	950	n	086	Ω	1,000	n	1,100	n+	1,000	D
Phenanthrene	380 U	380	Ω	390	Ω	400	D	460	n+	410	Ω
Phenol	380 U	380	Ω	390	Ω	400	n	460	n+	410	Ω
Pyrene	380 U	380	U	390	Ω	400	U	460	Λţ	410	Ω
1,2,4-Trichlorobenzene	380 U	380	Ω	390	Ω	400	n	460	T+	410	Ŋ
2,4,5-Trichlorophenol	N 096	950	D	086	n	1,000	Ω	1,100	T+	1,000	n
2,4,6-Trichlorophenol	380 U	380	Ω	390	Ŋ	400	Ω	460	r A	410	$\Box$

TABLE C.14d (Cont.)

Compound	TBNPBOR3 (0-2 ft)	TBNPBOR (2-4 ft)	TBNPBOR3 (4-6 ft)	TBNPBOR3 (6-8 ft)	TBNPBOR (8-10 ft)	TBNPBOR3 (10-12 ft)
Acenaphthene	393 U	446 UJ	465 11	393 11	11 C17	
Acenaphthylene	393 U	446 111			412 0	·
Anthracene	393 U	•	·			
Benz[a]anthracene	363	_				
Benzo[a]pyrene	393 11		465 U			
Benzo[b]fluoranthene	393 11	·				
Benzolo h ilnemlane	0 666				412 U	406 U*J
Donne [1-14]	393 U		465 U	393 U	412 U	406 U*J
Denzo[k]Huoranthene	51 J	446 UJ	465 U	393 U	412 U	406 U*J
Bis(2-chloroethoxy)methane	393 U	446 UJ	465 U	393 U	412 U	406 11*1
Bis(2-chloroethyl)ether	393 U	446 UJ	465 U	393 U		_
Bis(2-ethylhexyl)phthalate	65 J	446 UJ	1,300	1,900		
4-Bromophenyl phenyl ether	393 U	446 UJ	465 U	393 U		
Butylbenzylphthalate	393 U	446 UJ	465 U			
Carbazole	393 U	446 UJ	465 11			
4-Chloro-3-methylphenol	393 U			,		
4-Chloroaniline	393 U	446 UJ			·	
2-Chloronaphthalene	393 U				·	
2-Chlorophenol	393 U	·		·	412 O	
4-Chlorophenyl phenyl ether	393 U					
Chrysene	393 U		·			·
di-n-Butylphthalate	393 11	·		·	417 O	
di-n-Octvlnhthalate	202	,		393 U	63 J	406 U*J
in Coty puniatate	393 U		465 U	393 U	412 U	406 U*J
Diversial	393 U		465 U	393 U	412 U	406 U*J
Dibenzoruran	393 U	446 111	11 397	202		

TABLE C.14d (Cont.)

Compound	TBNPBOR3 (0-2 ft)		TBNPBOR (2-4 ft)	OR (	TBNPBOR3 (4-6 ft)	NPBOR3 (4-6 ft)	TBNPBOR3 (6-8 ft)	BOR3 ft)	TBNPBOR (8-10 ft)	BOR ) ft)	TBNI (10	TBNPBOR3 (10-12 ft)
1,2-Dichlorobenzene	393 U	4	446 L	m	465	Ω	393	Þ	412	n	406	I*(1)
1,3-Dichlorobenzene	393 U	4	446 L	UJ	465	U	393	Ω	412	n	406	1*11
1,4-Dichlorobenzene	393 U	4	446 L	T)	465	Ω	393	U	412	· D	406	(*)
3,3'-Dichlorobenzidine	393 U	4	446 L	UJ	465	Ω	393	Ω	412	'n	406	[*N
2,4-Dichlorophenol	393 U	4	446 L	U	465	U	393	n	412	Ω	406	(*) (*)
Diethylphthalate	393 U	4	446 U	CI	465	ם	393	Ω	412	Ω	406	[*]
2,4-Dimethylphenol	393 U	4	446 U	U	465	Ŋ	393	D	412	n	406	U*J
Dimethylphthalate	393 U	4	446 U	U	465	Ω	393	n	412	Ω	406	[*]
4,6-Dinitro-2-methylphenol	N 886	1,1	,120 U	U	1,170	Ω	886	n	1,040	D	1.020	[*1]
2,4-Dinitrophenol	N 886	1,1	,120 U	U	1,170	U	886	Ω	1,040	Ω	1,020	[*N
2,4-Dinitrotoluene	393 U	4	446 UJ	Ţ	465	Ω	393	D	412	Ω	406	*
2,6-Dinitrotoluene	393 U	4	446 UJ	Ţ	465	Ω	393	n	412	n	406	U*J
Fluoranthene	393 U	4	446 UJ	į	465	Ω	393	n	412	Ω	406	[*]
Fluorene	393 U	4	446 UJ	Ţ	465	U	393	Ω	412	n	406	U*J
Hexachlorobenzene	170 J	4	446 UJ	ſ	465	Ω	393	Ω	412	D	406	U*J
Hexachlorobutadiene	393 U	4	446 UJ	سر	465	Ω	393	n	412	Ω	406	[*]
Hexachlorocyclopentadiene	393 U	4	446 UJ	Ţ	465	Ω	393	Ω	412	Ω	406	ſ*Ŋ
Hexachloroethane	393 U	4	446 UJ	J	465	Ŋ	393	Ω	412	Þ	406	f*N
Indeno[1,2,3-cd]pyrene	393 U	4	446 UJ	_	465	U	393	Ω	412	D	406	[*]
Isophorone	393 U	446	te uj	1	465	U	393	Ŋ	412	D	406	U*J
2-Methylnaphthalene	120 J	446	t6 UJ	1	580		2,100		412	D	1,700	*
2-Methylphenol	393 U	446	10 91	<b>-</b>	465	Ω	393	n	412	n	406	(*)
4-Methylphenol	393 U	446	16 UJ	_	465	Ω	393	D	412	D	406	U*J
Naphthalene	393 U	446	16 UJ	_	710		1,800		412	1	1,600	*

TABLE C.14d (Cont.)

			Son Concentrations (µg/kg) at Various Depth Intervals	arious Depth Inte	rvals	
Compound	TBNPBOR3 (0-2 ft)	TBNPBOR (2-4 ft)	TBNPBOR3 (4-6 ft)	TBNPBOR3 (6-8 ft)	TBNPBOR (8-10 ft)	TBNPBOR3 (10-12 ft)
2-Nitroaniline	N 886	1,120 UJ	1.170 11	11 886	1 040	
3-Nitroaniline	N 886					
4-Nitroaniline		•	·			
Nitrobenzene	•					
2-Nitrophenol						
4-Nitrophenol	·				412 U	406 U*J
N. Nitrogo di a mondification			1,170 U	O 886	1,040 U	1,020 U*J
IN-INITIOSO-GI-II-PIOPYIAMINE			465 U	393 U	412 U	406 U*J
IN-INITOSOdiphenylamine	393 U	446 UJ	465 U	393 U	412 U	406 U*J
2,2-Oxybis-(1-chloropropane)	393 U	446 UJ	465 U	393 U	412 U	
Pentachlorophenol	Ω 886	1,120 UJ	1,170 U	D 886		
Phenanthrene	93 J	446 UJ	465 U	880		
Phenol	393 U	446 UJ	465 U	393 U		_
Pyrene	62 J	446 UJ	465 U			•
1,2,4-Trichlorobenzene	393 U	446 UJ	465 U	393		_
2,4,5-Trichlorophenol	N 886	1,120 UJ	1,170 U			-
2,4,6-Trichlorophenol	393 U	446 UJ	465 U			•
		Soil Concentration	Soil Concentrations (µg/kg) at Various Depth Intervals	Depth Intervals		; ; ; ; ;
Compound	VXBOR3	VXBOR4	VXBOR4	VXBOR5	VXBOR5	
Y	(11 1. 7)	(11 4-7)	(4-0 It)	(4-6 ft)	(8-10 ft)	
Acenaphthene	7,920 U	8,350 UJ	8,050 U	428 U	407 U	
Acenaphthylene	7,920 U	8,350 UJ	8,050 U	428 U	407 U	
Anthracene	810	9 250 111			ı	

TABLE C.14d (Cont.)

	VXBOR3	OR3	IXA	VXBOR4	VXB	VXBOR4	VXBOR5	OR5	VXBOR5	OR5
Compound	(2-4 ft)	ft)	(2-	(2-4 ft)	(4-1	(4-6 ft)	(4-6 ft)	ft)	(8-10 ft)	) ft)
Benz[a]anthracene	7,920	n	1,200	U	8,050	U	428	Ω	407	n
Benzo[a]pyrene	7,920	Ŋ	1,300	ŗ	8,050	n	428	Ω	407	n
Benzo[b]fluoranthene	7,920	Ω	1,200	ŗ	1,020	ſ	428	n	407	n
Benzo[g,h,i]perylene	890	<b>⊢</b>	2,400	J	1,600	J	428	U	407	D
Benzo[k]fluoranthene	7,920	n	8,350	n	3,100	J	428	Ω	407	Ω
Bis(2-chloroethoxy)methane	7,920	Ω	8,350	UJ	8,050	Ω	428	D	407	n
Bis(2-chloroethyl)ether	7,920	Ω	8,350	UJ	8,050	Ω	428	Þ	407	ר
Bis(2-ethylhexyl)phthalate	7,920	Ω	8,350	UJ	8,050	Ω	428	n	407	D
4-Bromophenyl phenyl ether	7,920	n	8,350	U	8,050	Ω	428	n	407	n
Butylbenzylphthalate	7,920	Ω	8,350	U	8,050	n	428	D	407	n
Carbazole	7,920	n	8,350	UJ	8,050	n	428	D	407	n
4-Chloro-3-methylphenol	7,920	Ω	8,350	M	8,050	D	428	n	407	Ω
4-Chloroaniline	7,920	U	8,350	n	8,050	Ω	428	Ω	407	
2-Chloronaphthalene	7,920	U	8,350	rin	8,050	Ω	428	n	407	Þ
2-Chlorophenol	7,920	Ω	8,350	m	8,050	Ω	428	n	407	ח
4-Chlorophenyl phenyl ether	7,920	Ω	8,350	UJ	8,050	Ω	428	Ω	407	D
Chrysene	1,500	Ţ	2,100	homa	1,100	J	428	D	407	n
di-n-Butylphthalate	7,920	n	47,000	BJ	27,000	В	428	D	73	JB
di-n-Octylphthalate	7,920	Ω	8,350	Ω	8,050	Ω	428	Ω	407	Ω
Dibenz[a,h]anthracene	7,920	U	1,700	_	8,050	C	428	D	407	n
Dibenzofuran	7,920	Ω	8,350	Ü	8,050	Ω	428	n	407	n
1,2-Dichlorobenzene	7,920	n	8,350	m	8,050	Ω	428	n	407	D
(3-Dichlorobenzene	7,920	Ω	8,350	m	8,050	D	428	D	407	$\supset$
1,4-Dichlorobenzene	7,920	D	8,350	UJ	8.050		428	Ξ	707	-

TABLE C.14d (Cont.)

			Soil (	Concentra	Soil Concentrations (µg/kg) at Various Depth Intervals	at Variou	is Depth In	itervals			
	VXBOR3	OR3	VX	VXBOR4	VXE	VXBOR4	VXBOR5	OR5	VXBOR5	OR5	
Compound	(2-4 ft)	(£)	(2)	(2-4 ft)	(4-	(4-6 ft)	(4-6 ft)	ft)	(8-10 ft)	0 ft)	
3,3'-Dichlorobenzidine	7,920	n	8,350	M	8,050	Ω	428		407	i.	
2,4-Dichlorophenol	7,920	Ω	8,350	UJ	8,050	Ω	428	o Þ	407	) <u>=</u>	
Diethylphthalate	7,920	n	8,350	UJ	8,050	n	428	) =	407	) <u> </u>	
2,4-Dimethylphenol	7,920	n	8,350	U	8,050	) D	428	) <u> </u>	407	) <u>=</u>	
Dimethylphthalate	7,920	n	8,350	U	8,050	'n	428	î D	407	) <u>=</u>	
4,6-Dinitro-2-methylphenol	19,900	Ω	21,000	U	20,200	U	1080	D	1020	) <u> </u>	
2,4-Dinitrophenol	19,900	n	21,000	UJ	20,200	Ŋ	1080	'n	1020	) =	
2,4-Dinitrotoluene	7,920	n	8,350	U	8,050	Ö	428	n	407	) <u></u>	
2,6-Dinitrotoluene	7,920	Ω	8,350	UJ	8,050	D	428	n	407	) <u>=</u>	
Fluoranthene	7,920	Ω	2,600	F	1,900	۳-,	150	· —	407	î D	
Fluorene	7,920	Ω	8,350	UJ	8,050	Ω	428	ח	407	î D	
Hexachlorobenzene	7,920	D	8,350	UJ	8,050	D	428	Ω	407	n	
Hexachlorobutadiene	7,920	Ω	8,350	m	8,050	Ω	428	Ω	407	) =	
Hexachlorocyclopentadiene	7,920	Ω	8,350	Ω	8,050	Ω	428	D	407	<u> </u>	
Hexachloroethane	7,920	n	8,350	m	8,050	n	428	Ω	407	n	
Indeno[1,2,3-cd]pyrene	7,920	D	2,000	J	8,050	U	428	Ω	407	'n	
Isophorone	7,920	Ω	8,350	m	8,050	n	428	Ω	407	Π	
2-Methylnaphthalene	086	ſ	53,000	J	12,000		8100	*	407	· =	
2-Methylphenol	7,920	n	8,350	<b>UJ</b>	8,050	Ŋ	428	n	407	) <u> </u>	
4-Methylphenol	7,920	Ω	8,350	UJ	8,050	Ω	428	'n	407	) <u>=</u>	
Naphthalene	7,920	Ω	8,350	m	8,050	U	420	<u>ب</u>	407	n	
2-Nitroaniline	19,900	Ω	21,000	UJ	20,200	n	1080	D	1020	· =	
3-Nitroaniline	19,900	Ω	21,000	m	20,200	n	1080	n	1020	, D	
4-Nitroaniline	19,900	Ω	21,000	UJ	20,200	U	1080	n	1020	n	

TABLE C.14d (Cont.)

Compound											
Compound	VXBOR3	1R3	IXA	VXBOR4	VXE	VXBOR4	VXB	VXBOR5	VXBOR5	OR5	
	(2-4 ft)	(f)	(2)	(2-4 ft)	(4-	(4-6 ft)	(4-(	(4-6 ft)	(8-1	(8-10 ft)	
Nitrobenzene	7,920	Ω	8,350	5	8,050	D	428	Ω	407	Þ	
2-Nitrophenol	7,920	Ω	8,350	Ω	8,050	n	428	n	407	'n	
4-Nitrophenol	19,900	Ω	21,000	UJ	20,200	D	1080	Ω	1020	Ď	
N-Nitroso-di-n-propylamine	7,920	Ω	8,350	Ω	8,050	n	428	Ω	407	n	
N-Nitrosodiphenylamine	7,920	Ω	8,350	IJ	8,050	D	428	n	407	n	
2,2'-Oxybis-(1-chloropropane)	7,920	n	8,350	U	8,050	n	428	D	407	Þ	
Pentachlorophenol	19,900	U	21,000	Ω	20,200	n	1080	D	1020	Ω	
Phenanthrene	1,500	<b>-</b>	32,000	J	27,000		3100		407	Ω	
Phenol	7,920	n	8,350	UJ	8,050	D	428	D	407	D	
Pyrene	3,300	ĭ	8,350	U	8,050	n	170	-	407	'n	
1,2,4-Trichlorobenzene	7,920	Ω	8,350	Ω	8,050	U	428	n	407	Ω	
2,4,5-Trichlorophenol	19,900	Ŋ	21,000	UJ	20,200	Ω	1080	Ω	1020	Ω	
2,4,6-Trichlorophenol	7,920	n	8,350	UJ	8,050	D	428	Ω	407	Ω	
		Soil Co	Soil Concentrations (µg/kg) at 0-6 in.	s (µg/kg) a	ıt 0-6 in.		[ ] ] ] [ ] [	i ! !	 	 	
Compound	SQPS-1	<del></del>	SQPS-5	S-5	SQF	SQPS-8					
Acenaphthene	412	Ω	407	Ω	989	n	I				
Acenaphthylene	412	n	407	n	989	D					
Anthracene	412	U	407	Ω	989	n					
Benz[a]anthracene	412	Ŋ	407	D	989	Þ					
Benzo[a]pyrene	412	Ω	407	Ω	989	Ω					
Benzo[b]fluoranthene	412	Ω	407	מ	989	Ω					
Benzole h ilnervlene	717	11	107								

TABLE C.14d (Cont.)

Compound	SQPS-1	S-1	SO	SQPS-5	SQF	SQPS-8
Benzo[k]fluoranthene	412	n	42		989	=
Bis(2-chloroethoxy)methane	412	n	407	D	989	) <del>=</del>
Bis(2-chloroethy1)ether	412	Ω	407	n	989	) <u>=</u>
Bis(2-ethylhexyl)phthalate	412	n	407	n	989	) <u> </u>
4-Bromophenyl phenyl ether	412	n	407	n	989	) <u>;</u>
Butylbenzylphthalate	412	Ω	407	n	989	î
Carbazole	412	n	407	n	989	<u> </u>
4-Chloro-3-methylphenol	412	D	407	Ω	989	) <u>=</u>
4-Chloroaniline	412	Ω	407	n	989	) =
2-Chloronaphthalene	412	Ω	407	Ω	989	) <u>=</u>
2-Chlorophenol	412	U	407	Ω	989	n
4-Chlorophenyl phenyl ether	412	Ω	407	n	989	Ď
Chrysene	412	Ω	407	Ω	989	Ω
di-n-Butylphthalate	412	n	407	Ω	989	i D
di-n-Octylphthalate	412	n	407	Ω	989	'n
Dibenz[a,h]anthracene	412	Ω	407	Þ	989	n
Dibenzofuran	412	Ω	407	Ω	989	n
1,2-Dichlorobenzene	412	Ω	407	Ŋ	989	n
1,3-Dichlorobenzene	412	Ω	407	Ω	989	Ω
1,4-Dichlorobenzene	412	Ω	407	Ω	989	n
3,3'-Dichlorobenzidine	412	n	407	n	989	n
2,4-Dichlorophenol	412	Ω	407	n	989	'n
Diethylphthalate	412	n	407	11	707	1.1

TABLE C.14d (Cont.)

			concentrations (PE/RE) at 0-0 m.			
Compound	SQPS-1	S-1	SQF	SQPS-5	SQI	SQPS-8
2,4-Dimethylphenol	412	n	407	Ω	989	D
Dimethylphthalate	412	U	407	D	989	n
4,6-Dinitro-2-methylphenol	1,040	Ŋ	1,020	n	1,730	U
2,4-Dinitrophenol	1,040	Ω	1,020	n	1,730	n
2,4-Dinitrotoluene	412	Ω	407	Ω	989	Ω
2,6-Dinitrotoluene	412	n	407	U	989	n
Fluoranthene	412	Ω	407	Ω	989	Þ
Fluorene	412	Ω	407	n	989	D
Hexachlorobenzene	412	Ω	407	Ŋ	989	Ŋ
Hexachlorobutadiene	412	Ω	407	Ω	989	Þ
Hexachlorocyclopentadiene	412	Ŋ	407	Ω	989	U
Hexachloroethane	412	D	407	Ω	989	D
Indeno[1,2,3-cd]pyrene	412	Ω	407	U	989	Ω
Isophorone	412	U	407	U	989	Ŋ
2-Methylnaphthalene	412	U	407	Ω	989	n
2-Methylphenol	412	n	407	Ω	989	D
4-Methylphenol	412	Ω	407	Ω	989	D
Naphthalene	412	Ω	407	Ω	989	D
2-Nitroaniline	1,040	U	1,020	Ŋ	1,730	Ω
3-Nitroaniline	1,040	Ω	1,020	D	1,730	n
4-Nitroaniline	1,040	D	1,020	Ω	1,730	Ω
Nitrobenzene	412	Ω	407	Ω	989	D
2-Nitrophenol	412	Ω	407	n	989	D
4-Nitrophenol	1.040	=	1.020		1 720	-

TABLE C.14d (Cont.)

at 0-6 in.	SQPS-8	n 989	n 989	N 989	1,730 U	n 989	n 989	n 989	Ω 989	1,730 U	11 989
ıs (µg/kg)	S-5	Ω	U	U	Ŋ	n	Ω	n	U	n	n
Soil Concentrations (µg/kg) at 0-6 in.	SQPS-5	407	407	407	1,020	407	407	407	407	1,020	407
Soil C	S-1	n	Ω	U	Ω	Ω	Ω	Ω	U	Ω	U
	SQPS-1	412	412	412	1,040	412	412	412	412	1,040	412
•	Compound	N-Nitroso-di-n-propylamine	N-Nitrosodiphenylamine	2,2'-Oxybis-(1-chloropropane)	Pentachlorophenol	Phenanthrene	Phenol	Pyrene	1,2,4-Trichlorobenzene	2,4,5-Trichlorophenol	2,4,6-Trichlorophenol

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

\* = internal standard area counts were outside QC limits.

B = analyte was found in the associated blank.

TABLE C.15a Inorganic Analysis for Soil Borings, January 1993<sup>a</sup>

Analyte	JBP1-E-2' (2 ft)	JBP1-E-2'Dup (2 ft)	-2'Dup (t)	JBP1-E-4' (4 ft)	E-4'	JBP1-W-2' (2 ft)		JBP1-W-4' (4 ft)	JBP2-E-2' (2 ft)	3-2'	JBP2-E-4' (4 ft)	E-4'	JBP2-E-4'Dup (4 ft)	4'Dup t)
Aluminum	11,300	6,870		12,600		14,000	12,800	Q	10.900		3.810		6.480	-
Antimony	5.08	4.57	Ω	4.7	D	4.68 U		3	4.6	Ξ	4 59	Ĭ	7 57	11
Arsenic	5.02	3.22		4.18					3.5	)	1.32	)	733	)
Barium	125	45.3		34.1		47.5	22.4	4	32.7		21.5		29.5	
Beryllium	0.31	0.19	Ω	0.45		0.33	0.28	~	0.23		0.19	Ω	0.38	
Cadmium	3.21	0.57	n	0.59	Ω	0.59 U		5	0.58	Ω	0.57	ם י	0.57	n
Calcium	1,530	5,670		787		529	828	∞	786		618		3.640	)
Chromium	36.6	11.6		18.2		16.5	16.5	5	14.2		6.51		12.7	
Cobalt	5.64	5.43	n	5.58	Ω	5.56 U		n 9	5.47	Ŋ	5.45	Ω	5.43	Ξ
Copper	248	34.1		10.1		19.9	10.9	6	6.85		9.73		10.1	)
Iron	35,800	11,400		23,100		13,300	20,400	0	14,100		4,900		21.100	
Lead	340	36.4		10.5		13.2	9.31		15.9		23.6		96.3	
Magnesium	2,340	1,200		1,940		2,490	1,610	)	1,560		572		704	
Manganese	197	118		72.6		98.6	47	7	64.6		21.5		108	
Mercury	0.12 U	0.12	Ω	0.12	n	0.12 U	0.12	7 n	0.12	Ω	0.12	Þ	0.12	n
Nickel	20.4	9.34		13.8		15.4	10	•	8.19		6.91	Ω	6.88	D
Potassium	916	530		1,220		1,460	1,460	_	971		338		482	
Selenium	1.05 U	1.02	D	1.05	Ω	1.05 U	1.07	0 /	1.03	n	1.03	D	1.02	ņ
Silver	1.63	0.17	D	0.17	Ω	0.56	0.17	u 1	0.17	Ω	0.17	Ω	0.17	Ω
Sodium	46.5 U	45.2	n	46.5	n	46.3 U	47.2	n a	333		45.4	Ω	86.5	
Thallium	0.56 U	0.54	D	0.56	Ω	0.56 U	0.57	n ,	0.55	Ω	0.55	n	0.55	Þ
Vanadium	19.8	12.8		23.4		22.1	25.1		19.2		7.57		18.9	
Zinc	1.220	182		,,,,		(								

TABLE C.15a (Cont.)

Aluminum Antimony	JBP2-C-2' (2 ft)		JBP2-C-4' (4 ft)	-4	JBP2-W-2 (2 ft)	7-2'	JBP2-W-4' (4 ft)	V-4'	JHDP-C-4' (4 ft)	-C-4'	JHDP-C-6' (6 ft)	-C-6'	JHDP-C-6'Dup (6 ft)	C-6'Dup	JVXP-C-4'	C-4
Antimony	4,980	δ.	5,190		9000		008 8		000 8							3
	5 29	L	1 57	Ļ		:	0,00	,	0,020		8,200		5,620		4,130	
				<b>D</b>	70.4	<b>&gt;</b>	4.58	Þ	4.6	Ω	4.67		4.3	D	5.06	
Arsenic	7.91		6.9		26.6		3.98		3.94		6.29		3.6		1 10	
Barium	121		91.8		87.8		28.8		32.3		504		32.4		51.1	
Beryllium	0.24	-	0.19	J.	0.19		0.19	Ω	0.46		0.00		72.4		77	;
Cadmium	6.97	-	0.57	J	0.58	U	0.57	. 11	0.57	11	7 22		0.41		0.21	$\supset$
Calcium	2,660		803		535		222	)	1 \$40	)	7.43		7 6		0.63	Þ
Chromium	7.35	-	8.42		=		12.7		0+0,1		2,190		1,310		334	
Cobalt		1	5.42		, , , , , , , , , , , , , , , , , , ,	11	1.C.1	,	9.11		44.1		13.8		5.33	
00000				_	5.49	<b>5</b>	5.44	<b>-</b>	5.46	Þ	5.55	Ω	5.1	D	6.01	
Copper	413	7	42.7		17.5		17.9		4.12	Ω	123		18.9		4 53	11
Iron	10,700	∞ં	8,860	-1	10,400		12,200		10,400		42.100		17 400		002.3	)
Lead	403	(')	37.8		37.6		8.54		10.5		121		00T,11		00/6	
Magnesium	805		903		1.150		1 830		213		171		4.07		10.5	
Manganese	70	•	52.1		76.5		0.00,1		616		3,900		969		1,100	
Mercura	2.0	, c	72.1				57.2		31.4		157		23.7		45.8	
retenty Haled	0.14	י נ	0.12 0	_		b	0.12	Ω	0.1	n	0.11		0.11		0.1	Ω
inickei 7	9.91	9	6.87 U		8.04		10.3		6.92	Ω	31.8		6.46	Ω	292	
Potassium	501	••	529		730		1,160		432		260		568	)	70.7	)
Selenium	1.18 U	] 1	1.02 U		1.03	Ω	1.02	Ω	1.03	Ω	1.05	D	96.0	11	1 13	<u>;</u>
Silver	0.19 U	1 0	0.17 U		0.17	n	0.17	Ŋ	0.17	Ω	0.17	- 1	0.16	> <b>=</b>	010	) ;
Sodium	52.3 U	1	45.2 U		45.7	Ω	45.3	Ω	51.2		73	,	3 CV	) <b>:</b>	0.10	);
Thallium	0.63 U		0.55 U		0.55	n	0.55	Ω	0.55	11	24 0	1	C:7t	<b>)</b> ;	30.1	<b>)</b>
Vanadium	7.72	1	11.1		12.5		19.6	ı	16	)	000	)	0.32	<b>)</b>	0.61	$\supset$
Zinc	17,800	<b>∞</b>	834		516		167		2 20		7:47		7.01		5.95	

TABLE C.15a (Cont.)

Analyte	JVXP-C-6'	C-6'	JSDP-C-4' (4 ft)	C-4′	JSDP-C-6' (6 ft)	-C-6'	JBPM-A-3" (3 in.)		JBPM-A-3"Dup (3 in.)	-3"Dup 1.)	JBPM-A-1'	4-1'	JBPM-B-3" (3 in.)	3-3"	JBPM-B-1' (1 ft)	B-1'
Aluminum	2,230		4,700		4,120		8,480		7,860		8,320		8,680		11.900	
Antimony	5.41	Ω	5.12	U	5.03	n	4.58	Ω	4.55	Ω	4.57	Ω	13.7		4.49	Ξ
Arsenic	1.04	Ω	1.38		0.97	Ω	3.95		4.8		22.5		7.49		3.58	)
Barium	8.1		15.5		12.7		403		158		216		336		42.8	
Beryllium	0.23	Ω	2.14	Ω	0.21	Ω	0.43		0.30		0.26		0.34		0.40	
Cadmium	89.0	Ω	0.64	Ω	0.63	Ω	2.95		2.37		6.05		5.94		0.56	Ω
Calcium	178		240		269		1,170		1,150		1,930		1,710		228	
Chromium	3.15		4.09		5.54		37.8		34.9		62.9		53.4		15.6	
Cobalt	6.42		60.9	Ω	5.98	D	5.44	U	5.41	Ω	5.42	n	6.41		5.33	D
Copper	4.84	Ω	4.59	n	4.51	Ŋ	305		209		392		521		4.77	1
Iron	3,170		5,970		5,090		13,400		12,600		21,000		28,300		20,500	
Lead	2.6		3.37		8.22		145		129		542		79,800		11.2	
Magnesium	551		1,010		944		1,470		1,170		1,350		4,460		1.550	
Manganese	25.6		48.1		38.4		142		138		163		209		86.6	
Mercury	0.1	D	0.1	Ω	0.1	Ω	0.13		0.22		0.95		0.22		0.1	
Nickel	8.14	n	7.72	Ω	7.58	n	8.19		98.9	Ω	12.5		24.2		7.26	
Potassium	405		383		376		431		384		429		243		508	
Selenium	1.21	n	1.15	n	1.13	Ω	1.03	Ω	1.02	n	1.02	Ω	1.98		1.01	$\supset$
Silver	0.20	n	0.19	Ω	0.18	Ω	2.91		1.87		1.81		1.49		0.16	D
Sodium	53.5		50.7	Ω	49.8	Ω	51		91.7		91.9		206		50	
Thallium	0.65	Ω	0.61	Ω	09.0	Ω	0.55	n	0.55	Ω	0.55	n	0.52	Ω	0.54	Ω
Vanadium	9	Ω	6.74		5.58	D	15.1		15.8		13.8		10.4		26.3	
Zinc	13.1		13.5		200		0		000							

TABLE C.15a (Cont.)

Analyte	JBPM-C-3" (3 in.)	-C-3"	JBPM-C-1'	(-C-1'	JBPP-A-3" (3 in.)	A-3" n.)	JBPP-A-1 (1 ft)	A-1'	JBPP-B-3" (3 in.)	B-3" 1.)	JBPP-B-1' (1 ft)	B-1'	JBPP-C-3" (3 in.)	C-3" n.)	JBPP-C-1' (1 ft)	. C (1
Aliminim	10 400		7		1										,	
A = 4.	10,400		2,100		5,670		7,320		6,720		5,860		5,910		10.500	
Antimony	4.88	Þ	4.4	n	4.5	Ω	4.55	D	67.3		44.2		4 57	П	7 57	1
Arsenic	12.3		2.32		1.76		2.32		16.7		100		75.0	)		)
Barium	2,400		150		107		31.5		534		300		53.7		1.88	
Beryllium	0.20	Ω	0.30		0.21		0.19	Ω	0.22	11	0 19	<u>,</u>	23.50		52.4	
Cadmium	2.43		0.55	Ω	0.56	U	0.57	Ω	11.4	)	0.57	)	0.00		07:0	<b>:</b>
Calcium	1,240		155		445		526		27 900		1 200		6.5		0.57	$\supset$
Chromium	9.68		7.47		17		10.3		200,12		0,500		470		701	
Cobalt	5.79	n	5.22	1	5 34	Ļ		-	C77		103		.11.3		14.3	
Conner	365			)	+ 1	)	14.0	<b>&gt;</b>	1.1		108		5.43	Ω	5.43	n
copper I	COC		77		60.5		4.08	Ω	806		662		33.2		8.39	
Iron	18,600		7,470		11,300		11,900		24,900		47,100		9.790		17 400	
Lead	290		6.05		105		8.37		1,180		1.110		286			
Magnesium	22,400		3,020		874		666		13,500		2.220		730		1 200	
Manganese	229		73.3		75.6		61.7		270		240		858		000,1	
Mercury	0.1	Ω	0.1	Ω	0.12	Ω	0.12	Ω	0.42		3.6		0.50	Ξ	3 5	į
Nickel	20.9		6.62	Ω	6.77	Ω	98.9	U	103		43.3		1 [	)	71.0	>
Potassium	271		329		331		384	ı	677				/		7.53	
Selenium	1 09	Ξ	000	<u>.</u>	101	1	t 6	;	000		381		337		554	
Silver	37.0	)	71.0	);	1.01	<b>D</b>	1.02	$\supset$	1.18	Ω	7.12		1.02	Ω	1.0	Ω
Sodium	0.4.0		0.16	<b>)</b>	0.26		0.17	Ω	8.53		5.84		0.19		0.17	Ω
Sodium	34.3	;	43.5	D :	44.5	Ω	45	Ω	176		44.8	U	45.2	Ω	45.2	n
LIIAIIIUIII	0.58	<b>-</b>	0.53	D	0.54	n	0.55	Ω	0.63	Ω	0.54	Ω	0.55	D	0.55	Ω

TABLE C.15a (Cont.)

	JBPP-C-1' (1 ft)	23.4
	JBPP-C-3" (3 in.)	12 82.8
ths	JBPP-B-1'	12.3 5,820
kg) at Various Dep	JBPP-B-3" (3 in.)	25.2 1,640
Soil Concentrations (mg/kg) at Various Depths	JBPP-A-1' (1 ft)	15 20.2
Soil C	JBPP-A-3" (3 in.)	12.6
	JBPM-C-1' (1 ft)	10.6
	JBPM-C-3" (3 in.)	11.3
,	Analyte	Vanadium Zinc

<sup>&</sup>lt;sup>a</sup> Sampled by Weston, analyzed by GP Environmental Services; provisional data.

Data qualifier:

U = analyte was analyzed for but not detected; detection limit given.

TABLE C.15b Inorganic Analyses for Soil Borings, January 1994<sup>a</sup>

			Soil	Soil Concentrations (mg/kg) at Various Depth Intervals	/kg) at Various De	pth Intervals		
Parameter	CLP1 (0-6 in.)	CLP1 (6-24 in.)	CLP1 (24-40 in.)	CLP2 (0-6 in.)	CLP2 (6-24 in.)	CLP2 (24-48 in.)	CLP3 (0-6 in )	CLP3
Metals							(:::::2)	(77.74 m.)
Aluminum	8,480	8,480	9,510	11,500	8,640	7.530	17 500	10.800
Antimony	360	54.4	5.1	119	198	134	16.6	7.77
Arsenic	18.3	46.9	18.9	37.3	31.6	39.8	2.5	15.4
Barium	873	1,470	1,340	837	1,180	1,230	843	1.580
Beryllium	0.44	0.31	0.56	0.34	0.23	0.42	0.95	0.59
Cadmium	6.5	9.1	3.4	6.4	8.9	8.1	1.6	7.0
Calcium	9,730	2,320	1,510	1,500	1,640	1,650	2.330	3 880
Chromium	171	293	85.9	338	878	312	59.6	130
Cobalt	6.1	6.7	5.1	7.7	10.3	8.4	8.5	7.3
Copper	919	2,170	301	1,840	4,320	1,380	530	1.070
Iron	21,200	46,200	20,600	68,800	104,000	69,400	29,000	49,000
Lead	41,000	6,460	1,060	16,500	24,200	18,900	3,190	15,500
Magnesium	2,870	3,400	3,030	2,160	2,590	2,500	3,390	3,230
Manganese	209	333	142	374	499	289	146	293
Mercury	0.5	0.46	0.38	0.27	0.18	0.21	1.5	1.4
Nickel	23.1	34.8	19.5	41.5	59.6	42.4	28.3	40.6
Potassium	613	408	489	443	419	283	1.190	791
Selenium	0.52	0.63	69.0	0.39	0.43	1.7	0.54	0.40
Silver	41.9	14.3	2.2	23.4	40.3	20.4	2.9	10.7
Sodium	144	156	219	141	161	292	263	437
Thallium	0.36		U 0.3	U 1.5 U	1.6 U	1.8 U	0.45 U	11 65:0
Vanadium	17.8	14.1	15.1	14.3	14.5	12.7		
Zinc	2,760	5,760	1,530	5,330	5,730	5,950	2,490	4,780

TABLE C.15b (Cont.)

			Soil Con	Soil Concentrations (mg/kg) at Various Depth Intervals	g) at Various De	pth Intervals		
Parameter	CLP1 (0-6 in.)	CLP1 (6-24 in.)	CLP1 (24-40 in.)	CLP2 (0-6 in.)	CLP2 (6-24 in.)	CLP2 (24-48 in.)	CLP3 (0-6 in.)	CLP3 (6-24 in.)
Other Parameters Cyanide (total)	0.1	<u></u>	16	-	, ,	C		
Solids (%)	66.4	7.7.7	79.6	78.6	5.0 77	4.0 66	55.9	11 66.2
			Soil Conc	Soil Concentrations (mg/kg) at Various Depth Intervals	z) at Various De	oth Intervals		
	CLP3-Dup	CLP3	CLP4	CLP4	CLP4	CLP5	CLP5	CLP5
Parameter	(6-24 in.)	(24-48 in.)	(0-6 in.)	(6-24 in.)	(24-48 in.)	(0-6 in.)	(6-24 in.)	(24-48 in.)
Metals								
Aluminum	10,900	9,280	9,310	13,900	10,300	8,070	8,270	10.800
Antimony	76.8	14.6	29.8	24.8	13.4	5.3 U	4.7 U	5.5 U
Arsenic	20.5	21.1	18.3	16	4.6	12.5	5.7	
Barium	1,450	503	820	803	143	294	267	
Beryllium	0.44	0.78	0.56	0.58	1.3	0.39	0.41	0.57
Cadmium	9.5	5.8	5.5	5.0	2.2	2.4	2.2	0.71
Calcium	5,340	4,530	2,050	3,430	4,430	2,720	15,700	2,820
Chromium	146	47.9	59.6	73.5	18.6	73.5	33.7	17
Cobalt	0.6	5.3	5.3	4.7	6.1	4.2	4.8	6.4
Copper	1,120	320	809	552	72.8	092	252	22.6
Iron	83,900	24,200	22,100	27,200	15,800	14,600	13,600	13,900
Lead	11,100	2,800	4,580	5,040	371	791	458	21.3
Magnesium	3,290	3,980	2,530	3,010	2,430	1,440	2,420	1,690
Manganese	593	151	981	201	107	168	340	191
Mercury		69.0	1.2	1.5	0.24 U	0.48	0.13	0.23

TABLE C.15b (Cont.)

CLP4   CLP5   CLP5				Soil Con	Soil Concentrations (mg/kg) at Various Depth Intervals	) at Various Dept	h Intervals	, and the second	
kel         45         188         23.8         26.2         13.3         11.2         13         U           ensium         675         1,190         617         662         952         506         13         U         13         U         13         U         662         952         506         13         U         0.38         U         0.38         U         0.42         0.38         0.38         0.38         0.38         0.38         0.38         0.38         0.38         0.39         0.38         0.39         0.39         0.38         0.39 <t< th=""><th>Parameter</th><th>CLP3-Dup (6-24 in.)</th><th>CLP3 (24-48 in.)</th><th>CLP4 (0-6 in.)</th><th>CLP4 (6-24 in.)</th><th>CLP4 (24-48 in.)</th><th>CLP5 (0-6 in.)</th><th>CLP5 (6-24 in.)</th><th>CLP5 (24-48 in.)</th></t<>	Parameter	CLP3-Dup (6-24 in.)	CLP3 (24-48 in.)	CLP4 (0-6 in.)	CLP4 (6-24 in.)	CLP4 (24-48 in.)	CLP5 (0-6 in.)	CLP5 (6-24 in.)	CLP5 (24-48 in.)
assium 675 1,190 617 662 952 506 339  rest 12.5 3.1 4.6 3.5 10 0.33 U 0.66 U 0.42 0.38  rest 12.5 3.1 4.6 3.5 3.1 3.7 1.3  rium 52.1 1,400 189 284 537 88.7 158  lilium 1.9 U 0.85 U 2.0 U 1.9 U 0.76 U 0.31 U 0.29 U addium  15.7 13.6 14.8 1.5 15.2 13.8 13.6  cadium 15.7 13.6 14.8 1.5 15.2 13.8 13.6  cadium 15.7 13.6 14.8 1.1 14.0 0.3 U 0.20 U 0.8 U 0.7 U 0.8 U 0.8 U 0.7 U 0.8 U 0.8 U 0.7 U 0.8 U 0.8 U 0.7 U 0.8 U	Nickel	45	18.8	23.8	26.2	13.3	12.2		10.2
rer 12.5 3.1 4.6 3.5 10 0.66 U 0.42 0.38  riam 52.1 1.400 189 284 537 88.7 1.3  Illium 1.9 U 0.85 U 2.0 U 1.9 U 0.76 U 0.31 U 0.29 U adium 15.7 13.6 14.8 15.2 13.8 13.6  Sadium 1.5.7 13.6 14.8 15.2 13.8 13.6 13.8 13.6  Satismeter 1.5.7 13.6 14.8 15.2 13.8 13.6 13.8 13.6  Satismeter 1.5.8	Potassium	675	1,190	617	662	952	506		557
ert         12.5         3.1         4.6         3.5         3.1         3.7         1.3           ium         52.1         1,400         189         284         537         88.7         1.8           llium         1.9         0.85         0         2.0         0         1.9         0         0.76         0         0.31         0         0.29         0           standium         15.7         13.6         14.8         15         15.2         13.8         13.6         13.6         13.8         13.6         13.6         13.8         13.6         13.8         13.6         13.6         13.8         13.6         13.6         13.6         13.8         13.6	Selenium	0.36					0.42	38	755
ium         521         1,400         189         284         537         88.7         158           adium         1.9         0.085         U         2.0         U         1.9         U         0.76         U         0.31         U         0.29         U           adium         15.7         13.6         1,48         15         15.2         13.8         13.6         13.6         13.8         13.6         13.8         13.6         13.6         14.8         15         15.2         13.8         13.6	Silver	12.5	3.1	4.6	3.5		3.7	1.3	6.0
llium 1.9 U 0.85 U 2.0 U 1.9 U 0.76 U 0.31 U 0.29 U adium 15.7 13.6 14.8 15 15.2 13.8 13.6    2 parameter a control of the con	Sodium	521	1,400	189	284	537	88.7	158	125
adium 15.7 13.6 14.8 15 15.2 13.8 13.6  Parameter  inide (mg/kg) 120 16 0.8 U 1.0 2.0 U 0.8 U 0.7 U  ds (%) 34.4 67.3 68.8 74.8 38.3 78.6 83.8  CLP5-Dup CLP6 CLP6 CLP7 CLP7 CLP7  inum 8.820 7,990 8,640 17,000 7,030 8,930 5,080 6  inc 2.2 39.8 22.9 12.3 4.5 6.2 U 5.0 15.6 in.)  inum 62.1 756 828 157 87.7 245 11. 21  inum 0.62 0.54 0.42 0.98 0.55 0.67 11 2.1  inum 0.78 2.5 6.2 1.4 0.99 0.57 11 2.1  inum 0.78 2.5 6.2 1.4 0.99 0.57 11 2.1  inum 0.78 2.5 6.2 1.4 0.99 0.57 11 2.1  inum 0.78 2.5 6.2 1.4 0.99 0.57 11 2.1  inum 0.78 2.5 6.2 1.4 0.99 0.55 0.55 0.45 in.)  inum 0.78 2.5 6.2 1.4 0.99 0.57 11 2.1  inum 0.78 2.5 6.2 1.4 0.99 0.57 11 2.1  inum 0.78 2.5 6.2 1.4 0.99 0.57 11 2.1  inum 0.78 2.5 6.2 1.4 0.99 0.55 0.55 0.45 in.)	Thallium								11 650
Farameter nide (mg/kg) 120 16 0.8 U 1.0 2.0 U 0.8 U 0.7 U 68 (%) 34.4 67.3 68.8 U 1.0 2.0 U 0.8 U 0.7 U 68 (%) 34.4 67.3 68.8 T4.8 38.3 78.6 83.8 T4.8 38.3 78.6 83.8 T4.8 in.) CLP5-Dup CLP6 CLP7 CLP7 CLP7 CLP7 CLP7 CLP8 in.) (0-6 in.) (	Vanadium	15.7	13.6	14.8	15				
Parameter         nide (mg/kg)         120         16         0.8         U         1.0         2.0         U         0.8         U         0.7         U           ds (%)         34.4         67.3         68.8         74.8         38.3         78.6         83.8           Soil Concentrations (mg/kg)         34.4         67.3         68.8         74.8         38.3         78.6         83.8           Soil Concentrations (mg/kg) at Various Depth Intervals           Soil CLP6         CLP7         CLP7         CLP8         CLP8           CLP6-In.)         (6-24 in.)         (6	Zinc	06969	2,960	2,880	2,940	1,400	732	845	114
Soil Concentrations (mg/kg) at Various Depth Intervals   Soil Concentrations (mg/kg) at Various Depth Intervals     CLP5-Dup   CLP6   CLP6   CLP7   CLP7   CLP8	Other Parameter Cyanide (mg/kg)	120	16		1.0				80
CLP5-Dup         CLP6         CLP7         CLP7         CLP8           arameter         (24-48 in.)         (0-6 in.)         (6-24 in.)         (0-6 in.)         (6-24 in.)         (6-24 in.)         CLP7         CLP7         CLP8           sinum         8.820         7,990         8,640         17,000         7,030         8,930         5,080         6           nony         5.0         U         20.2         38.6         6.3         U         5.2         U         501           iic         2.2         39.8         22.9         12.3         4.5         6.2         41           m         62.1         756         828         157         87.7         245         1,560           ilium         0.62         0.54         0.98         0.55         0.65         0.45           ilum         0.78         2.5         0.45         0.45         1.7         2.1	Solids (%)	34.4	67.3	68.8	74.8	38.3			
CLP5-Dup         CLP6         CLP6         CLP7         CLP7         CLP7         CLP7         CLP8           ninum         8,820         7,990         8,640         17,000         7,030         8,930         5,080         6           nic         2.2         38.6         6.3         U         5.3         U         5.2         U         501           nic         2.2         39.8         22.9         12.3         4.5         6.2         41           n         62.1         756         82.8         157         87.7         245         1,560           lium         0.62         0.54         0.42         0.98         0.55         0.45         0.45				Soil Conc	entrations (mg/kg)	at Various Depth	Intervals		
trameter         (24-48 in.)         (0-6 in.)         (6-24 in.)         (6-24 in.)         (6-24 in.)         (6-24 in.)         (6-24 in.)         (6-24 in.)         (6-24 in.)         (6-6 in.)         (6-7 in.)         (6-7 in.)         (6-7 in.)         (6-7 in.)         (6-7 in.)         (6-7 in.)         (6-7 in.)         (6-7 in.)         (6-7 in.)	,	CLP5-Dup	CLP6	CLP6	CLP7	CLP7	CLP7	CLP8	CT.P9
innum         8,820         7,990         8,640         17,000         7,030         8,930         5,080           nony         5.0         U         20.2         38.6         6.3         U         5.3         U         5.0         U         501           nic         2.2         39.8         22.9         12.3         4.5         6.2         41           m         62.1         756         828         157         87.7         245         1,560           lium         0.62         0.54         0.42         0.98         0.65         0.45           nim         0.78         2.5         6.2         1.4         0.9         0.67         11         21	Parameter	(24-48 in.)	(0-6 in.)	(6-24 in.)	(0-6 in.)	(6-24 in.)	(24-48 in.)	(0-6 in.)	(0-6 in.)
n         8,820         7,990         8,640         17,000         7,030         8,930         5,080           7.0         5.0         U         20.2         38.6         6.3         U         5.3         U         5.2         U         501           2.2         39.8         22.9         12.3         4.5         6.2         41           62.1         756         828         157         87.7         245         1,560           0.62         0.54         0.42         0.98         0.55         0.52         0.45           0.78         2.5         6.2         1.4         0.9         0.67         11         21	Metals								
5.0 U 20.2 38.6 6.3 U 5.3 U 5.2 U 501 2.2 39.8 22.9 12.3 4.5 6.2 41 62.1 756 828 157 87.7 245 1,560 0.62 0.54 0.42 0.98 0.55 0.52 0.45 0.78 2.5 6.2 1.4 0.9 0.67 11 2.1	Aluminum	8,820	7,990	8,640	17,000	7,030	8.930	5.080	6.780
2.2     39.8     22.9     12.3     4.5     6.2     41       62.1     756     828     157     87.7     245     1,560       0.62     0.54     0.42     0.98     0.55     0.52     0.45       0.78     2.5     6.2     1.4     0.9     0.67     11     21	Antimony		20.2	38.6				501	19.5
62.1     756     828     157     87.7     245     1,560       0.62     0.54     0.42     0.98     0.55     0.52     0.45       0.78     2.5     6.2     1.4     0.9     0.67     11     2.1	Arsenic	2.2	39.8	22.9	12.3	4.5		. <del>.</del> 41	17.1
0.62 0.54 0.42 0.98 0.55 0.52 0.45 0.78 2.5 6.2 1.4 0.9 0.67 II 2.1	Barium	62.1	756	828	157	87.7	245	1.560	498
0.78 2.5 6.2 1.4 0.9 0.67 11 2.1	Beryllium .	0.62	0.54	0.42	0.98	0.55	0.52	0.45	0.31
7.7	Cadmium	0.78	2.5	6.2	1.4	6.0	0.67 U	2.1	2,5

TABLE C.15b (Cont.)

C	CLP5-Dup	CLP6	CLP6	CLP7	CLP7		CLP7	CLP8	CLP9
rarameter	(24-48 in.)	(0-6 in.)	(6-24 in.)	(0-6 in.)	(6-24 in.)		(24-48 in.)	(0-6 in.)	(0-6 in.)
Calcium	2,140	3,520	2,800	3,950	4,080	1	11,100	386	1.090
Chromium	10.8	65.2	79.3	28.6	11.8		14.4	118	82.5
Cobalt	9.9	6.1	0.6	9.1	4.2		4.8	5.6	3.8
Copper	15.3	439	559	61.5	28.6		26.4	1,040	449
Iron	9,830	28,600	006,09	22,700	8,370	_	11,100	90,800	16,500
Lead	23.1	2,440	5,340	163	135		107	94,200	2,050
Magnesium	1,500	2,170	2,350	2,560	1,090		1,400	765	3,880
Manganese	2.66	316	393	340	172		216	134	156
Mercury	0.13	0.79	0.46	0.2	0.11	Ω	0.13 U	9.0	0.22
Nickel	9.4	20.4	38.2	17.2	8.4		9.5	24.8	17.9
Potassium	417	507	909	870	384		505	1,130	308
Selenium	0.44	0.43	0.62	0.42	0.43		0.36	2.8	0.5
Silver	O.6 U	5.1	6.7	0.75 U		Ω	0.62 U	12.7	2.0
Sodium	102	156	250	101	81.9		133	504	78.3
Thallium	0.33 U	0.32 U	0.32 U	0.38 U	0.31	Ŋ	0.33 U	3.8 U	0.29
Vanadium	13.3	16.9	20.1	32.7	12.2		16.5	20.2	11.7
Zinc	87.9	3,720	4,450	176	92.2		103	1,420	1,250
Other Parameters									
Cyanide (total)	0.8 U	0.8 U	0.8 U	U 6:0	0.8	n	0.8 U	5.0	0.7
Solids (%)	75.3	( L	(	,					

<sup>a</sup> Sampled by ANL, analyzed by Weston Gulf Coast, CLP TAL metals and other inorganic parameters.

Data qualifier:

 $\mathbf{U}=\mathbf{a}$ nalyte was analyzed for but not detected; detection limit given.

TABLE C.15c Inorganic Analyses for Soil Borings, 1994-1995<sup>a</sup>

			Soil Concentrat	Soil Concentrations (mg/kg) at Various Depth Intervals	is Depth Intervals			
Parameter	TBSPBOR1 (0-2 ft)	TBSPBOR1 Dup (0-2 ft)	TBSPBOR1 (2-4 ft)	TBSPBOR1 (4-6 ft)	TBSPBOR1 (6-8 ft)	TBSPBOR1 (8-10 ft)	TBSPBOR1	OR1
Metals								
Aluminum	10,900	9,940	13,300	13,300	13.100	11 400	0030	
Antimony	6.2 N	6.0	2.9 UN	V. 9.4	N11 8 C	M11 0 C	0,520	, 41
Arsenic	28.2	20.2					8.7	Z O
Barium	212	283	124	147	143	63.5	3.0	
Beryllium	0.43	0.38	0.54	0.5	0.52	0.00	42.3	
Cadmium	2.1	3.0	3.2	1.0	1.5	0.79	0.52	
Calcium	2,130	3,900	1,580	2,250	1,790	2.000	1,650	
Chromium	34.3	53.8	24.2	25	20.8	19.7	0.0,1	
Cobalt	7.0	5.6	6.3	6.0	6.0	2.61	14.3	
Copper	277	366	151	71.1	363	0.0	9.9	
Iron	52,000	26,500	20,300	20.400	21.800	24.0	10.3	
Lead	@ 0 <u>5</u> 9	831	204 @	116 @	38.8			(
Magnesium	1,580	1,720	1,700			Jo.0 @		<b>B</b> )
Manganese	219	172	9.88	116	91.3	78.7	1,490	
Mercury	1.3	1.8	0.38	0.1 U	0.13 U	0.1		Ĺ
Nickel	24.4	19	14.2	13				<b>-</b>
Potassium	531	473	530	603	089	716	12.3	
Selenium	1.7	1.4	1.1	0.23 W	0.25 W	71.0		,
Silver	1.6	2.5	1.0					<b>×</b>
Sodium	179	186	136	91.9	+:7	0.03		<b>-</b>
Thallium	0.26 UW	0.25 U	0.27 UW	0. 26 UW	0.27 UW	55.4 MIW	35.8	11117
								<u>*</u>

TABLE C.15c (Cont.)

			Soil Concentrat	Soil Concentrations (mg/kg) at Various Depth Intervals	ıs Depth Intervals		
Parameter	TBSPBOR1 (0-2 ft)	TBSPBOR1 Dup (0-2 ft)	TBSPBOR1 (2-4 ft)	TBSPBOR1 (4-6 ft)	TBSPBOR1 (6-8 ft)	TBSPBOR1 (8-10 ft)	TBSPBOR1 (10-12 ft)
Metals (Cont.) Vanadium Zinc	17.8 757 E	17.6	20.6 450 E	25.5 192 E	26.2 79.6 E	21.7 61.3 E	14.5 38.1 E
Other parameter Cyanide (total)	1.9 U	1.9 U	2.0 U	U. 7.1	U.9 U	U. 7.1	1.8 U
		Soil C	oncentrations (mg/kg	Soil Concentrations (mg/kg) at Various Depth Intervals	ervals		
Parameter	TBSPBOR2 (0-2 ft)	TBSPBOR2 Dup (0-2 ft)	TBSPBOR2 (2-4 ft)	TBSPBOR2 (4-6 ft)	TBSPBOR2 (6-8 ft)	TBSPBOR2 (8-10 ft)	
Metals							
Aluminum	11,300	9,160	9,900	5,270	6,380	9.720	
Antimony	2.8 UN	2.7 U	2.9 UN	2.8 UN	2.8 UN	2.8 UN	
Arsenic	5.4	4.5	4.9				
Barium	53.1	40.5	39.3	24.6	36.5	59.3	
Beryllium	0.27	0.28	0.49	0.23	3.0	0.37	
Cadmium	0.95	0.7	0.51 U	0.49 U	0.5 U	0.5 U	
Calcium	934	937	887	716	876	1290	
Chromium	19.4	16.5	16.4	8.9	11.2	14.5	
Cobalt	0.9	5.2	5.2	2.6	4.6	6.4	
Copper	72.4	55.7	19.4	8.9	7.2	=	
Iron	16,200	15,200	19,700	7.870	13,400	11.900	
Lead	28.2 (@	22.5	9.6	11 @	11.8	16.2 @	

TABLE C.15c (Cont.)

Parameter	TBSPBOR2 (0-2 ft)	TBSPBOR2 Dup (0-2 ft)	TBSPBOR2 (2-4 ft)	TBSPBOR2 (4-6 ft)	TBSPBOR2 (6-8 ft)	TBSPBOR2 (8-10 ft)	1
Metals (Cont.)							
Magnesium	2,130	1,870	1,660	778	1 100	0	
Manganese	99.3	90.1	6.99	27	67.5	1,810	
Mercury	0.09 U	0.1 U	0.1	0 11			
Nickel	13	10.8			0.1	0.11 U	
Potassium	748	561	909	524	6.7	12.6	
Selenium	0.22 W	0.24	0.16 UW	0.16	73. CO		
Silver	0.38 U	0.36 U					
Sodium	141	123	303			0.38	
Thallium	0.25 UW	0.26 U	0.26 UW	0.26 ITW	0.05 TW		
Vanadium	21	19.5	20.8			0.27 UW	
Zinc	97 E	77.4	48.7 E	17.8 E	23 E	11.9 44.5 E	
Other parameter							
Cyanide (total)	2.0 U	1.9 U	1.8 U	2.1 U	1.8 U	2.1 U	
			Soil Concentration	Soil Concentrations (mg/kg) at Various Depth Intervals	Depth Intervals		
Parameter	HBOR1 (0-2 ft)	HBOR1 (2-4 ft)	HBORI (4-6 ft)	HBOR1 (6-8 ft)	HBOR1 (8-10 ft)	HBOR2	HBOR2
Metals							(11 4-7)
Aluminum	11,600	8,470	6,920	4.880	3 530	0.100	
Antimony	21.5 @N	5.7 U@N	5.7 U@N	4.5 U@N	5.3 U@N	5.5 U@N	8,440 5.5 U@N
Sellic	13.7	4.6	2.7	1.8	1.7	8.3	

TABLE C.15c (Cont.)

					Solic	Olicentrat	Son Concenuations (mg/kg) at various Depth Intervals	) at various	s Depth Inte	ervals				
Parameter	HBOR1 (0-2 ft)	OR1	HB(	HBOR1 (2-4 ft)	HB (4-	HBOR1 (4-6 ft)	9) 	HBOR1 (6-8 ft)	HB (8-	HBOR1 (8-10 ft)	HB(	HBOR2 (0-2 ft)	HE (2	HBOR2 (2-4 ft)
Metals (Cont.)													:	
Barium	380	@	41.4	<b>@</b>	40.5	@	23	@	16.1	@	108	@	156	@
Beryllium	0.75		0.45		0.45		0.35		0.33	)	0.59	)	0.57	
Cadmium	1.7		0.57		0.53	Ω	0.42	Ω	0.5	Ω	0.64		0.66	
Calcium	5,320		2,050		1,690	,	917		613	)	2.590		2.510	
Chromium	33.5	@E	13	@E	12.7	@E	11.4	@ E	8.6	@E	18.2	@ T	243	(G)
Cobalt	5.8		1.9		3.0		2.1		1.4		4.9	1		ן ע
Copper	204		16.3		14.4		6.1		4.3		55.2		85.5	
Iron	17,400		10,600		068'6		6,690		5,390		13,800		13.700	
Lead	4,960	@	165	<b>©</b>	110	@	21	@	13.8	<b>@</b>	859	<b>@</b>	256	@
Magnesium	2,810		941		1,380		867		547		1,620		1.460	)
Manganese	213		57.8		61.1		35.4		28.9		177		108	
Mercury	0.12		0.11	Ω	0.12	Ω	0.1		0.11	n	0.14		0.22	
Nickel	16.6		4.2		6.7		3.6		5.1		10		10.1	
Potassium	703		470		453		342		369		511		588	
Selenium	0.29		0.19	n	0.17	Ω	0.16	U	0.16		0.33		0.34	
Silver	0.77		0.53	Ω	0.53	Ω	0.42	D	0.5	Ω	0.86		66 0	
Sodium	127		95.1		65.8		439		40		61.1		65.7	
Thallium	1.2	Ω	0.31	n	0.28	Ω	0.26	Ω	0.23	n	0.28	Ω	0.26	Ξ
Vanadium	20.3		4		12		8.2		7.9		17.2	ı	15.7	)
Zinc	968	@	82	<b>©</b>	58.5	<b>©</b>	23.6	<b>@</b>	18.5	<b>@</b>	287	в	185	@
Other parameter														
Cyanide (total)	1.7	ල	27	(6	7.1	(	•	(	,	;		,		

TABLE C.15c (Cont.)

Parameter	HB (4-	HBOR2 (4-6 ft)	HBOR2 (6-8 ft)	OR2 : ft)	HB(	HBOR2 (8-10 ft)	FTI (8-	FTBOR1 (8-10 ft)	HBOR3 (0-2 ft)	OR3 2 ft)	HBOR3 (2-4 ft)	JR3 ft)	HH O	HBOR4 (0-2 ft)
Metals													,	
Aluminum	8,220		6,970		2,980		4.780		8.440		7.640		0.110	ū
Antimony	5.5	U@N	5.2	U@N	0.9	U@N	6.0	N@N	8.1	(e)	5.3	Z@[]	0.20	INWI
Arsenic	5.4		1.8		2.0		1.8		8.1	!	10.0	; )	3.48	
Вагіит	41.3	@	27.5	<b>®</b>	13.7	<b>©</b>	20	<b>(b)</b>	116	@	137	@	55.1	
Beryllium	0.52		0.37		0.45		0.54		0.76		0.79	)	030	
Cadmium	0.52		0.49	Ω	0.57	D	0.56	Ω	0.53	Ω	0.94		0.00	1
Calcium	3,000		1,290		532		488		3.400	)	2,340		3.540	)
Chromium	15	@E	12.8	@E	6.4	@E	8.1	@ E	20.2	@ H	22.2	@ H	10.7	
Cobalt	3.5		3.1		1.5		2.3		4.2	ļ	4.3	1	50.4	
Copper	17.2		8.1		3.7		5.1		69.2		06	11@	10.0	
Iron	12,300		9,280		4,880		7,080		14.500		14,000	)	13,600	
Lead	43	@	12.4	<b>@</b>	6.5	(9)	5.4	<b>©</b>	484		419	@	13.1	
Magnesium	1,300		1,380		969		917		1,580		1.460	)	1310	9
Manganese	106		50.1		28.9		57.7		152		142		93.7	
Mercury	0.08	Ω	0.1	Ω	0.1	U	0.13	Ω	0.15		0.21		0.07	
Nickel	8.4		8.2		2.8	Ω	4.3		9.2		8.7		7.5	
Potassium	629		495		295		468		503		444		307	
Selenium	0.19		0.16	Ω	0.19	U	0.2	n	0.28		0.37		0.18	INII
Silver	0.57		0.49	Ω	0.57	n	0.56	n	0.80		1.3		0.08	; 5 =
Sodium	68.3		50.9		36.2		42.6		63.7		65.1		48.6	)
Thallium	0.24	Ω	0.25	Ω	0.32	U	0.33	n	0.29	Ω	0.26		0.32	1
Vanadium	16.2		11.4		5.1		8.0		19.0		14.6		17.4	)
Zinc	57.5	<b>©</b>	29.2	<b>@</b>	15.6	<b>®</b>	25.3	@	402	@	7			

TABLE C.15c (Cont.)

			1	Soil Concer	Soil Concentrations (mg/kg) at Various Depth Intervals	g) at Various	Depth Inter	rvals		
Parameter	HBOR2 (4-6 ft)	HBOR2 (6-8 ft)	2	HBOR2 (8-10 ft)	F1 (8)	FTBOR1 (8-10 ft)	HBOR3 (0-2 ft)	HBOR3 (0-2 ft)	HBOR3 (2-4 ft)	HBOR4 (0-2 ft)
Other parameter Cyanide (total)	1.4 U@	1.4	n @	1.3 U@	0.1.5	U@	1.3	D(@	1.4 U@	1.15 U
		Soil Con	centration	Soil Concentrations (mg/kg) at Various Depth Intervals	rious Depth Int	ervals				·
Parameter	TBNPBOR1 (0-2 ft)	TBNPBOR1 (2-4 ft)	R1	TBNPBOR1 (4-6 ft)		TBNPBOR1 (6-8 ft)	TBNPBOR1 (8-10 ft)	BOR1 0 ft)		
Metals										
Aluminum	22,600	10,300		5,990	4,280		4.710			
Antimony	19.4 @N	15.2	@N	5.4 U@N		N@N	5.6	N@D		
Arsenic	1,440	2,290		247	199		16.7			
Barium	649 @	1,140 (	@	414 @	305	<b>@</b>	104	<b>©</b>		
Beryllium	0.76	0.78		0.59	0.39		0.44			
Cadmium	35.5	12.9		44.1	77.4		2.8			
Calcium	4,890	1,550		440	293		247			
Chromium	240 @E	81.7	@E	18.4 @E		@E	9.7	@ E		
Cobalt	10.8	6.4		2.6	2.9		1.8			
Copper	2,240	7,120		1,120	544		90.7			
Iron	61,900	66,500		26,500	21,200		8,940			
Lead	1,910 @	1,630	<b>@</b>	117 @	90.3	@	48.4	@		
Magnesium	2,890	1.980		759	800		952			
Manganese	633	253		70.3	81.9		40.1			
Mercury	2.2	5.2		0.85	0.34		0.00	n		
Nickel	76.4	51.6		19.1	17.8		7.7			

TABLE C.15c (Cont.)

ł										
Parameter	TBNPBORI (0-2 ft)	3NPBOR1 (0-2 ft)	TBNPBOR1 (2-4 ft)	BOR1 ft)	TBNP	TBNPBOR1 (4-6 ft)	TBNF (6-	TBNPBOR1 (6-8 ft)	TBNPBOR1 (8-10 ft)	BOR1 0 ft)
Metals (Cont.)										
Potassium	802		1,050		685		478		508	
	0.61		0.17	Ω	0.31		0.2	n	0.18	Π
	24.8		14.8		4.4		2.9		0.53	) <u> </u>
	254		153		46.2		41.3		32.3	<b>;</b>
	7.9		3.4		1.6	Ω	0.32	Ω	0.3	11
Vanadium	16.6		6.0		6.9		3.1		5.8	)
	9,840	<b>©</b>	6,530	(9)	4,360	<b>©</b>	3,650	@	1.150	@
Other parameter										)
Cyanide (total)	1.4	U@	1.4	1.4 U@	1.5	1.5 U@	1.4	011	-	011

<sup>a</sup> Sampled by ANL, analyzed by Weston Gulf Coast or ANL/ACL, TCL TAL metals and other inorganic parameters. Data qualifiers:

N = spike and spike duplicate recovery of percent RPD were outside the control limits.

@ = percent RPD of sample duplicate was outside the control limits.

U = analyte was analyzed for but not detected; detection limit given.

W = postdigestion spike for furnace AAS analysis was outside the control limits.

E = serial dilution percent difference was outside the QC limit.

TABLE C.15d Inorganic Analysis for Soil Borings, May 1995<sup>a</sup>

TABLE C.15d (Cont.)

	Soil Co	Soil Concentrations (mg/kg) at Various Depth Intervals	(mg/kg) at V	arious Do	epth Inter	vals				
Parameter	TBNPBOR2 (0-2 ft)	TBNPBOR2 (2-6 ft)	OR2 ft)	TBNPBOR2 (6-8 ft)	3OR2 ft)	TBNPBOR2 (10-12 ft)				
Metals (Cont.) Vanadium	16.3	12		8.69		4.33	ı			
Zinc	2,690 EN@	46.1		149		198				
Other Parameters										
Cyanide (total)	1.21 UNJ	1.16	Ω	1.22	U	1.23 U				
Hq	7.46	NA		5.66						
Solids (%)	82.6	85.8		81.7		81.2				
ı		Sc	oil Concentra	tions (mg	g/kg) at V	Soil Concentrations (mg/kg) at Various Depth Intervals	rvals	 		
Parameter	TBNPBOR3 (0-2 ft)	TBNPBOR3 (2-4 ft)	DR3	TBNPBOR3 (4-6 ft)	OR3 ft)	TBNPBOR3 (6-8 ft)	TBNPBOR3 (8-10 ft)	TBNPBOR3 (10-12 ft)	3OR3 2 ft)	
Metals										
Aluminum	4,430	4,190	<b>-</b>	5,420		6.710	5 160	2 570		
Antimony	20.4	23.7	N	3.44	Z	0.202 U	0.212	0.202	IIVII	
Arsenic	278	63.5		316	@ <u>J</u>	7.63		10.5	@ I	
Barium	237	246	(@)	1,240		61.8	30.6	38.2	<b>.</b> )	
Beryllium	0.132 U	0.149	U	0.204	J	0.169	0.253	0.14	<b>—</b> ,	
Cadmium	5.13	3.55	@ J	5.71	N@J	0.583	0.698	0.70	N@J	
Calcium	2,080	946	@E	811	田	193	164	168	<u>н</u>	

TABLE C.15d (Cont.)

	TBNPBOR3	BOR3	TBNPBOR3	30R3	TBNPBOR3	30R3	TBNPBOR3	OR3	TBNPBOR3	OR3	TBNPBOR3	BOR3
Parameter	(0-2 ft)	2 ft)	(2-4 ft)	ft)	(4-6 ft)	ft)	(6-8 ft)	ft)	(8-10 ft)	ft)	(10-12 ft)	2 ft)
Metals (Cont.)												
Chromium	72.1		37.2	(@)	23.6	N@J	9.23		8.79		5.24	N@J
Cobalt	12.7		9.48		7.33		3.98		2.99		2.75	ŗ
Copper	814		916	@)	191		15.3		7.98		11.9	
Iron	154,000		120,000	ш	55,400	(@)	9,840		13,100		4,660	@ ]
Lead	3,230		459	(@)	92.7		18.6		69.9		22.7	
Magnesium	1,090		720		1,850	ш	1,470		1180		636	田
Manganese	470		480	(@)	186	EN@	43.3		51		32.3	EN@
Mercury	0.931		0.074	ŗ	0.069	J	0.044		0.062	Ω	0.053	Ω
Nickel	84.5		51.9		8.89	N@J	8.03		7.12		6.13	N@J
Potassium	366		711		<i>L</i> 99	<del>, ,</del>	781		650		288	<b>-</b>
Selenium	1.64		1.77		0.424	WJ	0.19	Ω	0.2	Ω	0.197	U
Silver	4.54		0.147		0.099	Ω	0.083	Ω	0.087	Ω	980.0	U
Sodium	83		125		113	ſ	40.9		48.6		47.9	-
Thallium	0.335	n	0.378	D	0.395	n	0.333	n	0.35	n	0.344	Ω
Vanadium	4.29		89.9		8.09		8.23		10.2		3.25	ŗ
Zinc	13,900		13,800	@NE	3,880		283		113		205	
Other Parameters												
Cyanide (total)	1.2	U	40.8		11.1		1.19	Ω	1.25	n	1.23	D
	6.81		6.22		6.22		7.05		7.09		6.5	
Solids (%)	3 60		i									

TABLE C.15d (Cont.)

Parameter							
	VXBOR1 (0-2 ft)	VXBOR1 (4-6 ft)	VXBOR1 (6-8 ft)	VXBOR2 (0-2 ft)	VXBOR2	VXBOR2	VXBOR2
Metals						(at o t)	(10-011)
Aluminum	NA	NA	Z Y	Ϋ́	Ą	Š	
Antimony	NA	NA	NA	NA	X X	C Z	¥ Z
Arsenic	13 U	10 U	10 U	15	10 11	10 11	11 01
Barium	224	62	30	166	51	39	0 OI 37
Beryllium	NA	NA	NA	NA	X A	Ϋ́ AZ	) Z
Cadmium	10 U	10 U	10 U	10 U	10 U	10 11	10 11
Calcium	NA	NA	NA	NA			
Chromium	37	42	13	41	25	17	13
Cobalt	NA	NA	NA	NA	NA N	. V	CI Z
Copper	170	192	10 U	124	10 U	40	10 11
Iron	NA	NA	NA	NA		î V	
Lead	210	. 49	10 U	161	: =	10 11	10 11
Magnesium	NA	NA	NA	N A	Y A		
Manganese	NA	NA	NA	, AN	Y N	C Z	¥ ×
Mercury	10 U	10 U	10 U	10 U	10 11	10 11	10 11
Nickel	11	11	10 U	23			
Potassium	NA	NA	NA	NA	NA		O Y
Selenium	10 U	10 U	10 U		10 U	10 U	10 11
Silver	10 U	10 U	10 U	10 U		10 U	10 11
Sodium	NA	NA	NA	NA	Ϋ́		
Thallium	NA	NA	NA	ĄN	A Z	VΔ	NA
Vanadium .	24	27	29	39	41	36	35
Zinc	432	7.1	19	361	17	3. 5	9 6

TABLE C.15d (Cont.)

•		Sc	Soil Concentrations (mg/kg) at Various Depth Intervals	/kg) at Various De	pth Intervals		
Parameter	VXBOR1 (0-2 ft)	VXBOR1 (4-6 ft)	VXBOR1 (6-8 ft)	VXBOR2 (0-2 ft)	VXBOR2 (2-4 ft)	VXBOR2 (4-6 ft)	VXBOR2 (6-8 ft)
Other Parameters							
Cyanide (total)	NA	NA	NA	NA	NA	NA	Z
Hq	NA	NA	NA	NA	ĄZ	Y X	i ∀ Z
Solids (%)	NA	NA	NA	NA	NA	NA	NA
ţ	Soil Concentratio	Soil Concentrations (mg/kg) at Various Depth Intervals	18 Depth Intervals				
į	VXBOR2	VXBOR2	VXBOR2-Dup				
Parameter	(8-10 ft)	(10-12 ft)	(10-12 ft)				
Metals							
Aluminum	NA	NA	NA				
Antimony	NA	NA	NA				
Arsenic	10 U	10 U	10 U				
Barium	38	45	42				
Beryllium	NA	NA	NA				
Cadmium	10 U	10 U	10 U				
Calcium	NA	NA	NA				
Chromium	10 U	17	13				
Cobalt	NA	NA	NA				
Copper	10 U	10 U	10 U				
Iron	NA	NA	NA				
Lead	10 U	10 U	10 U				
Magnesium	NA	NA AN	NA				
Manganese	NA AN	Z Z	V.				

TABLE C.15d (Cont.)

	Soil Concentrat	Soil Concentrations (mg/kg) at Various Depth Intervals	us Depth Intervals	ı		
Parameter	VXBOR2 (8-10 ft)	VXBOR2 (10-12 ft)	VXBOR2-Dup (10-12 ft)	!		
Metals (Cont.)				1		
Mercury	10 U	10 U	10 U			
Nickel	10 U	10 U	10 U			
Potassium	NA	NA	NA			
Selenium	10 U	10 U	10 U			
Silver	10 U	10 U				
Sodium	NA	NA				
Thallium	NA	NA	NA			
Vanadium	21	25	19			
Zinc	42	63	55			
Other Parameters						
Cyanide (total)	NA	NA	NA			
Hd	NA	NA	NA			
Solids (%)	NA	NA	NA			
I		Soil Conce	Soil Concentrations (mg/kg) at Various Depth Intervals	arious Depth Int	ervals	
Parameter	VXBOR4 (2-4 ft)	VXBOR4 (4-6 ft)	VXBOR5 (0-2 ft)	VXBOR5 (4-6 ft)	VXBOR5 (6-8 ft)	VXBOR5 (8-10 ft)
Metals Aluminum Antimony Arsenic	22,200 E 112 NJ 18.9	3,360 0.691 0.905	6,890 0.206 UNJ 6.09	6,950 0.485 4.54	1,810 0.209 U 0.568	4,420 0.211 UNJ 1.94

TABLE C.15d (Cont.)

Parameter	VXBOR4 (2-4 ft)	OR4 ft)	VXBOR4 (4-6 ft)	OR4 ft)	VXBOR5 (0-2 ft)	OR5 ft)	VXBOR5 (4-6 ft)	S	VXBOR5 (6-8 ft)	R5 ft)	VXB	VXBOR5 (8-10 ft)
(240]2 ((277)												
Metals (Cont.) Perium	c	⊢ ⊢	ŗ		6	(						
Barium	2.93	)	17		235	(@)	1,410		7.75		17.6	(@)
Beryllium	0.14	Ω	0.135	n	0.195		0.143	n	0.135	Ω	0.217	
Cadmium	2.65		0.49	Ω	1.72	@ J	3.33		0.368	Ω	0.372	U@J
Calcium	14,800		377		648	E@	1,680		175		373	$E_{@}$
Chromium	37.9		6.28		15.3	@ <u>J</u>	43.3		1.76		9.22	@ J
Cobalt	3.99		2.17		5.2		5.32		3.01		3.86	
Copper	1,110		33.7		51.4	(@)	218		2.82	n	3.78	@ <u>J</u>
Iron	16,900	Щ	4,930		6,600	Щ	13,800		3,040		10,200	П
Lead	1,690	@ <u>J</u>	13.7		27.5	(@)	26.3		2.17		2.77	@ ]
Magnesium	2,460		838		2,480		11,100		504		925	
Manganese	223		36.1		61.4	@ J	146		21.5		37.6	@
Mercury	0.426		0.021	Ω	0.34		0.263		0.052	Ω	0.054	Ω
Nickel	55.2		3.89		6.13		16		3.23		4.3	
Potassium	611		332		385		372		152		476	
Selenium	0.34	Ź	0.196	Ω	0.29		0.208	Ω	0.196	n	0.198	Ω
Silver	2.51		1.02		0.085	UWJ	0.11		0.086	n	0.087	Ω
Sodium	102		20.8	Ω	41.6		55		28		42.7	
Thallium	0.356	Ω	0.343	Ω	0.338	Ω	0.364	Ω	0.344	D	0.347	Ω
Vanadium	28.8		5.02		9.74		8.83		3.56	n	7.61	
Zinc	2,320		37.8		223	EN@	342		11.9		17 3	EN BN

TABLE C.15d (Cont.)

		Soil Con	Soil Concentrations (mg/kg) at Various Depth Intervals	Various Depth Int	ervals	
Parameter	VXBOR4 (2-4 ft)	VXBOR4 (4-6 ft)	VXBOR5 (0-2 ft)	VXBOR5 (4-6 ft)	VXBOR5	VXBOR5
					(30.5.5)	(11 01 -0)
Other Parameters						
Cyanide (total)	2.63	1.22 U	1.21 UNJ	1.3 U	1.23	1 24 IINI
hД	7.35	7.54				
Solids (%)	78.6	81.7	82.7	76.8	81.4	80.6
,		6	Soil Concentrations (mg/kg) at 0-6 in.	2/kg) at 0-6 in.		
Parameter	SOPS 1	c sacs	5 2400			
	10170	2 CJAC	3QF3 3	SQPS 4	SQPS 5	SQPS 6
Metals						
Aluminum	4,240	8,600	13,300	13,000	7.890	11.000
Antimony	0.212 U	0.559	1.04	0.598	0.94	0.445
Arsenic	0.224 U	2.96	4.96	3.59	2.71	3.02
Barium	94.9	183	323	148	124	991
Beryllium	1.38	0.419	0.539	0.411	0.298	0.517
Cadmium	1.8	1.43	1.99	2.06	1.23	1.41
Calcium	6,320	1,210	974	1,250	385	847
Chromium	34.4	67.4	85	70.5	92	38.9
Cobalt	11.1	4.45	4.78	4.86	2.19	4.9
Copper	98.3	216	267	215	355	120
Iron	11,200	13,600	21,400	17,000	12,300	12.300
Lead	229	65.8	91.6	95.1	8.69	70.4
Magnesium	2,740	1,540	1,710	1,700	912	1.480
Manganese	81.6	170	262	158	6.66	234

TABLE C.15d (Cont.)

Parameter	SQPS 1	SQPS 2	SQPS 3	SQPS 4	SQPS 5	SQPS 6
Metals (Cont.)						
Mercury	0.086	0.123	0.163	0.166	0.244	0.104
Nickel	3.96	8.77	13.4	8.3	6.05	8.87
Potassium	1,100	396	360	393	298	432
Selenium	0.2 U	0.73	0.438	0.227 U	0.263	0.46
Silver	2.18	3.97	3.15	3.32	4.17	1.92
Sodium	47	71.8	73.5	84	60.1	59.5
Thallium	0.349 U	0.339 U	0.347 U	0.397 U	0.347 U	0.344 U
Vanadium	4.98	14	18.5	22.1	14	18.1
Zinc	78.6	233	603	542	148	353
Other Parameters						
Cyanide (total)	1.25 U	NA	NA	NA	NA	Z
Hd	5.39	NA	NA	NA	5.12	Z
Solids (%)	80.2	82.6	80.6	70.6	81.6	. 81.5
ı	Soil	Soil Concentrations (mg/kg) at 0-6 in.	at 0-6 in.			
Parameter	SQPS 7	SQPS-8	6 SQPS 9			
Metals						
Aluminum	9,620	6,610	13100			
Antimony	0.239	2.72	2.7			
Arsenic	3.77	21.2	60.5			
Darium	( -		:			

TABLE C.15d (Cont.)

at 0-6 in.	SQPS 9		0.309	19.2	1,360	42.8	4.12	758	32,700	289	2,120	372	0.25	18.1	316	0.484	15.1	53	0.386	10.8	5,340		NA	NA	72.5
Soil Concentrations (mg/kg) at 0-6 in.	SQPS-8		0.298	20.7	4,610	42.2	3.82	561	31,800	165	1,580	254	0.242	14	547	0.714	5.56	80.8	0.589 U	12.3	7,970		2.1 U	4.69	45.0
Soil Co	SQPS 7		0.527	1.75	1,330	25.6	4.69	65.2	14,700	63.5	1,140	270	60.0	7.8	361	0.474	1.84	77.1	0.364 U	15.9	420		NA	NA	76.9
·	Parameter	Metals (Cont.)	Beryllium	Cadmium	Calcium	Chromium	Cobalt	Copper	Iron	Lead	Magnesium	Manganese	Mercury	Nickel	Potassium	Selenium	Silver	Sodium	Thallium	Vanadium	Zinc	Other Parameters	Cyanide (total)	pH .	Solids (%)

TABLE C.15d (Cont.)

			Soil C	Concentrati	/gm) suo	Soil Concentrations (mg/kg) at 0-6 in.	The state of the s			
Parameter	TPDGS 1	TPDGS 2	2	TPDGS 3	3	TPDGS 4	TPD	TPDGS 5	TPD	TPDGS 6
Metals										
Aluminum	8,650	11,600		8,490		13,900	11,400	_	15,700	
Antimony	0.921	0.512		0.212	n	0.213 U		n ,	0.209	ב
Arsenic	3.36	4.02		3.17		3.63	3.34		4.18	
Barium	82.3	120		71.4		6.06	75.9		77.4	
Beryllium	0.524	0.539		0.61		0.546	0.51		0.494	
Cadmium	1.1	0.918		0.682		0.926	0.898		1.07	
Calcium	25,300	22,200		4,300		2,410	36,000	_	2,590	
Chromium	12	17.5		9.11		13.5	13.4		15.1	
Cobalt	3.2	3.54		3.78		4.35	5.05		6.05	
Copper	32.9	128		11.7		18	15		10.7	
Iron	11,600	9,130		10,400		20,800	17,300		21,100	
Lead	57.1	53.3		25.2		23.5	11.7		17.6	
Magnesium	1,080	1150		944		1,220	1,880		1,600	
Manganese	253	202		222		201	189		209	
Mercury	0.06 U	0.055	Ω		Ω	0.052	90.0	Ω	90.0	n
Nickel	7.2	16.2		6.46		8.38	7		9.53	
Potassium	367	307		300		394	459		695	
Selenium	0.244	0.185	U	0.292		0.293	0.206		0.341	
Silver	0.085 U	0.081	Ω	0.118		0.088 U	0.085	Ŋ	0.086	Ω
Sodium	151	691		101		101	204		72.8	
Thallium	0.34 U	0.324	U	0.35 U	1	0.35 U	0.341	D	0.344	Ω
Vanadium	13.9	12.2		11.8		18.5	19.4		21.9	
Zinc	94.4	79.3		45.5		89	44.4		49.8	

TABLE C.15d (Cont.)

i		Š	Soil Concentrations (mg/kg) at 0-6 in.	g/kg) at 0-6 in.		
Parameter	TPDGS 1	TPDGS 2	TPDGS 3	TPDGS 4	TPDGS 5	TPDGS 6
Cyanide (total)	1.21 U	1.16 U	Ϋ́	Ϋ́	Ϋ́	V.
	NA	NA	NA	NA	Y AN	K Z
Solids (%)	82.4	86.3	80	79.9	82	81.4

<sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

Data qualifiers:

N = spike and spike duplicate recovery of percent RPD were outside the control limits.

= estimated value.

= percent RPD of sample duplicate was outside the control limits.

= serial dilution percent difference was outside the QC limit.

ш

U = analyte was analyzed for but not detected; detection limit given.

NA = not analyzed.

TABLE C.16a Pesticide and PCB Analyses for Soil Borings, January 1993<sup>a</sup>

Compound Aldrin alpha-BHC	4															
Aldrin alpha-BHC	JBFI-E-2	3-2'	JBP1-E-	P1-E-2'Dup	JBP1-E-4	E-4'	JBP	JBP1-W-2'	JBP1	JBP1-W-4	JBP2-E-2	-E-2'	JBP.	JBP2-E-4'	JBP2-E-4'Dup	-4'Du
alpha-BHC	819	U	80	Ω	∞	Ω	2	Ω	2	n	2	n	2	n	2	Ω
A 11. 1017	617	n	09	Ŋ	9	Ω		D	-	Ω		D	_	N	-	n
Arocior 1016	10,200	n	966	D	102	n	20	Ω	21	Ω	20	n	20	D	20	n
Aroclor 1221	10,200	n	966	Ω	102	n	20	Ω	21	Ŋ	20	U	20	n	20	Ω
Aroclor 1232	10,200	Ω	966	U	102	Ω	20	n	21	D	20	n	20	Ŋ	20	Ω
Aroclor 1242	13,300	Ω	1,300	n	133	Ŋ	27	Ω	27	D	26	n	26	Ω	26	Ω
Aroclor 1248	143,000		14,900		1,180		41	D	42	n	40	D	40	U	40	Þ
Aroclor 1254	20,500	n	1,990	Ω	204	Ω	41	Ω	42	Ω	224		40	U	40	D
Aroclor 1260	20,500	Ω	1,990	Ω	204	n	41	n	42	U	40	n	40	U	40	n
beta-BHC	1,230	Ω	119	Ω	12	n	2	n	7	D	7	Ω	7	n	2	n
Chlordane	2,870	n	279	n	29	n	9	Ω	9	U	9	n	9	U	9	n
4,4'-DDD	2,260	Ω	219	Ω	23	Þ	5	Ω	5	n	4	Ω	4	D	4	Ω
4,4'-DDE	819	Ω	80	U	∞	D	2	D	2	Ω	2	n	7	Ω	2	D
4,4'-DDT	2,460	Ω	239	D	25	Ω	5	U	5	n	5	n	S	U	\$	n
delta-BHC	1,840	D	179	Ω	18	Ω	4	Ω	4	n	4	n	4	D	4	D
Dieldrin	409	Ω	40	Ω	4	D	8.0	Ω	8.0	Ω	8.0	D	8.0	n	0.8	D
Endosulfan I	2,870	Ω	279	D	29	D	9	Ŋ	9	U	9	D	9	n	9	Ω
Endosulfan II	819	Ω	80	Ω	∞	n	7	Ω	2	D	7	n	2	מ	2	n
Endosulfan sulfate	13,500	Ω	1,320	Ω	135	Ω	27	Ω	27	Ŋ	26	D	27	Ω	26	D
Endrin	1,230	Ω	119	U	12	Ω	7	Ω	7	Ω	7	D	7	n	2	n
Endrin aldehyde	4,710	Ω	458	Ω	47	n	6	Ω	10	Ω	6	n	6	n	6	D
gamma-BHC (Lindane)	819	Þ	80	Ω	∞	n	2	Ω	2	Ω	2	n	7	n	2	D
Heptachlor	617	D	09	n	9	n		n	-	Ω		Ω	-	n	_	D
Heptachlor epoxide	17,000	n	1,660	Ω	691	D	34	n	35	n	33	n	33	n	33	D
Methoxychlor	36,000	Ω	3,510	Ω	360	Ω	72	n	73	ם	70	Ω	71	Ω	70	Ω
Toxaphene	49,100	ח	4,780	Ω	491	n	86	D	001	n	96	n	16	Ω	96	D

TABLE C.16a (Cont.)

					Boreh	ole Con	centrat	ions (µg	g/kg) a	Borehole Concentrations (µg/kg) at Various Depth Intervals	s Depth	Interva	Is			
Compound	JBP	JBP2-C-2'	IBI	JBP2-C-4'	JBP2-W	-W-2'	JBP2	JBP2-W-4'	JHD]	JHDP-C-4	JHDP-C-6	-C-6'	JHDP-	JHDP-C-6'Dup	JVXI	JVXP-C-4'
Aldrin	7	Ω	2	Ω	2	Ω	2	n	2	11	C	Ξ	C	11		:
alpha-BHC	1	n		n			-	· 1	۰ –	) <u> </u>	1 -	> =	۰ 4	) ;	7 ,	<b>)</b> ;
Aroclor 1016	23	Ω	20	11	20	) [	, כל	) [	- 6	)	- 6	<b>)</b> ;	<b>→</b> ;	<b>)</b>	-	<b>-</b>
Aroclor 1221	23	) <u> </u>	2 6		2 6	) <u>;</u>	07 6	<b>&gt;</b> ;	07	); 	07.	<b>O</b>	19	Þ	22	n
Aroclor 1232	£ 2	) <u> </u>	7 6		07 6	) :	07	⊃ ;	20	D :	70	n	19	Ω	22	Ω
Aroclor 1242	£ 0€	) <u> </u>	7 2	) <u>;</u>	70	); 	50	<b>)</b> ;	20	)	20	Ω	19	n	22	D
Aroclor 1248	7	) <u>;</u>	07	> ;	97	<b>)</b>	70	)	26	D	27	n	24	D	29	n
A #30[5= 1254	40	⊃ ;	40	)	40	D	40	n	40	n	11		178		44	D
Aroclor 1254	46	D	40	þ	40	D	40	Ω	40	n	41	Ω	38	Ω	4	n
Aroclor 1260	46	n	40	þ	40	Ω	40	Ω	40	Ω	41	Ω	38	Ω	44	11
beta-BHC	3	n	2	Ω	2	n	2	n	7	Ω	7	n	2	D	, co	<u> </u>
Chlordane	9	Ω	9	Ω	9	Ω	9	Ω	9	n	9	Ω	V	-	<i>د</i>	) =
4,4'-DDD	5	D	4	Ω	4	U	4	n	4	Ω	7		, 4	) <u> </u>	) <b>v</b>	) <b>=</b>
4,4'-DDE	2	Ω	7	Ω	7	Ω	7	Ω	7	n	E		. بر	)	, ,	) <u> </u>
4,4'-DDT	9	D	2	Ω	S	D	5	Ω	∞		179		, 9		1 V	) <u> </u>
delta-BHC	4	Ω	4	Ω	4	Ω	4	Ω	4	Ω	4	1	) (f	1	) ~	) <b>=</b>
Dieldrin	6.0	n	0.8	Ω	8.0	Ŋ	8.0	Ω	0.8	) D	80	) <u>=</u>	) «	) <u> </u>	4 0	) <u> </u>
Endosulfan I	9	Þ	9	Ω	9	Ω	9	Ω	9	Ω	9	) D	· ·	) <u> </u>	) Y	) <u> </u>
Endosulfan II	7	D	2	Ω	7	n	2	Ω	7	Ω	2	n	2	) <u> </u>	, ,	) <u>=</u>
Endosulfan sulfate	31	n	26	Ω	27	Ω	26	Ω	27	n	27	Ω	25	'n	<b>5</b>	)
Endrin	3	Ω	7	Ω	7	Ω	2	n	2	D	7	Ω	6	- 11	) (°	) <b>=</b>
Endrin aldehyde	11	Ω	6	Ω	6	Ω	6	Ω	6	Ω	6	D	0	ם	) <u>C</u>	) <u> </u>
gamma-BHC (Lindane)	2	Ω	2	Ω	2	n	2	Ω	2	Ω	7	U	7	Ω	2 6	) <u>=</u>
Heptachlor	1	Ω	-	Ŋ	-	U	-	n		Ω		Ω	-	• =	ı <b>–</b>	) <u> </u>
Heptachlor epoxide	38	Ω	33	Ω	33	Ω	33	Ω	33	U	34	Ω	31	, D	37	) <u> </u>
Methoxychlor	81	n	70	Ω	71	U	70	U	7.1	U	72	Ω	99	n	78	n
Toxaphene	110	n	96	U	26	Ω	96	Ŋ	26	n	86	Ω	06	Ω	106	· =
		i	!							1				1 1 1 1 1 1	7	֧֓֟֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓

TABLE C.16a (Cont.)

Compound	IXXI	WYPCK	15 D dast	7	יאים מתיזו	Ţ	TDDAY A 2"		, Ardai		, , ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	<del>-</del>	, ta ar	ā		
			7700	5	17700		TAT TOTC		7-101 LULC	dnct c-x	JDF IVI-	1-4	JDFIM-D-3	C-Q-	JBFINI-B-1	- <u>p-</u>
Aldrin	2	Ŋ	2	Ω	7	Ω	2	D	2	Ŋ	2	Ω	2	Ŋ	2	Ω
alpha-BHC	-	Ω		n	1	Ω	-	D	_	D	<b></b>	Ω	1	Ω		n
Aroclor 1016	24	U	22	Ω	22	Ω	20	Ω	20	U	20	Ω	19	Ω	20	Ω
Aroclor 1221	24	Ω	22	Ω	22	Ω	20	Ω	20	Ŋ	20	D	19	n	20	D
Aroclor 1232	24	Ω	22	Ω	22	D	20	Ω	20	U	20	D	19	Ω	20	Ω
Aroclor 1242	31	n	29	n	29	n	26	Ω	26	D	26	Ω	25	Ω	26	Ω
Aroclor 1248	47	Ω	45	Ω	44	U	40	Þ	40	n	40	Ω	38	Ω	39	D
Aroclor 1254	47	Ω	45	D	44	Ω	1,700		1,230		3,160		38	Ω	39	D
Aroclor 1260	47	Ω	45	Ω	44	n	40	n	40	n	40	Ω	38	Ω	39	D
beta-BHC	3	n	33	Ω	3	Ω	7	Ω	2	U	2	Ω	2	n	2	
Chlordane	7	D	9	n	9	Ω	9	Ω	9	Ω	9	Ω	5	Ω	5	Ω
4,4'-DDD	S	n	<b>v</b> )	n	Ŋ	n	4	Ω	4	n	4	Ω	4	Þ	4	Þ
4,4'-DDE	2	n	2	n	2	n	2	n	2	U	56		9		2	D
4,4'-DDT	9	Ω	2	n	5	Ω	5	D	5	Ω	144		7		5	D
delta-BHC	4	Ω	4	Ω	4	Ω	4	n	4	U	4	D	4		4	Ω
Dieldrin	6.0	Ω	6.0	Ω	6.0	Ω	8.0	D	0.8	U	0.8	Ω	0.8	Ω	0.8	n
Endosulfan I	7	Ω	9	Ω	9	U	9	Ω	9	Ŋ	9	D	5	D	5	Ω
Endosulfan II	5	Ω	2	n	2	U	7	Ŋ	2	U	2	Ω	2	Ω	2	Ω
Endosulfan sulfate	31	Ω	30	Ω	29	Ω	26	Ω	26	Ω	26	D	25	n	26	$\supset$
Endrin	$\varepsilon$	Ω	3	D	С	Ω	7	Þ	2	D	2	n	2	Ω	2	D
Endrin aldehyde	11	Ω	10	n	10	Ω	6	n	6	Ŋ	6	Ω	6	n	6	D
gamma-BHC (Lindane)	2	Ω	2	n	2	n	2	n	2	Ω	2	Ω	2	Ω	2	D
Heptachlor		Ω	-	Ω	-	n	_	n		n	-	Ŋ	-	Ω		ב
Heptachlor epoxide	39	n	37	n	36	U	33	n	33	n	33	n	32	Ω	32	$\Box$
Methoxychlor	83	Ω	79	n	77	n	70	D	70	D	70	D	19	n	69	$\Box$
Townsham																

TABLE C.16a (Cont.)

Compound	JBPN	JBPM-C-3"	JBPN	JBPM-C-1'	JBPP-A-3"	-A-3"	JBPP	JBPP-A-1'	JBPP	JBPP-B-3"	JBPI	JBPP-B-1'	JBPP-C-3"	-C-3"	JBPP-C-1'	-C-1.
Aldrin	2	Ω	7	Ω	2	n	2	Ω	19	D	2	Ω	C	11	C	=
alpha-BHC	_	Ω	_	Ω	_	Ω	-	D	4	Ω	۰ -	) [	۱	) <u> </u>	۰ -	) <u> </u>
Aroclor 1016	21	Ω	19	Ω	20	n	20	D	231	n	20	> =	20	> <b>=</b>	30	)
Aroclor 1221	21	D	19	Ω	20	n	20	Ω	231	n	20	) <u> </u>	3 5	> <b>=</b>	2 0	) <u> </u>
Aroclor 1232	21	Ω	19	Ω	20	Ω	20	Ω	231	n	30	n	20	) <u>=</u>	2 6	) =
Aroclor 1242	28	Ω	25	Ω	26	Ω	26	U	300	n	26	'n	<u>26</u>	) D	22	) <u> </u>
Aroclor 1248	43	Ω	39	Ω	39	Ω	40	Ω	3,820		40	n	40	'n	5 4	) =
Aroclor 1254	43	Ω	39	Ω	39	Ω	40	Ω	462	Ω	40	n	40	n	9	<u> </u>
Aroclor 1260	43	Ω	39	n	39	n	40	Ω	462	U	40	Ω	40	· D	40	n
beta-BHC	B	Ω	2	Ω	2	n	7	Ω	28	U	2	n	2	n	7	n
Chlordane	9	Ω	5	Ω	S	n	9	Ω	65	D	9	n	9	þ	9	· 12
4,4'-DDD	5	ם	4	Ω	4	Ω	4	Ω	51	n	4	Ω	4	n	, 4	) D
4,4'-DDE	12		2	Ω	2	Ω	7	Ŋ	19	n	42		7	n	. 2	î
4,4'-DDT	5	ב	2	Ω	2	Ω	5	Ω	56	Ω	5	D	5	Ω	S	n
delta-BHC	28		3	Ω	4	Ω	4	Ω	42	Ω	4	n	4		4	n
Dieldrin	6.0	Ω	8.0	Ω	8.0	Ω	8.0	Ω	6	U	8.0	n	0.8	Ω	0.8	ם ב
Endosulfan I	9	Ω	5	Ω	2	Ω	9	D	65	U	9	Ω	9	Ω	9	'n
Endosulfan II	2	Ω	2	Ω	2	Ω	2	Ω	19	n	7	D	2	Ω	7	'n
Endosulfan sulfate	28	Ω	25	n	26	Ω	26	n	305	Ω	26	n	79	n	26	D
Endrin	33	Ω	7	Ω	2	Ω	2	Ω	28	n	3		2	Ω	7	Ω
Endrin aldehyde	10	Ω	6	Ω	6	Ω	6	Ω	106	Ω	6	n	6	Ω	6	n
gamma-BHC (Lindane)	2	n	7	Ω	2	Ω	2	Ω	19	Ω	2	n	2	U	7	n
Heptachlor	1	Ω	_	Ω	_	n	-	n	14	Ω	20		4		1	n
Heptachlor epoxide	35	Ω	32	Ω	33	Ω	33	Ω	383	Ω	33	Ω	33	Ω	33	n
Methoxychlor	75	Ω	89	n	69	Ω	70	Ω	813	n	70	D	70	Ω	70	n
Toxaphene	102	Ω	93	Ω	94	n	96	D	1.110	П	95	Ξ	90	<u>;</u>	,	-

See next page for footnotes.

## TABLE C.16a (Cont.)

<sup>a</sup> Sampled by Weston, analyzed by GP Environmental Services, TCL pesticides and PCBs-SW846 8080; provisional data.

Data qualifier:

U = analyte was analyzed for but not detected.

TABLE C.16b Pesticide and PCB Analyses for Soil Borings, May 1995<sup>a</sup>

			Bore	hole Cor	centration	ns (μg/k	g) at Vari	ous Dep	oth Interv	als		
Compound		IPBOR2 )-2ft)	TBNP (2-6	BOR2	TBNP1 (0-2		TBNP (2-4			PBOR3 6 ft)	VXB (2-4	
Aldrin	2.06	U	1.99	U	2.04	U	2.3	U	2.4	U	40.6	U
alpha-BHC	2.06	UJ(+)	1.99	U	2.04	U	2.3	U	2.4	UJ	40.6	U
alpha-Chlordane	2.06	UJ(+)	1.99	U	2.04	U	2.3	U	2.4	UJ	40.6	U
Aroclor 1016	39.9	UJ(+)	38.6	U	39.5	U	44.6	U	46.5	UJ	789	U
Aroclor 1221	81.1	UJ(+)	78.4	U	80.3	U	90.5	U	94.5	UJ	1,600	U
Aroclor 1232	39.9	UJ(+)	38.6	U	39.5	U	44.6	U	46.5	UJ	789	U
Aroclor 1242	39.9	UJ(+)	38.6	U	39.5	U	44.6	U	46.5	UJ	789	U
Aroclor 1248	570	J(+)	38.6	U	39.5	U	44.6	U	46.5	UJ	789	U
Aroclor 1254	39.9	UJ(+)	38.6	U	39.5	U	44.6	U	46.5	UJ	789	U
Aroclor 1260	39.9	UJ(+)	38.6	U	39.5	U	44.6	U	46.5	UJ	789	U
beta-BHC	2.06	UJ(+)	1.99	U	2.04	U	2.3	U	2.4	UJ	40.6	U
4,4'-DDD	3.99	UJ(+)	3.86	U	3.95	U	4.46	U	4.65	UJ	78.8	JВ
4,4'-DDE	3.99	UJ(+)	3.86	U	1.9	JP	4.46	U	4.65	UJ	18.2	JB
4,4'-DDT	3.99	UJ(+)	3.86	U	3.95	U	4.46	U	4.65	UJ	78.9	U
delta-BHC	2.06	UJ(+)	1.99	U	2.04	U	2.3	U	2.4	UJ	40.6	U
Dieldrin	3.99	UJ(+)	3.86	U	3.95	U	4.46	U	4.65	UJ	78.9	U
Endosulfan I	2.06	UJ(+)	1.99	U	2.04	U	2.3	U	2.4	UJ	40.6	U
Endosulfan II	3.99	UJ(+)	3.86	U	3.95	U	4.46	U	4.65	UJ	78.9	U
Endosulfan sulfate	3.99	UJ(+)	3.86	U	3.95	U	19	P	4.65	UJ	78.9	U
Endrin	3.99	UJ(+)	3.86	U	3.95	U	4.46	U	4.65	UJ	78.9	U
Endrin aldehyde	3.99	UJ(+)	3.86	U	3.95	U	4.46	U	4.65	UJ	78.9	U
Endrin ketone	3.99	UJ(+)	3.86	U	3.95	U	4.9	P	4.65	UJ	78.9	U
gamma-Chlordane	2.06	UJ(+)	1.99	U	2.04	U	2.3	U	2.4	UJ	40.6	U
Heptachlor	2.06	UJ(+)	1.99	U	2.04	U	2.3	U	2.4	UJ	40.6	U
Heptachlor epoxide	2.06	UJ(+)	1.99	U	2.04	U	12.2		2.4	UJ	40.6	U
Lindane (gamma-BHC)	2.06	UJ(+)	1.99	U	2.04	U	2.3	U	2.4	UJ	40.6	U
Methoxychlor	20.6	UJ(+)	19.9	U	20.4	U	23	U	24	UJ	406	U
Toxaphene	206	UJ(+)	199	U	204	U	230	U	240	UJ	4,060	U

TABLE C.16b (Cont.)

	Bore	hole Co	oncentration	s (µg/kg	) at Vario	us Dept	h Interval	S
Compound	VXB(2-4		VXB0 (4-6		VXB0 (4-6		VXB0 (8-10	
Aldrin	43.2	U	20.8	U	2.21	U	2.11	U
alpha-BHC	43.2	U	20.8	U	1.3	JP	2.11	U
alpha-Chlordane	43.2	U	20.8	U	2.21	U	2.11	U
Aroclor 1016	838	U	404	U	43	U	40.9	U
Aroclor 1221	1700	U	820	U	87.2	U	83.1	U
Aroclor 1232	838	U	404	U	43	U	40.9	U
Aroclor 1242	838	U	404	U	43	U	40.9	U
Aroclor 1248	838	U	404	U	43	U	40.9	U
Aroclor 1254	838	U	404	U	43	U	40.9	U
Aroclor 1260	838	U	404	U	43	U	40.9	U
beta-BHC	43.2	U	20.8	U	2.21	U	2.11	U
4,4'-DDD	83.8	U	40.4	U	4.6		1.5	JB
4,4'-DDE	83.8	U	40.4	U	4.3	U	0.97	JВ
4,4'-DDT	83.8	U	40.4	U	4.3	U	4.09	U
delta-BHC	43.2	U	20.8	U	2.21	U	2.11	U
Dieldrin	83.8	U	40.4	U	4.3	U	4.09	U
Endosulfan I	43.2	U	20.8	U	2.21	U	2.11	U
Endosulfan II	83.8	U	40.4	U	4.3	U	4.09	U
Endosulfan sulfate	83.8	U	40.4	U	4.3	U	4.09	U
Endrin	83.8	U	40.4	U	4.3	U	4.09	U
Endrin aldehyde	83.8	U	40.4	U	4.3	U	4.09	U
Endrin ketone	83.8	U	40.4	U	2.9	JP	4.09	U
gamma-Chlordane	43.2	U	20.8	U	2.21	U	2.11	U
Heptachlor	43.2	U	20.8	U	2.21	U	2.11	U
Heptachlor epoxide	43.2	U	20.8	U	2.21	U	2.11	U
Lindane (gamma-BHC)	43.2	U	20.8	U	2.21	U	2.11	U
Methoxychlor	432	U	208	U	22.1	U	21.1	U
Toxaphene	4,320	U	2,080	U	221	U	211	U

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

## Data qualifiers:

U = analyte was analyzed for but not detected.

J = estimated value.

(+) = value obtained from a 1:10 dilution.

B = compound was found in the associated blank.

P = percent difference between the results from two GTC columns is greater than 25%, and the lower of the two values is reported.

TABLE C.17 On-Site Analytical Suite Results for Soil Borings, December 1993–May 1994<sup>a</sup>

0		Average	Bore	ehole Concen	trations (mg	/kg)
Sample No. (Depth Interval)	Collection Date	Depth (ft)	РАН	ТРН	РСВ	TNT
Sediment						
BOR1-1 (6-8 ft)	04/18/94	7	1 U	10 U	1 U	0.7 U
BOR2-1 (3-5 ft)	04/19/94	4	1 U	100 U10	1 U	0.7 U
BOR3-1 (6-8 ft)	04/19/94	7	1 U	100 G	1 U	0.7 G
BOR4-1 (5-8 ft)	04/19/94	7	1 U	100 U10	1 U	
BOR5-1 (2-4 ft)	04/20/94	3	1 U	100 U10	1 U	0.7 U 0.7 U
BOR6-1 (2-4 ft)	04/20/94	3	1 U	100 010	1 U	0.7 G
BOR7-1 (2-4 ft)	04/20/94	3		100 U10	1 U	0.7 U
BOR8-1 (0-2 ft)	04/21/94	1	1 U	10 U	1 U	0.7 U
Soil						
TBSPBOR1 (0-2 ft)	05/23/94	1	10 U1	100 U10	10 U1	0.7 U
TBSPBOR1 (2-4 ft)	05/23/94	3	1 U	100 U10	1 U	0.7 U
TBSPBOR1 (4-6 ft)	05/23/94	5	10 U1	100 U10	1 U	0.7 U
TBSPBOR1 (6-8 ft)	05/23/94	7	1 U	10 U	1 U	0.7 U
TBSPBOR1 (8-10 ft)	05/23/94	9	1 U	100 U10	1 U	0.7 U
TBSPBOR1 (10-12 ft)	05/23/94	11	1 U	100 U10	1 U	0.7 U
TBSPBOR2 (0-2 ft)	05/23/94	1	1 U	100 U10	10 G	0.7 U
TBSPBOR2 (2-4 ft)	05/23/94	3	1 U	100 U10	10 U1	0.7 U
TBSPBOR2 (4-6 ft)	05/23/94	5	1 U	10 U	1 U	0.7 U
TBSPBOR2 (6-8 ft)	05/23/94	7	1 U	100 G	1 U	0.7 U
TBSPBOR2 (8-10 ft)	05/23/94	9	1 U	100 G	1 U	0.7 U
TBNPBOR1 (0-2 ft)	05/24/94	1	1 U	100 U10	10 U1	NA
TBNPBOR1 (2-4 ft)	05/24/94	3	1 U	100 G	10 U1	NA
TBNPBOR1 (4-6 ft)	05/24/94	5	1 U	100 U10	10 U1	NA
TBNPBOR1 (6-8 ft)	05/24/94	7	1 U	10 U	NA	NA
HBOR1 (0-2 ft)	05/24/94	1	1 U	100 U10	10 U1	NA
HBOR1 (2-4 ft)	05/24/94	3	1 U	10 U	1 U	NA
HBOR1 (4-6 ft)	05/24/94	5	1 U	10 U	10 U1	NA
HBOR1 (6-8 ft)	05/24/94	7	1 U	10 U	NA	NA
HBOR1 (8-10 ft)	05/24/94	9	NA	100 U10	NA	NA
HBOR2 (0-2 ft)	05/25/94	1	1 U	10 U	1 U	NA
HBOR2 (4-6 ft)	05/25/94	3	1 U	10 U	1 U	NA
VXBOR1 (0-2 ft)	05/16/94	1	1 U	100 G	1 U	0.7 U
VXBOR1 (4-6 ft)	05/16/94	5	10 G	100 G	1 U	0.7 U
VXBOR1 (6-8 ft)	05/16/94	7	10 G	100 G	1 U	0.7 U
VXBOR1 (8-10 ft)	05/16/94	9	10 G	100 G	1 U	0.7 U
VXBOR2 (0-2 ft)	05/16/94	1	1 U	100 G	1 U	0.7 U
VXBOR2 (2-4 ft)	05/16/94	3	1 U	100 G	1 U	0.7 U
VXBOR2 (4-6 ft)	05/16/94	5	1 U	100 G	1 U	0.7 U
VXBOR2 (6-8 ft)	05/16/94	7	1 U	100 G	1 U	0.7 U

TABLE C.17 (Cont.)

g	<b></b> .	Average	Bore	hole Concent	rations (mg/	kg)
Sample No. (Depth Interval)	Collection Date	Depth (ft)	РАН	ТРН	PCB	TNT
Soil (Cont.)						
VXBOR2 (8-10 ft)	05/16/94	9	l U	100 U10	1 U	0.7 U
VXBOR2 (10-12 ft)	05/16/94	11	1 U	100 G	ΙU	0.7 U
CLP1 (0-6 in.)	01/24/94	0.25	1 U	10 U	10 U1	0.7 U
CLP1 (6-24 in.)	01/24/94	1.25	I U	10 U	1 U	0.96
CLP2 (0-6 in.)	01/24/94	0.25	1 U	10 U	ΙÜ	0.7 U
CLP2 (6-24 in.)	01/24/94	1.25	1 U	10 U	1 U	0.7 U
CLP2 (24-48 in.)	01/24/94	3	10 U1	100 U10	1 U	0.7 U
CLP3 (0-6 in.)	01/25/94	0.25	10 U1	100 U10	10 U1	0.7 U
CLP3 (6-24 in.)	01/25/94	1.25	10 U1	10 U	1 U	0.7 U
CLP3 (6-24 in.) Dup	01/25/94	1.25	1 U	10 U	1 U	0.7 U
CLP3 (24-48 in.)	01/25/94	3	1 U	10 U	1 U	0.7 U
CLP4 (0-6 in.)	01/25/94	0.25	1 U	10 U	l U	0.7 U
CLP4 (6-24 in.)	01/25/94	1.25	1 U	100 U10	1 U	0.7 U
CLP4 (24-48 in.)	01/25/94	3	1 U	100 U10	1 U	0.7 U
CLP5 (0-6 in.)	01/25/94	0.25	1 U	10 U	ΙÜ	0.7 U
CLP5 (6-24 in.)	01/25/94	1.25	1 U	100 U10	1 U	1.95
CLP5 (24-48 in.)	01/25/94	3	1 U	100 G	1 U	1.67
CLP5 (24-48 in.) Dup	01/25/94	3	1 U	100 G	1 U	0.99
CLP6 (0-6 in.)	01/25/94	0.25	1 U	10 U	1 U	0.7 U
CLP6 (6-24 in.)	01/25/94	1.25	10 U1	100 U10	l U	0.7 U
CLP7 (0-6 in.)	01/25/94	0.25	1 U	10 U	l U	0.7 U
CLP7 (6-24 in.)	01/25/94	1.25	1 U	10 U	l U	0.7 U
CLP7 (24-48 in.)	01/25/94	3	1 U	10 U	1 U	0.7 U
CLP8 (0-6 in.)	01/26/94	0.25	1 U	10 U	1 U	0.7 U
CLP9 (0-6 in.)	01/26/94	0.25	1 U	10 U	1 U	0.7 U
OT4A (0-6 in.)	11/09/93	0.25	10 U	10 U	1 U	0.7 U
OT6A (0-6 in.)	11/09/93	0.25	10 U	10 U	1 U	0.7 U
OT8A (0-6 in.)	10/13/93	0.25	NA	NA	NA	0.7 U
OT9A (0-6 in.)	10/13/93	0.25	NA	NA	NA	0.7 U
OT15B (6-12 in.)	11/08/93	0.75	100 U10	100 U10	1 U	NA
OT16B (6-12 in.)	11/08/93	0.75	10 U	10 U	1 <b>U</b>	NA
OT18A (0-6 in.)	11/08/93	0.25	NA	NA	1 U	NA
OT18B (6-12 in.)	11/08/93	0.75	1 U	10 U	NA	NA
OT19A (0-6 in.)	11/08/93	0.25	NA	NA	1 U	NA
OT19B (6-12 in.)	11/08/93	0.75	1 U	NA	NA	NA
OT29A (0-6 in.)	10/27/93	0.25	1 U	100 U10	10 U1	NA
OT29B (6-12 in.)	10/27/93	0.75	NA	10 U	1 U	NA
OT30A (0-6 in.)	10/27/93	0.25	NA	NA	10 U1	NA
OT31A (0-6 in.)	10/13/93	0.25	NA	NA	1 U	NA
OT32A (0-6 in.)	10/13/93	0.25	NA	NA	1 U	NA

See next page for footnotes.

## TABLE C.17 (Cont.)

<sup>a</sup> Sampled and analyzed on-site by ANL.

Data qualifiers:

U = < value given.

10 U1 = <10 mg/kg but > 1 mg/kg.

100 U10 = <100 mg/kg but > 10 mg/kg.

G = > value given.

NA = not analyzed.

TABLE C.18 Explosive Compounds Analyses for Soil Samples, May 1995<sup>a</sup>

												1
Compound	TBNPBOR2 (0-2 ft)	OR2	VXBOR4 (2-4 ft)	1R4 ft)	VXBOR4 (4-6 ft)	1R4	VXBOR5 (0-2 ft)	R5 ft)	VXBOR5 (4-6 ft)	oR5	VXBOR5 (8-10 ft)	ļ
2-Amino-4,6-dinitrotoluene	46.7	Ω	46.7	Ω	46.7	n	46.7	Ω	46.7	Ω	1 46.7	٣.
4-Amino-2,6-dinitrotoluene	40.8	Ω	40.8	n	40.8	n	40.8	n	40.8	Ω	40.8	, <u>I</u>
1,3-Dinitrobenzene	37.2	Ω	37.2	n	37.2	Ω	37.2	U	37.2	n		<u> </u>
2,4-Dinitrotoluene	51.6	n	51.6	n	51.6	D	51.6	Ω	51.6	n		n
2,6-Dinitrotoluene	47.6	D	47.6	n	47.6	D	47.6	n	47.6	n		, D
HMX	70.5	Ω	70.5	n	70.5	n	70.5	D	70.5	n		, <b>-</b>
Nitrobenzene	35.2	Ω	35.2	D	35.2	n	35.2	n	35.2	Ŋ	35.2 U	I
Nitroglycerin	10,000	Ω	50,000	n*	50,000	<u>n</u> *	10,000	Ω	50,000	$\Omega_*$	10,000	J
2-Nitrotoluene	81.4	D	81.4	n	81.4	n	81.4	D	81.4	n	81.4	<u></u>
3-Nitrotoluene	81.7	Ω	81.7	n	81.7	Ω	81.7	n	81.7	D	81.7 U	Í
4-Nitrotoluene	87.2	Ω	87.2	Ω	87.2	U	87.2	Ω	87.2	Ω		Ţ
Pentaerythritol tetranitrate	250	Ü	1,670		1,250	n *	250	Ω	250	n	250 U	Í
RDX	50.9	Ω	50.9	Ω	50.9	n	50.9	n	50.9	n	50.9 L	_
Tetryl	163	n	163	n	163	n	163	n	163	n	163 L	-
1,3,5-Trinitrobenzene	40.2	Ω	40.2	Ω	40.2	D	40.2	n	40.2	D	40.2 U	-
2,4,6-Trinitrotoluene	35.6	n	35.6	ם	35.6	n	35.6	n	35.6	n	35.6 U	
					Soil	Conce	Soil Concentrations (µg/kg) at 0-6 in.	ıg/kg) a	ıt 0-6 in.			
Compound	SQPS 4	4	SQPS 5	5	SQPS 8	∞	8QPS	6	TPDGS 3	5.3	TPDGS 4	TPDGS 5
2-Amino-4,6-dinitrotoluene	46.7	n	46.7	n	46.7	Þ	46.7	n	46.7	D	46.7 U	1 46.7
4-Amino-2,6-dinitrotoluene	40.8	Ω	40.8	Ω	40.8	Ω	40.8	D	40.8	n	40.8 U	
1,3-Dinitrobenzene	37.2	D	37.2	D	37.2	D	37.2		37.2	13	11 648	

TABLE C.18 (Cont.)

					So	il Conc	Soil Concentrations (µg/kg) at 0-6 in.	µg/kg)	at 0-6 in.					
Compound	SQPS 4	4	SQPS 5	\$5	SQPS 8	8 S	SQPS 9	6.6	TPDGS 3	3S 3	TPDGS 4	4	TPDGS 5	3.5
2,4-Dinitrotoluene	51.6	Ω	51.6	Ω	51.6	Ω	51.6	Ŋ	51.6	11	516	11	\$18	=
2,6-Dinitrotoluene	47.6	Ω	47.6	U	47.6	n	47.6	Ω	47.6	n	47.6		47.6	) <b>=</b>
HMX	70.5	D	70.5	Ŋ	70.5	n	70.5	Ω	70.5	Ω	70.5	'n	70.5	) D
Nitrobenzene	35.2	Ω	35.2	n	35.2	Ω	35.2	Ω	35.2	Ω	35.2	D	35.2	'n
Nitroglycerin	10,000	Ω	10,000	Ω	20,000	n	10,000	n	10,000	Ω	15,300		10.000	· 🗅
2-Nitrotoluene	81.4	Ω	81.4	n	81.4	n	81.4	n	81.4	Ω	81.4		814	) <u>=</u>
3-Nitrotoluene	81.7	Ω	81.7	n	81.7	Ω	81.7	n	81.7	U	81.7	n	81.7	) <u>=</u>
4-Nitrotoluene	87.2	Ω	87.2	Ω	87.2	Ω	87.2	Ω	87.2	n	87.2		87.2	) <u>=</u>
Pentaerythritol tetranitrate	250	Ω	250	Ω	200	Ω	500	Ω	250	Ω	250	· =	250	) <u>=</u>
RDX	50.9	Ω	50.9	Ω	50.9	Ŋ	50.9	D	50.9	D	50.9	) =	50.9	) <u>=</u>
Tetryl	163	Ω	163	Ω	163	Ω	163	n	163	D	163	<u> </u>	163	) <u>=</u>
1,3,5-Trinitrobenzene	40.2	n	40.2	Ω	40.2	n	40.2	Ω	40.2	Ω	40.2	Þ	40.2	) <u> </u>
2,4,6-Trinitrotoluene	35.6	U	35.6	Ω	35.6	Ω	35.6	n	35.6	Ω	35.6	n	35.6	) <u> </u>
												i		,

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

<sup>\* =</sup> internal standard area counts were outside the QC limits.

TABLE C.19 CWA Degradation Products and Organosulfur Compound Analyses for Soil, May 1995

								-				
Compound	HBOR4 (6-8 ft)	<b></b>	HBOR4 (8-10 ft)	)R4 ) ft)	HB(	HBOR4 (10-12 ft)	HB(	HBOR4 (14-16 ft)	TBNPBOR2	BOR2	TBNPBOR2	OR.
CWA degradation products												3
Diisopropyl methylphosphonate (DIMP)	0.09	~	0.01	_	0.03	ר	0.141	Þ	0.138	Π	0.133	_
Dimethyl methylphosphate (DMMP)	0.157	D	0.152	n	0.16	Þ	0.165	) D	0.161	> =	0.156	) =
Isopropyl methylphosphonic acid (IMPA)	0.978	ב	1.14	n	1.18	D	1.28	) =	1 02	) <u> </u>	1.35	) =
Methyl phosphonic acid	0.978	Ω	1.14	ם	1.18	o D	1.28	ם מ	1.02	) <u>=</u>	1.35	) <u></u>
Thiodiglycol (TDGCL)	4.65	n	4.49	Ω	4.73	ח	4.88	n	4.77	n	4.61	)
Organosul fur compounds												
Benzothiozole (BTZ)	6.35	n	2.45	n	2.59	n	1.34	D	1.31	Ξ	1 26	Ξ
p-Chlorophenylmethyl sulfide (CPMS)	6.35	n	2.45	n	2.59	Ω	1.34	D	1.31	) )	1.26	) =
p-Chlorophenylmethyl sulfone (CPMSO2)	13.9	Ω	5.38	⊃	5.69	Ω	2.94	⊃	2.87	n	2.77	-
p-Chlorophenylmethyl sulfoxide (CPMSO)	13.2	n	5.11	n	5.4	ב	2.79	n	2.72	· ⊃	2.63	î
Dimethyl disulfide (DMDS)	4.07	Ω	1.57	Ω	1.66	Ω	0.858	n	0.837	Ω	0.81	$\supset$
1,4-Dithiane	2.65	_	1.94	Þ	1.03	۳-,	1.82	Ω	1.78	n	1.72	$\supset$
1,4-Oxathiane	5.03	ח	0.59	- I	2.05	ב	1.06	n	1.04	Ω	-	$\supset$
			Soil	Concer	itrations (1	mg/kg)	Soil Concentrations (mg/kg) at Various Depth Intervals	Depth Ir	ıtervals			i
	TBNPBOR2	\$2	TBNPBOR2	3OR2	VXBOR4	OR4	VXBOR4	OR4	VXBOR5	OR5	VXBOR5	RS
Compound	(6-8 ft)		(10-12 ft)	2 ft)	(2-4 ft)	3	(4-6 ft)	ft)	(0-2 ft)	<b>(3)</b>	(4-6 ft)	12
CWA degradation products												
Diisopropyl methylphosphonate (DIMP)	1.39	n	0.137	D	0.725	n	0.279	D	0.138	D	0.148	$\supset$
Dimethyl methylphosphate (DMMP)	0.162	ב	0.16	ם	0.846	n	0.326	n	0.161	D	0.173	
Isopropyl methylphosphonic acid (IMPA)	1.43	n	1.1	n	1.36	ם	0.828	D	1.09	<b>-</b>	1 40	_
Methyl phosphonic acid	1.43	ם	-	Þ	1.36	=	0.828	_	1 00	) =	\	) [
						)	11.11.11.11.11.11.11.11.11.11.11.11.11.	٠	///-		1	

TABLE C.19 (Cont.)

			Soi	Concen	trations	(mg/kg) a	Soil Concentrations (mg/kg) at Various Depth Intervals	Depth I	ntervals			
Compound	TBNPBOR2 (6-8 ft)	2	TBNPBOR2 (10-12 ft)	3OR2 2 ft)	VXI	VXBOR4 (2-4 ft)	VXX	VXBOR4 (4-6 ft)	VXBOR (0-2 ft)	VXBOR5 (0-2 ft)	VXBOR5 (4-6 ft)	85
Organosulfur compounds Benzothiozole (BTZ)	1.32	n	1.33	n	68.7	E	26.4	E	1 31	=	-	:
p-Chlorophenylmethyl sulfide (CPMS)	1.32	n	1.33	Ω	68.7	5 15	26.4	n n	1.31	) <u>=</u>	4. T	o <u>=</u>
p-Chlorophenylmethyl sulfone (CPMSO2)	2.89	Ω	2.92	n	151	Ω	58	n	2.87	) D	3.08	) D
P-Chlorophenylmethyl sulfoxide (CPMSO)	2.74	n	0.851	Ω	143	D	55.1	n	2.72	D	2.92	, ,
Dimethyl disulfide (DMDS)		n		U	4	m	16.9	n	0.837	Ω	0.0	) <u>-</u>
1,4-Duniane		n	1.81	n	93.5	n	36	ח	0.46	~	1.91	ם ה
1,4-Oxaminane	1.04	n	1.05	Ω	54.4	B	21	n	1.04	n	1.11	n
		Soil	Concent	rations (	mg/kg) a	t Various	Soil Concentrations (mg/kg) at Various Depth Intervals	itervals				
	VXBOR5		VXBOR5	R5	VXB	VXBOR5	VX	VXBOR5				
Compound	(8-10 ft)		(10-12 ft)	£)	(12-14 ft)	4 ft)	(14-	(14-16 ft)				
CWA degradation products												
Diisopropyl methylphosphonate (DIMP)	0.141	Ω	0.14	Ω	0.142	Ω	0.141	n				
Dimethyl methylphosphate (DMMP)	0.165	n	0.17	n	0.166	Ω	0.165	Ω				
Isopropyl methylphosphonic acid (IMPA)		n	1.16	n	1.36	Ω	1.14	Ω				
Methyl phosphonic acid		n	1.16	Ω	1.36	Ω	1.14	מ				
i niodigiycol (TDGCL)	4.88	n	4.92	n	4.92	Ω	4.88	Ω				
Organosulfur compounds												
Benzothiozole (BTZ)	1.34	n	1.35	Ω	1.35	Ξ	1 34	Ξ				
p-Chlorophenylmethyl sulfide (CPMS)	1.34	n	1.35	n	1.35	n	1.34	> =				
p-Chlorophenylmethyl sulfone (CPMSO2)	2.94	Ω	2.96	n	2.96	n	2.94	n				
p-Chlorophenylmethyl sulfoxide (CPMSO)	2.79	n	2.81	Ω	2.81	Ω	2.79	n				
Dimethyl disulfide (DMDS)		Ω	0.87	n	0.865	n	0.858	Ω				
1,4-Dithiane		n		Ω	0.23	ĭ	1.82	Ω				
1,4-Oxathiane	1.06 U	J	1.07	Ω	1.07	Ω	1.06	n				
			ļ									

<sup>a</sup> Sampled by ANL, analyzed by GP Environmental Services.

Data qualifiers: J = estimated value. U = analyte was analyzed for but not detected; detection limit given.

TABLE C.20 Dioxin and Total Petroleum Hydrocarbons Analyses for Soil Samples, May 1995<sup>a</sup>

		Soil Concen	trations (	(μg/kg) at V	ariou:	s Depth Ir	iterva	ıls	
Analyte	TBNPBOR3 (4-6 ft)	VXB0 (2-4		VXBOR (4-6 ft		VXBOI (4-6 fi			30R5 10 ft)
Heptachlorodibenzo-p-dioxins	2.9	0.16	U	0.15	U	0.11	U	0.03	U
Heptachlorodibenzo-p-furans	2.9	0.07		0.02	U	0.02	U	0.01	Ü
Hexachlorodibenzo-p-dioxins	1.6	0.06	U	0.06	U	0.05	U	0.04	U
Hexachlorodibenzo-p-furans	3.1	0.03		0.03	U	0.01	U	0.01	Ü
Octachlorodibenzo-p-dioxin	4.3	0.32		0.28		0.1		0.03	U
Octachlorodibenzo-p-furan	1.7	0.04	U	0.04	U	0.04	U	0.02	U
Pentachlorodibenzo-p-dioxins	0.26	0.03	U	0.03	U	0.04	U	0.03	U
Pentachlorodibenzo-p-furans	3.4	0.14	U	0.09	U	0.05	U	0.02	U
2,3,7,8-TCDD	0.18	0.01	U	0.02	U	0.04	U	0.02	U
2,3,7,8-TCDF	0.49	0.04		0.04		0.02	U	0.01	U
Tetrachlorodibenzo-p-dioxins	0.07	0.01	U	0.02	U	0.04	U	0.02	U
Tetrachlorodibenzo-p-furans	1.5	0.04		0.04		0.02	U	0.01	U
Total petroleum hydrocarbons (mg/kg)	19,000 J	35,400	J(+)	18,200		666		31	UJ(+)

<sup>&</sup>lt;sup>a</sup> Sampled by ANL and analyzed by GP Environmental Services.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

+ = value obtained from a 1:10 dilution.

TABLE C.21a Volatile Organics Analyses for Groundwater, May 1994<sup>a</sup>

Compound							(-0-)				ĺ				
		TH-4		P-1		P-2		P-3		P-4			P-9	ſ	JF-43
Acetone	10	D	10	n	10	n	10			9	<u> </u>	2	=		*11
Benzene	10	Ω	10	D	10	D	4		•			2 5	) <b>=</b>	2 5	* *
Bromodichloromethane	10	Ω	10	Ω	10	n	10	n n	<del>, -</del>	. 2	. :	2 0	) <u>=</u>	2 2	· *
Bromoform	10	D	10	Ω	10	Ω	101		· -	10	) <u></u>	10	) <u> </u>	2 2	t <del>*</del>
Bromomethane	10	n	10	U	10	D	10	n		10	ם מ	2 0	) <u> </u>	2 0	† † 5
2-Butanone	10	Ω	10	Ω	10	U	10	Ω	10	0		20	î D	2 2	- <del> </del>
Carbon disulfide	10	n	10	Ω	10	Ω	10	n	10	,	D	10	ò	10	- *
Carbon tetrachloride	10	Ω	10	Ω	10	n	10	Ω	10		D	10	D	10	+*1
Chlorobenzene	10	Ω	10	Ω	10	D	170		10	1 0	_	10	D	9	*1
Chloroethane	10	D	10	Ω	10	Ω	10	Ω	10		Ŋ	10	n	01	* * 1
Chloroform	10	Ω	10	D	10	U	10	Ω	10	(	J	10	n	10	* * * * * * * * * * * * * * * * * * * *
Chloromethane	10	Ω	10	Ω	10	Ω	10	Ω	10		b	10	Ω	10	* *
Dibromochloromethane	10	D	10	Ω	10	n	10	Ω	10		J	10	Ω	10	**N
1,1-Dichloroethane	10	Ω	10	n	10	n	10	D	10	) (	_	10	Ω	10	1*1
1,2-Dichloroethane	10	Ω	10	n	10	n	10	Ω	10	) (	J	10	D	10	* * N
1,1-Dichloroethene	10	Ω	10	Ω	10	D	27		9	J		10	D	10	U*+
1,2-Dichloroethene (total)	10	Ω	10	Ω	10	Ω	1,100	Q	13,000	Д	_	10	n	10	*
1,2-Dichloropropane	10	n	10	Ω	10	n	10	Ω	10	n	_	10	Ω	10	* D
cis-1,3-Dichloropropene	10	n	10	Ω	10	Ω	10	D	10	<u> </u>	_	10	n	10	*
trans-1,3-Dichloropropene	10	Ω	10	Ω	10	Ω	10	Þ	10	) )	_	10	Ω	10	*
Ethylbenzene	10	D	10	Ω	10	n	10	Ω	10	D	_	10	n	10	**
2-Hexanone	10	Ω	10	Ω	10	D	10	Ω	10	,		10	Ω	10	†*0
4-Methyl-2-pentanone	10	Ω	10	n	10	n	10	n	10	D		10	Ω	10	+ ^
Methylene chloride	10	n	10	Ω	10	Ω	10	Ω	10	D		10	n	10	**
Styrene	10	n	10	Ω	10	Ω	10	Ω	10	Ω		10	U	10	$ m U^*_{+}$

TABLE C.21a (Cont.)

(			Concent	Concentrations (µg/L) in Groundwater	Groundwater		
Compound	TH-4	P-1	P-2	P-3	P-4	6-d	JF-43
1,1,2,2-Tetrachloroethane	10 11	10 11	11 01	1 /	3 500	1 01	
Tetrachloroethene	10 U			7 400 D			10 0:
Toluene	10 U				11 01	·	
1,1,1-Trichloroethane	10 U	10 U		î 01	·	·	10 0:4
1,1,2-Trichloroethane	10 U	10 U	10 U			·	
Trichloroethene	10 U	10 U	10 U	390 D	1,800 D		10 11*+
Vinyl chloride	10 U	10 U	10 U	570 D	74		
Xylenes (total)	10 U	10 U	10 U	3 J	10 U		
			Concentr	Concentrations (µg/L) in Groundwater	roundwater		
Compound	JF-53	JF-63	JF-73	JF-83	JF-173	JF-183	JF-52
Acetone	10 U	10 U	10 U	10 U	10 U	10 U	10 17
Benzene	10 U	10 U	10 U	10 U	10 U	3 J	
Bromodichloromethane	10 U	10 U	10 U	10 U	10 U	10 U	•
Bromoform	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Bromomethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U
2-Butanone	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Carbon disulfide	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Carbon tetrachloride	10 U	10 U	10 U	10 U	10 U	3 J	10 U
Chlorobenzene	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Chloroethane	10 U	10 U	10 U	O 01	10 U	10 U	10 U
Chloroform	10 U	9 J	10 U	44	10 U	12	
Chloromethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U

TABLE C.21a (Cont.)

Compound	JF	JF-53	JF	JF-63	IF	JF-73	JI	JF-83	JF	JF-173	JF.	JF-183	H	JF-52
Dibromochloromethane	10	Ω	10	n	10	U	10		10	ш	01	=	2	=
1,1-Dichloroethane	10	Ŋ	10	D	10	Ξ	101	) <u> </u>	10	) <u>=</u>		ָב כ	2 5	) ;
1,2-Dichloroethane	10	Ω	10	D	10	) <u> </u>	Q Y	) <u>-</u>		) =	01	)	2 :	⊃ ;
1,1-Dichloroethene	10	n	10		15	•	. 21	•	0 0	)	o <u>t</u>		2 ;	) ;
1,2-Dichloroethene (total)	2,200	Q	44	DI	8,900	Q	5,800	Щ	1.400	ם כ	10 000	ב	01 6	$\supset$
1,2-Dichloropropane	10	Ω	10	n	10	Ω	10		10		10	) <u> </u>	3 9	I
cis-1,3-Dichloropropene	10	Ŋ	10	D	10	Ω	10	D	10		10	) D	2 2	) <u>=</u>
trans-1,3-Dichloropropene	10	n	10	Ω	10	Ω	10	D	10		10	1	2 2	) <u>=</u>
Ethylbenzene	10	D	10	D	10	Ω	10	Ω	10	n	10	) D	2 2	) <u>=</u>
2-Hexanone	10	n	10	Ω	10	D	10	n	10	Ω	10	11	10	<del>-</del>
4-Methyl-2-pentanone	10	Ω	10	□	10	Ω	10	Ω	10	Ω	102	· 1	2	) <u> </u>
Methylene chloride	10	Ω	10	Ω	10	n	10	D	10		01	) <u>=</u>	2 4	ם בו
Styrene	10	Ω	10	n	10	n	10	n	10	n	01	) =	, 5	G =
1,1,2,2-Tetrachloroethane	550	Q	50		13,000	Q	160,000	Q	12,000	Ω	39,000	, _	2 9	=
Tetrachloroethene	7	_	32		250	D	1,100	Щ	170	l	8.300	ì	2 0	=
Toluene	10	Ω	10	Ω	10	Ω	10	D	10	Ŋ	10	Ξ	1 0	) <u> </u>
1,1,1-Trichloroethane	10	Ω	10	D	10	Ω	10	Ω	10	n	10	· =	2 2	=
1,1,2-Trichloroethane	58		10	Ω	200	Щ	066	Щ	230	Щ	009		2	· =
Trichloroethene	390	Q	089	Ω	4,100	Q	21,000	Q	5,800	Ω	13.000		0	) <u> </u>
Vinyl chloride	41		12		230	Щ	34		13				30	•
Xylenes (total)	10	11	2	11	•	;	•	i					) }	

TABLE C.21a (Cont.)

I			-			Concent	Concentrations (µg/L) in Groundwater	(L) in Gr	oundwate	1.					
Compound	JF	JF-62	JF	JF-72	JF	JF-82	JI	JF-51	JI	JF-61	JF	JF-71	Tr	JF-81	1
Acetone	10	Ω	52		10	U	10	þ	10	Þ	10	Ω	10	Ω	
Benzene	10	Ω	10	Ω	10	Ŋ	10	U	10	n	10	Ω	10	n	
Bromodichloromethane	10	U	10	Ω	10	U	10	Ω	10	U	10	D	10	Ω	
Bromoform	10	Ω	10	Ω	10	Ω	10	Ω	10	Ω	10	D	10	n	
Bromomethane	10	n	10	Ω	10	Ŋ	10	Ω	10	n	10	Ω	10	Ω	
2-Butanone	10	D	10	Ω	10	Ω	10	Ω	10	n	10	n	10	Ω	
Carbon disulfide	10	Ω	10	Ω	10	U	10	U	10	n	10	n	10	D	
Carbon tetrachloride	10	n	10	U	10	Ω	10	Ŋ	10	U	10	Ω	10	D	
Chlorobenzene	10	ņ	10	n	10	Ω	10	Ω	10	Ω	10	n	10	n	
Chloroethane	10	D	10	Ω	10	Ω	10	Ω	10	Ω	10	D	10	n	
Chloroform	10	Ω	10	Ω	10	n	10	b	10	n	10	Þ	10	D	
Chloromethane	10	D	10	Ω	10	n	10	n	10	n	10	n	10	D	
Dibromochloromethane	10	D	10	Ω	10	Ŋ	10	Ω	10	Ω	10	Ω	10	n	
1,1-Dichloroethane	10	Ω	10	D	10	n	10	Ω	10	Ω	10	Ω	10	ם	
1,2-Dichloroethane	10	Ω	10	Ω	10	n	10	Ω	10	Ω	10	Ω	10	þ	
1,1-Dichloroethene	10	n	10	Ω	11		9	ſ	10	Ω	10	Ω	28		
1,2-Dichloroethene (total)	10	Ω	ж	r	170	D	650	D	10	n	10	Ω	190		
1,2-Dichloropropane	10	n	10	n	10	Ω	10	Ω	10	n	10	Ω	10	D	
cis-1,3-Dichloropropene	10	n	10	Ω	10	Ω	10	Ω	10	D	10	n	10	n	
trans-1,3-Dichloropropene	10	n	10	N	10	Ω	10	n	01	D	10	Ω	10	D	
Ethylbenzene	10	Ω	10	n	10	Ω	10	n	10	n	10	n	10	Ω	
2-Hexanone	10	Ω	01	n	10	D	10	U	10	n	10	n	10	D	
4-Methyl-2-pentanone	10	Ω	10	Ω	10	Ω	10	b	01	n	10	Ω	10	D	
Methylene chloride	9	ВЈ	10	Ω	10	n	10	Ω	01	D	10	D	10	n	
Styrene	01	n	10	n	01	D	10	Ω	10	Ω	01	D	10	Ω	

TABLE C.21a (Cont.)

						Concentr	Concentrations (µg/L) in Groundwater	(L) in Gro	undwate	r				
Compound	JF-62	-62	JF	JF-72	JF-	JF-82	JI	JF-51	JF-61	-61	JF-71	71	H H	JF-81
1,1,2,2-Tetrachloroethane	10	Ŋ	10	n	10	Ω	0		2	11	2	=	-	:
Tetrachloroethene	10	U	10	n	17		10	) D	2 0	) <u> </u>	2 2	o	27	)
Toluene	10	n	10	U	10	n	10	n	10	) <u> </u>	10	) <u> </u>	t =	-
1,1,1-Trichloroethane	10	Ω	10	n	10	Ω	10	Ω	2 0	) [	1 2	) <u> </u>	2 5	ם כ
1,1,2-Trichloroethane	10	Ω	10	Ω	10	Ω	10	n D	21	) <u> </u>	2 2	> =	2 2	ם כ
Trichloroethene	10	Ω	10	n	1,500	D	850	o Q	2 9	) <b>-</b>	01	> =	1,600	ם כ
Vinyl chloride	10	D	10	Ω	23		10	n	10	. D	10	) <u>=</u>	00,1	ב ב
Xylenes (total)	10	D	10	n	10	Ω	10	Ω	10	11	2 2	) <u> </u>	2 5	) =

<sup>a</sup> Sampled by ANL, analyzed by EA Laboratories, CLP/HSL volatile organics. Wells JF-173 and JF-183 sampled in June and December 1994, respectively.

Data qualifiers:

U = analyte was analyzed for but not detected; detection limit given.

J = estimated value.

D = analyte was identified at given concentration with a secondary dilution factor.

\* = internal standard area counts were outside QC limits.

+ = surrogate recoveries were outside QC limits.

E = analyte concentration exceeded calibration range of instrument; concentration estimated.

TABLE C.21b Volatile Organics Analyses for Groundwater in Piezometer Located in Marsh, September 1994<sup>a</sup>

			Conc	centrat	ions (µg/L	) in Gr	oundwate	r		
Compound	JBPM	-1A <sup>b</sup>	JBPM	-1B <sup>c</sup>	JBPM	I-2A	JBPM	1-2B	JBPM	I-3A
Acetone	10	U	10	U	10	U	10	U	10	U
Benzene	2	J	10	U	10	U	10	Ü	10	U
Bromodichloromethane	10	U	10	U	10	U	10	Ū	10	U
Bromoform	10	U	10	U	10	U	10	U	10	U
Bromomethane	10	U	10	U	10	U	10	U	10	U
2-Butanone	10	U	10	U	10	U	10	U	10	U
Carbon disulfide	10	U	10	U	10	U	10	U	10	U
Carbon tetrachloride	10	U	10	U	10	U	10	U	10	U
Chlorobenzene	10	U	10	U	10	U	10	U	10	U
Chloroethane	10	U	10	U	10	U	10	U	10	U
Chloroform	13		10	U	10	U	10	U	10	U
Chloromethane	10	U	10	U	10	U	10	U	10	U
Dibromochloromethane	10	U	10	U	10	U	10	U	10	U
1,1-Dichloroethane	10	U	10	U	10	U	10	U	10	U
1,2-Dichloroethane	13		4	J	27		10	U	12	Č
1,1-Dichloroethene	24		2	J	5	J	10	U	6	J
1,2-Dichloroethene (total)	37,000	D	7,900	D	1,900	D	6	J	1,400	D
1,2-Dichloropropane	10	U	10	U	10	U	10	U	10	U
cis-1,3-Dichloropropene	10	U	10	U	10	U	10	U	10	U
trans-1,3-Dichloropropene	10	U	10	U	10	U	10	U	10	U
Ethylbenzene	10	U	10	U	10	U	10	Ū	10	U
2-Hexanone	10	U	10	U	10	U	10	U	10	U
4-Methyl-2-pentanone	10	U	10	U	10	U	10	U	10	U
Methylene chloride	10	U	10	U	10	U	10	U	10	U
Styrene	10	U	10	U	10	U	10	Ü	10	U
1,1,2,2-Tetrachloroethane	8,600	D	50		720	D	10	U	240	DJ
Tetrachloroethene	29		10	U	29		10	U	26	175
Toluene	10	U	10	U	10	U	10	U	10	U
1,1,1-Trichloroethane	10	U	10	U	10	U	10	U	10	U
1,1,2-Trichloroethane	440	DJ	10		210	DJ	10	U	170	O
Trichloroethene	1,200	DJ	18		4,300	D	10	U	7,300	D
Vinyl chloride	3,200	D	1,300	D	32		5	J	25	D
Xylenes (total)	10	U	10	U	10	U	10	U	10	U